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1 **On the glucoside analysis: simultaneous determination of free and**
2 **esterified steryl glucosides in olive oil. Detailed analysis of standards as**
3 **compulsory first step.**

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12

13

14 **ABSTRACT**

15 This work covers two important gaps in the field of micronutrient databases:
16 herein we describe a short and easy protocol that allows the analysis of both
17 free and esterified steryl glucosides in olive oil. By utilizing accurate
18 quantitative methods we achieve a better understanding of olive oil composition
19 and health promoting properties. The procedure consists of isolating the fraction
20 of interest through solid phase extraction, and using gas chromatography-flame
21 ionization detection for both identification and quantification of the derivatised
22 species. Additionally, mass-spectrometry detection has been utilized for
23 confirming the identity of the individual esterified steryl glucosides in some
24 cases.

25 The method's limit of detection has been set at 0.37 mg/kg for each free
26 steryl glucoside and 0.20 mg/kg for each esterified steryl glucoside, whereas
27 the recoveries are around 96% and 77%, respectively.

28 Finally, we provide a complete analysis of the commercial standard for
29 esterified steryl glucosides, since such information was not yet available.

30

31 Keywords: edible oils, olive oil, Solid Phase Extraction (SPE), steryl glucosides,
32 esterified steryl glucosides, acylated steryl glucosides.

33

34 **1. Introduction**

35 Plant glycolipids form a group of compounds consisting mainly of steryl
36 glucosides, sphingoglycolipids and glyceroglycolipids (Sugawara, & Miyazawa,
37 1999). Those steryl glucosides can be found either as free molecules or as
38 esterified compounds (Grille, Zaslowski, Thiele, Plat, & Warnecke, 2010).

39 Free steryl glucosides (SG) are phytosterol conjugates where one glucose
40 moiety binds at the C3 position of the sterol residue via an acetyl bond. When
41 this kind of sterol is esterified to a fatty acid at the C6 position of the sugar
42 moiety, it gives rise to esterified (also named as acylated) steryl glucosides
43 (ESG).

44 The positive effects of plant sterols regarding their cholesterol lowering ability
45 have been known for almost more than sixty years and this drove the use of
46 free sterol solutions for the treatment of pathologies related to elevated
47 cholesterol in blood (Pollak, 1953). Later on margarine spreads, cheese,
48 chocolates, etc. –much taster than the former pharmaceutical formulations-
49 were manufactured as functional foods where the added free sterols (FS) had

50 been substituted by esterified sterols (ES) by means of increasing their solubility
51 in the lipidic matrices (MacKay, & Jones, 2011). Interestingly, the observed
52 positive effects were not only due to the added phytosterols, but also due to
53 sterols naturally occurring in the food (Ostlund, 2002).

54 Concerning phytosterol conjugates (SG and ESG), the lack of knowledge
55 about steryl glucosides and their metabolism in humans has resulted in them
56 being cast aside in the field of food enrichment. However, between 2009 and
57 2011 Lin and co-workers demonstrated the potential health benefits of
58 glycosylated phytosterols both in their free and esterified forms (Lin, Ma,
59 Racette, Anderson Spearie, & Ostlund, 2009; Lin, Ma, Moreau, & Ostlund,
60 2011). For instance, reductions of around 37% in cholesterol absorption were
61 observed in the gut of human subjects thanks to the action of SG; this decrease
62 was comparable to 30% obtained with ES (Lin, Ma, Racette, Anderson Spearie,
63 & Ostlund, 2009).

64 Although it is well known that the richest natural sources of dietary
65 phytosterols are edible vegetable oils, like those from corn, palm, rapeseed or
66 sunflower (Piironen, Lindsay, Miettinen, Toivo, & Lampi, 2000), relatively little
67 information is available regarding their glycosylated sterol content. In this sense,
68 Nyström et al. published detailed data on the content and sterol composition of
69 glycosylated sterols in more than forty plant matrices. This was to facilitate
70 comparisons of the potential contribution of different foods as sources of steryl
71 glucosides when included in the diet (Nyström, Schär, & Lampi, 2012).

72 Regarding the methods of analysis, research on this field has been constant
73 since ipuranol –the first steryl glucoside relative- was isolated in 1908 (Power, &
74 Salway, 1913). We have summarized the progression of such work in a

75 previous publication, where we have described the development of a specific
76 protocol for the characterization of the SG fraction in olive oil (Gómez-Coca,
77 Pérez-Camino, & Moreda, 2012).

78 The aim of this work is to cover two important gaps: to provide a short and
79 easy protocol that allows the analysis of both SG and ESG directly, and to
80 provide a methodology that allows greater understanding of the chemical
81 compositions of olive oil and other edible oils. Besides, it will contribute to set up
82 nutrient databases of specific bioactive compounds, present in edible matrices.

83

84 **2. Experimental**

85 *2.1. Reagents*

86 The internal standard cholesterol β -D-glucoside (ChSG) was purchased from
87 Sigma-Aldrich Co. LLC (St. Louis, Missouri, USA). The commercial mixtures of
88 ESG and SG used as references were purchased from Matreya LLC (Pleasant
89 Gap, Pennsylvania, USA). Chlorotrimethylsilane, hexamethyldisilazane, iodine,
90 pyridine and sulfuric acid were from Fluka (Buchs, Switzerland). Ethanol and
91 methanol were supplied by Romil Ltd. (Waterbeach, Cambridge, GB), formic
92 acid, heptane and hexane by VWR International, LLC (West Chester,
93 Pennsylvania, USA), sodium chloride and trichloromethane by Panreac
94 (Montcada I Reixac, Barcelona, Spain), sodium methoxide by Acros Organics
95 (Thermo Fisher Scientific, Geel, Belgium), and *tert*-butyl methyl ether by
96 Scharlau Chemie SA (Barcelona, Spain). Talc was from Talcoliva (Boñar León,
97 Spain). All chemical reagents were analytical grade.

98

99 *2.2. Samples and standard solutions*

100 Samples of blended olive oils were purchased from local markets. These
101 samples belonged to the extra virgin olive oil (EVOO) and virgin olive oil (VOO)
102 categories, and were named as EVOO blend I to VI, and VOO blend I to V,
103 respectively. Varietal EVOO (arauco, arbequina, coratina, cornicabra,
104 hojiblanca, and picual cv.) was obtained directly from producers, who also
105 provided us with chemically refined olive oil, chemically refined olive-pomace
106 oils, and oils at different stages of a physical refining process –lampante (crude,
107 virgin olive oil not fit for consumption without further processing), bleached and
108 deodorized olive oils. Varietal EVOO of blanqueta, frantoio, karolla, uovo di
109 picchione, and verdial cultivars were directly prepared in the laboratory using an
110 Abencor® system (Section 2.3).

