

**Supplementary Information File for**  
**“An Experimental and Theoretical Study of the O(<sup>1</sup>D) + HD Reaction”**

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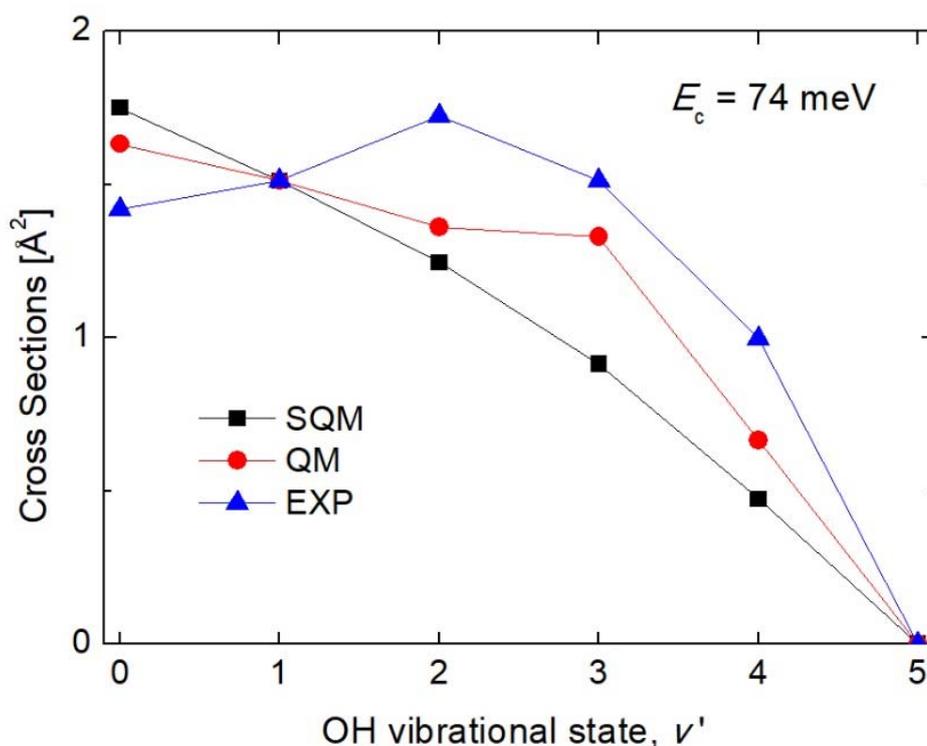
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The supplementary information file contains:

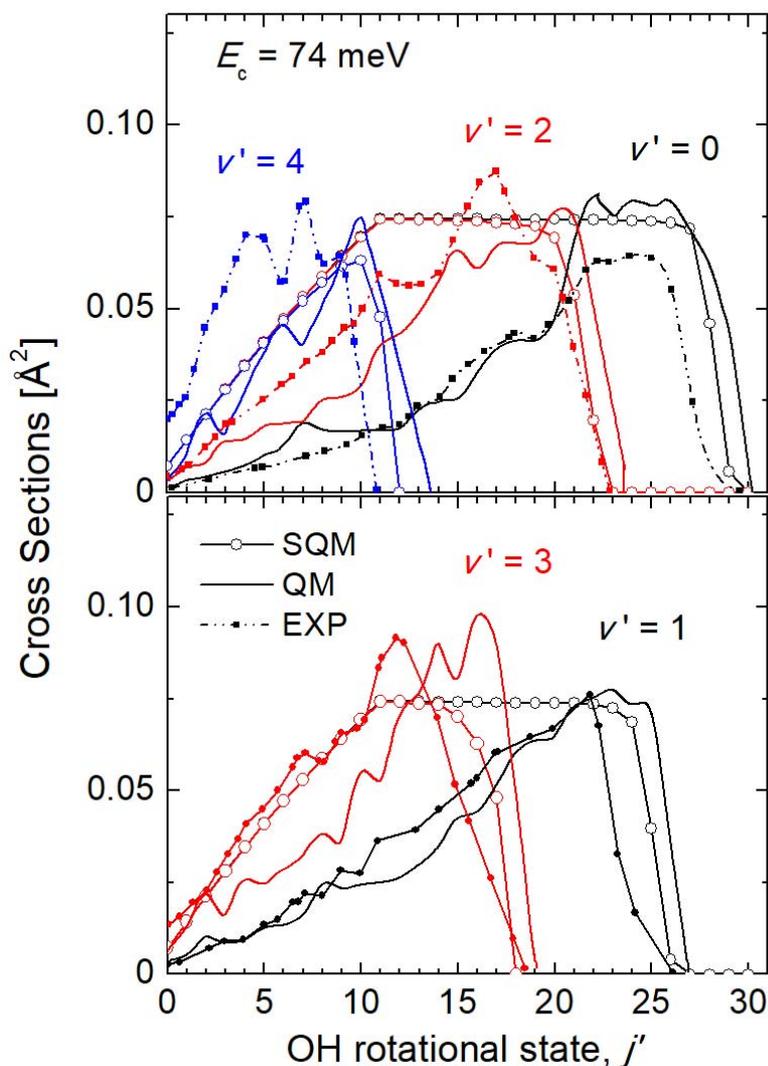
Supplemental discussion of the reaction dynamics. Supplemental figures S1-S4. Supplemental tables S1-S3.

