Se-methylselenocysteine protects human hepatoma HepG2 cells against oxidative stress induced by tertbutyl hydroperoxide.

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Running title: Se-methylselenocysteine protects HepG2 cells from oxidative stress.

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1 ABSTRACT

2 Se-methylselenocysteine (Se-MeSeCys) is a common selenocompound in the 3 diet with a tested chemopreventive effect. This study investigated the potential 4 protective effect of Se-MeSeCys against a chemical oxidative stress induced by tert-5 butylhydroperoxide (t-BOOH) on human hepatoma HepG2 cells. Speciation of 6 selenium derivatives by LC-ICP-MS depicts Se-MeSeCys as the only selenocompound 7 in the cell culture. Cell viability (lactate dehydrogenase) and markers of oxidative 8 status: concentration of reduced glutathione (GSH) and malondialdehyde (MDA), 9 generation of reactive oxygen species (ROS) and activity of antioxidant enzymes 10 glutathione peroxidase (GPx) and glutathione reductase (GR) were evaluated. Pre-11 treatment of cells with Se-MeSeCys for 20 h completely prevented the enhanced cell 12 damage, MDA concentration and GR and GPx activity and the decreased GSH induced 13 by t-BOOH but did not prevent increased ROS generation. The results show that 14 treatment of HepG2 cells with concentrations of Se-MeSeCys in the nanomolar-15 micromolar range confers a significant protection against an oxidative insult.

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18 <u>Keywords</u>: antioxidant defences, biomarkers for oxidative stress, dietary antioxidants,
19 selenium compounds

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21 INTRODUCTION

22 Human exposure to potentially toxic chemicals, either through an occupational environment, or as the result of plant and foodstuff pyrolysis (e.g. tobacco smoke, 23 24 charbroiled foods) is almost unavoidable. In many instances, increased exposure to 25 these hazardous chemicals, many of which are prooxidants or procarcinogens, is linked 26 to an increased incidence of cardiovascular disease and cancer [1,2]. This has prompted 27 a search for diets or chemical supplements that might mitigate or prevent the toxic 28 outcome of exposure. There is substantial evidence that antioxidative food components 29 have a protective role against oxidative stress-induced atherosclerosis, degenerative and 30 age-related diseases, cancer and aging [3,4]. Among dietary compounds considered for 31 chemopreventive activity, selenium showed early and continued promise [1,2,4-8]. 32 Selenium is a trace element essential to human health found in fish, meat and plants 33 such as garlic, onion and broccoli, and a deficiency of this element induces pathological 34 conditions, such as cancer, coronary heart disease, and liver necrosis [6,9]. Garlic is the 35 most popular and well-researched Allium plant that is known to accumulate Se as 36 selenoamino acid derivatives, including Se-methyl-l-selenocysteine (Se-MeSeCys), one 37 of the major forms of selenium in the diet, and glutamylmethylselenocysteine 38 (GluMeSeCys) [9,10].

39 have been widely reported effective Selenium compounds to be 40 chemopreventive agents against multiple models of tumorigenesis [4-8,11-14]. These 41 protective effects of selenium seem to be primarily associated with its presence as a 42 cofactor in the enzymes glutathione peroxidase (GPx) and thioredoxin reductase, which 43 are known to protect cellular components from oxidative damage [1,2]. Although these 44 properties indicate that Se-MeSeCys may favourably affect the antioxidant defence

45 system [1,2], little is known about the potentially beneficial role of Se-MeSeCys against
46 oxidative damage in vivo, both in cultured cells and live animals.

47 Human hepatocarcinoma HepG2 is widely used for biochemical and nutritional 48 studies as a cell culture model of human hepatocytes since they retain their morphology 49 and most of their function in culture [15]. In addition, HepG2 is a reliable model where 50 many dietary antioxidants and conditions can be assayed with minor interassay 51 variations [16-20]. Previous studies from our laboratory have shown that the plant 52 flavonol quercetin [18], the olive oil phenol hydroxytyrosol [19], and or a digested 53 coffee melanoidin [20] elicit a favorable response of the antioxidant defense system in 54 cultured human hepatoma HepG2 cells. Therefore, the potential protective effect of 55 different concentrations of the dietary compound Se-MeSeCys against an oxidative 56 stress chemically induced by a potent prooxidant, tert-butyl hydroperoxide (t-BOOH), 57 has now been tested in cultures of HepG2. Cell integrity and several markers of 58 oxidative status such as concentration of reduced glutathione (GSH), generation of 59 reactive oxygen species (ROS), evaluation of the activity of antioxidant enzymes GPx 60 and glutathione reductase (GR) and determination of malondialdehyde (MDA) as a 61 biomarker of lipid peroxidation, were measured to estimate the effect of Se-MeSeCys in 62 cell survival and the response of the antioxidant system of HepG2 to t-BOOH.