111 Blended EVOO, blended VOO and five out of the eight samples of lampante
112 olive oil (LOO) analysed in this study were classified accordingly by a panel of
113 tasters from the Instituto de la Grasa in Seville, Spain. This institution is
114 recognized by the IOC (International Olive Council, 2004), and the panel
115 followed the methodology on sensory analysis of olive oil provided by IOC
116 (International Olive Council, 2011) and EU Regulation (European Commission
117 Regulation, 1991) and also applied in other studies (Gómez-Coca, Moreda, &
118 Pérez-Camino, 2012). The oil was graded by comparing the median value of
119 the defects and the median for the fruity attribute with reference ranges.

120 According to the European reference ranges (European Commission
121 Regulation, 1991) the category EVOO refers to crude, virgin olive oil in which
122 the median of the defects is 0, and the median of the fruity attribute is above 0;
123 the category VOO refers to crude, virgin olive oil in which the median of the
124 defects is above 0 but not higher than 3.5 and the median of the fruity attribute

125 is above 0; finally, the category LOO refers to crude, virgin olive oil in which the
126 median of the defects is above 3.5.

127 Crude oil chemical refining consists of degumming (for the removal of
128 phospholipids), neutralization with hot caustic soda (for the removal of free fatty
129 acids), bleaching, and deodorization. Crude oil physical refining consists of
130 degumming, bleaching and steam stripping to remove free fatty acids, odour
131 and volatile organic compounds, all in one step.

132 Stock solutions of ESG were prepared by dissolving the standard mixture in
133 chloroform. Stocks solutions of SG and ChSG were made with a
134 chloroform:methanol 2:1 (v/v) blend. In all cases the concentration was 100
135 µg/ml.

136 Samples were prepared just before the analysis in the following way: 100 µl
137 of ChSG stock solution were introduced into a 25 ml flask and evaporated to
138 dryness in a current of nitrogen. Then 2 g of oil were weighed into the same
139 flask and dissolved in 4 ml chloroform.

140 To derivatise the hydroxyl groups, solutions consisting of
141 pyridine:hexamethyldisilazane:chlorotrimethylsilane 9:3:1 (v/v/v) were prepared
142 periodically and let to decant at 13 °C before use. Once the samples had been
143 extracted, they were dried under a gentle flux of nitrogen, dissolved with 200 µl
144 of the aforementioned mixture, and kept for one hour at 60 °C before taking
145 them to the gas chromatograph. Complete silylation was monitored as
146 described previously (Gómez-Coca, Pérez-Camino, & Moreda, 2012).

147

148 *2.3. Oil extraction*

149 Olive fruits were obtained from an irrigated orchard (drip irrigation) in the
150 southern part of Spain, under optimal cultivation parameters. In this region
151 winters are mild, January being the coldest month with temperatures between 5
152 and 16 °C, and 179 sunshine hours. Precipitations vary from 400 to 800 mm per
153 year, and they concentrate from October to April, with December being the
154 wettest month (95 mm). The olive plantation was in classical frame with 300
155 trees per hectare. Fruits from five different cultivars (blanqueta, frantoio, karolla,
156 uovo di pichone, and verdial) were hand-picked in January, 2012. To assure
157 maximum quality oil was extracted, an Abencor® system and a small-quantity
158 mill simulating commercial oil-extraction system (MC2 Ingeniería de Sistemas,
159 Sevilla, Spain) were used. The olives were crushed with an Abencor® hammer
160 mill equipped with a 4 mm sieve. 700 g of paste were processed using the
161 system's malaxer and centrifuge with the conditions: malaxation temperature,
162 30 °C; malaxation time, 20 min, plus 10 min after water addition; amount of
163 distilled water added to the paste, 100 ml; amount of talc added, 40 g. The
164 mixtures were left to decant before proceeding to oil separation.

165

166 *2.4. Instrumentation*

167 Gas chromatography (GC) analyses of the sterol glucosides were carried out
168 with an Agilent 6890N Gas Chromatograph (Agilent Technologies, Santa Clara,
169 California) equipped with an Agilent 7683B Automatic Liquid Sampler and a
170 flame ionization detector (FID). Data was acquired with the Agilent ChemStation
171 for GC System program. The conditions for the GC assays were: TRB-5HT
172 column (5% diphenyl-95% dimethylpolysiloxane; 30 m x 0.32 mm ID x 0.10 µm
173 film; Teknokroma, Sant Cugat del Vallés, Barcelona, Spain), 1.0 µl injection

174 volume, hydrogen carrier gas at 3 ml/min and ECP cool on-column injection.

175 The oven temperature program was: 80 °C (1 min), 50 °C/min to 320 °C (12
176 min), and 40 °C/min to 360 °C (19 min). The detector temperature was 360 °C.

177 GC analyses of the fatty acids at the C6 position of the ESG were carried out
178 under the following conditions: SP2380 column (poly 90% biscyanopropyl/10%
179 cyanopropylphenyl siloxane), 60 m x 0.25 mm ID x 0.20 µm film (Sigma-Aldrich
180 Co. LLC, St. Louis, Missouri, USA), 4.0 µl injection volume and split injection.
181 The oven temperature program was: 170 °C (10 min), and 2 °C/min to 200 °C
182 (10 min). The injector and detector temperatures were 210 °C and 240 °C,
183 respectively.

184 An Ultimate 3000 LC System (Thermo Scientific, Vertex Technics S.L.,
185 Hopitalet de Llobregat, Barcelona, Spain) equipped with an Ultimate 3000 NS
186 pump and an Ultimate 3000 NS autosampler, was also used during the analysis
187 of the ESG. A first stage separation of the ESG was performed on a
188 LiChrospher Si60 column (250 x 4 mm ID, 5 µm; Merk Chemicals Spain, Mollet
189 del Vallès, Barcelona, Spain), using an isocratic mode of 80% chloroform-20%
190 methanol for 15 min. The flow rate was 0.4 ml/min and the column temperature
191 was kept at 30 °C. The injection volume was 30 µl. This system was coupled to
192 a microTOF II mass spectrometer paired to an APCI source (Bruker Daltonics,
193 Bruker Corporation in Spain, Rivas-Vaciamadrid, Madrid, Spain). The APCI inlet
194 conditions were: heated vaporization temperature 400 °C; curtain gas 2 bar, 8
195 ml/min; dry temperature 200 °C. Further separation of the individual ESG was
196 performed on a Kinetex C18 column (50 x 2.10 mm ID, 2.6 µm; Phenomenex,
197 Torrance, California, USA). The elution solvent was a mixture of water+1%
198 formic acid: methanol+1% formic acid, beginning with a 50:50 solution (10 min),

199 then 15:85 (10 min) and finally 0:100 (11 min). The flow rate was 0.8 ml/min and
200 the column temperature was 30 °C. The injection volume was 30 µl. Here the
201 APCI inlet conditions were: heated vaporization temperature 300 °C; curtain
202 gas, 2.5 bar, 4 l/min; dry temperature 230 °C.