## Supplementary discussion of the reaction dynamics

Supplementary evidence for a complex-forming mechanism for the  $O(^1D) + HD$  reaction is obtained when the vibrational distributions for the product  $OH(v')$  fragments are analysed. The corresponding statistical predictions can be compared with those cross sections from the work by Yuan *et al.*<sup>14</sup> As shown in Figure S1 both QM and experimental populations of the  $OH(v')$  states are not far from a monotonical decrease as the vibrational excitation  $v'$  increases.



**Figure S1.** Cross sections measured in  $\text{\AA}^2$  for the  $OH(v') + D$  formation process at  $E_c = 74$  meV. (Black squares) Present SQM distributions are compared with (red circles) QM and (blue triangles) experimental results both reported by Yuan *et al.*<sup>14</sup> QM and experimental distributions, originally in arbitrary units in Ref. (14) have been scaled here for comparison with the statistical cross sections.

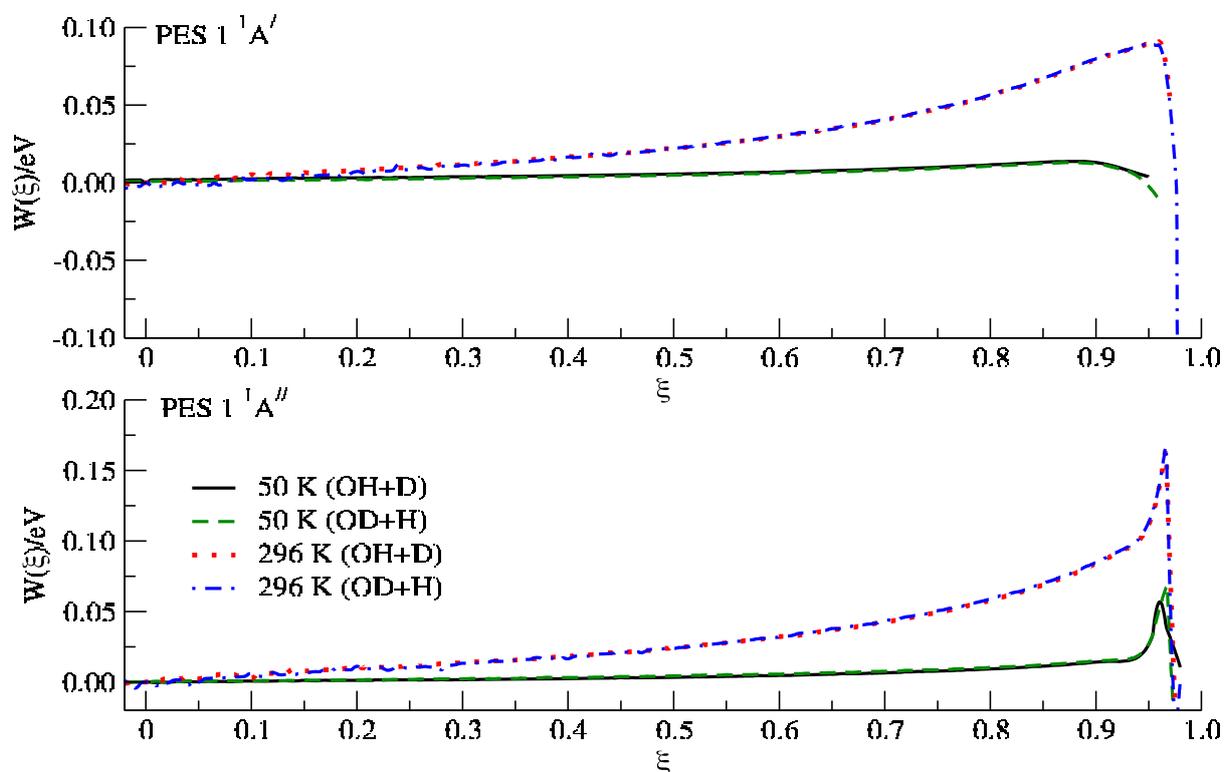