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64 MATERIALS AND METHODS

65 **Reagents and samples**

Se-MeSeCys (Sigma Chemicals, St. Louis, MO, USA) was dissolved in Milli-Q
water. Inorganic selenium solution was obtained by dissolving sodium selenite (Merck,
Darmstadt, Germany) in deionised Milli-Q water. Stock solutions of 10 mg/L were
stored in the dark at 4 °C and working standard solutions were prepared daily by

70 dilution. For hydride generation atomic fluorescence spectroscopy (HG-AFS) studies, 71 sodium borohydride (Sigma-Aldrich, Steinheim, Germany) was prepared in 0.3 % 72 sodium hydroxide. For high-performance liquid chromatography (LC)-inductively-73 coupled plasma mass spectrometer (ICP-MS) studies, hepta-fluorobutiric acid (HFBA), 74 trifluoroacetic acid (TFA) and ammonium citrate from Sigma Chemical (St. Louis, MO, 75 USA) and methanol (Sharlab, Barcelona, Spain) were used in the different 76 chromatographic mobile phases. For the enzymatic extraction procedure the non-77 specific protease XIV (Sigma) was used to prepare the extracts. H_2O_2 and HNO_3 were 78 used for acid digestion of samples. The Bradford reagent was from BioRad Laboratories 79 S.A. All other chemicals, including glutathione reductase, reduced and oxidized 80 glutathione, NADPH, o-phthaldehyde (OPT), dichorofluorescin (DCFH) and 81 dinitrophenylhydrazine (DNPH) were purchased from Sigma Chemical Co. Other 82 reagents were of analytical or chromatographic quality.

83 Instrumentation

84 For total selenium determination, samples were microwave-assisted acid 85 digested in doubled-walled advanced composite vessels using a 1000 W MSP 86 (Microwave Sample Preparation system) microwave oven (CEM, Matthews, NC, USA). 87 A Sonoplus ultrasonic homogenizer (Bandelin, Germany) equipped with a titanium 3-88 mm-diameter microtip and fitted with an HF generator of 2200 W at a frequency of 20 89 KHz was used for the extraction of selenium species. An ICP-MS HP-4500 Plus 90 (Tokyo, Japan), fitted with a Babington nebuliser and Scott double-pass spray chamber 91 cooled by a Peltier system was used for total selenium determination after 92 chromatographic separation. A PU-2089 HPLC pump (Jasco Corporation, Tokyo, 93 Japan) fitted with a six-port injection valve (model 7725i, Rheodyne, Rohner Park, CA, 94 USA) with a 100 µL injection loop was used for the chromatographic experiments.

Anionic exchange chromatography was performed using a Hamilton PRP-X100 (Reno,
NE, USA) column (10 µm particle size, 250 mm x 4.1 mm i.d.). Reversed phase
chromatography was performed using a C-18 Gemini column (10 µm particle size, 150
mm x 2.0 mm i.d) Phenomenex (Torrance, CA, USA). For molecular weight
fractionation, 10 kDa cut-off filters (Millipore, Bedford, MA, USA) and an Eppendorf
(Hamburg, Germany) centrifuge 5804, F34-6-38 were used [21].

101 Cell Culture

102 Human hepatoma HepG2 cells were maintained in a humidified incubator 103 containing 5 % CO₂ and 95 % air at 37°C. They were grown in DMEM-F12 medium 104 from Biowhitaker (Innogenetics, Madrid, Spain), supplemented with 2.5 % Biowhitaker 105 foetal bovine serum (FBS) and 50 mg/L of each of the following antibiotics: 106 gentamicin, penicillin and streptomycin (all from Sigma, Madrid, Spain). Plates were 107 changed to FBS-free medium before the beginning of the assay. The serum added to the 108 medium favours growth of most cell lines but might interfere in the running of the 109 assays and affect the results. Moreover, it has been observed a fairly good growth of 110 HepG2 cells in FBS-free DMEM-F12 [17].

111 Cell treatment

112 Two sets of experiments were designed for this study: A) experiments of plain 113 treatment of cells with Se-MeSeCys for 2 or 20 h to test for a direct effect of the 114 selenocompound and B) experiments of pretreatment of cells with Se-MeSeCys for 2 or 115 20 h before submitting the cells to an oxidative stress by t-BOOH to test for a protective 116 effect against an oxidative insult. In order to infer the effect of the time of treatment to 117 the different concentrations of Se-MeSeCys, two experimental terms of short (2 h) and 118 long (20 h) treatment with the compound were selected according to previous studies 119 [17,18]. Lactate dehydrogenase (LDH), GSH, MDA and ROS were evaluated in both experimental conditions and, in addition, GPx and GR were also determined in
experiment B. The different concentrations of Se-MeSeCys were dissolved in serumfree culture medium. For further details see Material and Methods in references 18-20.

123 **Procedure for selenium determination and speciation**

124 Cultured cells were washed with phosphate buffered saline (PBS), collected by 125 scraping and resuspended in PBS. One mL of cell sample was digested in an analytical 126 microwave oven with 1 mL of nitric acid and 0.5 mL of hydrogen peroxide following 127 cycles of 5 min pressure/15 min break. Se (VI) was reduced to Se (IV) after the acid 128 digestion. Hydrochloric acid (6 M final concentration) and heating at 95°C for 1 h are 129 needed to convert Se (VI) in Se (IV). The solutions were then diluted to 10 mL with 130 Milli-Q-water. Total selenium concentration was determined by the continuous 131 selenium hydride system connected to an AFS equipment. Hydrochloric acid (3 M) and 132 1% sodium tetrahydroborate (w/v) were used to generate the selenium hydride. For Se 133 speciation, 1 mL of cell or culture media samples, 2 mL of Milli-Q-water and 10 mg of 134 the non-specific protease S. griseus (protease XIV) were placed in a 10 mL Teflon vial. 135 The mixture was sonicated for 120 s at 20 W of power intensity. In order to enhance 136 sample clean-up after enzymatic hydrolysis, samples were centrifuged and the 137 supernatants were removed and passed through a 10 kDa cut-off filter (centrifugation at 138 7500 g and 20°C). LC coupled with an ICP-MS was used for the measurement of 139 selenium species with the operating conditions given in Table 1.