203

204 *2.5. Analysis of standards*

205 *2.5.1. Purity of the commercial products*

206 According to the manufacturers, the standards for both SG and ESG were of
207 natural origin, whereas ChSG was a synthetic product.

208 The determination of the purity (and composition) of the SG mixture had
209 been done in previous work (Gómez-Coca, Pérez-Camino, & Moreda, 2012).
210 We applied the same procedure to the ChSG and ESG standards. In short, we
211 carried out thin layer chromatography (TLC) tests in chloroform:methanol:water
212 70:15:2 (v/v/v), spraying with concentrated sulfuric acid and developing at 190
213 °C. We repeated the tests eluting with chloroform:methanol:water 80:20 (v/v),
214 and developing in a chamber containing I₂.

215

216 *2.5.2. Determination of the molecular composition of the individual ESG*

217 The commercially available standard for ESG consisted of a mixture of
218 esterified phytosterol glucosides. To identify those glucosides, two approaches
219 were carried out: hydrolysis of the ESG with GC-FID analysis of the fatty acids
220 and SG released, and direct analysis of the individual molecular species by
221 normal phase (NP) and reverse phase (RP) high performance liquid
222 chromatography (HPLC) with MS detection.

223 Hydrolysis of the ESG was achieved according the following procedure: 3 ml
224 of a 100 µg commercial ESG/ml solution were evaporated to dryness under
225 nitrogen flux. We added 3 ml of a 0.119 M CH₃NaO solution, and incubated for
226 one hour at 50 °C. Next we added 3 ml of heptane and shook cautiously. After
227 phase separation, the upper layer was evaporated until a 200 µl volume was
228 reached. The unbound fatty acids as fatty acid methyl esters (FAME's) were
229 chromatographed according to the conditions given in Section 2.4.

230 Isolation of the resulting SG was carried out following the procedure
231 published by Breinhölder and coworkers (Breinhölder, Mosca, & Lindner, 2002),
232 with some modifications. After removing the upper layer, to the lower layer we
233 added 4 ml water, 0.5 ml saturated sodium chloride solution and 5 ml
234 chloroform. The mixture was shaken vigorously for one minute, 0.5 ml ethanol
235 was added and centrifuged at 2000 rpm for 10 minutes. Immediately afterwards
236 the lower phase was evaporated to dryness under nitrogen flux. We
237 chromatographed the trimethyl silyl (TMS) ether derivatives of the SG (Section
238 2.2) according to the conditions given in Section 2.4.

239 In parallel we also chromatographed an ESG solution identical to the starting
240 one (3 ml of a 100 µg/ml solution evaporated and derivatised) by means of
241 control.

242 HPLC-MS analysis was carried out as described in Section 2.4, by direct
243 injection of the ESG standard solutions.

244

245 *2.5.3. Composition of the ESG mixture*

246 The composition of the ESG mixture was determined by 13 doubled runs of
247 GC analysis (Section 2.4) of the TMS ether derivatives of the standard at three

248 different concentrations (10, 20, and 25 mg commercial mixture/kg oil). The
249 relative amount of each peak was calculated on the basis of the corresponding
250 area as a percentage of the area equivalent to the total ESG.

251 The ESG peaks were identified by their absolute and relative retention time
252 (Rt), which were the result of fourteen injections of standard solutions at three
253 different concentrations (3.92, 7.84, and 9.80 mg ESG/kg oil). The changes in
254 the chromatographic conditions, with respect to those applied previously
255 (Gómez-Coca, Pérez-Camino, & Moreda, 2012), made us also recalculate the
256 absolute and relative Rt of SG. In this case, the values given are the result of
257 seven injections of standard solutions at two different concentrations (9.80 and
258 24.50 mg SG/kg). The absolute Rt was always measured to three decimal
259 places.

260 The Rt window for each target analyte was established by multiple GC
261 injections of the blank sample (Section 2.7) spiked with ChSG, SG and ESG
262 standard solutions (4.90 mg ChSG/kg; 2.45, 4.90, 7.35, 9.80, and 24.50 mg
263 SG/kg, and 3.92, 7.84, 9.80, 11.76, and 15.68 mg ESG/kg) to compensate for
264 shifts in absolute Rt as a result of chromatographic variability.

265

266 *2.6. Evaluation of the method*

267 Recovery tests to assess the reproducibility of the method, and trials to
268 establish the limit of detection (LOD) and the limit of quantitation (LOQ) were
269 performed using in-house blank oil labelled as *refined olive-pomace oil* (oil
270 comprising exclusively olive-pomace oils that have undergone classical
271 refining), which had showed no chromatographic peaks within the Rt windows
272 of any of the glucosides under study.

273 The sensitivity (the minimum concentration of analyte that could be
274 measured and reported with an acceptable confidence that it was higher than
275 zero) was determined by spiking six samples of refined olive-pomace oil with
276 SG and ESG standard solutions at increasingly lower concentrations. The
277 accepted concentration values were those that produced sharp, symmetrical
278 analyte peaks with a signal-to-noise ratio of at least 2 and with no tailing or
279 shoulders.

280 To obtain recovery values, fourteen more samples of refined olive-pomace oil
281 were spiked with the standard mixtures at different concentrations (1.43, and
282 2.86 mg β -sitosteryl glucoside/kg, and 4.13, and 8.26 mg esterified β -sitosteryl
283 glucoside/kg).

