

Supporting information

Supercritical CO₂ synthesis of porous metalloporphyrin frameworks: application in photodynamic therapy

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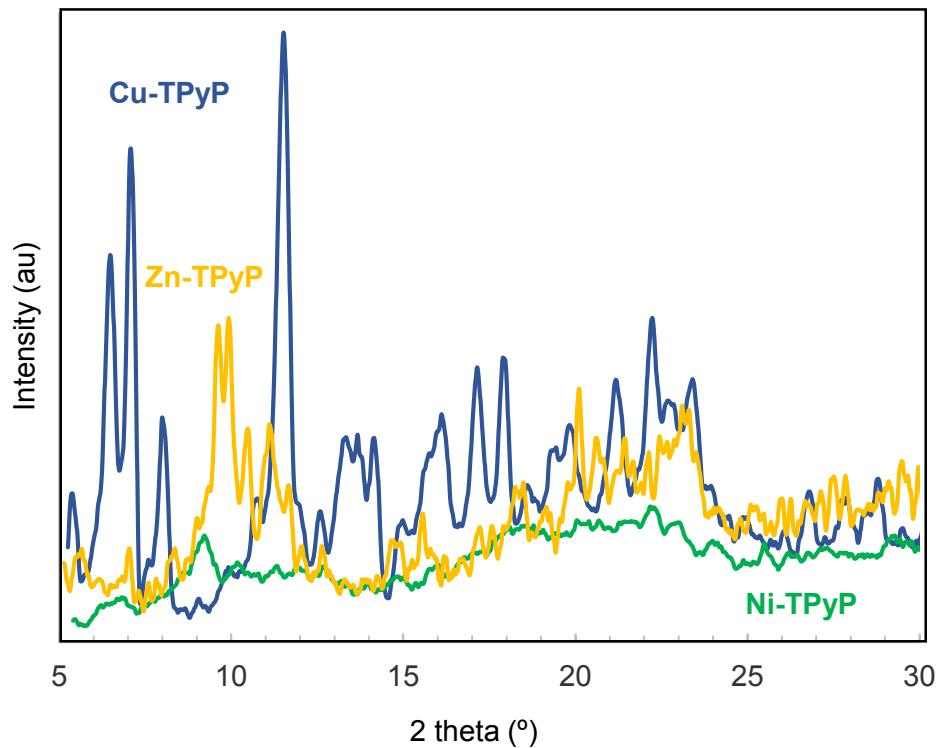


Figure S1. Capillary PXRD patterns of reaction products from the layering runs of Cu(II), Zn(II) and Ni (II) reacted with H₂TPyP.

