

Anharmonic potential functions as derived from Raman intensities: Methane

Cite as: J. Chem. Phys. **81**, 3835 (1984); <https://doi.org/10.1063/1.448166>

Submitted: 30 November 1983 . Accepted: 19 June 1984 . Published Online: 31 August 1998

D. Bermejo, and S. Montero



View Online



Export Citation

ARTICLES YOU MAY BE INTERESTED IN

[Calculation of converged rovibrational energies and partition function for methane using vibrational-rotational configuration interaction](#)

The Journal of Chemical Physics **121**, 2071 (2004); <https://doi.org/10.1063/1.1759627>

[Vibrational zero-point energies and thermodynamic functions beyond the harmonic approximation](#)

The Journal of Chemical Physics **120**, 3059 (2004); <https://doi.org/10.1063/1.1637580>

[Vibrational Energy Transfer in Methane](#)

The Journal of Chemical Physics **49**, 1111 (1968); <https://doi.org/10.1063/1.1670199>

Lock-in Amplifiers
up to 600 MHz



Anharmonic potential functions as derived from Raman intensities: Methane^{a)}

D. Bermejo and S. Montero

Instituto de Estructura de la Materia, CSIC, Serrano 119, Madrid (6), Spain

(Received 30 November 1983; accepted 19 June 1984)

The cubic force field of methane has been investigated on the basis of the analysis of Raman scattering cross sections of binary overtones and combination bands. Seven absolute cross sections, ten symmetry cubic force constants, and the second derivative of the mean CH bond polarizability are the original results of present work. Some of these cubic force constants are determined with high accuracy. Most \hat{H}_{22} rovibrational constants calculated by means of present anharmonic force field are in good agreement with the experimental values reported by various authors.

INTRODUCTION

Deriving the anharmonic vibrational potential function of polyatomic molecules is one of the most difficult problems of present day spectroscopic research. For most small polyatomic molecules, attempts have been made to establish the first coefficients of the potential function, namely, the quadratic, cubic, and quartic force constants. The problem is so complex that only in very few cases, CO₂ for instance,¹ these force constants are firmly established.

In the case of methane, the cubic force field has been investigated by various authors,²⁻⁸ but the lack of agreement is considerable for most cubic force constants. Besides that, for the quadratic force field still are doubts arising from the $\nu_2:\nu_4$ Coriolis resonances in CH₄ and CD₄ and $\nu_2 + \nu_4:\nu_1$ in CD₄. About the quartic force constants almost nothing is known.

With few exceptions,⁹ the force field has been hitherto investigated according to two different procedures, a theoretical one, based on *ab initio* calculations, and an empirical one based on the analysis of the experimental wave numbers of the rotation-vibration bands, in connection with an approximate expression for the Hamiltonian operator. For methane, the first method was used in Refs. 2-4 and the second in Refs. 6 and 7.

A different approach has been followed in present work, since the main source of information has been the experimental Raman scattering cross sections of first and second order vibrational spectrum, using the bond polarizability model in connection with the anharmonic treatment of Raman intensities to the cubic term. With little theoretical effort, the method might be extended to the quartic term, but the necessary experimental intensity data are not yet available.

The method here proposed seems to be quite efficient, in so far as the Raman cross sections of overtones and combination bands are extremely sensitive to the vibrational interactions (mainly Fermi resonances) governed by cubic force constants. For these bands, changes of up to three orders of magnitude with respect to the calculated unperturbed cross section are observed. In such cases, the cross section of the Fermi diad $\nu_a:2\nu_b$ or $\nu_a:\nu_b + \nu_c$ yields highly localized information on the cubic force constant k_{abb} or k_{abc} , respec-

tively.

Ten cubic force constants in symmetry coordinates F_{ijk} common to all isotopic derivatives of methane have been derived here by using the absolute cross sections of 14 overtone or combination bands of the species ¹²CH₄, ¹²CD₄, ¹²CH₃D, and ¹²CH₂D₂. Since the possible number of useful intensity data on methane isotopic species is well over 200, the possibilities of the present method for establishing the complete set of 13 symmetry cubic constants F_{ijk} are still very large.

As an intermediate result of present work, all second derivatives of the mean molecular polarizability of methane species with respect to symmetry and normal coordinates were obtained. They may be useful for further refinement of the cubic force field by including new intensity data. Those of ¹²CH₄, ¹³CH₄, ¹²CD₄, and ¹³CD₄ are reported in Table II.

EXPERIMENTAL

Commercial samples of methane deuterated derivatives (Stohler, Isotopical Center) were used for recording most spectra. For CD₄, a new sample was prepared by us, in order to verify some features in the $\nu_3 + \nu_4$ band, which in a preliminary analysis of the data were attributed to impurities of methylacetylene or some of its deuterated derivatives.

The spectra were recorded with a 25-100 Jarrell Ash double monochromator by using a 90° single pass configuration. The relative cross sections of the second order bands were measured with respect to an appropriate fundamental of the same sample and were then scaled by using the absolute values of Refs. 10 and 11. Exciting laser power of up to 9 W at 488 nm was necessary to measure the very weak $2\nu_1$ and $2\nu_3$ overtones of ¹²CH₄ and ¹²CD₄. The gas samples were contained in a cylindrical glass cell of low fluorescent material.

The experimental cross sections and wave numbers are reported in Table I, and the spectra of the $2\nu_1$, $2\nu_3$, and $\nu_3 + \nu_4$ bands of ¹²CH₄ are shown in Figs. 1 and 2.

THEORETICAL BACKGROUND

In order to establish a relationship between the differential Raman scattering cross sections and the cubic force constants several intermediate steps are necessary.