284

285 *2.7. Extraction procedure and steryl glucoside quantification*

286 The SPE procedure developed previously to isolate SG (Gómez-Coca,
287 Pérez-Camino, & Moreda, 2012) was utilized, with some modifications. The
288 cartridges (ExtraBond Si-1 g; Scharlab S.L., Barcelona, Spain) were
289 conditioned with 5 ml *tert*-butyl methyl ether, 5 ml hexane, 5 ml chloroform, and
290 10 ml of a newly prepared blend consisting of chloroform:methanol 4:1 (v/v).
291 Samples were then quantitatively transferred to the cartridges using 1 ml
292 chloroform to wash the walls of the containers. The loaded cartridges were
293 washed with 10 ml hexane, 10 ml *tert*-butyl methyl ether and 10 ml chloroform,
294 applied one after the other. The adsorbed SG (including the internal standard)
295 and ESG were eluted with 5 ml of the chloroform:methanol 4:1 (v/v) solution.
296 The procedure was carried out under negative pressure (1 in.Hg). The eluted

297 fractions were evaporated to dryness under nitrogen flux, derivatised and
298 chromatographed using cool on-column injection.

299 The quantitative evaluation of the SG was carried out using ChSG as internal
300 standard. The response coefficient for ChSG equalled 1. Once the areas of the
301 ChSG and SG peaks had been determined using the integrator, we calculated
302 the concentration of each individual SG in mg/kg oil as follows:

$$303 \quad \text{SG } x = A_x \cdot m_{IS} \cdot 1000/A_{IS} \cdot m_S$$

304 Where:

305 A_x is the peak area for steryl glucoside x ,

306 A_{IS} is the area of the ChSG peak,

307 m_{IS} is the mass of the ChSG added, in milligrams,

308 m_S is the mass of the sample used for the determination, in grams.

309 Regarding the ESG, the composition of the commercially available standard
310 was a mixture of the same ESG expected in the real samples, therefore
311 confirming the need to use external standardization during the quantification
312 stage. The calibration curves were carried out by progressively diluting one
313 ESG standard solution. The final derivatised volumes were of 500 μl . These
314 critical volumes were measured with a calibrated 200-1000 μl precision pipette
315 (-1.00 μl systematic error). The curves consisted of three points that
316 encompassed the range from 15.36 to 59.09 mg ESG mixture/kg
317 corresponding, i.e. to 3.17 to 12.20 mg esterified β -sitosteryl glucoside/kg. One
318 specific calibration curve was prepared for each ESG.

319 Two separately prepared standard solutions of ESG (corresponding to 3.92
320 and to 7.84 mg ESG/kg) were also analysed in each GC run to check the
321 suitability of quantitation by the external calibration curve (data not shown).

322

323 **3. Results and discussion**324 *3.1. Analysis of standards*325 *3.1.1 Purity of the commercial products*

326 TLC analysis reported that the ChSG standard contained 98% ChSG
327 minimum, whereas in the case of the ESG commercial mixture, analysis
328 showed only 39% purity, which was in agreement with the results observed
329 during the chromatography analysis (data not shown).

330

331 *3.1.2 Determination of the molecular composition of the individual ESG*

332 ESG are bulky molecules whose complexity deserves some considerations.
333 From a structural point of view one can split the moiety in two parts, an SG
334 fragment and a fatty-acyl residue. According to the manufacturer, the ESG
335 mixture purchased was based on β -sitosteryl (6'-O-palmitoyl)- β -D-glucoside,
336 and its origin was natural (from soybean extract). Therefore it was reasonable to
337 assume that not only β -sitosteryl (6'-O-palmitoyl)- β -D-glucoside was to be
338 expected in the commercial product. This was supported by observing not only
339 one, but up to three (even four), GC-peaks (Figure 1). If one analyses the sterol
340 composition and content of soybean oil (Moreda, Pérez Camino, & Cert, 2004),
341 the following desmethylsterols are mainly present: β -sitosterol (35-72%),
342 campesterol (16-34%) and stigmasterol (10-28%). There may also be small
343 quantities of Δ 7-stigmastenol (1-5.2%), Δ 7-avenasterol (1-5%), and Δ 5-
344 avenasterol (1-4%). When these sterols carry a sugar residue attached at the
345 C3 position, the order of elution from a GC column is the same as when free
346 sterols are injected (Gómez-Coca, Pérez-Camino, & Moreda, 2012). However,

347 in the case of ESG, one could expect some changes depending on the fatty-
348 acyl residue esterifying the C6 position. On the other hand, the fatty acid
349 content of soybean oil consist of 50-62% linoleic acid (C18:2), 17-26% oleic
350 acid (C18:1), 8-13% palmitic acid (C16:0), 4-10% linolenic acid (C18:3), and 3-
351 5% stearic acid (C18:0) (Zamora, & Hidalgo, 2004), thus the number of possible
352 ESG combinations.

353 After hydrolysing the ESG present in the standard mixtures, it was clear that
354 under the working conditions, the molecules broke at the ester bond formed
355 between the fatty acid at C6 position and the sugar residue. Actually, GC-
356 results showed four unique SG peaks and four major fatty acid signals.

357 The profile of the released SG corresponded to that obtained previously in
358 the analysis of the SG standards (Gómez-Coca, Pérez-Camino, & Moreda,
359 2012) therefore, we concluded that the sterol residues of the ESG commercial
360 mixture were campesterol, stigmasterol, β -sitosterol and Δ 5-avenasterol. On the
361 other hand, the relative proportions of the four SG released after hydrolysis
362 were the same, within the error limits, to those of the four ESG showed in the
363 gas chromatograms; that is ca. 25%, 20%, 52% and 3% for (esterified)
364 campesteryl glucoside ((E)CSG), (esterified) stigmasteryl glucoside ((E)SSG),
365 (esterified) β -sitosteryl glucoside ((E)BSSG), and esterified Δ 5-avenasteryl
366 glucoside ((E)AvSG), respectively. This indicates that the ESG elute in the
367 same order as the free sterols and SG, and that there is no influence (beyond
368 that of increasing the temperature of evaporation) of the attached C6-fatty acid
369 under the applied GC conditions.

370 Regarding the analysis of the fatty-acyl residues, four fatty acids were
371 predominantly present: linoleic acid ($30.44 \pm 0.22\%$), palmitic acid ($28.76 \pm$

372 0.05%), stearic acid ($12.40 \pm 0.34\%$) and oleic acid ($12.96 \pm 0.39\%$). Small
373 amounts of linolenic acid and of behenic acid (C22:0) were also detected.
374 These results are the average of three independent measures.

375 To confirm the identity of the four GC peaks corresponding to the esterified
376 sterol glucosides in the standard mixture, a $39.19 \mu\text{g ESG/ml}$ solution was taken
377 to both NP and RP HPLC-MS, where all spectra were acquired by APCI in the
378 positive ion mode (Section 2.4).

