Femtosecond XUV induced dynamics of the methyl iodide cation

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Abstract. XUV wavelength-selected pulses obtained with high harmonic generation are studies of cation dynamics with state-by-state resolution. We demonstrate this by pump-probe experiments on CH_3I^+ cations and identify both resonant and non-resonant dynamics.

1 Introduction

Ultrashort extreme ultraviolet (XUV) pulses obtained from high harmonic generation (HHG) allow for studies of dynamics in molecular cations. Typical experiments using HHG have excellent time-resolution and can access processes such as charge migration [1], evolution of electronic and vibrational wavepackets [2], and fast decays through conical intersections [3]. However, in experiments using the full bandwidth of HHG sources, signals from different states often are mixed, which makes interpretation difficult. Here, we employ a time-delay-compensating XUV monochromator [4] which defines the photon energy and allows to study the dynamics state by state and also to distinguish processes in cations from those in highly excited neutrals, because the former are present with all harmonics of sufficient energy and the latter are resonant and thus only present for a certain XUV photon energy. The monochromator delivers selected harmonics, obtained from the output of an HHG-source, with a spectral bandwidth of 300 meV and a time duration of 20-25 fs. To perform pump-probe experiments, the wavelength-selected XUV beam is recombined with 25 fs IR pulses. In the present contribution we investigate cationic states of the CH₃I molecule. The dissociation of the CH₃I⁺ cation along the C-I axis is mainly accompanied by a geometry change in the methyl part from pyramid to planar and the spin-

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orbit splitting in the iodine atoms [5]. The latter induces avoided crossings in the manifold of potential energy surfaces which make CH₃I⁺ dissociation complex and interesting.

2 Results and Discussion

In the experiments, we used three harmonics: 7, 9, and 11 of the 800 nm fundamental driving pulse. With $E_{ph} = 10.9 \text{ eV}$ (harmonic 7), only the spin-orbit split ground state (X ${}^{2}E_{3/2,1/2}$) can be populated. With the next harmonic $E_{ph} = 14.0 \text{ eV}$ (harmonic 9), the weakly bound first excited state (A ${}^{2}A_{1}$) is populated as well, while harmonic 11 ($E_{ph} = 17.1 \text{ eV}$) also reaches the repulsive B ${}^{2}E$ state. Note that at these photon energies also neutral Rydberg series converging to the respective excited cations can be resonantly excited, leading to fluorescence, auto-ionization, or neutral dissociation processes. For each harmonic we record ion velocity map images for both the CH₃I⁺ and I⁺ fragment, observing both dynamics in the cation as well as dynamics of resonant neutral states.

With $E_{ph} = 10.9$ eV photons, no decay dynamics in the cation are expected upon ionization to the X ${}^{2}E_{3/2,1/2}$ ground state as the states are bound and their vibrational activity is negligible [6]. However, in the experiment we observe an enhancement of the signal at temporal overlap followed by a decay. This behavior cannot be explained by cation dynamics, and therefore must come from dynamics in the neutral molecule upon resonant absorption. This is supported by the fact that these dynamics are not observed for higher harmonics. Previous photoabsorption studies indeed have identified Rydberg series converging to the A ${}^{2}A_{1}$ and B ${}^{2}E$ potential energy surfaces [7]. Calculations are currently in progress to establish with certainty which of these states are responsible for the observed behavior and to explain the observed 90 fs decay timescale.



Fig 1. Time-dependent kinetic energy release maps of the CH_3^+ (a) and I⁺ (b) obtained with a 17.2 eV XUV pump pulse and an IR probe pulse. A positive delay means that the IR pulse comes after the XUV pulse.

For photon energies of 14.0 and 17.1 eV we obtain very comparable data, both different from 10.9 eV. Figure 1 displays maps of the total kinetic energy release (KER) as a function of time delay for dissociation into $I + CH_3^+$ and $CH_3 + I^+$, obtained for the $E_{ph} = 17.1$ eV pump and IR probe scheme. When IR comes after XUV, the yield of CH_3^+ fragments decreases while the yield of I⁺ fragments is enhanced. On top of this long-lived contribution, clear oscillatory signals are observed in both ionic fragment yields. The period of this oscillation is 127 ± 3 fs and the oscillations in the CH_3^+ and I^+ yield are in opposite phase with respect to each other.

