

Freeze-dried phosphatidylcholine liposomes encapsulating various antioxidant extracts from natural waste as functional ingredients in surimi gels

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Abstract

Three antioxidant extracts (collagen hydrolysate, pomegranate peel extract, shrimp lipid extract) were encapsulated in soy phosphatidylcholine liposomes with the addition of glycerol. The particle size of the fresh liposomes ranged from 75.7 to 81.0 nm and zeta potential from –64.6 to –88.2 mV. Freeze-drying increased particle size (199–283 nm), and slightly decreased zeta potential. The lyophilized liposomes were incorporated in squid surimi gels at 10.5 % concentration. An alternative functional formulation was also prepared by adding 2 % of non-encapsulated bioactive extract. The gels were characterized in terms of colour, texture and oxidative stability (TBARS) after processing and also after frozen storage. The incorporation of the freeze-dried liposomes caused a slight decrease in gel strength and contributed to maintaining the stability of the gels during long-term frozen storage. The antioxidant properties of the bioactive extracts, liposomes and *in vitro* digested surimi gels were determined.

Keywords: soy phosphatidylcholine, liposomes, freeze-drying, antioxidants, surimi gels, storage.

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25 **1. Introduction**

26 Growing consumer demand for a healthier lifestyle means that more attention is paid to a
27 balanced diet which guarantees a good nutritional status and prevention of chronic diseases.
28 Some years ago the food industry began incorporating healthy ingredients in prepared foods,
29 so-called “functional foods”, with the aim of providing important nutrients and other bioactive
30 substances in the diet. Food products with antioxidant properties are indicated for the
31 prevention of numerous diseases related to oxidative stress, such as cardiovascular and
32 neurodegenerative diseases, diabetes and cancer (Pham-Huy, He, & Pham-Huy, 2008). Raw
33 materials from animal and plant waste represent an important source of antioxidant
34 compounds that could be included in the composition of functional foods. There has been
35 growing interest in the recovery of fish waste by enzymatic hydrolysis of protein residues. The
36 tunics of giant squid are an excellent source of antioxidant peptides that may protect living
37 cells against free radical-mediated oxidative damage (Mendis et al., 2005; Giménez et al.,
38 2009). The high levels of hydroxyproline, hydrophobic residues and glycosylated peptides in
39 squid tunic gelatin hydrolysates are related to their noticeable antioxidant capacity (Alemán,
40 Giménez, Pérez-Santín, Gómez-Guillén, & Montero, 2011; Alemán, Gómez-Guillén, & Montero,
41 2013). Large amounts of cephalothorax and cuticle waste produced by the crustacean
42 processing industry may also provide a source of bioactive compounds with health-promoting
43 effects. Shrimp (*L. vannamei*) lipid extract, with high proportions of ω -3 polyunsaturated fatty
44 acids (DHA and EPA), astaxanthin and α -tocopherol, also presents antioxidant properties
45 (Gómez-Estaca, Montero, Fernández-Martín, Calvo, & Gómez-Guillén, 2016; Gómez-Estaca,
46 Calvo, Álvarez-Acero, Montero, & Gómez-Guillén, 2017). Among plant waste, pomegranate
47 peel and pulp extracts, rich in polyphenols, mainly ellagitannins, ellagic acid and anthocyanins,
48 are well recognized for their health-promoting effects based on their high antioxidant and

antiproliferative activities (Masci et al., 2016). Surimi gels offer an acceptable food matrix for the incorporation of functional ingredients leading to products with high versatility of presentation. They provide a source of muscle proteins which, after gastrointestinal digestion, could be capable of releasing peptides with biological activities (Ueki, Wan, & Watabe, 2014). The activities, however, may be decreased by thermal degradation during the gelling process or by excessive interactions with the protein matrix (Montero, Giménez, Pérez-Mateos, & Gómez-Guillén, 2005), which could affect their bioaccessibility. A way of enhancing the efficacy and stability of bioactive substances in food systems could be encapsulation in nanoliposomes (Mozafari, Johnson, Hatziantoniou, & Demetzos, 2008). Liposomes are colloidal vesicular structures composed of one or more lipid bilayers that have the ability to act as target delivery carriers for both lipophilic and hydrophilic compounds. The hydrodynamic properties, encapsulation efficiency and stability of the loaded liposomes might be affected, depending on the nature and concentration of the loading material (Tan et al., 2013). In food applications, the use of non-synthetic lecithins for liposomal encapsulation, such as soy lecithin, has a reasonably low cost, does not raise any food legislation concerns and provides nutritional value due to the high essential polyunsaturated fatty acid profile (Laye, McClements, & Weiss, 2008). Moreover, natural soybean phospholipids are able to interact with cellular membranes, changing their fatty acid compositions; therefore, they could act as active ingredients for the treatment of various diseases, such as inflammatory processes, cardiovascular risks, neurological disorders, cancer, etc. (Küllenberg, Taylor, Schneider, & Massing, 2012). The use of polyunsaturated phospholipids for liposomal encapsulation and their addition in the surimi gels contributes greatly to reinforcing the overall functional design of the product. Owing to the chemical instability of liposomes for long-term storage, lyophilization is the main approach used to extend their shelf life. The incorporation of dry liposomes brings about the technological possibility of adjusting the moisture content in the resulting gel products, which is of extreme importance to modulate their textural properties. However, freeze-drying may

damage lipid bilayers by ice crystal formation during freezing, vesicle fusion/aggregation following dehydration and changes in phase transition during rehydration (Chen et al., 2010). To prevent these drawbacks, the use of cryoprotectants, such as glycerol or oligosaccharides, has been proposed (Mozafari, 2005; Stark et al., 2010).