379 In NP HPLC-MS the predominant ions in the mass spectra were $[\text{M}-\text{H}_2\text{O}+\text{H}]^+$.
380 In these spectra five major ions (those with intensity higher than $5 \cdot 10^{-5}$) could be
381 identified, which were tentatively assigned to: β -sitosteryl (6'-*O*-linoleoyl)- β -D-
382 glucoside, ($m/z = 821$), β -sitosteryl (6'-*O*-palmitoyl)- β -D-glucoside ($m/z = 797$),
383 stigmasteryl (6'-*O*-linoleoyl)- β -D-glucoside ($m/z = 819$), campesteryl (6'-*O*-
384 linoleoyl)- β -D-glucoside ($m/z = 807$) and stigmasteryl (6'-*O*-stearoyl)- β -D-
385 glucoside ($m/z = 823$). The ions with intensities between $5 \cdot 10^{-5}$ and $2.5 \cdot 10^{-5}$
386 were: campesteryl (6'-*O*-palmitoyl)- β -D-glucoside ($m/z = 783$), stigmasteryl (6'-
387 *O*-palmitoyl)- β -D-glucoside ($m/z = 795$), β -sitosteryl (6'-*O*-stearoyl)- β -D-
388 glucoside, ($m/z = 821$), and campesteryl (6'-*O*-oleoyl)- β -D-glucoside ($m/z =$
389 807). The presence of Δ^5 -avenasteryl (6'-*O*-palmitoyl)- β -D-glucoside ($m/z =$
390 795) and of Δ^5 -avenasteryl (6'-*O*-oleoyl)- β -D-glucoside ($m/z = 821$) could be, so
391 far, neither confirmed nor rejected. These ions have been given in decreasing
392 order of signal intensity.

393 In RP HPLC-MS we scanned the $[\text{M}+\text{H}_2\text{O}]^+$ and the $[\text{aglycone}-\text{H}_2\text{O}+\text{H}]^+$ ions
394 from those ESG which were to be present according to the preliminary results.
395 Those ESG may consist by different combinations between CSG, SSG and
396 BSSG, and stearic acid, oleic acid and linoleic acid. We assumed the presence

397 of EAvSG, since this has the same molecular weight as the corresponding
398 ESSG. This analysis showed that our commercial standard had the following
399 composition for the ECSG: 7.0% campesteryl (6'-*O*-stearoyl)- β -D-glucoside (m/z
400 = 818 and m/z = 383, respectively), 10.9% campesteryl (6'-*O*-oleoyl)- β -D-
401 glucoside (m/z = 844 and m/z = 383, respectively) and 7.4% campesteryl (6'-*O*-
402 linoleoyl)- β -D-glucoside (m/z = 842 and m/z = 383, respectively); for the ESSG
403 (and EAvSG): 7.8% stigmasteryl (6'-*O*-stearoyl)- β -D-glucoside (m/z = 830 and
404 m/z = 395, respectively), 4.7% stigmasteryl (6'-*O*-oleoyl)- β -D-glucoside (m/z =
405 856 and m/z = 395, respectively), and 11.7% stigmasteryl (6'-*O*-linoleoyl)- β -D-
406 glucoside (m/z = 854 and m/z = 395, respectively); finally, the EBSSG is a
407 mixture of: 15.8% β -sitosteryl (6'-*O*-stearoyl)- β -D-glucoside (m/z = 832 and m/z
408 = 397, respectively), 3.4% β -sitosteryl (6'-*O*-oleoyl)- β -D-glucoside (m/z = 858
409 and m/z = 397, respectively), and 31.2% β -sitosteryl (6'-*O*-linoleoyl)- β -D-
410 glucoside (m/z = 856 and m/z = 397, respectively).

411

412 3.1.3 Composition of the ESG mixture

413 From HPLC-MS results we could only be sure of the presence of three ESG
414 (ECSG, ESSG and EBSSG) whereas the GC-FID analysis might show four
415 peaks, which we identified as ECSG, ESSG, EBSSG and EAvSG based on the
416 results obtained with the SG standard mixture (Figure 1). The proportions
417 (taking into account 39% purity) were: 9.97% (2.03% Relative SD), 7.09%
418 (8.46% Relative SD), 20.64% (7.49% Relative SD) and 1.49% (10.37% Relative
419 SD), respectively. These quantities were utilized to calculate the concentration
420 of each glucoside in the standard solutions and to build the calibration curves.

421 The relative Rt values were kept constant all over the study. Those values
422 and the Rt windows for both SG and ESG are shown in Table1 and Table 2,
423 respectively.

424

425 *3.2. Evaluation of the method*

426 In previous work we proved that, in olive oil, the lowest concentration of
427 BSSG detectable to fulfil a number of predetermined conditions was 0.37 mg/kg
428 (Gómez-Coca, Pérez-Camino, & Moreda, 2012). Consequently we focused on
429 BSSG as it was the only SG present in olive oil, but the same limit has proven
430 to be true for CSG and SSG (present in other edible oils such as soybean),
431 which was to be expected since all three different SG should have the same
432 response factor. In this way, hundred per cent of the samples (n = 6) spiked
433 with such SG concentration gave signals within the acceptance criteria. In the
434 cases of the ESG, LOD was 0.20 mg/kg for the three species tested (ECSG,
435 ESSG, and EBSSG). That concentration always (n = 6) gave signals clearly
436 distinguishable from the background.

437 We set the empirical LOQ for the ESG at 0.37 mg/kg. According to the
438 definition of empirical LOQ (Armbruster, Tillman, & Hubbs, 1994), 0.37 mg/kg
439 will be the lowest concentration at which the acceptance criteria are met, and
440 the lowest analyte concentration (0.46 mg/kg) is (at the minimum) 20% above
441 such limit. Following the same reasoning, the LOQ for SG was established at
442 0.49 mg/kg (Gómez-Coca, Pérez-Camino, & Moreda, 2012).

443 The global recovery of the method was assessed with fourteen samples of
444 refined olive-pomace oil spiked with SG (Table 3) and ESG (Table 4) standard
445 solutions at different concentrations. In the cases of SG there was an improved

446 recovery with respect to that obtained previously: $96.33 \pm 2\%$ (3SD) vs. $90.13 \pm$
447 6% (3SD) (Gómez-Coca, Pérez-Camino, & Moreda, 2012), probably due to the
448 optimization of the extraction procedure and the use of internal standardization.
449 However, recoveries of the ESG do not reach 80%, the actual value being
450 $76.58 \pm 0.21\%$ (3SD). This recovery was independent of the presence of the SG
451 in the media, and it did not seem related to incomplete silylation.