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Evaluation of LDH, GSH and MDA

141 Cells were plated in 60 mm diameter plates at a concentration of 1.5×10^6 per 142 plate and the assay was carried out two days later. Cells were treated as described in the 143 section above and LDH leakage to the culture medium was estimated from the ratio 144 between the LDH activity in the culture medium and that of the whole cell content 145 [16,18]. The content of GSH was quantitated by the fluorometric assay of Hissin and 146 Hilf [22]. The method takes advantage of the reaction of reduced glutathione with OPT 147 at pH 8.0. Fluorescence was measured at an emission wavelength of 460 nm and an 148 excitation wavelength of 340 nm. The precise protocol has been described elsewhere 149 [16,18]. Cellular MDA was analyzed by HPLC as its 2,4-dinitrophenylhydrazone 150 (DNPH) derivative [23]. Cells were treated as in the LDH assay and then collected. For 151 MDA, values are expressed as nmol of MDA/mg protein; protein was measured by the 152 Bradford kit.

153 **Determination of ROS**

154 Cellular reactive oxygen species were quantified by the DCFH assay using 155 microplate reader [24]. After being oxidized by intracellular oxidants, DCFH will 156 become dichorofluorescein (DCF) and emit fluorescence. By quantifying fluorescence 157 over a period of 90 min, a fair estimation of the overall oxygen species generated under 158 the different conditions was obtained. This parameter gives a very good evaluation of 159 the degree of cellular oxidative stress. The assay has been described elsewhere [16,18].

160 **Determination of GPx and GR Activity**

161 For the assay of the GPx and GR activity, cells previously treated as in LDH, 162 GSH and MDA assays were suspended in phosphate buffered saline (PBS) and 163 centrifuged at low speed for 5 min. Cell pellets were resuspended in 20 mM Tris, 5 mM 164 EDTA and 0.5 mM mercaptoethanol, sonicated, and centrifuged at 3000 g for 15 min. The enzyme activity was measured in the supernatants. The determination of GPx 165 166 activity is based on the oxidation of reduced glutathione by GPx, using tert-butyl 167 hydroperoxide as a substrate, coupled to the disappearance of the reduced form of 168 nicotinamide adenine dinucleotide phosphate (NADPH) by GR [25]. GR activity was 169 determined by following the decrease in absorbance due to the oxidation of NADPH

170 utilized in the reduction of oxidized glutathione [26]. The methods have been previously

171 described [16,18]. Protein was measured by the Bradford kit.

172 Statistics

173 Statistical analysis of data was as follows: prior to analysis the data were tested 174 for homogeneity of variances by the test of Levene; for multiple comparisons, one-way 175 ANOVA was followed by a Bonferroni test when variances were homogeneous or by 176 Tamhane test when variances were not homogeneous. The level of significance was p < 177 0.05. A SPSS version 12.0 program has been used.

178

179 **RESULTS AND DISCUSSION**

180 **Total selenium content and speciation**

181 Analytical systems developed for the speciation of selenium species employ a 182 powerful LC coupled to a specific detector with a high efficiency sample introduction 183 system. Nowadays, the most favoured instrument combination for this purpose is ICP-184 coupled to various LC techniques, such as anion-, cation-exchange MS 185 chromatography, reversed-phase, ion-pair, size exclusion and chiral chromatography. 186 Identification of species is achieved by retention time matching with available standards 187 utilized in a standard addition mode [21,27,28]. This analysis has not been previously 188 applied to cell culture studies.

Speciation of selenium was carried out in order to observe potential biotransformation of Se-MeSeCys. For separation and unambiguous identification of Se species two chromatographic techniques (reversed-phase chromatography and anionic exchange chromatography) coupled to ICP-MS were used. The addition of a standard to the sample was necessary to confirm the species that appear in chromatograms. Duplicates of samples were spiked with 5 (cell samples) or 30 (culture medium

195 samples) $\mu g/L$ of Se-MeSeCys at the end of the sample treatment. The single peak that 196 appears in chromatograms of cell homogenates and collected medium indicates that Se-197 MeSeCys is the only relevant selenium species (98 % of the total selenium content) in 198 cultures of HepG2 treated with this selenium derivative, suggesting no 199 biotransformation of the compound during the length of the assay (Figure 1). 200 Preservation of the original structure of Se-MeSeCys indicates that the protective effect 201 should be bestowed to the whole molecule. Although the presence of the SH group in 202 the cysteine moiety could greatly help to increase the antioxidant potential and, thus, the 203 protective effect against an oxidative insult, the SH by itself does not necessarily confer 204 antioxidant capacity nor protection against oxidative stress to a molecule since this SH 205 group and its containing aminoacids are widely present in all proteins and most 206 biological peptides and none has shown a protective effect against an oxidative damage. 207 This fact suggests that the reducing group (SH) must be in the proper template or 208 adequate biological presentation to exert the antioxidant protective effect. Further 209 experimentation focussed separately on pure selenium and cysteine would be necessary 210 to delineate the effect of the different moieties.