**Figure S2.** Rotationally resolved cross sections measured in  $\text{\AA}^2$  for the  $O(^1D) + HD \rightarrow OH(v', j') + D$  process at  $E_c = 74$  meV. (Open circles and solid lines) SQM distributions are compared with (solid) QM<sup>14</sup> and (solid circles and dashed-dotted lines) experimental<sup>14</sup> results for (top panel)  $v' = 0, 2, 4$ , and (bottom panel)  $v' = 1, 3$ . Results from Ref. (14) have been scaled here.

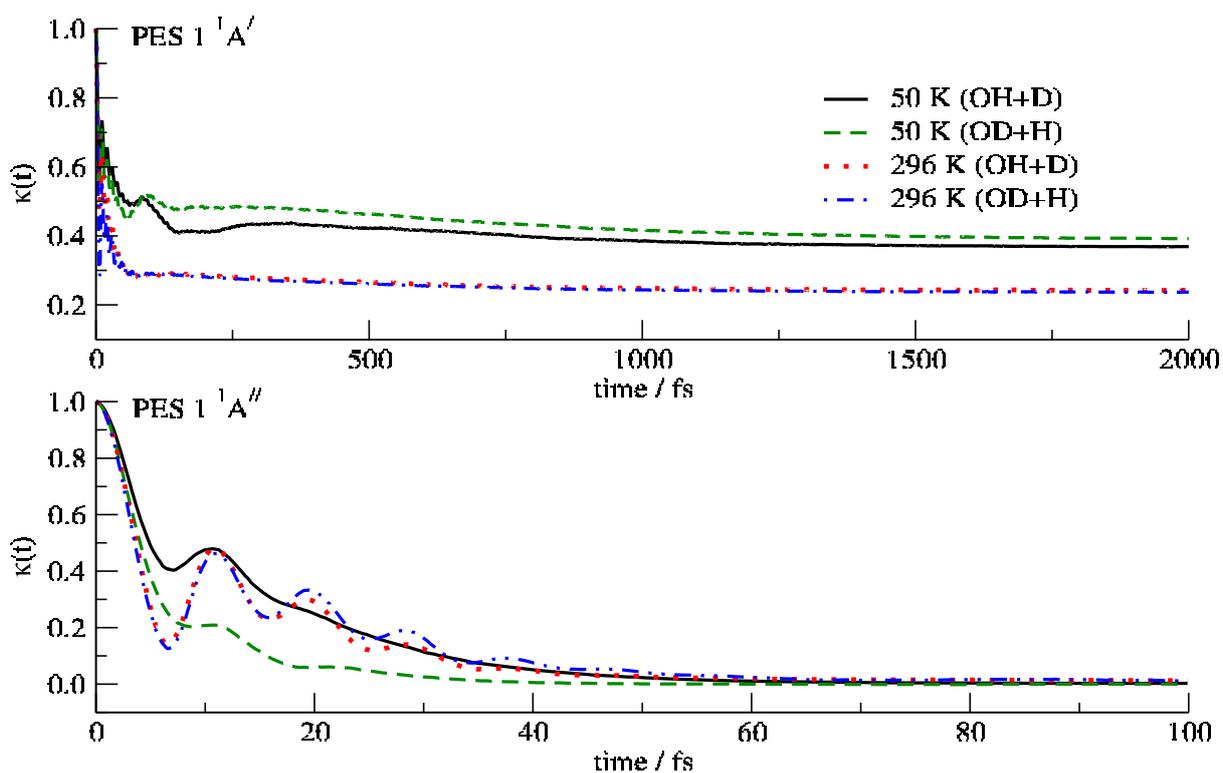
A much more demanding comparison can be established at the  $O(^1D) + HD \rightarrow OH(v', j') + D$  state to state level. Thus, we have calculated the rovibrational distributions  $(v', j')$  of the product OH by means of the SQM method in Figure S2. The corresponding comparison has been made with the QM and experimental results by Yuan *et al.*<sup>14</sup> Although the measured cross sections extend up to slightly lower rotational  $j'$  states for all possible final  $v'$  states than the theoretical calculations, the present SQM predictions are possibly in a better agreement with experiment than the QM values. However the almost equally distributed cross sections seen in