The observed frequency closely corresponds to the frequency of the C-I stretching mode in the A ${}^{2}A_{1}$ state [8]. Therefore we interpret the observation as follows. The A ${}^{2}A_{1}$ state is populated at the inner turning point of the potential well, launching a wavepacket composed of several vibrational states. In the XUV-only experiment the most common

dissociation pathway would be a non-adiabatic coupling to hot vibrational bands in the spin-orbit split ground state of the cation X ${}^{2}E_{3/2,1/2}$. The molecule then statistically dissociates producing CH₃⁺. The latter mechanism is consistent with the discrepancy between lifetime of the A ${}^{2}A_{1}$ state (10⁻¹⁰ s) and the dissociation rate constant 10⁻⁷ s) [8,9]. The low total KER for the dissociation into I + CH₃⁺ is also consistent with this scenario.

The IR pulse couples the A ${}^{2}A_{1}$ state population to a repulsive potential energy surface leading to dissociation into CH₃ + I⁺. As this state serves as a common final state for the coherently populated vibrational levels in the A ${}^{2}A_{1}$ state, these pathways interfere with a beating frequency equivalent to the energetic separation between those states. The fact that dissociation into CH₃ + I⁺ occurs with a nonzero KER corresponds to a prompt dissociation process. Inherently, the I + CH₃⁺ dissociation channel should exhibit the same oscillation, but with opposite sign.

To complete the picture, it is necessary to know the exact probing mechanism induced by the IR. The potential energy surfaces of reference [5] can qualitatively describe the probing step. Further calculations are underway to obtain further insight into the role of the IR field on the coupling to dissociative states.

3 Conclusion

We used wavelength-selected XUV pulses to study the lowest states of the CH_3I^+ cation. We observed dynamics upon resonant absorption by members of Rydberg series converging to these states as well as a vibrational wavepacket that can be used to steer the dissociation to either the $I + CH_3^+$ or the $CH_3 + I^+$ channel depending on the phase of the wavepacket at the arrival time of the probe pulse. In the future, we aim to quantify this control process with the help of potential energy surface calculations on which a wavepacket will be propagated in the presence of the IR field.

References

[1] F. Calegari, D. Ayuso, A. Trabattoni, L. Belshaw, S. De Camillis, S. Anumula, F. Frassetto, L. Poletto, A. Palacios, P. Decleva, J. B. Greenwood, F. Martin, and M. Nisoli, Science **346**, 336-339 (2014).

[2] F. Kelkensberg, C. Lefebvre, W. Siu, O. Ghafur, T. T. Nguyen-Dang, O. Atabek, A. Keller, V. Serov, P. Johnsson, M. Swoboda, T. Remetter, A. L'Huillier, S. Zherebtsov, G. Sansone, E. Benedetti, F. Ferrari, M. Nisoli, F. Lepine, M. F. Kling, and M. J. J. Vrakking, Phys. Rev. Lett. **103**, 123005 (2009).

[3] M. C. E. Galbraith, S. Scheit, N. V. Golubev, G. Reitsma, N. Zhavoronkov, V. Despre, F. Lepine, A. I. Kuleff, M. J. J. Vrakking, O. Kornilov, H. Koeppel, and J. Mikosch, Nature Communications. **8**, 1018 (2017).

[4] M. Eckstein, C.-H. Yang, M. Kubin, F. Frassetto, L. Poletto, H.-H. Ritze, M. J. J. Vrakking, and O. Kornilov, J. Phys. Chem. Lett. **6**, 419–425 (2015).

[5] S. Marggi Poullain, D. V. Chicharro, J. Gonzalez-Vazquez, L. Rubio-Lago, and L. Banares, Phys. Chem. Chem. Phys. **19**, 7886–7896 (2017).

[6] L. Karlsson, R. Jadrny, L. Mattsson, F. Chau, and K. Siegbahn, Phys. Scr. 16, 225–234 (1977).

[7] R. Locht, B. Leyh, H. W. Jochims, and H. Baumgaertel, Chem. Phys. **365**, 109–128 (2009).

[8] S. Goss, D. Mcgilvery, J. Morrison, and D. Smith, J. Chem. Phys. **75**, 1820–1828 (1981).

[9] D. Mintz and T. Baer, J. Chem. Phys. 65, 2407–2415 (1976).