The aim of this work was to encapsulate three antioxidant extracts (collagen hydrolysate, pomegranate peel extract and shrimp lipid extract) in soy phosphatidylcholine liposomes, which were freeze-dried and incorporated in surimi gels. The particle and antioxidant properties of liposomes and the colour, texture and oxidative stability (TBARS) of the resulting gels (newly prepared and after 3 and 7 months of frozen storage) were analysed. The surimi gels incorporating the freeze-dried liposomes were subjected to *in vitro* simulated gastrointestinal digestion (sGID) in order to determine the antioxidant potential of the corresponding digests. An alternative functional formulation was also tested by including the non-encapsulated bioactive extracts in the gels.

2. Materials and methods

2.1. Materials

Frozen cleaned tunics and surimi blocks of giant squid (*Dosidicus gigas*) were provided by PSK Océanos S.A. (Madrid, Spain). Peel and albedo of pomegranate (*Punica granatum*) were collected from a local market and transported in ice to the laboratory. Shrimp waste was obtained from frozen shrimp (*L. vannamei*) kindly provided by Angulas Aguinaga Burgos (Burgos, Spain). Soybean lecithin was purchased from Manuel Riesgo S.A. (Madrid, Spain). Alcalase® 2.4L, pepsin and pancreatin enzymes were obtained from Sigma Aldrich, Inc. (St. Louis, MO, USA). Epigallocatechin, epicatechin-3-gallate, epigallocatechin-gallate, catechin, rutin, hyperoside and kaempferol-3-O-glucoside were purchased from Extrasynthese (Genay,

Cedex, France). Quercetin, gallic acid, ellagic acid, caffeic acid, punicalagin, chlorogenic acid, coumaric acid, ferulic acid, syringic acid, vitamin C, vitamin B12, aprotinin, hippuryl histidyl leucine (HHL), dimethyl sulfoxide and glycine were from Sigma Aldrich. All other reagents were of analytical grade.

2.2. Extraction and characterization of active compounds

2.2.1. Squid collagen hydrolysate (HC)

Giant squid tunics were thawed, profusely washed in water and slowly stirred with 0.5 M acetic acid (1:20 w/v) at 3 °C for 96 hours. The mixture was filtered with Whatman No. 4 filter paper and the filtrate, which constitutes the acid-soluble collagen, was then freeze-dried. Dried collagen was suspended with distilled water (30 % w/v) and subjected to enzymatic hydrolysis in a pH-stat (TIM 856, Radiometer analytical, Villeurbanne, France) using Alcalase® 2.4L, as described previously (Alemán et al., 2011), with an enzyme substrate ratio of 1:60 (w/w). The collagen hydrolysate (HC) was freeze-dried and stored at –20 °C until use. The molecular weight distribution and the amino acid composition of HC were determined by using, respectively, size-exclusion chromatography and ion exchange chromatography, as described by Aleman et al. (2011).

2.2.2. Pomegranate extract (PG)

Peel and albedo of pomegranate (*Punica granatum*) were dried in an oven (FD 240, Binder, Tuttlingen, Germany) at 50 °C and ground in an Osterizer (Sunbeam Par, 4153-50, Mexico). The resulting dry powder was suspended in ethanol/water (70/30) at a ratio of 1:20 (w/v), gently stirred at 40 °C for 4 h and left to stand at 21 °C for 24 h. The mixture was centrifuged at 12000 g for 15 min at 4 °C and the supernatant was filtered through Whatman No. 1 paper, rota-evaporated at 40 °C and stored at –20 °C until use. Total phenolic compound content was

determined by the Folin–Ciocalteu method, using gallic acid as standard. To identify the main phenolic compounds, PG was dissolved in dimethyl sulfoxide and analysed by reverse phase high performance liquid chromatography (RP-HPLC), following the procedure described in a previous work (Giménez, Moreno, López-Caballero, Montero, & Gómez-Guillén, 2013).

2.2.3. Shrimp lipid extract (SL)

The preparation and chemical characterization of the ethyl acetate-soluble lipid extract from shrimp waste (SL) used in the present study was described in a recent work (Gómez-Estaca et al., 2017). The SL was dried by rotary evaporation at 60 °C and nitrogen aspiration, and stored at –20 °C until use. Fatty acid profile was analysed by gas chromatography with FID detector, carotenoids by RP-HPLC with diode array detector, and α -tocopherol and cholesterol by HPLC coupled to a quadrupole mass spectrometer.

2.3. Encapsulation in liposomes

2.3.1. Purification of phosphatidylcholine

Partially purified phosphatidylcholine (PC) from commercial soybean lecithin (Manuel Riesgo, S.A., Madrid, Spain) was obtained by dissolving lecithin in ethyl acetate (1:5, w/v) and subsequently performing five washes with acetone (1:2, w/v). The preparation and chemical characterization of the PC powder used in the present study were described in a previous work (Taladrid et al., 2017). The PC powder was stored at –20 °C until use.

2.3.2. Preparation of liposomes

The bioactive extracts (HC, PG and SL) (2.5 g) were first suspended in 250 mL of 0.1 M phosphate buffer (pH 7). PC (62.5 g) was mixed with each bioactive solution and it was kept in

a water bath at 80 °C for 1 h with gentle stirring. Then 212.5 mL of the phosphate buffer and 37.5 mL of glycerol were added, and the mixture was kept in a water bath at 80 °C for 1 h with gentle stirring. The volume of the suspension was completed with 750 mL of the phosphate buffer, and then vortexed at 60 °C to produce multilamellar vesicles. The preparations were subsequently sonicated in an ultrasonic cell disrupter (Model Q700, Qsonica sonicators, Newton, CT, USA) at 90 % amplitude (power rating of 700 W) for 5 min, with a 60 s stop every min to allow sample cooling. The composition of the liposomes, expressed in dry weight (dw), was 2 % bioactive extract, 56 % PC and 42 % glycerol). Part of the newly prepared liposomal dispersions (L-HC, L-PG and L-SL) was used for particle size, zeta potential and entrapment efficiency determinations. The rest was freeze-dried and stored at –20 °C. Rehydration of freeze-dried liposomes was performed by suspending 14 g of the liposomal pastes in 100 mL of distilled water at 20 °C for 30 min under magnetic stirring.