211 In order to predict the potential bioavailability and metabolism of selenium and 212 its species from Se-enriched radish, Pedrero et al. [29] performed an in vitro 213 gastrointestinal process, concluding that concentration of the species found, 214 selenomethionine and Se-MeSeCys, remains almost unaltered after simulated 215 gastrointestinal digestion. Although other authors have found intense metabolism of Se-216 MeSeCys to methyl-seleninic acid in all tissues containing beta-lyase enzyme activity 217 [30], the absence of any structural transformation in the Se-MeSeCys molecule 218 throughout the GI tract ensures preservation of the complete biological activity in a 219 strenuous metabolic environment. However, the present findings in cultured live cells should be considered as a first necessary step in the research on the protective effects of selenium compounds against oxidative stress and future research including experiments in live animals should be delineated in order to address the in vivo metabolic fate of this compound.

224 Total selenium content was measured in cells treated for 24 h with Se-MeSeCys 225 at two different concentrations (10 and 100 µM) to evaluate uptake capability of 226 selenium by HepG2 cells. Data of selenium concentration was validated by applying the 227 method to a certified reference material: bovine liver CRM 185R (1.68 + 0.14 mg/Kg). 228 The result obtained, 1.8 + 0.2 mg/Kg, was in agreement with the reference value. Final 229 concentration of Se-MeSeCys inside the cells remains relatively constant regardless of 230 the selenium dose (13.2 \pm 0.2 μ g/mL homogenate in cells treated with 10 μ M vs. 12.1 \pm 231 1.1 µg/mL in those treated with 100 µM), probably due to saturation of uptake 232 mechanisms. In fact, in human trabecular meshwork cell (HTM) treated with 233 methylselenic acid, saturation of selenium uptake was observed at the same doses [31]. 234 No amount of any selenium compound was found in either the culture medium or cell 235 homogenate in HepG2 untreated cultures.

236 Cell viability

237 As a trace element, selenium appears_at low concentrations in the diet but 238 chemopreventive (antioxidant and antitumorigenic) levels of selenium appear to reside 239 at doses greater than those regarded as necessary for supporting adequate expression of 240 selenoenzymes [32,33]. Selenium requirements in human diet are in a very narrow range: consumption of food containing less than 0.1 mg kg⁻¹ of this element will result 241 in its deficiency, whereas dietary levels above 1 mg kg⁻¹ will lead to toxic 242 manifestations [34,35]. Although selenite has been reported to protect human 243 244 endothelial cells from oxidative damage induced by a potent prooxidant [36], it has

245 been suggested that the cancer chemopreventive effect of Se-MeSeCys in tumor cells 246 may reflect pro-oxidant rather than antioxidant activity of this compound [14]. 247 Therefore, before aiming for the protective effect of the tested antioxidant it is necessary 248 to ensure that no direct damage is caused to the cell by the compound. Thus, in 249 experiment A, cell toxicity and basal cellular redox status were determined in cells 250 treated for short (2 hours) and long (20 h) terms with different concentrations of Se-251 MeSeCys in the nM-µM range. The concentration range is fairly realistic in order to 252 evaluate the effect at the physiological level since steady-state concentrations around 253 0.7 µM of all selenium species have been reported in human serum [37]. Our results 254 show that treatment with concentrations of Se-MeSeCys up to 10 µM for 20 h evoked 255 no toxicity in HepG2 (Figure 2a). Indeed, the complete inhibition of cell damage in 256 experiment B when human hepatoma HepG2 cells were pretreated with Se-MeSeCys 257 for 2 or 20 h prior to being submitted to t-BOOH (Figure 3a), indicates that integrity of 258 the Se-MeSeCys-treated cells was largely protected against the oxidative insult.

259 **GSH concentration**

260 Cells are naturally provided with an extensive array of protective enzymatic and 261 non-enzymatic antioxidants that counteract the potentially injurious oxidizing agents 262 [2,16-20,32,36]. But even this multifunctional protective system cannot completely 263 prevent the deleterious effects of ROS, and consequently oxidatively damaged 264 molecules accumulate in cells. Reduced glutathione is the main non-enzymatic 265 antioxidant defense within the cell and plays an important role in protection against 266 oxidative stress, as a substrate in glutathione peroxidase-catalysed detoxification of 267 organic peroxides, by reacting with free radicals and by repairing free-radical-induced 268 damage through electron-transfer reactions [2,16-20]. It is usually assumed that GSH 269 depletion reflects intracellular oxidation. On the contrary, an increase in GSH

concentration could be expected to prepare the cell against a potential oxidative insult[16-20,38,39].

