

The surface chemistry of graphene oxide: Enhancing the performance of ZnO – GO photocatalysts

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We demonstrate that the surface chemistry of graphene oxide (GO) is of critical importance for the photocatalytic performance of semiconducting metal oxides such as TiO₂ or ZnO. To this end we employ GO with distinct and well-defined oxidation degree for the preparation of ZnO-GO hybrid materials. We show that the surface chemistry of GO is key for facilitating favorable interface interactions. This results in a lowering of the alignment of energy levels at the ZnO-GO interface and enables the transfer of photo-excited electrons from ZnO to GO. The remaining positive charge at the valence band of ZnO then easily can promote oxidation reactions with the analyte and/or the solvent. In this way, GO reduces the recombination rate of the photogenerated electron-hole pair and enhances the photocatalytic activity of ZnO. We show that by solely controlling the oxidation degree of GO and by optimizing the loading fraction highly efficient ZnO-GO photocatalysts can be prepared. Most favorable conditions are achieved for intermediate oxidation degrees of GO. Optimized ZnO-GO hybrids thus reveal photocatalytic degradation of methylene blue at conversion rates of up to 80%, within times as short as 70 minutes at concentrations as low as 0.045 mg/mL, accompanied by a good recyclability behavior. The results are of general character and extendable to semiconducting metal oxides used in photocatalysis [1].

References

1. S. Victor-Román, E. García-Bordejé, J. Hernández-Ferrer, J.M. González-Domínguez, A. Ansón-Casaos, A.M.T. Silva, W.K. Maser, A.M. Benito, *Catalysis Today* 2019. DOI: 10.1016/j.cattod.2019.05.049

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