In present work, the experimental cross sections related

^{a)} Issued as CSIC No. FM49

TABLE I. Experimental and calculated differential cross sections, $(\partial\sigma/\partial\Omega)$, for the mean polarizability Raman scattering of vibrational transitions in methane species. $(\partial\sigma/\partial\Omega)$ at 488 nm and 300 K, in units of $10^{-39} [\text{m}^2 \text{sr}^{-1}]$.

		$\nu[\text{cm}^{-1}]$	Experimental	Calculated		
				(a)	(b)	(c)
$^{12}\text{CH}_4$	$2\nu_1(A_1)$	{ 5769	54 ± 10	53	58	231
		5790				
		5804				
	$2\nu_2(A_1)$	3066	11000 ± 3000^e	{ 9420 9467(P)	{ 9420 9467(P)	1746
	$2\nu_3(A_1)$	{ 5945 5969	81 ± 16	82	82	61
	$2\nu_4(A_1)$	2587	3900 ± 800	{ 9252 4983(P)	{ 9225 4989(P)	5810
$\nu_3 + \nu_4(A_1)$	{ 4314 4342	180 ± 90	117	126	71	
	$\nu_1(A_1)$	2917	439000 ± 24000			
$^{12}\text{CD}_4$	$2\nu_1(A_1)$	{ 4153 4188	(15-60) ^d	40	44	176
	$2\nu_2(A_1)$	2184	7160 ± 2000^e	{ 5750 8129(P)	{ 5750 8129(P)	1069
	$2\nu_3(A_1)$	{ 4455 4486	(15-60) ^d	10	165	381
	$2\nu_4(A_1)$	1964	27000 ± 5000^e	{ 101300 27588(P)	{ 102120 27966(P)	3724
	$\nu_3 + \nu_4(A_1)$	3877	920 ± 230	425	407	150
	$\nu_1(A_1)$	2108	313000 ± 16000			
$^{12}\text{CH}_3\text{D}$	$2\nu_{4b}(A_1)$	2316	20800 ± 4000^e	{ 2608 15800(P)	{ 2608 15800(P)	3270
	$2\nu_2(A_1)$	2910	205000 ± 40000^e	213000(P)	213000(P)	2020
	$2\nu_{4a}(A_1)$	2596	2800 ± 500^e	2870	2870	1900
	$\nu_{3a}(A_1)$	2967	326000 ± 33000			
	$\nu_1(A_1)$	2200	148000 ± 8000			
$^{12}\text{CH}_2\text{D}_2$	$2\nu_{2a}(A_1)$	2860	9500 ± 2000^e	{ 8970 12050(P)	{ 8970 12050(P)	1295
	$\nu_{3a}(A_1)$	2975	237000 ± 48000			

^a With cubic field (a) of Table III and $(2\alpha'' + \gamma'')_{\text{CH}} = (4.81 \pm 1.0) \times 10^{-20} [\text{C V}^{-1}]$.

^b With cubic force field (b) of Table III and $(2\alpha'' + \gamma'')_{\text{CH}} = (8.0 \pm 1.2) \times 10^{-20} [\text{C V}^{-1}]$.

^c Harmonic approximation.

^d Lower and upper limits.

^e Reference 11.

(P): Values calculated with the theory of perturbations [Eqs. (4) and (5)]. Otherwise results are from the contact transformation method [Eqs. (2) and (3)].

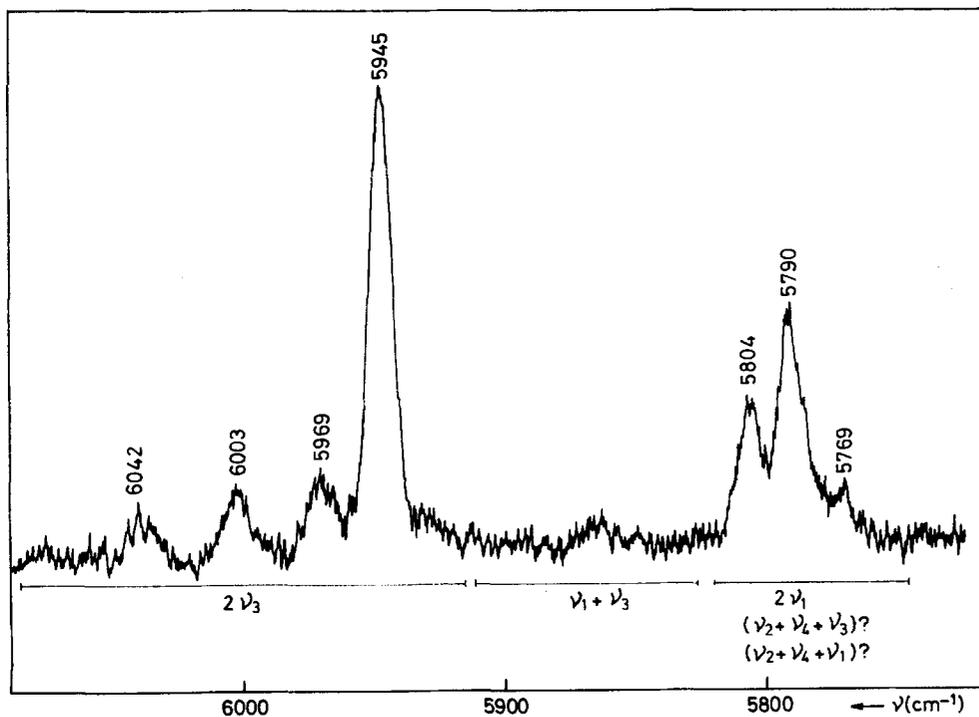


FIG. 1. Gas phase Raman spectrum of stretching overtones of $^{12}\text{CH}_4$. Cross section proportional to $45\bar{\alpha}^2 + 7\gamma^2$ (no analyzer); pressure ~ 2 atm; laser power ~ 8 W; exciting line = 488 nm. Peaks at 5969, 5945, 5804, and 5790 cm^{-1} are strongly polarized. Peaks at 6003 and 6042 cm^{-1} are depolarized.