452

453 *3.3. Extraction procedure and steryl glucoside quantification*

454 The procedure for isolating SG from olive oil samples had been previously
455 developed. Initially the method had been applied to a small number of olive oil
456 samples of different categories and origin, the conclusion reached was that the
457 only SG present was BSSG, whose concentration in olive oil was never higher
458 than 3 mg/kg. It was also clear that in olive oil, there was no correspondence
459 between the FS profile and its SG composition (Gómez-Coca, Pérez-Camino, &
460 Moreda, 2012), as Nyström and colleagues had pointed out in the cases of
461 other matrices (Nyström, Schär, & Lampi, 2012).

462 In olive oil, the FS profile and content is of the utmost importance since both
463 are species-characteristic. The lack of a direct and clear relationship between
464 FS composition and SG content is of the utmost importance, as this kind of oil is
465 constantly under surveillance due to the economic repercussions that fraud may
466 cause.

467 A second point of interest was the presence of those SG in which the C6 of
468 the glucose moiety had been esterified to a fatty acid. If there was not
469 parallelism between FS and SG contents and compositions, why should one
470 expect such relationship between any of them and the ESG profile? Besides,

471 was such profile specific of a particular cultivar or olive oil category? If so, were
472 the predominant defects in the lowest categories related to a specific steryl
473 glucoside composition? How were they affected during the refining processes?
474 And, finally, were SG and ESG affected in the same way?

475 In order to answer all these questions we analysed the steryl glucoside
476 content of olive oil samples from different origins and cultivars, classified under
477 different categories and treated with two different refining processes.

478 Table 5 shows the results of the analysis. In the case of the SG, figures are
479 due to BSSG, although the presence of AvSG is also possible. However, the
480 same is not true for the ESG counterparts. Mass spectrometry results confirmed
481 that in most of the samples, the only ESG present was that corresponding to β -
482 sitosterol (possibly with $\Delta 5$ -avenasterol derivatives), but in some cases (e.g.
483 VOO cv. uovo di pichone) ECSG and ESSG may also contribute to the total
484 ESG content. When considering mono-varietal oils, that is those oils made just
485 from olives from one cultivar, the content of SG in EVOO ranged from $1.68 \pm$
486 0.08 mg/kg to 2.95 ± 0.14 mg/kg, and in VOO from 1.05 ± 0.01 mg/kg to $2.14 \pm$
487 0.10 mg/kg, meaning that they are almost the same within the error limits. As far
488 as the ESG are concern, the concentration ranges are from 2.54 ± 0.04 mg/kg
489 to 9.20 ± 0.36 mg/kg for EVOO, and from 3.14 ± 0.06 mg/kg to 11.95 ± 2.75
490 mg/kg for VOO, showing their similarity.

491 As far as the blends are concerned, the glucoside content depends on the
492 oils of origin.

493 Interestingly, there is no relationship between the SG content and the ESG
494 concentration. In this way, for instance, the oil with the lowest SG concentration

495 is not the one with the lowest ESG content, and vice versa. The ESG:SG ratio is
496 not always above 1, although it seems to be the usual trend.

497 Among all samples analysed, those of LOO exhibited (again) the highest SG
498 contents, although we could not find a relationship with the different defect.

499 Finally, we found the effect of the physical refining processes, whose main
500 difference with the classical (chemical) refining is that there is no treatment with
501 hot caustic soda. The sample labelled as 'LOO III' had been subjected to
502 deodorization and bleaching by the manufacturer. After analysing the steryl
503 glucosides in each of the stages, we saw the drastic effect that physical refining
504 has on SG, whereas such an effect is much softer on the ESG. Actually, there is
505 a decrease of around 88% in the case of SG and of 18-28% in the ESG.

506 However, the effect of the chemical treatment is drastic in all cases, making all
507 glucosides disappear. From this data it was clear to us that it is the acidic media
508 created during the decolourization step that is responsible for the acetyl bond
509 breakage, and therefore for the SG decrease. There seems to be a protective
510 effect of the fatty-acyl residue at the C6 position of the ESG over the acetyl
511 bond, probably due to specific molecule folding, although more studies in this
512 area must be done.

513

514 **4. Conclusion**

515 The steryl glucoside composition and content may be utilized as a further
516 chemical characterization parameter in olive oil. These glucoside compositions
517 may help explain in more detail, the beneficial health properties of olive oil.

518 The only free steryl glucoside clearly present in olive oil is that derived from
519 β -sitosterol: BSSG, whereas in the cases of ESG, other glucosides may
520 contribute to the global content.

521 There is no direct relationship between the FS profile and the composition
522 and content of SG and of ESG. The FS profile does not reflect the ESG
523 composition.

524 There is no relationship between olive oil category and origin, and the ESG
525 profile.

526 Refining process, especially classical refining, destroys ESG.

527 There are still some unanswered questions:

528 a) Is there any relationship between the ESG composition and content of
529 olive oil, and any of the other legislated parameters? (e.g. free fatty acid, wax
530 content, etc.)

531 b) How is it possible that the acetyl bond between the sterol moiety and the
532 sugar residue gets hydrolysed under *in vitro* conditions but resists the extreme
533 media of the gastro-intestinal tract? It is clear that studies on steryl glucoside
534 metabolism remains a gap in this field.

535

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543

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545

546 **References**

547 AOAC (1998). Association of Official Analytical Chemists Guidelines for single
548 laboratory validation of chemical methods for dietary supplements and
549 botanicals. In: *AOAC peer-verified methods program, manual on policies and*
550 *procedures* (1-38). Arlington, Va.

551 Armbruster, D. A., Tillman, M. D., & Hubbs, L. M. (1994). Limit of detection
552 (LOD)/limit of quantitation (LOQ): comparison of the empirical and the
553 statistical methods exemplified with GC-MS assays of abused drugs. *Clin.*
554 *Chem.*, 40, 1233-1238.

555 Breinhölder, P., Mosca, L., & Lindner, W. (2002). Concept of sequential analysis
556 of free and conjugated phytosterols in different plant matrices. *J. Chromatogr.*
557 *B*, 777, 67-82.

558 European Commission Regulation (1991). EEC 2568/91 of 11 July 1991 on the
559 characteristics of olive oil and olive-residue oil and on the relevant methods
560 of analysis, and subsequent amendments. *Off. J. Eur. Commun.*, L248, 1-
561 102.

562 Gómez-Coca, R. B., Pérez-Camino, M. C., & Moreda, W. (2012). Specific
563 procedure for analysing steryl glucosides in olive oil. *Eur. J. Lipid Sci.*
564 *Technol.* DOI:10.1002/ejlt.201200181.

