

1 **Oral tribology: Update on the relevance to study astringency in wines**

2

3 Laura Laguna^a and Anwesha Sarkar^{b*}

4

5

6 ^a Institute of Food Science Research (CIAL), CSIC-UAM C/ Nicolás Cabrera 9, 28049,

7 Madrid, Spain

8 ^b Food Colloids and Processing Group, School of Food Science and Nutrition, University of

9 Leeds, Leeds LS2 9JT, UK

10

11

12 *Corresponding author:

13 Dr. Anwesha Sarkar

14 Food Colloids and Processing Group,

15 School of Food Science and Nutrition, University of Leeds, Leeds LS2 9JT, UK.

16 E-mail address: A.Sarkar@leeds.ac.uk (A. Sarkar).

17

18

19 **Oral tribology: Update on the relevance to study astringency in wines**

20

21 **Abstract**

22 Oral tribology is emerging as a new paradigm to quantify friction and lubrication of food-saliva
23 mixtures in the oral mucosa. Recently, oral tribology has captured research attention in
24 quantifying wine astringency, a characteristic “dryness feeling”, which strongly impacts
25 consumer preference. Hence, this paper aims to provide a concise review of oral tribology in the
26 context of wine astringency. Firstly, the important roles of “biolubricant” saliva, salivary proteins
27 and current tribo-pairs used in oral tribology measurements are reviewed. Then, we have
28 discussed the key mechanisms of wine astringency involving polyphenol-salivary protein
29 interactions (hydrogen bonding, hydrophobic interactions), rupture of the lubricating salivary film
30 and oral sensation of discrete particles. Studies employing Stribeck curve analysis and
31 microstructural characterization to understand polyphenol-salivary protein interactions are
32 reviewed. Finally, we highlighted the need for bio-relevant tribo-pairs, simulated oral conditions
33 and tribology-sensory correlation, before such quantification can be used to characterize wine
34 astringency at a commercial level.

35

36 Keywords: Oral tribology; Wine; Astringency; Mucin; Saliva; Lubrication; Tannins;
37 Proline-rich proteins

38

39 **Introduction**

40 Astringency is defined as “the complex of sensation due to shrinking, drawing or puckering of
41 the oral epithelium as a result of exposure to substances, such as alums or tannins”¹.
42 Astringency is a very frequent sensory experience perceived upon consumption of various food
43 and beverages, such as some unripe fruits (e.g. persimmon, chokecherry), soy-based foods,
44 green and black tea, some herbs and spices (e.g. turmeric, marjoram, sage) and red wine. In

45 wine, astringency can be associated with different components, such as metals ions, alcohols,
46 organic acids, but polyphenols are generally agreed to play the most important role. These
47 polyphenols in wine come from the grapes (hydroxybenzoic and hydroxycinnamic acids,
48 flavonol glycosides, flavan-3-ols/procyanidins, and stilbenes) and the oak barrels (hydrolysed
49 tannins) in which the wine is stored for ageing.

50 Particularly, astringency is an important wine texture quality parameter. Till now,
51 wine astringency research has mainly focused on identifying appropriate analytical methods,
52 such as chromatography². Although chromatographic tests have enabled successful
53 identification of the relevant wine components that cause astringency, they do not allow
54 quantifying the intensity or the evolution of the “astringent feeling”. That is why, the gold
55 standard method of assessment of the astringency in wine is “tasting” by trained sensory panels
56 using set of reference compounds and descriptors³. However, training a sensory panel is time-
57 consuming and expensive. Furthermore, astringency is a complex sensory attribute as it builds
58 in intensity over repeated exposure. Thus, it is difficult to clean the mouth between the samples
59 with astringent components, latter can cause fatigue in sensory panel members and
60 consequently assessments errors^{4, 5}.

61 From mechanistic viewpoint, the term astringency comes from the Latin phrase “ad
62 stringere” meaning ‘to bind, which is believed to be related to the ability of astringent
63 substances, such as wine polyphenols to bind to and precipitate salivary proteins⁵. Although
64 there have been several hypotheses on interactions between wine polyphenols and salivary
65 proteins, the predominant mechanism by which solutions containing polyphenols are perceived
66 as astringent is still not clear. Using psychophysical methods, Green⁶ suggested that oral friction
67 is the key underlying physiological mechanism behind the sensation of astringency. This oral
68 friction has been postulated to be resulting from the loss of oral mucosal lubrication of the
69 salivary film, on exposure to the polyphenol components⁷. Therefore, “tribology” i.e. the

70 science of friction, wear and lubrication appears as a promising approach that can be used
71 quantify coefficient of friction in oral environment, former has gathered recent research
72 attention in understanding astringency perception.