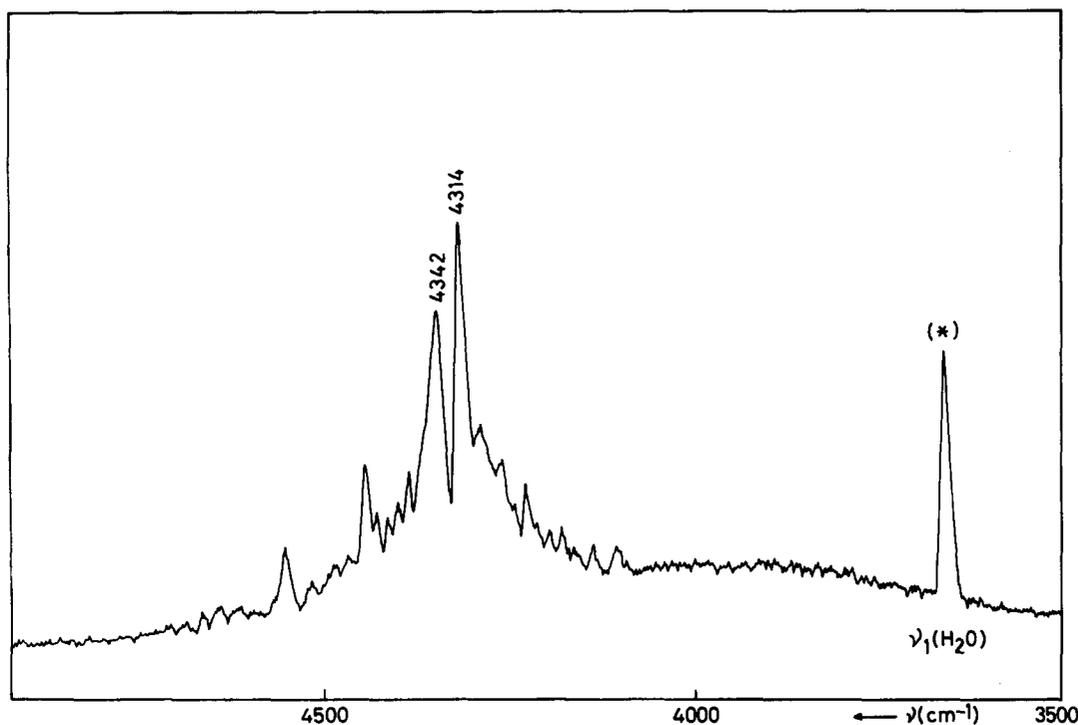


FIG. 2. Gas phase Raman spectrum of $^{12}\text{CH}_4$ in the region of $\nu_3 + \nu_4$. Experimental conditions as in Fig. 1. (*) H_2O impurity of the $^{12}\text{CH}_4$ sample. The broad background centered at about 4000 cm^{-1} is due to the fluorescence of the cell.

to the mean polarizability scattering were used, since they usually appear as sharp bands not obscured by overlapping of the rovibrational structures.

In the SI system,¹² the cross section for the $i \rightarrow j$ vibrational transition is

$$\left(\frac{\partial\sigma}{\partial\Omega}\right)_{i \rightarrow j}^{\text{mean pol.}} = (\pi/\epsilon_0)^2 (\nu_0 - \nu_j)^4 \langle i|\bar{\alpha}|j\rangle^2 [Q_{\text{vib}}(T)]^{-1} \times \exp(-E_i/kT), \quad (1)$$

where $\epsilon_0 = 8.8542 \times 10^{-12} [\text{C V}^{-1} \text{ m}^{-1}]$ is the permittivity of vacuum and ν_0 and ν_j the wave numbers of the exciting radiation and of the Raman shift, respectively. $Q_{\text{vib}}(T)$ is the vibrational partition function of the molecule at the temperature T of the sample. $|i\rangle$ and $|j\rangle$ are the vibrational wave functions for the initial (i) and final (j) vibrational states of the transition, and $\bar{\alpha}$ is the mean molecular polarizability operator. For methane species only the transition $0 \rightarrow j$ give an appreciable contribution to the vibrational scattering cross section at room temperature.

For practical reasons, the relevant transition moments of the type $\langle 0|\bar{\alpha}|j\rangle$ have been classified here in two groups:

Weak vibrational resonances

In these cases, the final level of an overtone or combination transition j is far from any fundamental level of the molecule and the observed cross section, $(\partial\sigma/\partial\Omega)_{0 \rightarrow j}^{\text{mean pol.}}$, is usually below about 15% of the cross section of the closest fundamental bands.