2.4. Characterization of liposomes

2.4.1. Particle size and zeta potential

Particle size measurement (expressed as z-average in intensity), polydispersity (PDI) and zeta potential of liposomes (fresh and rehydrated) were performed using a Zetasizer Nano ZS (Malvern Instruments Ltd, Worcestershire, UK), as described previously (Mosquera et al., 2016).

2.4.2. Water content

The moisture of freeze-dried liposomes was determined according to method 950.46 (A.O.A.C., 2005).

2.4.3. Water solubility

Freeze-dried liposomes were mixed with distilled water (1 % w/v) under agitation (100 rpm) for 150 min at 20 °C and centrifuged at 3500 rpm (Multifuge 3 L-R, Heraeus, Madrid, Spain) for

5 min. The supernatant was dried at 105 °C for 24 h and the water solubility, expressed as a percentage, was calculated by weight differences.

2.4.4. Entrapment efficiency

The entrapment efficiency (EE) of the various encapsulated compounds was determined by measuring: i) protein content with a LECO FP-2000 nitrogen/protein analyser (LECO Corp., St. Joseph, MI, USA) for L-HC; ii) phenolic content by the Folin–Ciocalteu method (Slinkard and Singleton, 1977) for L-PG, and iii) astaxanthin concentration as reported in Gómez-Estaca et al. (2016) for L-SL. Encapsulated and non-encapsulated bioactive compounds were separated by mixing the liposomes with acetone (120 mg/mL) for L-HC, ethanol:methanol (1:1 v/v; 120 mg/mL) for L-PG, and hexane (80 mg/mL) for L-SL. Then all the mixtures were centrifuged at 5000 g for 30 min at 4 °C and filtered through a 0.45 µm filter device. EE was calculated using the following equation:

$$\% \text{ EE} = \text{encapsulated bioactive} / \text{total bioactive} * 100$$

2.5. Preparation of squid surimi gels

Prior to use (24 h), the surimi was thawed in a 4 °C refrigerator for easy cutting into smaller pieces. Then the surimi was homogenized at 2 °C in a vacuum homogenizer (FD112M10-72D, Stephan Machinery GmbH, Hameln, Germany) with 1 % NaCl and the amount of water (as ice) needed to obtain a moisture of 80.5 % in the final gels. Freeze-dried liposomal pastes were then incorporated to obtain a final concentration of 10.5 % w/w in the resulting gel (28 % on a dry basis, db), and the mixture was again homogenized for 5 min. Final concentrations of antioxidant extract, PC and glycerol in the surimi gels containing the freeze-dried liposomes were then 0.15 % (0.6 %, db), 3.8 % (15.6 %, db) and 2.9 % (11.8 %, db), respectively. To allow proper comparisons, the concentrations of each liposome constituent (antioxidant compound, PC and glycerol) were also provided on a dry basis in order to ignore the presence of water in both the surimi and the dried liposomal pastes. An alternative functional formulation without

PC and glycerol was prepared by the sole addition of the corresponding extracts to the surimi at a concentration of 2 % (9 %, db). The resulting pastes were stuffed into 35 mm plastic cellulose casings (Viscase SA, Bagnold Cedex, France) and heated in an Eller oven (Unimatic 1000, Micro 40, Eller, Merano) at 90 °C for 40 min. Then the gels were dipped into ice water to cool them quickly and stored overnight at 4 °C. For long-term storage, gels were frozen and stored at –20 °C for seven months.

2.6. Characterization of gels

2.6.1. Colour

A Konica Minolta CM-3500d colorimeter (Konica Minolta, Madrid, Spain) was used to determine L*a*b* parameters using D65 illuminant (daylight) and D10 standard observer. Results are the average of at least 10 replicates and are expressed as Hue angle and Chroma values.

2.6.2. Mechanical properties

A puncture test was performed to determine the breaking force (in N) and breaking deformation (in mm) of gels using a TA-XT plus Texture Analyser (Texture Technologies Corp., Scarsdale, NY, USA) with a cylindrical stainless steel plunger (5 mm diameter) attached to a 5 kg load cell, at 1.0 mm/s crosshead speed. Results were the mean values of at least five measurements. Gel strength (N·mm) was the product of multiplying the breaking force by the breaking deformation.

2.6.3. Soluble protein content

Surimi gels (1 g) were mixed with 25 mL of sodium chloride (5 % w/v) using an Omni-Mixer model 17106 homogenizer (Omni Intl., Waterbury, Conn., USA). The resulting homogenates were stirred for 30 min at 4 °C and then centrifuged at 6000 g for 30 min in a Sorvall RT60008 centrifuge. The protein content in the supernatant (soluble protein) and in the whole surimi

gel (total protein) was determined with a LECO-FP 2000 nitrogen/protein analyser (LECO Corp., St. Joseph, MI, USA) using a nitrogen-to-protein conversion factor of 6.25. Soluble protein content was expressed as the percentage of protein solubilized with respect to total protein in the surimi gel.

2.6.4. Thiobarbituric acid reactive substances (TBARS)

Surimi gels were solubilized with 7.5 % trichloroacetic acid (1:3 w/v) and centrifuged at 5100 g for 5 min at 4 °C. The supernatant was mixed with 0.02 M thiobarbituric acid (1:2 v/v) and kept at 20 °C for 15 hours in darkness. Finally, absorbance was measured at 532 nm. Results were expressed as µg of malondialdehyde (MDA) equivalents per kg of gel, based on a 1,1,3,3-tetraethoxypropane (TEP) standard curve.