73 Oral tribology is the study of friction and lubrication between two interacting surfaces, such
74 as teeth–teeth, tongue–palate, tongue–teeth, tongue–food, lips, lips–food, bolus–palate, food
75 particles–oral surfaces that are in relative motion in the oral cavity^{8, 9}. Coefficient of friction
76 and its relation to sensory smoothness and slipperiness in food research domain was first
77 detailed by Kokini and co-workers¹⁰ in 1977. The term “lubrication” as a determinant of food
78 bolus formation and swallowing was used by Hutchings and Lillford^{11, 12} after nine years.
79 Lubrication in mouth was proposed to be dependent on saliva coating the oral surfaces before
80 eating. Post food consumption, the changing properties of food and its interaction with the in-
81 mouth environment was hypothesized to be the driver of oral lubrication. However, it is only
82 recently that there has been an upsurge in research efforts in oral tribology, which can be
83 evidenced by a power-law behaviour in the distribution of citations received by scientific papers
84 over the last 10 years (Figure 1). Particularly, there has been some recent efforts to relate oral
85 friction to sensory characteristics of “astringency”¹³⁻¹⁶, latter is an important quality
86 characteristic in wine.

87 Hence, this review is aimed to provide a concise update on studies employing oral tribology
88 as a quantitative tool to predict wine astringency. Firstly, we have provided a brief introduction
89 on oral tribology with respect to definition and relevance of the tribo-pairs (i.e. pair of materials
90 used to create the contact surfaces), load (i.e. tongue pressure against the hard upper palate) and
91 chemistry of the “biolubricant” saliva. Then we have specifically focussed on wine and its
92 components (polyphenols), which interact with saliva. Finally, we have provided an update of
93 how tribology has been used as a tool to determine the loss of salivary lubricity on exposure to
94 wine polyphenols and highlighted the research gaps in this area.

95 **Oral tribology**

96 The key parameter of tribology measurement is the friction coefficient, calculated as the
97 ratio of the measured friction force against the normal load (Figure 2a)^{8, 9, 17-19}. When two
98 surfaces are in the relative motion at a steady speed of V , the frictional force (F_R) can be
99 expressed as $F_R = \mu \times F_L$, where μ is the friction coefficient (dimensionless) and F_L is the
100 normal force. Lubrication is a surface property, and the magnitude of μ thus depends on the
101 surface roughness and geometry of the interacting surfaces as well as nature of lubricant. A
102 typical tribometer with ball on a rotating disc configuration during sliding is illustrated in Figure
103 2b.

104 The friction coefficient is dependent on the lubricant film thickness (δ) between the two
105 moving surfaces and is typically presented in a Stribeck curve (Figure 2c)¹⁷. The distinct friction
106 scenarios that can occur between the tongue and palate is represented by three different regimes:
107 the boundary regime, the mixed regime and the hydrodynamic regime. Details of these regimes
108 can be found in previous reviews^{8, 17}.

109 **Role of tribo-pairs and loads**

110 In order to understand the complex oral system (oral surfaces, saliva or saliva-wine mixtures as
111 the lubricants), researchers have used different metallic, crystalline, polymeric and animal
112 tissue-based tribo-pairs to mimic the topologies of real human tongue and oral palate. Pin-on-
113 disc, ball-on-disc tribometers with tribo-pairs made up of steel²⁰, tetrafluorethylene and
114 zirconia²¹, glass²² surfaces in a sliding or rotating configurations have been used. However, as
115 one might imagine, contrasting to these surfaces, oral surfaces may vary significantly from
116 highly keratinized bony palate to soft and rough tongue with papillae being in of order 20-100
117 μm ^{12, 23}.
118 Innovative approaches, such as everted dried dead tongues of pigs/ piglets have been also used
119 in tribometers to represent human tongue surfaces^{12, 24}. Besides ethical constraints, lack of

120 information about surface chemistry and biological heterogeneity of using animal tissues,
121 papillae of the dried pig tongue ex vivo was not firm and erect during tribology measurements,
122 which might be attributed to the biochemical changes (post-mortem) or dehydration process.
123 Furthermore, the dead animal tissues were less hydrophobic and lubricating as compared to the
124 living surfaces^{12, 25}. It is also worth recognizing that the diameter of the hairs of the human
125 filiform papillae (27 μm) is larger than that of the pig tongue (18 μm)²⁶. Hence, the surface
126 roughness of these dried animal tissue surface used in the tribology measurement was not
127 representative of the real human tongue surface. Hence, the friction measurement interpretation
128 for human tongue needs to be taken with precaution.