Under these conditions the contact transformation of the polarizability operator¹³ yields a good convergence and the transition moment of the mean polarizability corresponding to an overtone $2\nu_i$ is given by

$$\langle 0|\bar{\alpha}|j\rangle_{2\nu_i} = \frac{1}{2\sqrt{2}} \left[\left(\frac{\partial^2 \bar{\alpha}}{\partial q_i^2} \right)_0 + \frac{2k_{iii}}{\omega_i} \left(\frac{\partial \bar{\alpha}}{\partial q_i} \right)_0 + \sum_{m \neq i} \frac{2k_{iim}\omega_m}{(4\omega_i^2 - \omega_m^2)} \left(\frac{\partial \bar{\alpha}}{\partial q_m} \right)_0 \right], \quad (2)$$

while for a combination $\nu_i + \nu_{i'}$ one obtains

$$\langle 0|\bar{\alpha}|j\rangle_{\nu_i + \nu_{i'}} = \frac{1}{2} \left[\left(\frac{\partial^2 \bar{\alpha}}{\partial q_i \partial q_{i'}} \right)_0 + \frac{2k_{iii}\omega_i}{\omega_i(2\omega_i + \omega_{i'})} \left(\frac{\partial \bar{\alpha}}{\partial q_i} \right)_0 + \frac{2k_{i'i'i}}{\omega_{i'}(2\omega_{i'} + \omega_i)} \left(\frac{\partial \bar{\alpha}}{\partial q_{i'}} \right)_0 + \sum_{m \neq i, i'} \frac{k_{iim}\omega_m}{[(\omega_i + \omega_{i'})^2 - \omega_m^2]} \left(\frac{\partial \bar{\alpha}}{\partial q_m} \right)_0 \right]. \quad (3)$$

Here, $\partial \bar{\alpha} / \partial q_i$ and $\partial^2 \bar{\alpha} / \partial q_i \partial q_{i'}$ are derivatives of the mean molecular polarizability with respect to dimensionless normal coordinates q ; ω_i are the harmonic wave numbers and k_{ijk} the cubic force constant in the representation of the coordinates q .

Provided the cubic constants k_{iim} or $k_{i'i'm}$ are not vanishing by symmetry, the resonant terms $(4\omega_i^2 - \omega_m^2)^{-1}$ and $[(\omega_i + \omega_{i'})^2 - \omega_m^2]^{-1}$ in Eqs. (2) and (3) clearly show that $2\nu_i$ and $\nu_i + \nu_{i'}$ always are in vibrational resonance with the Raman active fundamentals ν_m . These weak resonances have remarkable effects on the observed cross sections of second order transitions. Even in those cases where the difference $2\nu_i - \nu_m$ or $\nu_i + \nu_{i'} - \nu_m$ is larger than 1000 cm^{-1} , the observed cross sections may change by more than one order of magnitude with respect to the unperturbed value.¹⁴

Strong Fermi resonances

If the final level of an overtone transition $2\nu_l$ is close to a fundamental level of appropriate symmetry, its observed cross section may be larger than 15% of that of the fundamental band. For such a strong resonance Eq. (2) is no longer accurate because of the singularity $\nu_m = 2\omega_l$. Fortunately, only the interaction between these two levels needs to be taken into account in this case and the following expressions, derived from an exact diagonalization in the frame of the theory of perturbations¹⁵ may be safely used:

$$\langle 0|\bar{\alpha}|u\rangle = (1/2)\left[\left(\frac{\partial\bar{\alpha}}{\partial q_m}\right)_0\sqrt{1+t}\right. \\ \left.\pm(\sqrt{g}/2)\left(\frac{\partial^2\bar{\alpha}}{\partial q_l^2}\right)_0\sqrt{1-t}\right], \quad (4)$$

$$\langle 0|\bar{\alpha}|l\rangle = (1/2)\left[\left(\frac{\partial\bar{\alpha}}{\partial q_m}\right)_0\sqrt{1-t}\right. \\ \left.\mp(\sqrt{g}/2)\left(\frac{\partial^2\bar{\alpha}}{\partial q_l^2}\right)_0\sqrt{1+t}\right]. \quad (5)$$

Here, u and l are the two final levels of the transition at upper and lower wave numbers, respectively; $g = 1, 2$, or 3 is the degeneracy of mode q_l , and t is defined by the expression

$$t = \pm \left[1 - g\left(\frac{k_{ilm}}{X}\right)^2\right]^{1/2}, \quad (6)$$

where $|X|$ is the observed splitting of the Fermi diad. The upper signs set (\pm) in Eqs. (4) and (5) stand for the case $k_{ilm} > 0$, and the lower (\mp), for $k_{ilm} < 0$. The factor t must be taken positive if $\omega_m^* - 2\omega_l^* > 0$, and negative if $\omega_m^* - 2\omega_l^* < 0$; ω_m^* and ω_l^* are the "unperturbed" wave numbers, i.e., the harmonic wave numbers corrected by the anharmonicity constants $x_{ss'}$ according to the expression

$$\Delta E = \sum_{s < s'} x_{ss'} \left(\nu_s + \frac{g_s}{2}\right) \left(\nu_{s'} + \frac{g_{s'}}{2}\right) \quad (7)$$

for the energy increment of the unperturbed levels. For tetrahedral methane species the derivatives $\partial\bar{\alpha}/\partial q_l$ and $\partial^2\bar{\alpha}/\partial q_l\partial q_l$ in Eqs. (2) to (5) have been deduced from Table IV of Ref. 13 by using the transformation from mass weighted coordinates Q to dimensionless coordinates q ,

$$(\partial\alpha/\partial q_l) = \sqrt{2}b_l(\partial\alpha/\partial Q_l), \quad (8)$$

$$(\partial^2\alpha/\partial q_l\partial q_l) = 2b_l b_l' (\partial^2\alpha/\partial Q_l^2). \quad (9)$$

$b_l = (\hbar/8\pi^2\omega_l c)^{1/2}$ is the zero vibrational amplitude. For nontetrahedral species the corresponding derivatives were calculated with the FORTRAN program OVER by using the same set of bond polarizability parameters.²⁰