2.7. *In vitro* simulated gastrointestinal digestion (sGID)

In vitro simulated gastrointestinal digestion of surimi gels was carried out in a water bath at 37 °C, according to Aleman et al. (2013), using pepsin solution at pH 2 for 2 h, followed by incubation with pancreatin and biliary salts at pH 6.5 for 2 h. Enzyme activity was stopped by cooling in ice. Final pH was adjusted to pH 7.2 and samples were centrifuged at 13000 g and 4 °C for 30 min. The resulting supernatants were freeze-dried and stored at –20 °C for further assays.

2.8. Determination of antioxidant activity

The ABTS radical scavenging capacity (ABTS assay, for hydrophilic compounds) and Superoxide () free radical scavenging capacity (photochemiluminescence, PCL assay, for lipophilic compounds) were used to measure the antioxidant activity as described previously (Alemán et al., 2011; Pérez-Santín et al., 2013). The results of the ABTS assay were expressed as mg Vitamin C equivalent antioxidant capacity (VCEAC) per g of sample, based on a standard curve

of vitamin C. The results for the PCL assay were expressed as mg Trolox equivalent antioxidant capacity (TEAC) per g of sample, based on a standard curve of Trolox.

2.9. Statistical analysis

All determinations were carried out at least in triplicate. Statistical tests were performed using the SPSS® computer program (IBM SPSS Statistics 22 Software, Inc., Chicago, IL, USA). One-way analysis of variance was carried out. Differences between pairs of means were assessed on the basis of confidence intervals using the Tukey test, with a significance level set at $p \leq 0.05$.

3. Results and discussion

3.1. Chemical properties of antioxidant extracts

The peptide molecular weight distribution (expressed in Dalton, Da) and amino acid composition (expressed as number of residues per 1000 residues, ‰) of the acid-soluble collagen hydrolysate (HC) are shown in Figure 1. According to the SEC molecular profile, HC was mostly composed of peptide fractions with molecular weights ranging from 6.5 to 1.3 kDa (≈ 11 ‰), 1.3 to 0.5 kDa (≈ 47 ‰) and lower than 0.5 kDa (≈ 42 ‰). Thus the predominant molecular weight of the HC peptides is lower than 1.3 kDa (Figure 1a). The major constituent amino acid residues were glycine (253 ‰), glutamic acid (117 ‰), alanine (98 ‰) and aspartic acid (92 ‰) (Figure 1b). A noticeable proportion of hydroxylysine (20 ‰) was also detected. Total imino acid content (proline + hydroxyproline) was 106 ‰, considerably lower than that reported in a <3 kDa peptide fraction from fish sea bream scale collagen (Mosquera et al., 2014). The total proportion of hydrophobic amino acids in HC was low (≈ 28 ‰), as compared to the ≈ 62 ‰ reported in the previous work (Mosquera et al., 2014).

The pomegranate peel and albedo extract (PG) presented high total phenolic content (166 mg gallic acid eq./g dry extract). The main compounds identified in the pomegranate extract (PG) were, in descending order, β -punicalagin, ellagic acid, α -punicalagin, rutin and epigallocatechin (Figure 2). The predominant presence of ellagitannins (punicalagin anomers and ellagic acid) in pomegranate peel extract has been documented previously (Masci et al., 2016).

The shrimp lipid extract (SL) was composed of 804 mg/g fatty acids (saturated = 31 %, monounsaturated = 25 % and polyunsaturated = 44 %). Other compounds present in the lipid extract were α -tocopherol (126 mg/g), cholesterol (65 mg/g) and astaxanthin (7 mg/g, in the form of free astaxanthin, and also as mono- and diesters, which were tentatively identified by HPLC-MS). The DHA (C22:6) astaxanthin monoester was found to be the most abundant (Gómez-Estaca et al., 2017).

3.2. Vesicle characterization

3.2.1. Fresh liposomal dispersions

The particle size of freshly prepared liposomes loaded with squid collagen hydrolysate (L-HC), pomegranate peel polyphenolic extract (L-PG) and shrimp lipid extract (L-SL) yielded unimodal distributions with particle sizes in all cases close to the range of about 100 nm. The results of z-average (mean particle size), polydispersity index (particle size distribution) and zeta potential (membrane surface charge) of fresh and rehydrated freeze-dried liposomes are summarized in Table 1. Fresh liposomes containing the polyphenolic extract (L-PG) and the shrimp lipid extract (L-SL) showed similar z-average values, 81.0 and 80.3 nm, respectively, while the collagen-hydrolysate-loaded liposomes (L-HC) were slightly smaller ($p \leq 0.05$), 75.7 nm. Similar vesicle size was reported previously for PC-liposomes loaded with a <1 kDa squid tunic peptide fraction obtained by the film hydration method (Mosquera, Giménez, Montero, & Gómez-Guillén, 2016). Also α -tocopherol-loaded liposomes in the nanometre range were reported by Charcosset, Juban, Valour Urbaniak, and Fessi (2015) using the ethanol injection method, and

by Memoli, Palermiti, Travagli, and Alhaique (1995) for liposomes of soya phospholipids prepared by sonication. In the present study, working with the same encapsulating material and liposome production method, only slight differences in mean particle size were detected, despite the differences in the hydrophobic nature of the loaded material. No significant differences ($p>0.05$) were observed regarding the polydispersity index (0.218–0.238) in the three liposomal dispersions, and it was well within the range between 0.2 and 0.3 expected for systems prepared from biological materials (Malheiros, Sant'Anna, Micheletto, da Silveira, & Brandelli, 2011). The zeta potential in all the fresh liposomal preparations showed strong electronegative values, especially in the case of L-SL (–88.2 mV). The zeta potential is normally used as an indicator of particle stability, and when values are above 30 mV or below –30 mV the dispersions are considered physically stable (Müller, Jacobs, & Kayser, 2001).