129 Instead of “hard” metallic surfaces and animal tissues, soft elastomeric substrates, such
130 as polydimethylsiloxane (PDMS) that can be deformed by contact pressure are currently
131 preferred as tribo-pairs^{19, 27, 28}. Although tongue surface is significantly rougher than smooth
132 PDMS surfaces, PDMS surfaces can be modified in deformability, roughness and
133 hydrophobicity to represent tailored oral surfaces. For example, the hydrophobicity of PDMS
134 surfaces can be tuned using plasma oxidation, surface coating with functional groups or layer-
135 by layer²⁹⁻³¹.

136 “Loads” in oral tribology context can be defined as the normal force that the tongue
137 exerts on the hard upper palate. As compared to typical mechanical engineering context, a lower
138 range of loads (1-10 N) has been used in oral tribology studies^{19, 27}. Measurements of the loads
139 of the tongue against the upper hard palate generally ranges from 0.01-90 N³². It is worth noting
140 that the tongue pressure distribution is not uniform across different parts of the tongue-oral
141 palate contacts and the load distribution might also vary with time³³. Tongue pressure might
142 also differ depending upon the population used for study, for instance, elderly population show
143 significantly lower tongue pressures than younger adults group³⁴⁻³⁷. Hence, oral tribology study

144 for a particular wine consumer group needs to be carried out at a range of relevant loads rather
145 than a single-point load to represent different oral conditions.

146 **Saliva: The potent “bio-lubricant”**

147 Saliva is composed of water (99.5%), proteins (highly glycosylated mucins, proline-rich
148 proteins and enzymes, such as α -amylase) (0.3%), and inorganic substances (0.2%) with pH
149 around 6.8³⁸⁻⁴⁰.

150

151 Formation of salivary mucosal pellicle

152 Salivary mucosal pellicle is a viscoelastic gel that protects the oral mucosa from mechanical
153 and chemical damages, such as exposure to microorganism, toxic materials, environmental
154 insult, dehydration of oral mucosal epithelium and lubrication. The most prominent constituent
155 of oral pellicles are mucins, a high molecular weight glycoprotein^{41 42}. As Figure 3 shows,
156 salivary mucosal pellicle comprises of two phases, an immobile pellicle retained on epithelial
157 cells (membrane associated mucins: MUC1, MUC3, MUC4, MUC12) and a mobile salivary
158 film (secreted soluble mucins: MUC2, MUC5A, MUC5B, MUC6, MUC7)⁴³⁻⁴⁶.

159 The MUC5B (high molecular weight) and MUC7 (low molecular weight) are the most
160 important glycoproteins with regards to lubrication. Saliva is secreted to maintain saliva pellicle
161 thickness of ~ 70 - $100 \mu\text{m}$ ⁴⁷, but vary depending upon the oral location. The oral mucosa where
162 saliva pellicle is created is generally hydrophobic until the salivary proteins bind. Upon
163 adsorption to the tongue (hydrophobic), glycoproteins tend to bind with their hydrophobic sites
164 towards the tongue, whilst hydrophilic sites point outwards for water retention. Salivary film
165 reduces the “ μ ” in oral surfaces. Using AFM, human salivary pellicles have been shown to
166 reduce the μ by a factor of 20 between hard contact surfaces⁴⁸, having $\mu \approx 0.02$ i.e. two orders
167 of magnitude lower than that of water⁴⁹.

168

169 Use of saliva in oral tribology studies

170 Use of saliva is becoming popular in oral tribological measurements in food research as saliva
171 is a key “biolubricant” that can reduce “ μ ” significantly within the human oral surfaces.
172 However, such lubricating properties of saliva (ex vivo) can vary significantly depending upon
173 stimulation (unstimulated, mechanical, acid), collection (protein-binding properties and air
174 exposure) and usage (immediate use, freeze-thaw-induced precipitation)^{50, 51}. Also, within an
175 individual, salivary protein amount varies and acidic and glycosylated proline-rich-proteins
176 PRPs (gPRPs and aPRPs) may vary significantly throughout the day and is highly dependent
177 on the type of food ingested⁵². Other factors influencing interactions with wine are pH,
178 buffering capacity and concentrations of calcium and phosphate in saliva, latter shows huge
179 variation over a day in unstimulated whole saliva⁵³ and even depends on how saliva has been
180 handled after collection⁵¹.