It should be noticed that, in the limit of negligible anharmonic interactions, i.e., when all k_{ijk} cubic constants are negligibly small, Eq. (2) and Eqs. (4) or (5) are strictly equivalent. However, for finite values of the k_{ijk} cubic constants, the results may be different, even for weak resonances. This is expected to be so, since Eq. (2) includes the interactions with all vibrational levels and Eqs. (4) and (5) only include the interaction with the closest vibrational level. Furthermore, Eq. (2) originates from the first two terms of the contact transformed polarizability in the expansion

$$\alpha_{\text{eff}} = \alpha + i[S, \alpha] - \frac{1}{2}[S, [S, \alpha]] \\ - (i/6)[S, [S, [S, \alpha]]] + \dots, \quad (10)$$

where the contribution of the neglected terms $\frac{1}{2}[S, [S, \alpha]] + (i/6)[S, [S, [S, \alpha]]]$ is expected to be small, while Eqs. (4) and (5) arise from the exact diagonalization of a 2×2 perturbation Hamiltonian.¹⁵ The numerical results from both methods (a) and (b) are discussed in more detail in the final part of this work.

NUMERICAL CALCULATIONS

According to previous expressions and to Ref. 13, the observed Raman scattering cross sections of second order bands provide information about the quadratic and cubic terms of the force field and about the equilibrium and the first and second derivatives of the CH bond polarizability tensor.

For calculating the scattering cross section due to the mean polarizability of methane, only the bond parameters $(2\alpha' + \gamma')_{\text{CH}}$ and $(2\alpha'' + \gamma'')_{\text{CH}}$ are necessary. The parameter¹⁰

$$(2\alpha' + \gamma')_{\text{CH}} = 3.787 \times 10^{-30} [\text{C V}^{-1} \text{ m}],$$

and the quadratic force field⁷ have been taken from the literature and are supposed to be quite accurately determined. Its refinement has not been attempted here. The parameter $(2\alpha'' + \gamma'')_{\text{CH}}$ has been determined and 10 cubic force constants in symmetry coordinates have been refined by using experimental cross section data of Ref. 11 and of present work, according to the following procedure:

From Eq. (1), the square of the transition moment $\langle 0|\bar{\alpha}|j\rangle$ can be derived from the experimental cross section of the corresponding Raman band. By substituting the two possible experimental values of the transition moment (differing in sign) in Eqs. (2), (3), (4), or (5), it was possible to derive either two possible "experimental" values for each k_{rst} [Eqs. (4), (5), and (6)], or two possible relations between the various experimental k_{rst} 's appearing in Eqs. (2) or (3). From these values and relations, and by using the transformation

$$k_{rst} = \frac{2\sqrt{2}b_r b_s b_t}{(1 + \delta_{rs})(1 + \delta_{rt} + \delta_{st})} \\ \times [F_{ijk} L_i^r L_j^s L_k^t + F_{ij}(L_i^r L_j^t + L_i^s L_j^t + L_i^t L_j^s)], \quad (11)$$

it was possible to express the experimental information as a system of 14 linear equations with 10 symmetry force constants F_{111} , F_{122} , F_{133} , F_{134} , F_{144} , F_{234} , F_{244} , F_{334} , F_{344} , and F_{444} , and the parameter $(2\alpha'' + \gamma'')_{\text{CH}}$ as unknowns. Due to the uncertainty in the sign of the transition moments the system had a multiplicity of numerical solutions.

On the other hand, the structure of the system was such that the constants F_{111} , F_{122} , F_{133} , F_{134} , F_{144} , and the parameter $2\alpha'' + \gamma''$ were overdetermined, but only four constants among F_{234} , F_{244} , F_{344} , F_{444} , and F_{334} could be simultaneously determined from the available experimental data. In order to get a value for these five constants, the condition of minimal deviation from the previously reported values²⁻⁷ was

explicitly imposed. Under these conditions only two satisfactory solutions were found. They are reported in Table III.

It is worthwhile to point out that imposing external constraints was necessary just as a consequence of the reduced number of experimental data used in present work, but is not a concomitant limitation of the proposed method. In fact, over 200 significant intensity data are in principle observable, and should be used in the future refinement of the cubic force field of methane.

DISCUSSION

It is interesting to note that most of the refined F_{ijk} 's of Table III appear to be well determined, despite the relatively low accuracy of the experimental cross sections. This is due to the high sensitivity of the transition moments [Eqs. (2)–(5)] vs the cubic constants k_{rst} . This effect is enhanced by the fact that the scattering cross sections depend on the square of the transition moments, and so, variations of a few percent in the cubic force constants may induce changes in some of the calculated cross sections of more than one order of magnitude. However, it should be emphasized that the uncertainties quoted for F_{ijk} 's in Table III are not necessarily the actual uncertainty of the physical magnitude, since the method here used is based on the model of bond polarizabilities. We believe that the derivatives $(\partial\bar{\alpha}/\partial q_i)_0$ and $(\partial^2\bar{\alpha}/\partial q_i\partial q_{i'})_0$ on

Table II, which were calculated with the bond polarizability model, are accurate to nearly 10% but systematic deviations larger than this value cannot be excluded, at present. If this is the case, systematic errors in some of the $(\partial\bar{\alpha}/\partial q_i\partial q_{i'})_0$ should propagate in the numerical calculation yielding not too accurate values for the F_{ijk} 's but, probably, maintaining a good consistency for the overall system. Since the values here used for $(\partial^2\bar{\alpha}/\partial q_i\partial q_{i'})_0$ cannot be directly checked, for instance, by comparing them with *ab initio* results, the only way to estimate the accuracy of the F_{ijk} 's is either to compare them with values reported by other authors using a different methodology or by calculating spectroscopic constants that are known on the basis of the observed wave numbers of rovibrational bands.