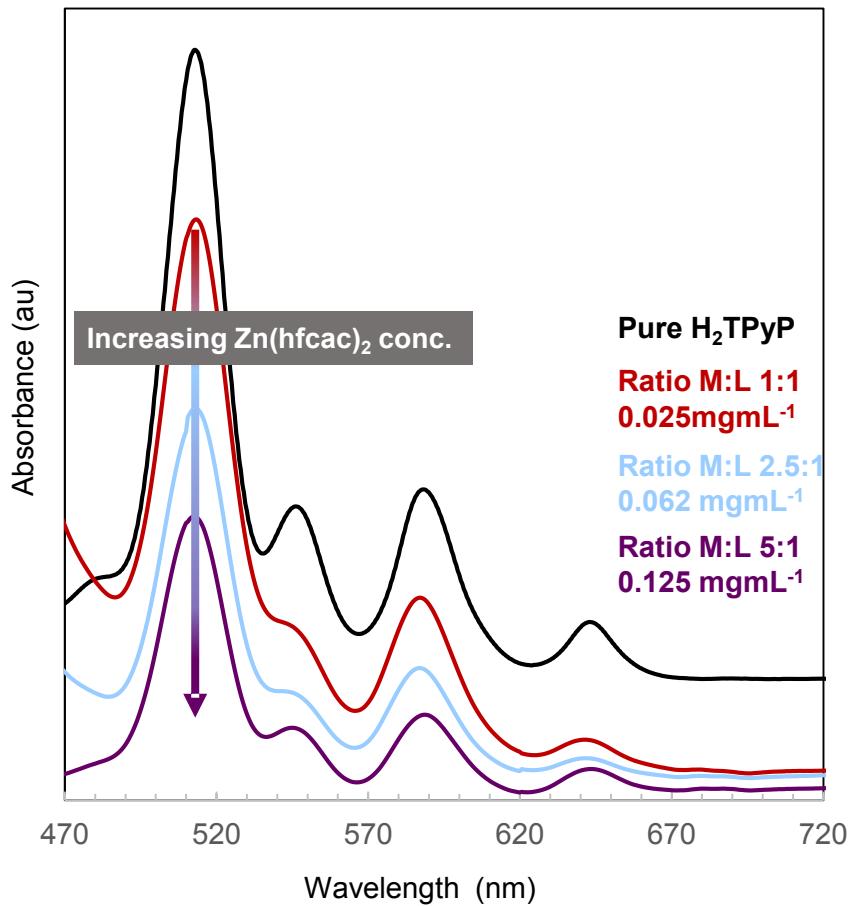


Figure S2. Titration experiments of H₂TPyP with Zn(hfac)₂. The porphyrin concentration was 0.025 mgmL⁻¹. The concentrations for the metal were 0.025 mgmL⁻¹ (red), 0.0625 mgmL⁻¹ (blue) and 0.125 mgmL⁻¹ (purple).

Table S1. Crystallographic data for compound $[\{\text{Co(hfac)}_2\}\text{H}_2\text{TPyP}]_n$ determined by single crystal diffraction.

Compound	$[\{\text{Co(hfac)}_2\}\text{H}_2\text{TPyP}]_n$
Empirical formula	$\text{C}_{15}\text{H}_{7.5}\text{CoF}_6\text{N}_2\text{O}_2$
Formula weight	391.19
Radiation	X-rays
T (K)	100(2)
Wavelength (Å)	0.72932
System, space group	Monoclinic, $C\ 2/m\ (12)$
<i>a</i> (Å)	22.913(10)
<i>b</i> (Å)	25.721(4)
<i>c</i> (Å)	6.524(1)
β (°)	105.60(2)
V (Å ³)	3703.25
Z	8
<i>D</i> calc (g cm ⁻³)	1.403
μ (mm ⁻¹)	0.597
<i>F</i> (000)	1560
Crystal size (mm ³)	0.100x0.060x0.040
<i>hkl</i> ranges	-31≤ <i>h</i> ≤31, -35≤ <i>k</i> ≤35, -8≤ <i>l</i> ≤8
2θ range (°)	1.248 to 29.967
Reflections collected/unique/[R _{int}]	125748/5021/[R _{int} = 0.0546]
Completeness	98.6%
Refinement method	Full matrix least-squares on F ²
Data/restrains/parameters	5021/0/234
Goodness of fit (GOF) on F ²	1.134
Final <i>R</i> indices [<i>I</i> >2σ(<i>I</i>)]	R1 = 0.0483, wR2 = 0.1401
<i>R</i> indices (all data)	R1 = 0.0500, wR2 = 0.1416
Largest. Diff. peak and hole (e Å ⁻³)	0.945 and -0.995