the statistical investigation for the most populated rotational states seem to be too crude in view of both the experimental and QM distributions obtained by Yuan *et al.*<sup>14</sup>

**Figure S3.** Ring polymer potentials of mean force ( $W(\xi)$ ) along the reaction coordinate for the  $O(^1D) + HD$  chemical reaction on the  $1^1A'$  (top panel) and  $1^1A''$  (bottom panel) PESs at 50 and 296 K. Legends correspond to both panels.



**Figure S4.** Ring polymer recrossing factor  $\kappa(t)$  along the reaction coordinate for the  $O(^1D) + HD$  chemical reaction on the  $1^1A'$  (top panel) and  $1^1A''$  (bottom panel) PESs at 50 and 296 K. Legends correspond to both panels.



**Table S1.** Input parameters for the RPMDrate calculations on the O(<sup>1</sup>D) + HD reaction.

Parameter	Product Channel		Explanation <sup>a</sup>
	OH + D	OD + H	
PES	<sup>1</sup> A'	<sup>1</sup> A''	[DK:MP1997, DK:FD98] <sup>b</sup>
Command Line Parameters			
temp		296 50	temperature (K)
N <sub>beads</sub>		128 (296 K); 296 (50 K)	number of beads
Dividing Surface Parameters			
R <sub>∞</sub>		60 a <sub>0</sub>	dividing surface parameter (distance)
N <sub>bond</sub>		1	number of forming and breaking bonds
N <sub>channel</sub>		1	number of equivalent product channels
thermostat		“Andersen”	thermostat option
Biased Sampling Parameters			
N <sub>windows</sub>		111	number of Windows
ξ <sub>1</sub>		-0.05	center of the first window
dξ		0.01	window spacing step
ξ <sub>N</sub>		1.05	center of the last window
dt		0.0001	time step (ps)
k <sub>i</sub>	20.41 (296 K); 53.06 (50 K)	20.41 (296 K); 54.42 (50 K)	umbrella force constant ((T/K) eV)
N <sub>trajectory</sub>		200	number of trajectories
t <sub>equilibration</sub>		20	equilibration period (ps)
t <sub>sampling</sub>		100	sampling period in each trajectory (ps)
N <sub>i</sub>		2 × 10 <sup>8</sup>	total number of sampling points
Potential of Mean Force Calculation			
ξ <sub>0</sub>		-0.02	start of umbrella integration
ξ <sup>‡</sup>	1 <sup>1</sup> A': 0.96 (296 K); 0.88 (50 K) 1 <sup>1</sup> A'': 0.96 (OH + D/50 K); 0.97		end of umbrella integration
N <sub>bins</sub>		5000	number of bins
Recrossing Factor Calculation			
dt		0.0001	time step (ps)
t <sub>equilibration</sub>		20	equilibration period (ps) in the constrained (parent) trajectory
N <sub>totalchild</sub>		100000	total number of unconstrained (child) trajectories
t <sub>childdata</sub>		2	sampling increment along the parent trajectory (ps)
N <sub>child</sub>		100	number of child trajectories per one initially constrained configuration
t <sub>child</sub>		2 (1 <sup>1</sup> A'); 0.1 (1 <sup>1</sup> A'')	length of child trajectories (ps)