Our two possible solutions (a) and (b) in Table III may be compared with F_{ijk} 's from Refs. 2–7. For solution (a), corresponding to the bond parameter $(2\alpha'' + \gamma'')_{\text{CH}} = (4.81 \pm 1.0) \times 10^{-20} [\text{C V}^{-1}]$, the constants F_{111} , F_{122} , F_{133} , F_{244} , and F_{444} agree well with the results reported by Gray and Robiette,⁷ but, for F_{134} and F_{144} our results are closer to those of the early work of Kuchitsu and Bartell.² For the remaining constants, the literature F_{ijk} values reported on Table III are so contradictory that no further conclusions about the quality of our results can be reached in

TABLE II. Derivatives ($D^{(n)} = \partial\bar{\alpha}/\partial q_i$ or $\partial^2\bar{\alpha}/\partial q_i\partial q_{i'}$) of the mean polarizability of T_d symmetry methane species with respect to the dimensionless normal coordinates of the harmonic force field of Ref. 7. $D^{(n)}$ in units of $10^{-40} [\text{C V}^{-1} \text{m}^2]$; ω 's are calculated wave numbers.

	¹² CH ₄			¹³ CH ₄			¹² CD ₄			¹³ CD ₄		
	ω_i	ω_j	$D^{(n)}$									
$\partial\bar{\alpha}/\partial q_1$	3025	...	0.2655	3025	...	0.2655	2140	...	0.2233	2140	...	0.2233
$\partial^2\bar{\alpha}/\partial q_1^2$	3025	3025	0.0177	3025	3025	0.0177	2140	2140	0.0125	2140	2140	0.0125
$\partial^2\bar{\alpha}/\partial q_{2\sigma}^2$ ($\sigma = a, b$)	1583	1583	0.0246	1583	1583	0.0246	1120	1120	0.0174	1120	1120	0.0174
$\partial^2\bar{\alpha}/\partial q_{3\sigma}^2$ ($\sigma = x, y, z$)	3157	3157	0.0190	3146	3146	0.0189	2336	2336	0.0147	2320	2320	0.0145
$\partial^2\bar{\alpha}/\partial q_{4\sigma}^2$ ($\sigma = x, y, z$)	1367	1367	0.0345	1359	1359	0.0343	1034	1034	0.0259	1025	1025	0.0257
$\partial^2\bar{\alpha}/\partial q_{3\sigma}\partial q_{4\sigma}$ ($\sigma = x, y, z$)	3157	1367	0.0033	3146	1359	0.0031	2336	1034	0.0042	2320	1025	0.0040

TABLE III. Cubic force constants of methane F_{ijk} in the representation of symmetry coordinates of Ref. 7.

	This work		Gray and Robiette (Ref. 7)	Hirota (Ref. 6)	Pulay, Mayer and Boggs (Ref. 4)	Bartell and Kuchitsu (Ref. 2)	Kuchitsu and Bartell (Ref. 3)
	(a)	(b)					
F_{111}^c	-15.32 ± 2.08	-12.06 ± 2.54	-15.30	-12.40	-15.08	-17.21	-19.30
F_{122}^c	-0.322 ± 0.020	-0.322 ± 0.022	-0.299	-0.33	-0.227	-0.193	-0.466
F_{133}^c	-15.58 ± 2.48	-15.80 ± 2.74	-15.69	-14.08	-15.47	-14.86	-15.74
F_{134}^d	0.347 ± 0.190	0.292 ± 0.200		0.25	0.066	0.180	0.637
F_{144}^e	-0.371 ± 0.020	-0.368 ± 0.022	-0.110	0.14	-0.196	-0.195	-0.466
F_{222}^f			0.094	0.058	0.098		
F_{233}^d				-0.22	-0.370		
F_{234}^e	0.258 ± 0.055	0.258 ± 0.063		0.29	0.160		
F_{244}^f	-0.319 ± 0.167	-0.319 ± 0.184	-0.311	-0.37	-0.386		
F_{333}^c			-15.87	-13.48	-15.65	-13.83	-13.96
F_{334}^d	-0.293 ± 0.755	-0.293 ± 0.831		0.18	-0.268		
F_{344}^e	0.045 ± 0.077	0.045 ± 0.083		-0.77	-0.101		
F_{444}^f	0.328 ± 0.245	0.328 ± 0.276	0.345	0.233	0.426		

^a For $(2\alpha'' + \gamma'')_{\text{CH}} = (4.81 \pm 1.00) \times 10^{-20} [\text{C V}^{-1}]$; (preferred solution).

^b For $(2\alpha'' + \gamma'')_{\text{CH}} = (8.00 \pm 1.20) \times 10^{-20} [\text{C V}^{-1}]$; (alternative solution).

^c aJ Å⁻³

^d aJ Å⁻²

^e aJ Å⁻¹

^f aJ

TABLE IV. Cubic force constants for T_d symmetry methane species, k_{rst} , in the representation of dimensionless normal coordinates, corresponding to solution (a) of Table III. k_{rst} in cm^{-1} .