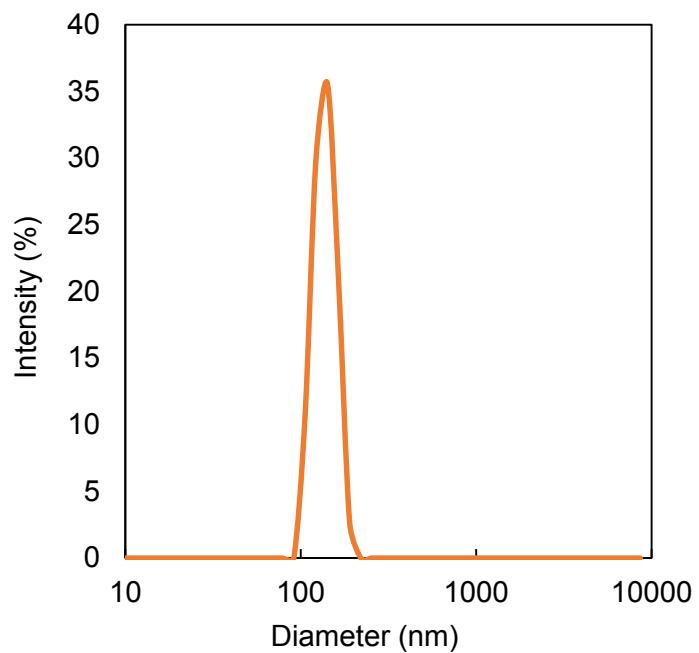


Figure S3. DLS measurement of $[Zn\text{-}TPyP]_n$ sample in water.

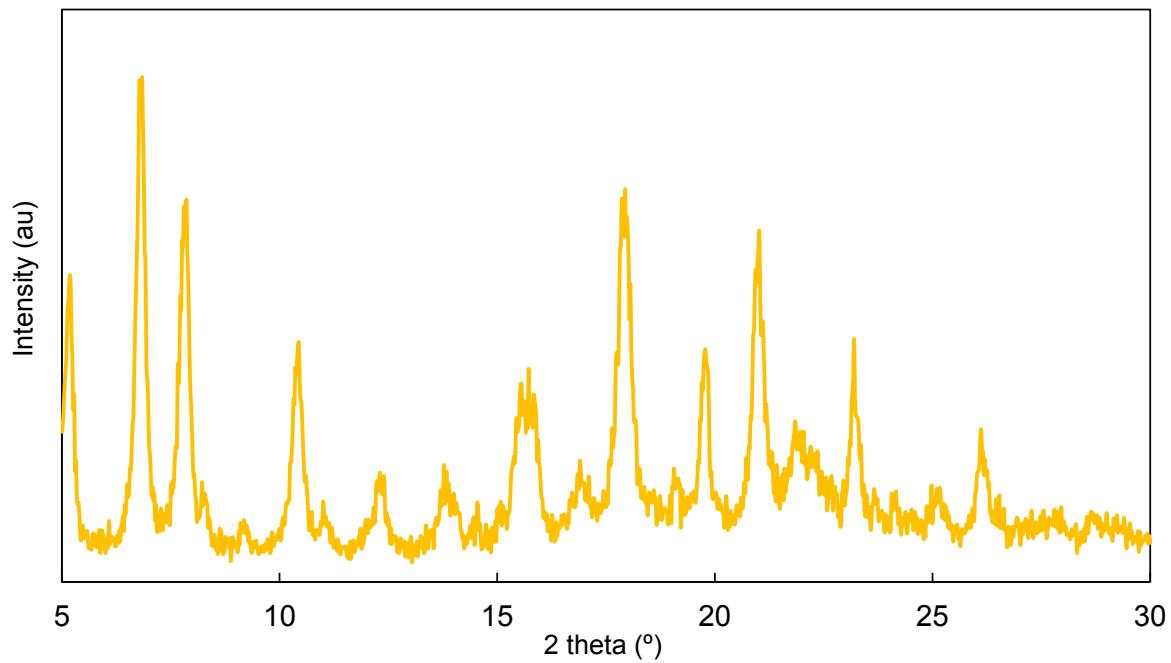


Figure S4. PXRD of the $[Zn\text{-}TPyP]_n$ sample after its dispersion in water achieved applying sonication.