272 In our experimental conditions, addition of Se-MeSeCys did not evoke changes 273 in GSH concentration whereas treatment of HepG2 cells with 200 µM t-BOOH induced 274 a remarkable decrease in the concentration of reduced glutathione indicative of 275 oxidative stress (Figure 2b). This decrease of GSH induced by t-BOOH was partly (2 h) 276 or completely (20 h) prevented by pretreatment with Se-MeSeCys (Figure 3b). This 277 result could explain the protected cell integrity reported above since maintaining GSH 278 concentration above a critical threshold while facing a stressful situation represents an 279 advantage for cell survival.

280 MDA concentration

281 An important step in the degradation of cell membranes is the reaction of ROS 282 with the double bonds of polyunsaturated fatty acids (PUFAs) to yield lipid 283 hydroperoxides. On breakdown of such hydroperoxides a great variety of aldehydes can 284 be formed; MDA, a three-carbon compound formed by scission of peroxidized PUFAs, 285 mainly arachidonic acid, is one of the main products of lipid peroxidation [40]. Since 286 MDA has been found elevated in various diseases thought to be related to free radical 287 damage, it has been widely used as an index of lipoperoxidation in biological and 288 medical sciences [41]. However, other than our previous results, determination of MDA 289 levels in cell culture conditions is extremely scant in the literature [42].

We have established a new method of evaluation of MDA in cultures of human hepatoma cells that is sensitive enough to detect a significant increase in MDA concentration in response to an oxidative stress induced by t-BOOH [23]. By using this method we have found that a 3 h treatment of HepG2 with 200 µM t-BOOH evoked a significant increase of about 35 % in the cellular concentration of MDA, indicating

295 damage to cell lipids (Figures 2c and 3c). In fact, a significant decrease of MDA was 296 observed in cells treated with 1 or 10 µM Se-MeSeCys for 20 h in experiment A (Figure 297 2c). In addition, the t-BOOH-induced increase of MDA was completely avoided when cells were pretreated for 2 or 20 h with 0.1-10 µM of Se-MeSeCys in experiment B 298 299 (Figure 3c). This protection by Se-MeSeCys against an induced lipid peroxidation in a 300 cell culture has not been previously reported and is in line with previous studies that 301 showed a similar effect by other dietary compounds including plant polyphenols such as 302 tea catechins [43,44], quercetin [18] and olive oil hydroxytyrosol [19], beta carotene or 303 lutein [43], and Maillard reaction products such as coffee melanoidin [20] in the same 304 cell line, human hepatoma HepG2.

305 **ROS Generation**

306 Accumulation of ROS in several cellular components is thought to be a major 307 cause of molecular injury leading to cell aging and to age-related degenerative diseases 308 such as cancer, brain dysfunction and coronary heart disease [2,3,32,36]. Direct 309 evaluation of ROS yields a very good indication of the oxidative damage to living cells 310 [24]. Based upon the fact that nonfluorescent DCFH crosses cell membranes and is 311 oxidized by intracellular ROS to highly fluorescent DCF [45], the intracellular DCF 312 fluorescence can be used as an index to quantify the overall oxidative stress in cells [16-313 20]. A prooxidant such as t-BOOH can directly oxidize DCFH to fluorescent DCF, and 314 it can also decompose to peroxyl radicals and generate lipid peroxides and ROS, thus 315 increasing fluorescence.

316 A significant increase in ROS generation was observed in HepG2 cells treated 317 with 200 μ M t-BOOH for the 90 min of the assay, whereas ROS levels in cells treated 318 with 0.01-10 μ M Se-MeSeCys in experiment A were below those of control non-319 stressed cells (Figure 4). Similar to what observed with other competent antioxidants

320 [18,19], the presence of Se-MeSeCys in the cell culture appears to reduce the 321 progressive formation and accumulation of oxygen radicals by the mitochondrial 322 respiratory chain [18,19]. The increased ROS generation induced by t-BOOH was not 323 prevented in cultured cells pretreated for 2 or 20 h with Se-MeSeCys in experiment B 324 (Figure 5), contrary to a previous report with quercetin [18], but similar to what 325 observed with other dietary antioxidants such as the olive oil phenol hydroxytirosol [19] 326 and coffee melanoidin [20]. In the 2 h pre-treatment condition, this phenomenon is 327 consistent with the partial recovery of the levels of reduced glutathione which may be 328 explained by an increased consumption of GSH in the enzymatic and non-enzymatic 329 quenching of reactive oxygen species generated by t-BOOH. In any case, these data 330 suggest that high levels of ROS generated during the stress period are being more 331 efficiently quenched in cells pretreated with Se-MeSeCys resulting in a reduced cell 332 damage and lipid peroxidation.

333 GPx and GR activity

In the defense against oxidative stress, the cellular antioxidant enzyme system plays a crucial role and changes in the activity of antioxidant enzymes can be considered as biomarkers of the antioxidant response [16-20,46,47]. GPx catalyses GSH oxidation to GSSG at the expense of H_2O_2 or other peroxides [47] and GR recycles oxidized glutathione back to reduced glutathione [3,26], therefore, their activities are essential for the intracellular quenching of cell-damaging peroxide species and the effective recovery of the steady-state concentration of reduced glutathione.