	This work			Gray and Robiette (Ref. 7)	
	$^{12}\text{CH}_4$	$^{12}\text{CD}_4$	$^{13}\text{CH}_4$	$^{12}\text{CH}_4$	$^{12}\text{CD}_4$
k_{111}	-149.23	-88.77	-149.20	-149.33	-88.83
k_{122}	17.52	10.42	17.52	21.00	12.50
k_{133}	-480.40	-296.80	-478.80	-487.00	-302.50
k_{134}	13.18	41.19	8.68	-19.00	25.00
k_{144}	81.44	50.82	80.97	114.50	72.50
k_{222}	-8.87	-5.28	-8.87	-8.83	-5.33
k_{233}	-44.20	-34.21	-43.14	-46.00	-37.00
k_{234}	-119.17	-74.41	-118.59	-147.00	-92.00
k_{244}	-18.84	-9.02	-19.13	-18.00	-8.00
k_{333}	-171.62	-112.54	-170.25	-171.83	-112.83
k_{334}	-61.84	-33.23	-62.29	-63.00	-36.00
k_{344}	-56.61	-37.55	-56.09	-76.50	-50.00
k_{444}	-21.61	-12.73	-21.60	-21.17	-12.00

this way. Solution (b), for $(2\alpha'' + \gamma'')_{\text{CH}} = (8.0 \pm 1.2) \times 10^{-20} [\text{C V}^{-1}]$, yields slightly larger uncertainties. Consequently, we shall provisionally prefer solution (a).

The cubic constants k_{rst} in the representation of dimensionless normal coordinates corresponding to solution (a) are reported in Table IV for some tetrahedral methane species. They have been calculated with Eq. (11), and the constants F_{222} , F_{233} , and F_{333} , not refined in present work, were kept fixed to the *ab initio* values of Pulay, Mayer, and Boggs.⁴ The corresponding k_{rst} constants, reported by Gray and Robiette,⁷ are reproduced in the same table, the only significant differences being localized on the constants k_{134} , k_{144} , k_{234} , and k_{344} .

The \hat{H}_{22} spectroscopic constants Y_1 , Y_2 , Y_3 , Y_4 , Z_{3s} , Z_{4s} , Z_{3t} , Z_{4t} , and Z_2 according to the nomenclature of Hecht,¹⁶ were calculated for the derivatives $^{12}\text{CH}_4$, $^{12}\text{CD}_4$, $^{13}\text{CH}_4$, and $^{13}\text{CD}_4$ by using the k_{rst} force constants of Table IV. They are shown in Table V, together, with experimental results originating from different sources [Refs. 7, 8, and references therein]. Our \hat{H}_{22} constants are not too different from those calculated by Gray and Robiette,⁷ due to the fact that they mainly depend on k_{rst} constants that are similar in both force fields. On the other hand, for some of the \hat{H}_{22} constants, as for instance Z_{3t} in $^{12}\text{CH}_4$ and $^{13}\text{CH}_4$, the difference between theoretical and experimental results is quite large for any of the two independently determined cubic force fields. These discrepancies suggest that the use of \hat{H}_{22} coefficients might not always be safe for determining the cubic force field.

We have also calculated the F_2 , F_{3s} , F_{3t} , F_{4s} , and F_{4t} coefficients by using the expressions derived by Aliev and Watson.¹⁷ Our results are nearly the same as those reported by Gray and Robiette,⁷ since the influence of those terms depending on k_{134} , k_{144} , k_{234} , and k_{344} constants, which are different in both force fields, is quite small. Thus, the discrepancies between calculated and experimental F coefficients that were pointed out by Gray and Robiette cannot be satisfactorily explained on the basis to the inaccuracy of the cubic k_{rst} constants. More likely, the F coefficients depend on higher order vibration-rotation terms not included in the expressions of Aliev and Watson.

It has been already pointed out that weak Fermi diads might be alternatively described in two ways, with similar results. Among the spectra utilized in present work the following weak Fermi diads are present: $2\nu_2:\nu_1$ in $^{12}\text{CH}_4$ and $^{12}\text{CD}_4$; $2\nu_4:\nu_1$ in $^{12}\text{CH}_4$ and $^{12}\text{CD}_4$; $2\nu_{4b}:\nu_1$ in $^{12}\text{CH}_3\text{D}$ and $2\nu_{2a}:\nu_{3a}$ in $^{12}\text{CH}_2\text{D}_2$. The cross sections calculated accord-

TABLE V. Calculated and experimental \hat{H}_{22} coefficients for $^{12}\text{CH}_4$, $^{13}\text{CH}_4$, $^{12}\text{CD}_4$, and $^{13}\text{CD}_4$ (in cm^{-1}).

e	$^{12}\text{CH}_4$		$^{13}\text{CH}_4$		$^{12}\text{CD}_4$		$^{13}\text{CD}_4$
	calc ^a	expt	calc ^a	expt	calc ^a	expt	calc ^a
Y_1	-0.0382	-0.0377 ^d	-0.0382		-0.0136		-0.0136
Y_2	0.0877	{ 0.0879 ^b 0.0910 ^c	0.0822		0.0717	0.0639 ^b	0.0628
Y_3	-0.0347	-0.0357 ^d	-0.0342	{ -0.0400 ^b -0.0399 ^c	-0.0146	{ -0.0132 ^b -0.0134 ^c	-0.0142
Y_4	-0.0682	{ -0.0647 ^b -0.0635 ^c	-0.0649	{ -0.0598 ^b -0.0598 ^c	-0.0498	-0.0543 ^b	-0.0441
Z_{3s}	-0.0053	{ -0.0075 ^b -0.0076 ^c	-0.0047	{ -0.0066 ^b -0.0067 ^c	-0.0037	{ -0.0036 ^b -0.0038 ^c	-0.0030
Z_{4s}	0.0133	{ 0.0138 ^b 0.0135 ^c	0.0142	{ 0.0159 ^b 0.0158 ^c	-0.0023	-0.0060 ^b	-0.0006
Z_{3t}	-0.0019	{ -0.0064 ^b -0.0031 ^c	-0.0017	{ -0.0051 ^b -0.0026 ^c	-0.0013	-0.0009 ^b	-0.0012
Z_{4t}	-0.0293	{ -0.0303 ^b -0.0151 ^c	-0.0277	{ -0.0292 ^b -0.0146 ^c	-0.0221	-0.0201 ^b	-0.0195
Z_2	-0.0102	0.0396 ^c	-0.0138		0.0341		0.0282

^a Calculated using the k_{rst} force constants of present work [solution (a)] and the expressions of Hecht (Ref. 16).