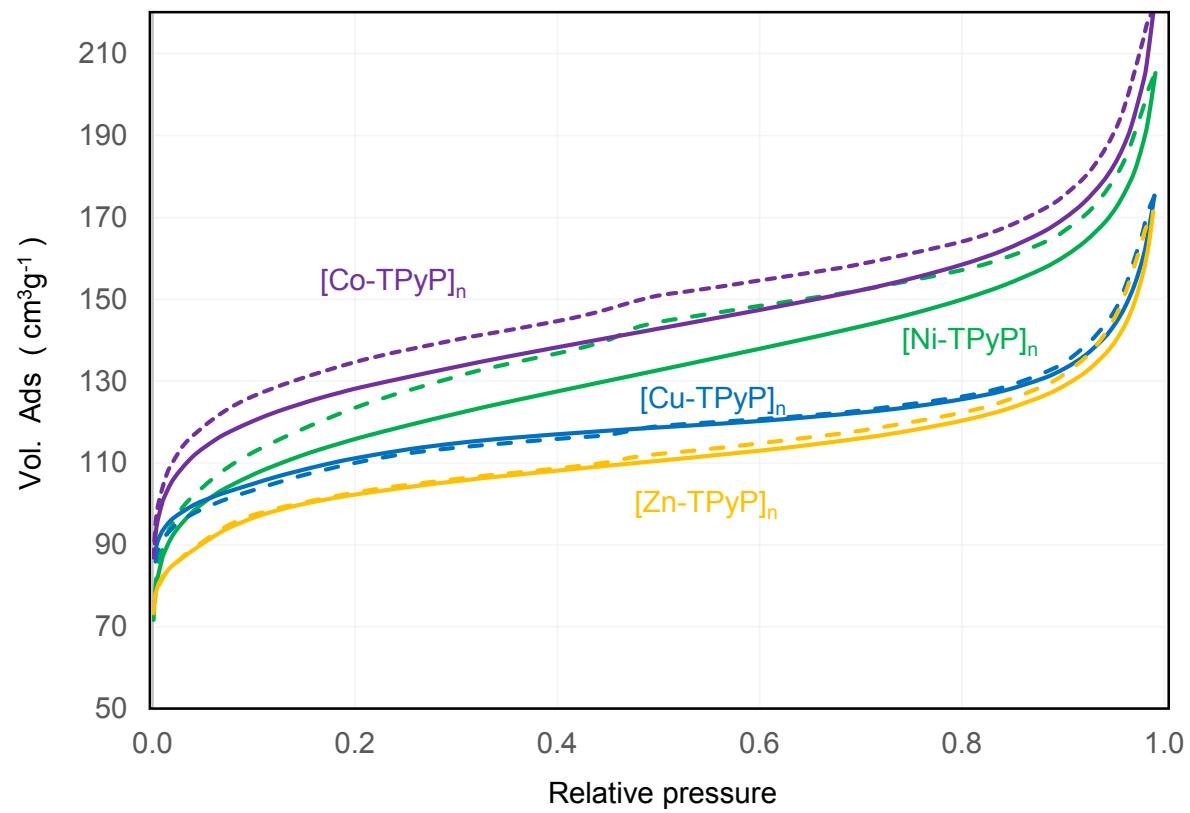


Figure S5. N₂ adsorption-desorption isotherms at -196 °C for all samples.

Table S2. Experimentally measured C, H and N molar percentages obtained by elemental analysis, presented together with metal molar percentages by ICP for the samples precipitated in scCO₂.

Sample	Atom	Exp. data	Derived viable formula
[Cu-TPyP] _n	C	46.44	$[(\text{Cu}(\text{hfac})_2)_2\text{Cu}_{0.32}(\text{H}_2)_{0.67}\text{TPyP}]_n$
	H	1.96	1.0 % H ₂ TpyP
	N	6.99	
	Cu	9.22	
[Zn-TPyP] _n	C	46.74	$[(\text{Zn}(\text{hfac})_2)_2\text{Zn}_{0.30}(\text{H}_2)_{0.70}\text{TPyP}]_n$
	H	1.94	
	N	7.40	
	Zn	9.31	
[Co-TPyP] _n	C	44.45	$[(\text{Co}(\text{hfac})_2)_2\text{Co}_{0.70}(\text{H}_2)_{0.30}\text{TPyP}]_n$
	H	2.00	
	N	6.40	
	Co	9.24	
[Ni-TPyP] _n	C	44.64	$[(\text{Ni}(\text{hfac})_2)_2\text{NiTPyP}]_n$
	H	1.67	3.5 % Ni(hfac) ₂
	N	6.63	
	Ni	14.31	