181 The friction coefficient of stimulated and unstimulated saliva measured between two
182 mucosal surfaces using loads (0.34-2.20 N) showed decrease of μ with increase in load and
183 speed for both types of saliva^{54, 55}. The differences in μ were due to the protein content and
184 rheological properties of saliva, particularly, stimulated saliva produced by sublingual and
185 submandibular gland had a higher protein content and lower viscosity as compared to
186 unstimulated saliva⁵⁴.

187 Saliva also changes its composition along the salivary film (Figure 3), and until now,
188 the “mobile salivary phase” has only been studied. However, the most important lubricating
189 proteins (MUC5B and MUC7) still remain attached to the mucosal epithelia even if the salivary
190 film is ruptured. As these mucins may be important to understand “astringency”, it might be
191 worth to consider collecting saliva from parotid glands or gently scraping the immobile salivary
192 pellicle from the oral surfaces of the participants after ethics approval for tribological
193 measurements.

194 Finally, the use of “artificial saliva” i.e. fluid mimicking the ionic composition, mucin
195 and rheological properties of unstimulated human saliva has been quite common due to its ease
196 of preparation and reproducibility³⁸⁻⁴⁰. However, the term “artificial saliva” has been argued by
197 several authors as there has been no bio-mimetic that accurately simulates all of the properties
198 of saliva⁵⁰. In a recent study by Laguna and coworkers¹⁹, μ of artificial saliva was measured in
199 a PDMS-PDMS ball-dics set-up and the Stribeck pattern was found to be similar to real human
200 saliva (unstimulated)⁵⁴. Hence, use of at least mucin in a mimicked ionic composition can be a
201 good starting point to understand wine-saliva interaction as compared to that without
202 consideration of any aspects of salivary lubrication.

203

204 **Wine and astringency**

205 Wines, derived from fermented grapes⁵⁶ (*Vitis Vinifera*) are essentially composed of 80-85%
206 water, 9-20% ethanol and other minor compounds, such as phenolic compounds, esters, acids,
207 nitrogenous compounds, volatiles, lipids, mineral salts etc. A well balanced-wine should
208 contain optimum primary taste components (i.e. balance between sweetness and acidity), tactile
209 elements (i.e. astringency) and flavour⁵⁷. Among the different textural attributes, astringency
210 has been considered to be one of the most important sensory characteristic in red wines.

211 Since astringency can be perceived in the mouth where no taste receptors are present,
212 it is considered to be tactile rather than a taste stimulation, contrary to the initial speculations⁶.
213 Different phenolic compounds show different affinities towards human salivary proteins⁵⁸.
214 Polyphenols with extended structure have been reported to have higher affinity to PRPs^{58,59}. In
215 other words, smaller polyphenols can bind with one phenolic ring, whilst larger polyphenols
216 interact in a multi-dentate fashion, occupying two or three consecutive prolines increasing the
217 degree of salivary protein precipitation. Despite the chemical differences in phenolic
218 components, the astringency of polyphenols mixtures with different structures, such as,

219 phenolic acids and catechins were perceived to be of same astringency by a trained sensory
220 panel⁶⁰. The total phenolic content of wines depends on many factors and it can vary from 900-
221 1400 mg/L in young red wines and 1600-2500 mg/L in aged red wines⁶¹. Astringency feeling
222 evolves during aging, and it is generally higher in young wines and decreases over time,
223 “softening” the wine. This is caused presumably by the soluble pectin fragments, associated
224 with the grapes that might inhibit protein-tannin interactions and pectin might aggregate or
225 encapsulate the tannins making the latter unavailable to the salivary proteins^{62, 63}.

226 Three different mechanisms of wine astringency has been hypothesized that
227 complement each other: protein precipitation, rupture of the lubricating salivary film and
228 formation of mouth debris⁶⁴ (Figure 3). Firstly, wine polyphenolic compounds form complexes
229 with salivary proteins, specially PRPs⁶⁵ due to hydrophobic interactions and hydrogen bonding,
230 precipitating the salivary proteins and decreasing its viscosity⁶⁶, latter affecting the integrity of
231 the salivary film. Hydrogen bonding occurs between hydroxyl groups of phenolic compounds
232 and carbonyl and amide group of the salivary protein, whereas hydrophobic interactions occur
233 between the benzoic ring of phenolic compounds and the apolar side chains of amino acids such
234 leucine, lysine or proline in the salivary proteins⁶² (Figure 3). The rupture of the lubricating
235 saliva film activates the mechanoreceptors, located within the mucosa connected with the
236 trigeminal nerve that then transmits to brain the perception of astringency⁶⁷. Furthermore, the
237 increase in precipitated salivary proteins and other debris in saliva increases the sense of
238 “discrete particles” in the mouth, which essentially relates to roughness and oral friction²³. Due
239 to the strong correlation between astringency perception and formation of insoluble salivary
240 protein-wine polyphenol complexes, research has focused in finding analytical methods for
241 quantification/qualification of these complexes. In the next section, we only focus on recent
242 studies that used Stribeck curves to quantify astringency.