^b Experimental values from different sources quoted in Ref. 7.

^c Experimental values from different sources quoted in Ref. 8.

^d Reference 19.

^e Equivalence between Hecht's (Ref. 16) and Herranz's (Ref. 18) nomenclature: $Y_i = -\alpha_i$; $i = 1, 2, 3, 4$; $Z_{jt} = -\delta_j/10$; $Z_{js} = 3\beta_j$, $j = 3, 4$.

ing to both methods are shown in Table I. There, P stands for the exact diagonalization. Otherwise results are from the contact transformation method. It may be seen that the results are similar only for $2\nu_2:\nu_1$ in $^{12}\text{CH}_4$ and $^{12}\text{CD}_4$ and for $2\nu_{2a}:\nu_{3a}$ in $^{12}\text{CH}_2\text{D}_2$ but, for the other resonances, serious discrepancies are evident. This might be due to two independent factors: First, in the contact transformation method, the transition moments were defined from the first term of the contact transformed polarizability, namely $\alpha_{\text{eff}} = \alpha_0 + i[S,\alpha]$, neglecting the higher order terms; second, some of the harmonic wave numbers appearing in Eqs. (2) and (3), which were taken from Ref. 7 might still have considerable errors, perhaps larger than 10 cm^{-1} . Such inaccuracy in the wave numbers may introduce appreciable errors in the cross sections of second order bands, especially in the case of vibrational resonances. Thus, the expressions derived from the exact diagonalization [Eqs. (4) and (5)], which appear as a function of the observed wave numbers rather than of the harmonic ones, were preferentially used in the computational refinement of the cubic force constants.

Finally, it is worthwhile to consider the scattering cross sections of second order transitions calculated in the harmonic approximation [Table I, row (c)]. It is evident that the harmonic approximation is almost useless for interpreting such spectroscopic features since deviations of more than one order of magnitude with respect to the experimental value arise commonly from this approximation.

CONCLUSION

In view of the results here presented, we believe that the use of second order vibrational Raman intensities represents an interesting approach towards the evaluation of the cubic force field of a molecule, provided the harmonic force field is well established.

In principle, this method yields very accurate cubic force constants, since the second order vibrational intensities are extremely sensitive to the anharmonicity. However, at present, this accuracy is limited by that of the second deriva-

tives of the molecular polarizability with respect to the normal coordinates. Since the uncertainty of the second derivatives of the polarizability calculated according to the bond polarizability model is not accurately known the cubic force field of methane here reported should be considered only as a first step of a new methodology for studying the vibrational anharmonicity.

ACKNOWLEDGMENTS

We are indebted to J. Brandmüller and H. W. Schrötter for moral support and to the Stiftung Volkswagenwerk of the Federal Republic of Germany for providing the high power laser source to obtain some of the results presented here.

- ¹Z. Cihla and A. Chedin, *J. Mol. Spectrosc.* **40**, 337 (1971).
- ²K. Kuchitsu and L. S. Bartell, *J. Chem. Phys.* **36**, 2470 (1962).
- ³L. S. Bartell and K. Kuchitsu, *J. Chem. Phys.* **68**, 1213 (1978).
- ⁴P. Pulay, W. Meyer, and J. E. Boggs, *J. Chem. Phys.* **68**, 5077 (1978).
- ⁵P. Pulay, G. Fogarasi, I. Pang, and J. E. Boggs, *J. Am. Chem. Soc.* **101**, 2550 (1979).
- ⁶E. Hirota, *J. Mol. Spectrosc.* **77**, 213 (1979).
- ⁷D. L. Gray and G. A. Robiette, *Mol. Phys.* **37**, 1901 (1979).
- ⁸Y. S. Makushkin, V. N. Savelev, and O. N. Ulenikov, *Opt. Spectrosc.* **48**, 264 (1980).
- ⁹W. F. Murphy and S. Montero, *Mol. Phys.* **44**, 187 (1981).
- ¹⁰S. Montero and D. Bermejo, *Mol. Phys.* **32**, 1229 (1976).
- ¹¹D. Bermejo, R. Escibano, and J. M. Orza, *J. Mol. Spectrosc.* **65**, 345 (1977).
- ¹²J. Martin and S. Montero, *J. Chem. Phys.* **80**, 4610 (1984).
- ¹³S. Montero, *J. Chem. Phys.* **77**, 23 (1982).
- ¹⁴C. Domingo, R. Escibano, W. F. Murphy, and S. Montero, *J. Chem. Phys.* **77**, 4353 (1982).
- ¹⁵S. Montero, *J. Chem. Phys.* **79**, 4091 (1983).
- ¹⁶K. T. Hecht, *J. Mol. Spectrosc.* **5**, 355 (1960).
- ¹⁷M. R. Aliev and J. K. G. Watson, *J. Mol. Spectrosc.* **75**, 157 (1979).
- ¹⁸J. Herranz, *J. Mol. Spectrosc.* **6**, 343 (1961).
- ¹⁹G. Poussiguet, E. Pascaud, J. P. Champion, and G. Pierre, *J. Mol. Spectrosc.* **93**, 351 (1982).
- ²⁰The program OVER performs the numerical calculation of Eqs. (8), (9), and (10) of Ref. 13.