<sup>a</sup>The explanation of the format of the input file can be found in the RPMDrate code manual (<http://rpmbrate.cyi.ac.cy>).

<sup>b</sup>[DK:MP97] A. J. Dobbyn and P. J. Knowles, *Mol. Phys.* **91**, 1107-1124 (1997). [DK:FD98] A. J. Dobbyn and P. J. Knowles, *Farad. Discuss.* **110**, 247 (1998).

**Table S2.** Results of RPMD calculations for the  $O(^1D) + HD$  reaction over the  $^1A'$  and  $^1A''$  PESs at 50 and 296 K, including ring polymer transmission coefficients ( $\kappa$ ) and final RPMD rate constants ( $k_{\text{RPMD}}$ ).<sup>a</sup>

T (K)	Product Channel	$\kappa$	$k_{\text{RPMD}}$ no correction (with correction) <sup>b</sup>
<b>PES 1 <math>^1A'</math></b>			
296	OH + D	$2.42 \times 10^{-1}$	$3.41 \times 10^{-10}$ ( $0.68 \times 10^{-10}$ )
	OD + H	$2.37 \times 10^{-1}$	$3.66 \times 10^{-10}$ ( $0.73 \times 10^{-10}$ )
50	OH + D	$3.69 \times 10^{-1}$	$3.16 \times 10^{-10}$ ( $0.63 \times 10^{-10}$ )
	OD + H	$3.92 \times 10^{-1}$	$4.06 \times 10^{-10}$ ( $0.81 \times 10^{-10}$ )
<b>PES 1 <math>^1A''</math></b>			
296	OH + D	$1.27 \times 10^{-2}$	$1.20 \times 10^{-12}$ ( $2.40 \times 10^{-13}$ )
	OD + H	$1.23 \times 10^{-2}$	$8.79 \times 10^{-13}$ ( $1.76 \times 10^{-13}$ )
50	OH + D	$2.94 \times 10^{-3}$	$1.11 \times 10^{-16}$ ( $2.22 \times 10^{-17}$ )
	OD + H	$3.42 \times 10^{-4}$	$2.04 \times 10^{-17}$ ( $4.08 \times 10^{-18}$ )

<sup>a</sup>All rate constants are given in  $\text{cm}^3 \text{s}^{-1}$ .

<sup>b</sup>Corrected by electronic partition functions  $Q_{\text{el}} = 5$ .

**Table S3.** Thermal rate constants  $k_{\text{RPMD}}$  for the  $O(^1D) + HD$  reaction and branching ratios (OD + H / OH + D) calculated by the RPMD method.

T (K)	$k_{\text{RPMD}}(\text{OH} + \text{D})^{a,b}$	$k_{\text{RPMD}}(\text{OD} + \text{H})^{a,b}$	$k_{\text{RPMD}}(\text{O}(^1D) + \text{HD})^{a,b}$	branching ratio <sup>b</sup>
296	$0.68 \times 10^{-10}$	$0.73 \times 10^{-10}$	$1.41 \times 10^{-10}$	1.07
50	$0.63 \times 10^{-10}$	$0.81 \times 10^{-10}$	$1.44 \times 10^{-10}$	1.28

<sup>a</sup>Rate constants are given in  $\text{cm}^3 \text{s}^{-1}$ .

<sup>b</sup>Corrected by electronic partition functions  $Q_{\text{el}} = 5$ .