The entrapment efficiency (EE) varied depending on the encapsulated material. The high EE shown by the collagen hydrolysate ($\approx 95\%$) could be explained by the low average molecular weight of HC (predominantly <1.3 kDa) and the relatively low amount of proline, hydroxyproline and total hydrophobic residues. The more abundant hydrophilic peptides would therefore be located preferentially in the inner aqueous phase of the liposome and also attached to the surface membrane polar heads. In the case of L-SL the EE was higher than 90 %, in agreement with Tan et al. (2013) for phosphatidylcholine-liposomes incorporating lutein (82.64–91.98 %). In this case, since the material is predominantly of hydrophobic nature (water-insoluble), it is expected to be located mostly inside the bilayer. L-PG liposomes showed noticeably lower EE (63 %) than L-HC and L-SL, indicating that a considerable amount of PG would remain non-encapsulated. Other authors showed 89 % EE for a pomegranate peel extracted procyanidine incorporated in liposomes of soybean lecithin and cholesterol (Hu, Yao, & Liu, 2010). In the present work, the main polar phenolic compounds present in PG, such as punicalagin anomers and rutin, would be preferentially located in the liposome aqueous core. However, ellagic acid, which is poorly soluble in polar solvents, would intercalate into the

aliphatic chain zone of the membrane lipids, presumably with relatively low efficiency. Phenolic compounds are known to interact with the polar head groups and also intercalate into the bilayer lipid membrane (Pawlikowska-Pawłęga et al., 2014). The EE depends greatly on the composition of the encapsulating material, the preparation method, the presence of cryoprotectants, and the chemical properties and concentration of the loaded material (Tan et al., 2014; Xia, Cheng, Cheng, & Xu, 2016). In the present study, working under identical conditions, the PG extract presented considerably lower entrapment capacity than HC and SL, which was mainly attributed to the particular chemical composition of the polyphenolic extract.

3.2.2. Freeze-dried liposomes

Residual moisture in all dried samples was almost the same ($p>0.05$), ranging between 34 and 36 %. This water content is much higher than values reported previously, such as 0.4–0.7 % (Van Winden, Zhang, & Crommelin, 1997). In order to protect the liposomal stability during freeze-drying, glycerol (42 %, dw) was included in the formulation to maintain vesicle integrity and prevent sedimentation and leakage of the entrapped bioactive materials (Mozafari, 2005). Glycerol has been reported to play a plasticizer role by inducing increased hydration of the lipid bilayers (Manca et al., 2013). The relatively high water content in the present lyophilized liposomes could therefore be attributed to the cryoprotectant added. The addition of glycerol inhibited the formation of a dry lyophilized powder, providing samples with a gluey and appearance, as observed before by Stark, Pabst, and Prassl (2010). Despite this, the lyophilized liposomal pastes could be easily rehydrated in water. The water solubility at 20 °C showed no significant differences ($p>0.05$) between the three liposomal pastes studied, with values ranging from 82.15 % in L-HC to 84.88 % in L-PG. Similar high water solubility was obtained for the L-SL preparation (83.11 %), despite the strongly lipophilic nature of the shrimp lipid extract. The poor bioavailability of astaxanthin has been shown to be improved by liposomal

encapsulation as a result of enhanced solubility in an aqueous system (Peng, Chang, Peng, & Chyau, 2010).

Freeze-drying caused significant variation ($p \leq 0.05$) of liposomes upon rehydration in practically all parameters (Table 1). This behaviour was typical for particle destabilization due to aggregation. Thus, a significant increase in z-average was observed in the three liposomal preparations, with L-HC yielding the smallest average vesicle size (198.9 nm). The highest variation in particle size was produced in the case of L-SL (3.5-fold increase). The polydispersity index also increased to values ranging from 0.46 to 0.52. Pronounced differences were observed in particle size distribution among the three rehydrated liposomal dispersions, where the initial unimodal distribution changed into bi- or trimodal distributions (Figure 3). Interestingly, after rehydration L-HC kept 62 % (expressed in intensity) of particles measuring 162 ± 77 nm, in contrast to L-PG and L-SL, which registered, respectively, 30.5 % of 121 ± 37 nm and 26 % of 111 ± 36 nm vesicles. Freeze-drying-induced an increase in mean particle diameter with respect to fresh liposomes, as has been previously reported, probably due to the breaking of hydrogen bonding between the water molecules and phospholipid head groups (Stark et al., 2010). During the process of freeze-drying, the liposomes are firstly concentrated by the propagation of the ice front. Subsequently, during water removal, the chance of membrane apposition increases, producing liposome fusion or aggregation (Chen et al., 2010). The present results indicate that, despite the incorporation of glycerol, a certain amount of large vesicles (0.5–1 μ m) was produced, resulting from freeze-drying-induced fusion or aggregation. In the present study, which tried to simulate a feasible industrial method for preparing the freeze-dried liposomes, large vesicles were formed, but a significant percentage still remained in the original nanoscale range (around 100 nm). In addition, although rehydrated particles became less electronegative in all formulations (Table 1), zeta potential values around –60 mV denoted very high vesicle stability. It should be noted that changes in liposome properties may occur to some extent during the rehydration process. In the present

work, the freeze-dried pastes were directly homogenized with the surimi, which, despite having 80.5 % moisture, presumably did not allow complete rehydration of the liposomes.