The chemopreventive activity of selenium has been attributed to its effect on a variety of molecular targets and cellular processes, although it appears that these targets and processes may differ with the prooxidant or tumorigenic agent and the chemical nature of the selenium compound [1,2,6,32,36]. Nevertheless, linkage of the antioxidant

and anti-tumorigenic activity with positive effects on "protective" enzymes remains a 345 346 recurrent theme and a common feature of most of the implicated enzymes is their 347 function in sequestering reactive oxygen species and/or maintaining the cell and cellular 348 components in their appropriate redox state [1,2,32,33,36]. Some protective enzymes 349 such as GPx are selenoproteins and are likely to be impacted by selenium 350 supplementation [1,36]. The significant increase in the activity of GPx and GR observed 351 after a 3 h treatment with 200 µM t-BOOH (Figure 6), clearly indicates a positive 352 response of the cell defense system to face an oxidative insult [16-20]. Other dietary 353 antioxidants have been tested by other authors and significant changes in the enzyme 354 activity of the antioxidant enzymes have been observed only at very high doses [3,48]. 355 Although only a previous cell culture study has shown a specific effect of a selenium 356 compound on the GPx response to an oxidative insult [36], in experimental conditions 357 similar to those of the present study we have shown that the flavonoid quercetin [18], 358 olive oil phenol hydroxytyrosol [19] and a coffee melanoidin [20] protect cell damage 359 by preventing the severely increased activity of antioxidant enzymes induced by t-360 BOOH. In line with those results, in experiment B of the present study we show, for the 361 first time, that a long-term treatment of human hepatoma cells with Se-MeSeCys 362 prevents the increase in the activity of GPx and GR induced by oxidative stress (Figure 363 6).

364

365 CONCLUSIONS

The results indicate that at the end of an induced stress period the antioxidant defense system of cells that had been pretreated with Se-MeSeCys has more efficiently returned to a steady-state activity diminishing, therefore, cell damage and enabling the cell to cope in better conditions with further oxidative insults.

In addition, our results support previous data on the chemoprotective effect of Se-MeSeCys and give more insight on its potential biological activity, showing that concentrations of Se-MeSeCys within the physiological range remain unaltered during the treatment and evoke a favourable response in cellular models. Therefore, Se-MeSeCys may contribute to the protection afforded by fruits, vegetables and plantderived beverages against diseases for which excess production of ROS has been implicated as a casual or contributory factor.

Acknowledgements

This work was supported by the grants AGL2000-1314, AGL2004-00302 and CTQ2005-02281 from the Spanish Ministry of Science and Technology (CICYT), and grant CAM-S505/AGR/0312 from the Comunidad Autónoma de Madrid. Susana Cuello has a predoctoral fellowship, Raquel Mateos is a post-doctoral fellow and Sonia Ramos has a Ramón y Cajal contract all from the Spanish Ministry of Education.

LITERATURE CITED

- 1. El-Sayed WM, Aboul-Fadl T, Lamb JG, Roberts JC, Franklin MR (2006) Toxicology 220:179-188.
- 2. Valko M, Rhodes CJ, Moncol J, Izakovic M, Mazur M (2006) Chem.-Biol Interact 160:1-40.
- 3. Duthie GG, Duthie SJ, Kyle JAM (2001) Nutr Res Rev 3:79-106.
- 4. Combs Jr. GF (2005) J Nutr 135:343-347.
- 5. Yu SY, Zhu YG, Li WG (1997) Biol Trace Elem Res 56:117-24.
- 6. Finley JW (2003) J Med Food 6:19-26.
- 7. El-Bayoumy K, Sinha R (2004) Mutation Res 551:181-197.
- 8. El-Bayoumy K, Sinha R, Pinto JT, Rivlin RS (2006) J Nutr 136:864S-869S.
- 9. Arnault I, Auger J (2006) J Chromatography A 1112:23-30.
- 10. Milner JA (2001) J Nutr 131:1027-31.
- 11. Duffield-Dillico AJ, Reid ME, Turnbull BW, Combs GF Jr, Slate EH, Fischbach
- LA, Marshall JR, Clark LC (2002) Cancer Epidemiol Biomarkers Prev 11:630-9.
- Reid ME, Duffield-Lillico AJ, Garland L, Turnbull BW, Clark LC, Marshall JR
 (2002) Cancer Epidemiol Biomarkers Prev 11:1285-91.
- 13. Duffield-Dillico AJ, Slate EH, Reid ME, Turnbull BW, Wilkins PA, Combs GF Jr,

Park HK, Gross EG, Graham SF, Stratton MS, Marshall JR, Clark LC (2003) J Natl Cancer Inst 95:1477-81.

14. Drake EN (2006) Medical Hypotheses 67:318-322.

15. Brandon EFA, Bosch TM, Deenen MJ, Levink R, Van der Wal E, Van Meerveld JBM, Bijl M, Beijnen JH, Schellens JHM, Meijerman I (2006) Toxicol Appl Pharmacol 211:1-10.

16. Alía M, Ramos S, Mateos R, Bravo L, Goya L (2005) J Biochem Mol Toxicol 19:119-128.

- 17. Alía M, Mateos R, Ramos S, Lecumberri E, Bravo L, Goya L (2006) Eur J Nutr 45:19-28.
- Alía M, Ramos S, Mateos R, Bravo L, Goya L (2006) Toxicol Appl Pharm 212:110-118.
- 19. Goya L, Mateos R, Bravo L (2007) Eur J Nutr 46:70-78.