243

244 **Relevance of oral tribology to unravel wine astringency**

245 Salivary proteins are widely separated from each other due to mutually repulsive forces of
246 negatively charged mucins³⁸ at neutral pH in saliva, latter is a highly diluted system²³. However,
247 when tannic acid was added, large flocs appeared in saliva (approx. 300 μm) (Figure 4a)²³. In
248 red wines-saliva mixtures, similar aggregates have been recently observed using light and
249 transmission electron microscopy (Figure 4b)¹⁴. Furthermore, the microstructure of such
250 aggregates varied depending upon the wine type and their polyphenol composition, specifically
251 proanthocyanidin (grape skin) and tannin (seeds). Cabernet Sauvignon wines presented
252 densely-packed aggregates whereas Carménère, Merlot wines showed smaller aggregates with
253 much more open structure (Figure 4b)¹⁴. However, irrespective of the type of wines¹⁴, wine-
254 saliva mixtures showed a significantly higher μ as compared with human saliva in the boundary
255 regime using a PDMS-steel contact surfaces (Figure 4c). Authors reported a high correlation
256 ($R^2=0.93$) between μ and sensory “astringency” at a sliding speed of 0.075 mm/s linking
257 astringency to salivary protein depletion by wine polyphenols.

258 In a separate study, tribological analysis in a soft PDMS ball/model mucin-adsorbed
259 glass surface¹⁶ indicated that astringency may arise from the temporary failure of the boundary
260 lubrication of the adsorbed mucins by tannic acid. This loss of boundary lubrication showed
261 concentration dependency on tannic acid. Authors suggested that interaction with tannic acid
262 molecules might result in the change in conformation and hydration of adsorbed mucin, both
263 leading to the marked rise in friction force. This is in agreement with a previous report, where
264 “chemically pure” polyphenol (epigallocatechin gallate) appeared to partially deplete the thin
265 lubricating human salivary film (mechanically stimulated whole saliva) from the smooth
266 PDMS-PDMS contact surfaces in a tribological experiment performed at 37 °C¹⁵. This induced
267 an increase in μ and was correlated to a certain extent with the astringency perception.

268 Besides phenolic compounds, there are other components in wine, which can also
269 contribute to astringency perception. For example, tartaric acid present in wines are known to
270 lower the pH of wine significantly, which precipitates the salivary proteins² as well as increase
271 the binding affinities of the salivary proteins with polyphenols. In contrast, the presence of
272 ethanol in wine has been reported to modify the degree of hydrogen bonding between
273 polyphenols and salivary proteins. This tend to modify the degree of protein folding and
274 solubility of tannins⁶⁸. Another key component in wines i.e. glycerol has been associated with
275 oiliness, persistence and mellowness⁶⁹. Interestingly, tribological measurements of aqueous
276 solutions of glycerol in steel tribo-pairs (ball/ disc) have suggested glycerol to be a potential
277 “green lubricant” with its lubricating properties being better than those of rapeseed oil. Hence,
278 contribution of wine components other than polyphenols in astringency should not be
279 underestimated and the complex interplay of polyphenol, pH, ethanol, glycerol in wine
280 astringency needs further investigation from tribological viewpoint⁷⁰.

281

282 **Conclusions**

283 In summary, astringency studies in wine essentially rely on sensorial methods so far. Interaction
284 between polyphenols in wine and salivary proteins is generally considered to be the main
285 mechanism inducing astringency sensation. Oral tribology is a relatively recent approach that
286 has been used to quantitatively study the loss of lubricity of saliva on exposure to polyphenols.
287 Measurement of coefficient of friction of wine and specific polyphenols at certain sliding
288 speeds have shown some correlation with sensory perception of astringency. This shows
289 potential of oral tribology measurement as a promising quantitative tool for analysing
290 astringency perception. However, lubrication is a surface property. Hence, the friction
291 coefficient not only depends on the mechanical properties of the lubricant (e.g. saliva) but also
292 on the surfaces used in tribology measurement to represent the tongue and the upper palate.