3.3. Surimi gel characterization

The freeze-dried liposomal pastes were homogenized with the salt-ground muscle at 10.5 % and heated at 90 °C to obtain the corresponding thermally-induced surimi gels. Another type of functional seafood product was also tested by direct addition of the bioactive extracts (non-encapsulated) to the gel (Free) at a concentration of 2 %. All the liposome-containing gels, as well as the control gel (without liposomes or bioactive extracts), were easy to remove from the casings and presented a smooth, cohesive and somewhat glossy appearance. The gels with liposomes showed protein solubility values of 7.81 ± 2.28 % in the control gel, 8.36 ± 1.50 % in L-HC, 8.42 ± 0.40 % in L-SL, and slightly higher, 14.42 ± 2.91 % in L-PG. All these properties were indicative of proper protein gelation. In contrast, gels made with non-encapsulated bioactive extracts were weak and crumbly, and presented strong water exudate, which was more intense with the pomegranate extract. Although the addition of the non-encapsulated extracts was at a lower concentration than the dried liposomes (2 % vs 10.5 %), they caused stronger interference in the formation of the gel network, probably by disturbing the myofibrillar protein hydration state. The strong reactivity of the phenolic compounds in the free mode of addition (Free-PG), or the amount which was not efficiently encapsulated in the liposomal mode (L-PG), caused higher water loss (in Free-PG) or lower protein thermal aggregation (with L-PG). Some properties of both types of functional formulations will be compared, despite their noticeable differences in chemical composition, especially regarding the concentration of bioactive extract. In this respect, it should be mentioned that the effective concentration of the extract when added in the liposomal form to the gel was 0.15 %, which means a 13-fold dilution effect with respect to the free form (2 %). From a technological point of view, direct addition of the bioactive extracts caused a failure in texture and water

binding. In order to ameliorate this detrimental effect, gelling and water binding ingredients would be necessary; however, their addition would result in dilution of the bioactive extract. In contrast, in the liposome-containing gels, no other ingredients were needed for proper gelation. In addition, the presence of soy phosphatidylcholine (comprising around 4 % of the final gel composition) provides a natural source of polyunsaturated fatty acids, contributing to the beneficial healthy properties of the functional product.

3.3.1. Colour

The colour polar plot of surimi gels containing the various freeze-dried liposomes is shown in Figure 4. The colour parameters of gels containing the free active compounds (Free-HC, Free-PG and Free-SL) are also represented. The addition of Free-HC and Free-PG caused a slight change in hue value as compared to the control gel. In contrast, Free-SL moved the resulting gel into the orange-reddish region, in accordance to the striking red colour of the astaxanthin-rich shrimp lipid extract. The colours of surimi gels with added L-HC or L-PG liposomes were very close to each other, however Free-PG led to significantly higher ($p<0.05$) chroma than L-PG. Both hue and chroma values in the L-SL-containing gel were considerably attenuated when compared with the corresponding gel with Free-SL, in part due to the considerably lower concentration of SL in the former. No significant differences ($p>0.05$) in colour parameters were observed in liposome-added surimi gels as a function of time over 7 months of frozen storage (data not shown).

3.3.2. Gel strength

The puncture test was carried out only on the liposome-containing surimi gels, as well as on the control gel without liposomes, since the formulations with free bioactive extracts did not gel adequately (Table 2). All unfrozen liposome-incorporated gels showed a significant decrease ($p\leq0.05$) in breaking force, breaking deformation and gel strength, as compared to the gel with no liposome addition. This effect was attributed to some interference in protein-

protein interactions. Despite the different nature of the encapsulated extracts, no significant differences ($p>0.05$) in gel strength were observed between L-HC and L-SL gels. In contrast, L-PG gels showed lower ($p\leq 0.05$) gel strength and breaking deformation values, coinciding with the slightly higher percentage of soluble protein found in this gel. This finding indicates that the addition of L-PG caused slightly more disturbance in the gel protein network, probably owing to the contribution of the non-encapsulated phenolic compounds. With regard to the stability of the gels during frozen storage, the control gel showed a strong tendency to harden ($p\leq 0.05$) at the end of the long-term frozen storage (7 months). In contrast, the gels with liposomes proved to be more stable from month 3 to 7, by preventing further cold-induced protein aggregation (Table 2). Frozen gels with L-PG exhibited a pronounced increase in both breaking force and deformation during the first three months of storage, as compared to the gels with L-HC and L-SL. This effect could be the result of the initial weaker gel structure in L-PG gels. All gels with added liposomes remained noticeably softer than the control at the end of the storage period, which indicates good tolerance to frozen storage.

3.3.3. Thiobarbituric acid reactive substances (TBARS)

The oxidative stability of surimi gels with liposomes was determined following the TBARS method, which quantifies the total amount of malondialdehyde (MDA) equivalent accumulation (Table 2). In general, very low values were recorded in all samples, largely due to the lean composition of the squid surimi. The addition of liposomes significantly ($p\leq 0.05$) increased the TBARS content in the surimi gels, more markedly in the case of gels with added L-HC, followed by, in descending order, L-SL and L-PG. The increase in TBARS was inversely related to the antioxidative capacity. The fatty acid composition of phosphatidylcholine from soybean lecithin is characterized by a high degree of unsaturation, which makes the resulting liposomes susceptible to being oxidized (Wang & Wang, 2008). Lipid oxidation might have occurred during the liposome production by the sonication step, and also during the thermal

gelation of the myofibrillar proteins. The three types of active compounds used in the present study are well known to have antioxidant properties, as will be discussed later. In spite of this, the TBARS of gels with the loaded liposomes was slightly higher than in the control gel (without liposomes), and tended to increase after long-term frozen storage. This finding suggests that the encapsulated extracts were not fully available to protect the gel matrix from lipid oxidation. The lower TBARS accumulation in gels with added L-PG could be ascribed to the lower encapsulation efficiency of PG in the liposomal preparation, and also to its higher antioxidant capacity.