20. Goya L, Delgado-Andrade C, Rufián-Henares JA, Bravo L, Morales FJ (2007) Mol Nutr Food Res 51:536-545.

- 21. Ipolyi I, Stefánka Z, Fodor P (2001) Anal Chim Acta 435:367-375.
- 22 Hissin PJ, Hilf R (1976) Anal Biochem 74:214-226.
- 23. Mateos R, Goya L, Bravo L (2004) J Chromatography B 805:33-39.
- 24. Wang H, Joseph JA (1999) Free Rad Biol Med 27:612-616.
- 25. Gunzler WA, Kremers H, Flohe L (1974) Klin Chem Klin Biochem 12:444-448.
- 26. Goldberg DM, Spooner RJ (1987) Glutathione reductase, In: Bergmeyer, H.V. ed.,
- Methods of Enzymatic Analysis (Weinheim, Verlag-Chemie), pp. 258-265.
- 27. Cankur O, Yathavakilla SKV, Caruso JA (2006) Talanta 70:784-790.
- 28. Grant TD, Montes-Bayón M, LeDuc D, Fricke MW, Terry N, Caruso JA (2004) J Chromatogr A 1026:159-166.
- 29. Pedrero Z, Madrid Y, Cámara C (2006) J Agric Food Chem 54:2412-2417.
- 30. Ip C, Thompson HJ, Zhu Z, Ganther HE (2000) Cancer Res 60:2882-2886.
- 31. Conley SM, McKay BS, Gandolfi AJ, Stamer WD (2006) Exp Eye Res 82:637-647.
- 32. El-Sayed WM, Aboul-Fadl T, Roberts JC, Lamb JG, Franklin MR (2007) Toxicol In Vitro 21:157-164.
- 33. Combs Jr GF, Gray WP (1998) Pharmacol Ther 79:179–192.

34. Wada O, Kurihara N, Yamazaki N (2003) J Nutr Assess 10:199-210.

35. Raisbeck MF (2000) Vet Clin N Am Food Anim Pract 16:465-480.

36. Miller S, Walker SW, Arthur JR, Nicol F, Pickard K, Lewin MH, Howie AF, Beckett GJ (2001) Clin Sci 100:543-550.

37. Look MP, Rockstroh JK, Rao GS, Kreuzer KA, Barton S, Lemoch H, Sudhop T, Hoch J, Stockinger K, Spengler U, Sauerbruch T (1997) Eur J Clin Nutr 51:266-272.

38. Scharf G, Prustomersky S, Knasmuller S, Schulte-Hermann R, Huber WW. (2003) Nutr Cancer 45:74-83.

39. Myhrstad MC, Carlsen H, Nordstrom O, Blomhoff R, Moskaug JO (2002) Free Radical Biol Med 32:386-393.

40. Pilz J, Meineke I, Gleiter CH (2000) J Chromatograpy B 742:315-325.

41. Suttnar J, Masova L, Dyr E (2001) J Chromatography B 751:193-119.

42. Courtois F, Delvin E, Ledoux M, Seidman E, Lavoie JC, Levy E (2002) J Nutr 132:1289-1292.

43. Chen L, Yang X, Jiao H, Zhao B (2002) Toxicol Sci 69:149-156.

44. Murakami C, Hirakawa Y, Inui H, Nakano Y, Yoshida H (2002) J Nutr Sci Vitaminol 48:89-94.

45. LeBel CP, Ishiropoulos H, Bondy SC (1992) Chem Res Toxicol 5:227-231.

46. Rodgers EH, Grant MH (1998) Chem Biol Interact 116:213-228.

47. Ursini F, Maiorino M, Brigelius-Flohé R, Aumann KD, Roveri A, Schomburg

D, Flohé L (1995) Methods Enzymol 252:38-114.

48. Röhrdanz E, Ohler S, Tran-Thi Q-H, Kahl R (2002) J Nutr 132:370-375.

FIGURE CAPTIONS

Figure 1.- Chromatogram of Se species by a) anionic exchange chromatography in cell homogenates, b) reversed-phase chromatography in cell homogenates and c) anionic exchange chromatography in collected media from cell cultures.

Figure 2.- <u>Experiment A</u>. Effect of Se-MeSeCys on cell viability and intracellular concentration of GSH and MDA. Cells treated with 200 μ M t-BOOH for 3 h (noted t-BOOH) are positive controls. a) LDH values are means \pm SD of 6-8 data. b) GSH values are means of 4-5 different samples per condition. c) MDA values are means \pm SD, n= 4. Groups of data within the same graphic with a different letter are statistically different (*P* < 0.05).

Figure 3.- Experiment B. Protective effect of Se-MeSeCys on cell viability and intracellular concentration of GSH and MDA. a) LDH values are means \pm SD of 6-8 data. b) GSH values are means of 4-5 different samples per condition. c) MDA values are means \pm SD, n= 4. Groups of data within the same graphic with a different letter are statistically different (*P* < 0.05).