293 Currently, the contact surfaces used in oral tribology range from steel to glass to PDMS. The
294 key requirement is the accurate development of bio-relevant tribo-contact surfaces that
295 effectively represent the soft, micro-patterned tongue and bony upper palate surfaces. Use of
296 accurate loads representative of real human tongue pressure values when consuming
297 polyphenol-rich food need to be used in such measurements. Use of relevant tribo-pairs and
298 loads need to be standardized across different laboratories to have comparable results. Most
299 importantly, these quantitative friction measurements need appropriate correlation with sensory
300 perception using trained sensory panel, before such quantification can be of use to characterize
301 astringency in wine and other polyphenol rich foods at a commercial level.

302

303 **Notes on Contributors**

304 **Dr. Laura Laguna** works for the Spanish National Research Council (CSIC) at the Institute of
305 Food Research (CIAL, Madrid). Over the last years, her research has been focused on the study
306 of food oral processing, nowadays with special emphasis on studying the sensory perception of
307 wine components.

308 **Dr. Anwasha Sarkar** is a Lecturer of Food Colloids at the University of Leeds. She has more
309 than 36 scientific papers and 4 patents. Research interests: colloidal design (emulsion, emulsion
310 gels, microgel, particles, protein complexes, coacervates), oral tribology, lubrication in soft
311 contacts, multi-scale structural analysis, in vitro digestion.

312

313 **Disclosure statement**

314 No potential conflict of interest was reported by the authors.

315

316 **References**

317 1. ASTM, Philadelphia: American Society for Testing and Materials, 2004, 2004.

- 318 2. A. Rinaldi, A. Gambuti, and L. Moio, *Journal of Food Science*, 2012, **77**(4), C485-
319 C490.
- 320 3. S. Vidal, P. Courcoux, L. Francis, M. Kwiatkowski, R. Gawel, P. Williams, E. Waters,
321 and V. Cheynier, *Food Quality and Preference*, 2004, **15**(3), 209-217.
- 322 4. J.-X. Guinard, R. M. Pangborn, and M. J. Lewis, *American Journal of Enology and*
323 *Viticulture*, 1986, **37**(3), 184-189.
- 324 5. B. J. Lyman and B. G. Green, *Chemical Senses*, 1990, **15**(2), 151-164.
- 325 6. B. G. Green, *Acta Psychologica*, 1993, **84**(1), 119-125.
- 326 7. H. L. Gibbins and G. H. Carpenter, *Journal of Texture Studies*, 2013, **44**(5), 364-375.
- 327 8. J. R. Stokes, M. W. Boehm, and S. K. Baier, *Current Opinion in Colloid & Interface*
328 *Science*, 2013, **18**(4), 349-359.
- 329 9. J. Chen and J. R. Stokes, *Trends in Food Science & Technology*, 2012, **25**(1), 4-12.
- 330 10. J. L. Kokini, J. B. Kadane, and E. L. Cussler, *Journal of Texture Studies*, 1977, **8**(2),
331 195-218.
- 332 11. J. B. Hutchings and P. J. Lillford, *Journal of Texture Studies*, 1988, **19**(2), 103-115.
- 333 12. H. Ranc, A. Elkhyat, C. Servais, S. Mac-Mary, B. Launay, and P. Humbert, *Colloids*
334 *and Surfaces A: Physicochemical and Engineering Aspects*, 2006, **276**(1-3), 155-161.
- 335 13. U. Rutuja, B. Natalia, and C. Jianshe, *Journal of Physics D: Applied Physics*, 2016,
336 **49**(10), 104003.
- 337 14. N. Brossard, H. Cai, F. Osorio, E. Bordeu, and J. Chen, *Journal of Texture Studies*,
338 2016, **47**(5), 392-402.
- 339 15. D. Rossetti, J. H. H. Bongaerts, E. Wantling, J. R. Stokes, and A. M. Williamson,
340 *Food Hydrocolloids*, 2009, **23**(7), 1984-1992.
- 341 16. S. Ma, H. Lee, Y. Liang, and F. Zhou, *Angewandte Chemie International Edition*,
342 2016, **55**(19), 5793-5797.
- 343 17. S. Prakash, D. D. Y. Tan, and J. Chen, *Food Research International*, 2013, **54**(2),
344 1627-1635.
- 345 18. C. Pradal and J. R. Stokes, *Current Opinion in Food Science*, 2016, **9**, 34-41.
- 346 19. L. Laguna, G. Farrell, M. Bryant, A. Morina, and A. Sarkar, *Food & Function*, 2017,
347 **8**, 563-573
- 348 20. I. Fernández Farrés and I. T. Norton, *Food Hydrocolloids*, 2015, **45**, 186-195.
- 349 21. S. Lee, M. Heuberger, P. Rousset, and N. D. Spencer, *Tribology letters*, 2004, **16**(3),
350 239-249.
- 351 22. K. Liu, Y. Tian, M. Stieger, E. van der Linden, and F. van de Velde, *Food*
352 *Hydrocolloids*, 2016, **52**, 403-414.
- 353 23. R. A. De Wijk and J. F. Prinz, *Journal of Texture Studies*, 2006, **37**(4), 413-427.
- 354 24. D. M. Dresselhuis, E. H. A. de Hoog, M. A. Cohen Stuart, and G. A. van Aken, *Food*
355 *Hydrocolloids*, 2008, **22**(2), 323-335.
- 356 25. J. Asserin, H. Zahouani, P. Humbert, V. Couturaud, and D. Mougín, *Colloids and*
357 *surfaces B: Biointerfaces*, 2000, **19**(1), 1-12.
- 358 26. A. Kullaa-Mikkonen, M. Hynynen, and P. Hyvönen, *Cells Tissues Organs*, 1987,
359 **130**(3), 280-284.
- 360 27. J. H. H. Bongaerts, K. Fourtouni, and J. R. Stokes, *Tribology International*, 2007,
361 **40**(10-12), 1531-1542.
- 362 28. D. M. Dresselhuis, H. J. Klok, M. A. C. Stuart, R. J. de Vries, G. A. van Aken, and E.
363 H. A. de Hoog, *Food Biophysics*, 2007, **2**(4), 158-171.
- 364 29. B. T. Ginn and O. Steinbock, *Langmuir*, 2003, **19**(19), 8117-8118.
- 365 30. D. Xiao, H. Zhang, and M. Wirth, *Langmuir*, 2002, **18**(25), 9971-9976.
- 366 31. T. Trantidou, Y. Elani, E. Parsons, and O. Ces, 2017, **3**, 16091.
- 367 32. J. L. Miller and K. L. Watkin, *Dysphagia*, 1996, **11**(2), 117-124.