3.4. Antioxidant activity

3.4.1. Free and encapsulated active compounds

The antioxidant activity of the collagen hydrolysate (HC), the pomegranate phenolic extract (PG) and the shrimp lipid extract (SL) is shown in Figure 5. The radical scavenging activities of PG were much greater than those of HC and SL. The hydrophobic nature of the shrimp lipid extract prevented the analysis of the ABTS assay in aqueous solution. However, the encapsulation of SL in liposomes made it water-soluble, leading to measurable ABTS values (Fig. 5a). Astaxanthin and α -tocopherol, both contained in a significant amount in SL, are well-known antioxidants (Gómez-Estaca et al., 2016; Gómez-Estaca et al., 2017). According to Hama et al. (2012), the terminal rings of astaxanthin, located preferentially on the liposomal membrane surface, as well as the conjugated polyene moieties at the interface of the lipid membranes, would be available to scavenge hydroxyl radicals. Given that the ABTS activity of an empty liposome was 2.07 ± 0.03 mg VCEAC/g sample, the activity of the liposome loaded with SL could be mostly attributable to the extract. Slight reduction in ABTS values in PG-loaded liposomes was found with respect to the corresponding phenolic extract (PG in aqueous solution). This loss of activity could be due to partial degradation of phenolic compounds during liposome preparation or to less availability of reactive groups resulting

from interactions with the bilayer membrane (Lopes de Azambuja et al., 2015). In contrast, HC became apparently more antioxidant when included in liposomes, with around a 4-fold increase in ABTS value (from 20.36 to 90.40 VCEAC mg/g bioactive). Other authors have attributed such an increase to the contribution of the encapsulating material (da Rosa Zavareze et al., 2014). In the present study, the low ABTS values found in the empty capsules suggests that this effect is probably attributable to enhanced peptide conformation for radical scavenging at the polar membrane surface.

The PCL assay was performed to consider the antioxidant property of the entire lipophilic material, including the encapsulating material (Fig. 5b). The results showed that the PCL activity of the lipophilic fraction of HC was low and close to that measured in the corresponding loaded liposomes (L-HC) (0.378 and 0.326 TEAC mg/g, respectively). The PCL activity of an empty liposome was 0.02 ± 0.00 TEAC mg/g sample, so the activity of L-HC was mostly attributable to the hydrolysate. The activity of SL was 60.89 TEAC mg/g, while L-SL showed only 0.502 TEAC mg/g. Given the high encapsulation efficiency in L-SL ($\approx 90\%$), the low PCL values therefore were mostly attributed to the encapsulating process, with a minimal contribution of SL. This low contribution might suggest that the amount of the entrapped material that could have leaked out as a result of freeze-drying was very low. These findings indicate that encapsulation of the lipophilic compounds in PC-liposomes would be useful to make them water-soluble in order to determine antioxidant properties in aqueous dispersions. However, no great contribution was found when measuring the radical scavenging capacity in a lipophilic medium, probably because a major part of the bioactive extract is entrapped inside the bilayer. The PG extract showed the highest superoxide radical scavenging activity (644.54 TEAC mg/g), and PCL values were also considerably diminished in the corresponding PG-loaded liposomes (L-PG) (38.474 TEAC mg/g). The higher antioxidant activities found in L-PG could be explained by the amount of non-encapsulated highly active phenolic compounds (EE=63 %), which remained partially available for radical scavenging in the liposomal dispersion. In

particular, ellagic acid has been reported as a main component in pomegranate peel extract yielding high antioxidant capacity (Masci et al., 2016).

3.4.2. Simulated gastrointestinal digestion of gels

In order to gain an insight into the potential bioaccessibility, the surimi gels incorporating the liposome-encapsulated extracts were subjected to simulated gastrointestinal digestion (sGID). The residual radical scavenging (ABTS method) was determined in the digests, expressed per g of surimi gel on a dry weight basis (Figure 6). For comparison purposes, formulations with free added extracts (non-encapsulated) were also tested, although, as discussed above, they did not show adequate gel properties. The radical scavenging capacity of the digests was lower when the bioactive extracts were added encapsulated in liposomes as compared to the free form. The ABTS value of the surimi gel digest without any extract was 6.53 ± 0.09 mg VCEAC/g, which was noticeably lower than with the added liposomes (around 14.5 mg VCEAC/g). The antioxidant activity found in the surimi gels after *in vitro* gastrointestinal digestion was related to the release of short peptides from the thermally polymerized myosin in the gel (Ueki et al., 2014). The ABTS values were increased by the addition of the various extracts, especially in the free mode, but slight differences depending on the type of extract were observed. A possible explanation for the lack of differences is that the radical scavenging activity attributable to the extracts was considerably masked by the digestibility of the protein gel. Protein digestion could have been favoured in the poorer and weaker gel matrices resulting from the distortion of protein-protein interactions by the presence of bioactive extracts or liposomes. The effect was more evident with the free-added extracts, which produced improper gels. The lower ABTS values found in all the liposome-containing gels would be also the result of the dilution effect produced by the liposome content (10.5 %) in the digestible myofibrillar protein. It should be considered that the activity of the various extracts might also have been modified as a result of

the sGID. In this respect, an increase in radical scavenging activity was found after subjecting HC to sGID, increasing the ABTS values from 22.36 to 33.17 VCEAC mg/g of dry extract. Similar results were obtained by Alemán et al. (2013) for digested peptide fractions from squid hydrolysate. This effect could contribute slightly to the ABTS values in the corresponding digested surimi gels. In contrast, the activity of PG after sGID decreased significantly ($p \leq 0.05$) from 360.01 to 188.30 VCEAC mg/g of dry extract, indicating noticeable inactivation of phenolic compounds after digestion (Surek & Nilufer-Erdil, 2016). This could be a reason why, although it had much more activity, the difference in the resulting PG-added gel was small in comparison with the HC- and SL-added gels. In the gels with liposomes, no significant differences as a function of extract were found, so it could be presumed that the encapsulated material was protected inside the vesicle from interactions with the gel matrix during the sGID.

4. Conclusions

The three different waste materials were successfully encapsulated in soy phosphatidylcholine nanoliposomes with high particle stability. Although freeze-drying caused an increase in the average particle size and polydispersity of rehydrated liposomes, a significant percentage still remained near the original nanoscale. Freeze-drying represents a feasible solution to standardize the moisture content in surimi gels. The incorporation of freeze-dried liposomes helped to provide gel integrity and to maintain stability during long-term frozen storage. The surimi gels subjected to sGID showed noticeable radical scavenging capacity, attributed predominantly to the digested myofibrillar protein. The encapsulated extracts were not available to protect the gel matrix from lipid oxidation, which in any case was very low. Therefore it was speculated that the active compounds were efficiently protected from process degradation or interactions in the gel matrix by the liposome membrane. However,

further *in vivo* studies would be needed to ascertain the bioavailability of the encapsulated bioactive extracts in the surimi gels.