Figure 4.- <u>Experiment A</u>. Effect of Se-MeSeCys on intracellular ROS generation. Cells treated with 200 μ M t-BOOH for 3 h (noted t-BOOH) are positive controls. Different letters upon symbols indicate statistically significant differences (*P* < 0.05) when that group or close groups of data are compared within the same time point. Values are

means \pm SD of 7-8 different samples per condition. SD values were not included due to intense bar overlapping.

Figure 5.- <u>Experiment B.</u> Protective effect of Se-MeSeCys on intracellular ROS generation. Different letters upon symbols indicate statistically significant differences (P < 0.05) when that group or close groups of data are compared within the same time point. Values are means ± SD of 7-8 different samples per condition. SD values were not included due to intense bar overlapping.

Figure 6.- <u>Experiment B.</u> Protective effect of Se-MeSeCys on the activity of GPx and GR. Groups of data within the same graphic with a different letter are statistically different (P < 0.05). Values are means ± SD of 4-5 different samples per condition.

 Table 1 Instrumental operating conditions for Se determination by

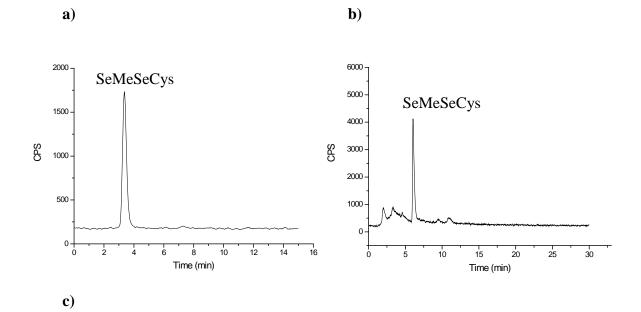
 HPLC-ICP-MS

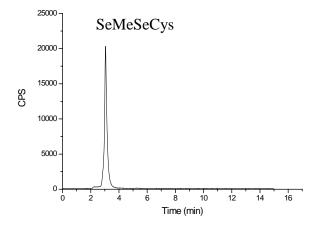
HPLC parameters		
Analytical column	PRP-X100	Phenomenex C-18
Eluent	10mM citrate buffer,	0.1% HFBA; 0.05% TFA;
	H ₂ O:MeOH (98:2)	2% MeOH
Eluent flow rate	1 ml min ⁻¹	0.2 ml min^{-1}
Elution programme	Isocratic	Isocratic
Injection volume	100 µ1	20 µl

ICP-MS conditions

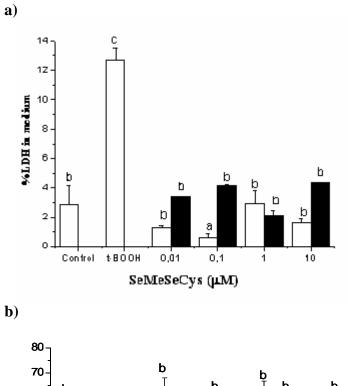
ici nib conditions	
Forward power	1350 W
Plasma gas (Ar) flow rate	15 l min ⁻¹
Auxiliary gas (Ar) flow	
rate	$1.3 \mathrm{lmin}^{-1}$
Carrier gas (Ar) flow rate	$1.4 \mathrm{l} \mathrm{min}^{-1}$
Nebulyser type	Babington
Isotopes monitored	⁷⁷ Se, ⁷⁸ Se, ⁸² Se
-	

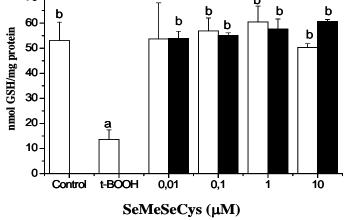
Figure 1





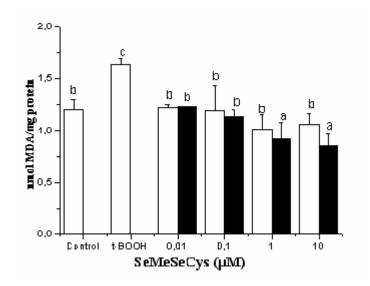






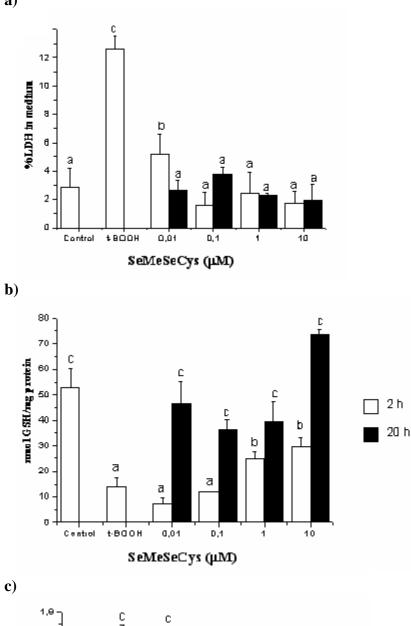












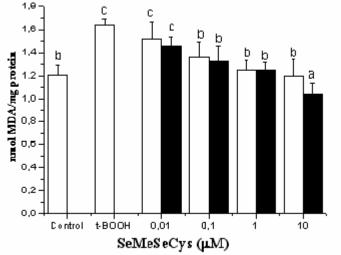


Figure 4