- 368 33. J. F. Prinz, R. A. de Wijk, and L. Huntjens, *Food Hydrocolloids*, 2007, **21**(3), 402-
369 408.
- 370 34. L. Laguna, M. M. Hetherington, J. Chen, G. Artigas, and A. Sarkar, *Food Quality and*
371 *Preference*, 2016, **53**, 47-56.
- 372 35. L. Laguna and A. Sarkar, *Food Hydrocolloids*, 2016, **61** (2), 286-299.
- 373 36. L. Laguna, A. Sarkar, G. Artigas, and J. Chen, *Physiology & Behavior*, 2015, **147**,
374 274-281.
- 375 37. L. Laguna, A. Sarkar, and J. Chen, *Proceedings of the Nutrition Society (Joint Royal*
376 *Society of Medicine-Nutrition Society Winter Meeting)*, 2015, **74**(OCE2), E167-E167.
- 377 38. A. Sarkar, K. K. T. Goh, and H. Singh, *Food Hydrocolloids*, 2009, **23**(5), 1270-1278.
- 378 39. A. Sarkar and H. Singh: 'Oral behaviour of food emulsions', in '*Food Oral Processing*',
379 111-137; 2012, Wiley-Blackwell.
- 380 40. A. Sarkar, A. Ye, and H. Singh, *Food & Function*, 2017, **8**(2), 511-521.
- 381 41. H. Gibbins, G. Proctor, G. Yakubov, S. Wilson, and G. Carpenter, *Oral diseases*,
382 2014, **20**(7), 707-713.
- 383 42. L. Becerra, R. V. Soares, L. S. Bruno, C. C. Siqueira, F. G. Oppenheim, G. D. Offner,
384 and R. F. Troxler, *Archives of Oral Biology*, 2003, **48**(2), 147-154.
- 385 43. S. J. Gendler, *Journal of mammary gland biology and neoplasia*, 2001, **6**(3), 339-353.
- 386 44. M. Cárdenas, U. Elofsson, and L. Lindh, *Biomacromolecules*, 2007, **8**(4), 1149-1156.
- 387 45. L. Macakova, G. E. Yakubov, M. A. Plunkett, and J. R. Stokes, *Tribology*
388 *International*, 2011, **44**(9), 956-962.
- 389 46. M. Morzel, T. Siying, H. Brignot, and J. Lherminier, *Microscopy research and*
390 *technique*, 2014, **77**(6), 453-457.
- 391 47. L. M. C. Collins and C. Dawes, *Journal of Dental Research*, 1987, **66**(8), 1300-1302.
- 392 48. I. C. Hahn Berg, M. W. Rutland, and T. Arnebrant, *Biofouling*, 2003, **19**(6), 365-369.
- 393 49. J. H. H. Bongaerts, D. Rossetti, and J. R. Stokes, *Tribology Letters*, 2007, **27**(3), 277-
394 287.
- 395 50. J. Chen and L. Engelen: '*Food oral processing: fundamentals of eating and sensory*
396 *perception*'; 2012, Chichester, Wiley-Blackwell.
- 397 51. C. A. Francis, M. P. Hector, and G. B. Proctor, *Archives of Oral Biology*, 2000, **45**(7),
398 601-606.
- 399 52. E. Brandão, S. Soares, N. Mateus, and V. de Freitas, *Food Research International*,
400 2014, **64**, 508-513.
- 401 53. M. J. Larsen, A. F. Jensen, D. M. Madsen, and E. I. F. Pearce, *Archives of Oral*
402 *Biology*, 1999, **44**(2), 111-117.
- 403 54. J. F. Prinz, R. A. de Wijk, and L. Huntjens, Load dependency of the coefficient of
404 friction of oral mucosa, in *Food Hydrocolloids*. 2007, Elsevier Ltd: OXFORD. p. 402-
405 408.
- 406 55. H. Olsson, V. Henricsson, and T. AxÉLI, A new device for measuring oral mucosal
407 surface friction - reference values, in *European Journal of Oral Sciences*. 1991,
408 Blackwell Publishing Ltd. p. 329-332.
- 409 56. R. S. Jackson: '*Table wines: sensory characteristics and sensory analysis*', 196-226;
410 2012, Wood Head Publishing.
- 411 57. R. Koone, R. J. Harrington, M. Gozzi, and M. McCarthy, *Journal of Wine Research*,
412 2014, **25**(3), 158-174.
- 413 58. O. Cala, E. J. Dufourc, E. Fouquet, C. Manigand, M. Laguerre, and I. Pianet,
414 *Langmuir*, 2012, **28**(50), 17410-17418.
- 415 59. N. J. Baxter, T. H. Lilley, E. Haslam, and M. P. Williamson, *Biochemistry*, 1997,
416 **36**(18), 5566-5577.