565 Gómez-Coca, R. B., Moreda, W., & Pérez-Camino, M. C. (2012). Fatty acid
566 alkyl esters presence in olive oil vs. organoleptic assessment. *Food Chem.*,
567 135, 1205-1209.

- 568 Grille, S., Zaslowski, A., Thiele, S., Plat, J., & Warnecke, D. (2010). The
569 functions of sterol glycosides come to those who wait: recent advances in
570 plants, fungi, bacteria and animals. *Prog. Lipid Res.*, *49*, 262-288.
- 571 International Olive Council (2004). IOOC recognition of laboratories undertaking
572 the sensory analysis of virgin olive oil. RES-2/90-IV/04, 1-3.
- 573 International Olive Council (2011). Sensory analysis of olive oil. Method for the
574 organoleptic assessment of virgin olive oil. COI/T. 20/Doc. No 15/Rev. 4, 1-
575 14.
- 576 Lin, X., Ma, L., Moreau, R. A., & Ostlund R. E. Jr. (2011). Glycosidic bond
577 cleavage is not required for phytosterol glycoside-induced reduction of
578 cholesterol absorption in mice. *Lipids*, *46*, 701-708.
- 579 Lin, X., Ma, L., Racette, S. B., Anderson Spearie, C. L., & Ostlund R. E. J.
580 (2009). Phytosterol glycosides reduce cholesterol absorption in humans. *Am.*
581 *J. Physiol. Gastrointest. Liver Physiol.*, *296*, G931-G935.
- 582 MacKay, D. S., & Jones, P. J. H. (2011). Phytosterols in human nutrition: type,
583 formulation, delivery, and physiological function. *Eur. J. Lipid Sci. Technol.*,
584 *113*, 1427-1432.
- 585 Moreda, W., Pérez Camino, M. C., & Cert, A. (2004). Analysis of neutral lipids:
586 unsaponifiable matter. In: L. M. L. Nollet (Ed.), *Handbook of food analysis:*
587 *physical characterization and nutrient analysis* 2nd ed. (pp. 313-348). New
588 York-Basel: Marcel Dekker, Inc.
- 589 Nyström, L., Schär, A., & Lampi, A.-M. (2012). Sterol glycosides in plant foods
590 reflect unique sterol patterns. *Eur. J. Lipid Sci. Technol.*, *114*, 656-669.
- 591 Ostlund, R. E. (2002). Phytosterols in human nutrition. *Annu. Rev. Nutr.*, *22*,
592 533-549.

- 593 Piironen, V., Lindsay, D. G., Miettinen, T. A., Toivo, J., & Lampi, A.-M. (2000).
594 Plant sterols: biosynthesis, biological function and their importance to human
595 nutrition. *J. Sci. Food Agric.*, *80*, 939-966.
- 596 Pollak, O. J. (1953). Reduction of blood cholesterol in man. *Circulation*, *7*, 702-
597 706.
- 598 Power, F. B., & Salway, A. H. (1913). The identification of ipuranol and some
599 allied compounds as phytosterol glucosides. *J. Chem. Soc.*, *103*, 399-406.
- 600 Sugawara, T., & Miyazawa, T. (1999). Separation and determination of
601 glycolipids from edible plant sources by high-performance liquid
602 chromatography and evaporative light-scattering detection. *Lipids*, *34*, 1231-
603 1237.
- 604 Zamora, R., & Hidalgo, J. H. (2004). Fatty acids. In: L. M. L. Nollet (Ed.),
605 *Handbook of food analysis: physical characterization and nutrient analysis*
606 2nd ed. (pp. 221-274). New York-Basel: Marcel Dekker, Inc.
607

608 **LEGEND**

609 **Figure 1.** GC-FID chromatograms for steryl glucosides in spiked chemically
610 refined olive pomace oil. This chromatogram has been obtained after analysing
611 the sample according to the proposed method.
612 I. S. Internal standard Cholesterol glucoside, 1. Campesteryl glucoside, 2.
613 Stigmasteryl glucoside, 3. β -Sitosteryl glucoside, 4. (presumptive) Δ^5 -
614 Avenasteryl glucoside, 5. Esterified campesteryl glucoside, 6. Esterified
615 stigmasteryl glucoside, 7. Esterified β -sitosteryl glucoside, 8. (presumptive)
616 Esterified Δ^5 -avenasteryl glucoside.

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617 Table 1. Relative retention time (Rt) values of cholesterol glucoside (ChSG),
618 campesteryl glucoside (CSG), stigmasteryl glucoside (SSG), and Δ^5 -
619 avenasteryl glucoside (AvSG), with respect to β -sitosteryl glucoside (BSSG),
620 with three times their SD. The corresponding Rt windows ($Rt \pm 3SD$) are also
621 given.

	ChSG	CSG	SSG	BSSG	AvSG
Relative Rt	1.201 ± 0.002	1.095 ± 0.002	1.064 ± 0.001	-	0.990 ± 0.002
Rt window [min]	12.274 ± 0.260	13.926 ± 0.100	14.338 ± 0.100	15.244 ± 0.102	15.423 ± 0.080

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623 Table 2. Relative retention time (Rt) values of cholesterol glucoside (ChSG),
 624 esterified campesterol glucoside (ECSG), esterified stigmasterol glucoside
 625 (ESSG), and esterified Δ^5 -avenasterol glucoside (EAvSG), with respect to
 626 esterified β -sitosterol glucoside (EBSSG), with three times their SD. The
 627 corresponding Rt windows ($Rt \pm 3SD$) are also given.

	ChSG	ECSG	ESSG	EBSSG	EAvSG
Relative Rt	2.342 ± 0.011	1.052 ± 0.001	1.037 ± 0.001	-	0.993 ± 0.001
Rt window [min]	12.274 ± 0.260	28.127 ± 0.100	28.535 ± 0.100	29.533 ± 0.135	29.703 ± 0.114

628

629 Table 3. Assessment of the recovery of β -sitosterol glucoside (BSSG) in olive
 630 oil. Samples of chemically refined olive-pomace oil were spiked with standard
 631 solutions at the concentrations indicated in each case.

		Sample + 1.43 mg/kg BSSG	Sample + 2.86 mg/kg BSSG
BSSG [mg/kg]	Trial 1	1.43	2.72
	Trial 2	1.51 [#]	2.79
	Trial 3	1.30	2.81
	Trial 4	1.39	-
	Trial 5	1.47 [#]	-
	Trial 6	1.28	-
	Trial 7	1.39	-
	Trial 8	1.21	-
Mean value [mg/kg]		1.37	2.77
SD		0.01	0.01
RSD [%]		0.73	0.36
Recovery [%]		95.80	96.85

[#] Recoveries of 107% and 103% (Trials 2 and 5, respectively) are within the

AOAC accepted limits according to which concentrations of the order of
 magnitude of ppm must show recovery values between 80% and 115% (AOAC,
 1998)

Table 4. Assessment of the recovery of esterified β -sitosteryl glucosides (EBSSG) in olive oil. Samples of chemically refined olive-pomace oil were spiked with standard solutions at the concentrations indicated in each case.

		Sample + 4.13 mg/kg EBSSG	Sample + 8.26 mg/kg EBSSG
EBSSG [mg/kg]	Trial 1	3.12	6.30
	Trial 2	3.13	6.32
	Trial 3	3.30	6.33
	Trial 4	3.23	6.30
	Trial 5	3.20	
	Trial 6	3.03	-
	Trial 7	2.55	-
	Trial 8	3.40	-
	Trial 9	3.38	-
	Trial 10	3.38	
Mean value [mg/kg]		3.17	2.77
SD		0.06	0.01
RSD [%]		1.89	0.36
Recovery [%]		76.76	76.39

Table 5. Steryl glucoside (SG) and esterified steryl glucoside (ESG) contents in mg/kg of olive oil from different cultivars and categories (extra virgin, virgin, and lampante olive oils: EVOO, VOO, and LOO, respectively). Three times the standard deviation is also given.

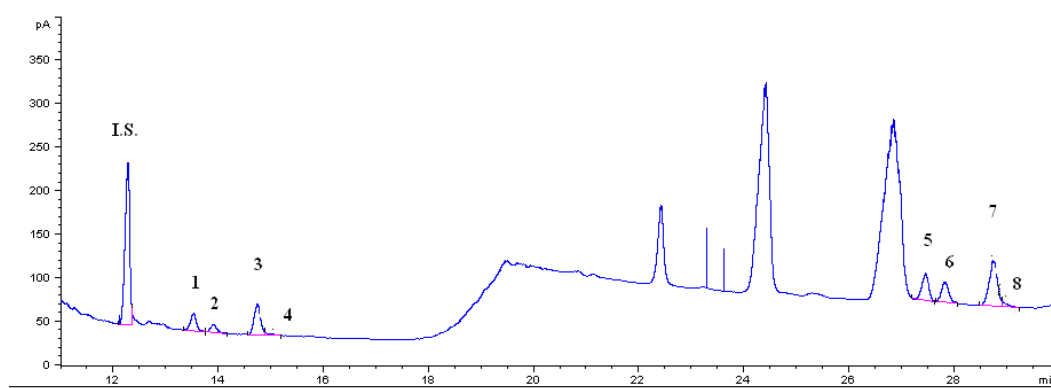
Oil	SG [mg/kg]	3SD (n)[§]	ESG [mg/kg]	3SD (n)[§]
EVOO blend I	1.77	0.06 (2)	6.16	0.29 (2)
EVOO blend II	2.06	0.01 (3)	2.73	0.15 (3)
EVOO blend III	1.38	0.04 (3)	4.28	0.01 (2)
EVOO blend IV	1.91	0.01 (3)	5.16	0.09 (3)
EVOO blend V	1.28	0.03 (4)	3.64	0.30 (3)
EVOO blend VI	2.08	0.01 (2)	4.94	0.03 (2)
EVOO cv. Arauco	1.68	0.08 (2)	4.90	0.33 (2)
EVOO cv. Arbequina	2.15	0.02 (2)	9.20	0.36 (2)
EVOO cv. Coratina	1.73	0.04 (2)	2.54	0.04 (2)
EVOO cv. Cornicabra	2.95	0.14 (2)	2.67	0.19 (3)
EVOO cv. Hojiblanca	1.88	0.01 (2)	3.79	0.01 (2)
EVOO cv. Picual	1.97	0.01 (2)	2.78	0.01 (2)
VOO blend I	2.60	0.03 (3)	4.17	0.30 (2)
VOO blend II	2.37	0.04 (2)	3.09	0.01 (2)
VOO blend III	1.67	0.01 (2)	2.42	0.01 (2)
VOO blend IV	0.90	0.01 (2)	1.91	0.01 (2)
VOO blend V	2.09	0.07 (3)	5.87	0.91 (3)
VOO cv. Blanqueta	1.34	0.01 (2)	9.63	0.71 (2)
VOO cv. Frantoio	1.47	0.01 (2)	4.32	0.07 (2)
VOO cv. Karolla	2.14	0.10 (4)	3.14	0.06 (4)
VOO cv. Uovo di Pichone	1.96	0.04 (3)	11.29	2.75 (2)
VOO cv. Verdial	1.05	0.01 (2)	4.34	0.13 (2)
LOO I	2.12	0.25 (2)	3.44	0.39 (2)
LOO II	2.23	0.01 (2)	8.50	1.35 (2)
LOO III	3.23	0.01 (3)	6.40	1.02 (3)
Deodorized OO III	0.36	0.01 (2)	5.24	0.28 (2)
Decolorized OO III	0.39	0.01 (2)	4.64	0.31 (2)
LOO (brine: 3.5) [#]	3.66	0.06 (4)	1.61	0.19 (2)
LOO (burnt: 7.0) [#]	1.95	0.01 (2)	1.19	0.05 (2)
LOO (muddy sediment: 7.4) [#]	1.52	0.04 (2)	2.05	0.02 (2)
LOO (rancid: 4.2) [#]	2.08	0.03 (3)	2.36	0.05 (3)
LOO (winey: 4.5) [#]	2.76	0.02 (3)	5.40	0.01 (2)
Refined olive-pomace oil	0.00	0.00 (3)	0.00	0.00 (3)
Chemically Refined Olive Oil	0.00	0.00 (5)	0.00	0.00 (5)

[§]n = number of independent measurements

[#] The main negative attributes are named according to the IOC vocabulary for sensory analysis (IOC, 2011)

632 Figure 1, Gómez-Coca et al.

633



635 **Highlights**

636 **Olive oil's free and esterified steryl glucosides can be analyzed**

637 **simultaneously**

638

639 **Commercial standards of esterified steryl glucosides were analysed by**

640 **HPLC-MS in detail**

641

642 **Free sterol composition and steryl glucoside profiles are independent**

643 **from each other**

644

645 **Olive oil (esterified) steryl glucoside content isn't cultivar- neither quality-**

646 **dependent**

647

648

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