- 417 60. R. Ferrer-Gallego, J. M. Hernández-Hierro, J. C. Rivas-Gonzalo, and M. T. Escribano-
418 Bailón, *Food Research International*, 2014, **62**, 1100-1107.
- 419 61. A. García-Ruiz, B. Bartolomé, A. J. Martínez-Rodríguez, E. Pueyo, P. J. Martín-
420 Álvarez, and M. Moreno-Arribas, *Food Control*, 2008, **19**(9), 835-841.
- 421 62. C. Santos-Buelga and V. De Freitas: 'Influence of phenolics on wine organoleptic
422 properties', in 'Wine chemistry and biochemistry', 529-570; 2009, Springer.
- 423 63. E. Haslam, E. Haslam, *Practical Polyphenol, from Structure to Molecular
424 Recognition and Physiological Action*, 1998.
- 425 64. L. Laguna, B. Bartolomé, and M. V. Moreno-Arribas, *Trends in Food Science &
426 Technology*, 2017, **59**, 49-59.
- 427 65. A. E. Hagerman and L. G. Butler, *Journal of Biological Chemistry*, 1981, **256**(9),
428 4494-4497.
- 429 66. J. Prinz and P. Lucas, *Journal of oral rehabilitation*, 2000, **27**(11), 991-994.
- 430 67. A. Rinaldi, A. Gambuti, and L. Moio, *J Food Sci*, 2012, **77**(4), C485-490.
- 431 68. C. Poncet-Legrand, D. Cartalade, J.-L. Putaux, V. Cheynier, and A. Vernhet,
432 *Langmuir*, 2003, **19**(25), 10563-10572.
- 433 69. S. Lubbers, C. Verret, and A. Voilley, *LWT - Food Science and Technology*, 2001,
434 **34**(4), 262-265.
- 435 70. Y. Shi, I. Minami, M. Grahn, M. Björling, and R. Larsson, *Tribology International*,
436 2014, **69**, 39-45.

437