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668

FIGURE CAPTIONS

Figure 1. Molecular weight distribution (a) and amino acid composition (b) of the giant squid acid-soluble-collagen hydrolysate (HC).

Figure 2. HPLC chromatographic profile of the phenolic compounds present in the pomegranate peel and albedo extract (PG).

Figure 3. Particle size distribution of rehydrated freeze-dried liposomes loaded with (a) collagen hydrolysate (L-HC), (b) pomegranate peel extract (L-PG) and (c) shrimp lipid (L-SL).

Figure 4. Polar plot representation of Hue and Chroma values of surimi gels with the addition of non-encapsulated collagen hydrolysate (Free-HC), pomegranate peel extract (Free-PG) and shrimp lipid extract (Free-SL), and of the respective freeze-dried liposomes (L-HC, L-PG and L-SL). Control = gel without additives.

Figure 5. Antioxidant activity of non-encapsulated (Extract) and liposome-encapsulated (Liposome) collagen hydrolysate (HC), pomegranate peel extract (PG) and shrimp lipid extract (SL). (a) Results of the ABTS assay; (b) results of Photochemiluminescence assay. Different letters (a, b, c, ...) indicate significant differences ($p \leq 0.05$) among samples.

Figure 6. Antioxidant activity (ABTS assay) of surimi gels with the addition of non-encapsulated (Free) collagen hydrolysate (HC), pomegranate peel extract (PG) and shrimp lipid extract (SL), and of the respective freeze-dried liposomes (Liposome). Different letters (a, b, c, ...) indicate significant differences ($p \leq 0.05$) among samples.

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693 Table 1.- Size (z-average), polydispersity and zeta potential of fresh and rehydrated freeze-dried
 694 liposomes loaded with collagen hydrolysate (L-HC), pomegranate peel extract (L-PG) and shrimp lipid
 695 (L-SL).

	Z-average (nm)		Polydispersity index		Zeta potential (mV)	
	Fresh	Freeze-dried	Fresh	Freeze-dried	Fresh	Freeze-dried
L-HC	75.7 ± 0.5 ^{a/x}	198.9 ± 2.8 ^{a/y}	0.231 ± 0.010 ^{a/x}	0.462 ± 0.035 ^{a/y}	-68.4 ± 1.8 ^{a/x}	-62.9 ± 1.2 ^{b/x}
L-PG	81.0 ± 0.6 ^{b/x}	274.2 ± 3.3 ^{b/y}	0.238 ± 0.012 ^{a/x}	0.518 ± 0.029 ^{a/y}	-64.6 ± 1.8 ^{a/x}	-58.5 ± 1.6 ^{a/y}
L-SL	80.3 ± 1.1 ^{b/x}	282.9 ± 2.3 ^{c/y}	0.218 ± 0.013 ^{a/x}	0.523 ± 0.037 ^{a/y}	-88.2 ± 2.7 ^{b/x}	-62.5 ± 1.7 ^{b/y}

696 Different letters (a, b, c) indicate significant differences ($p \leq 0.05$) among samples as a function of
 697 loaded substance. Different letters (x, y) indicate significant differences ($p < 0.05$) between fresh and
 698 freeze-dried samples.

699

Table 2.- Breaking force, breaking deformation, gel strength and TBARS of surimi gels without liposomes (G-control) and with liposomes loaded with collagen hydrolysate (L-HC), pomegranate peel extract (L-PG) and shrimp lipid extract (L-SL), determined after overnight storage at 4 °C (Unfrozen) and after 3 months and 7 months of frozen storage at –20 °C.

	Force N	Deformation mm	Gel strength N·mm	TBARS µg MDA/kg gel
Unfrozen				
G-control	1.25 ± 0.05 ^{C/X}	7.15 ± 0.16 ^{C/X}	8.94 ± 0.29 ^{C/X}	0.088 ± 0.000 ^{A/X}
L-HC	0.41 ± 0.01 ^{A,B/X}	6.39 ± 0.12 ^{B,C/X,Y}	2.62 ± 0.05 ^{B/X}	0.511 ± 0.042 ^{D/X}
L-PG	0.35 ± 0.03 ^{A/X}	5.21 ± 0.75 ^{A/X}	1.82 ± 0.20 ^{A/X}	0.276 ± 0.027 ^{B/X}
L-SL	0.47 ± 0.02 ^{B/X}	6.13 ± 0.06 ^{B/X}	2.88 ± 0.09 ^{B/X}	0.386 ± 0.042 ^{C/X}
3 months				
G-control	1.71 ± 0.13 ^{D/Y}	6.98 ± 0.04 ^{A/X}	11.94 ± 0.49 ^{D/Y}	0.015 ± 0.000 ^{A/X}
L-HC	0.76 ± 0.04 ^{B/Y}	6.91 ± 0.39 ^{A/Y}	5.25 ± 0.29 ^{B/Z}	0.508 ± 0.059 ^{D/X}
L-PG	0.94 ± 0.01 ^{C/Y}	7.76 ± 0.12 ^{A/Y}	7.29 ± 0.08 ^{C/Y}	0.220 ± 0.050 ^{B/X}
L-SL	0.57 ± 0.08 ^{A/X}	6.21 ± 1.17 ^{A/X}	3.54 ± 0.59 ^{A/X}	0.380 ± 0.072 ^{C/X}
7 months				
G-control	2.47 ± 0.03 ^{D/Z}	8.00 ± 0.00 ^{D/Y}	19.76 ± 0.10 ^{D/Z}	0.333 ± 0.074 ^{A/Y}
L-HC	0.74 ± 0.02 ^{A/Y}	6.01 ± 0.07 ^{A/X}	4.45 ± 0.08 ^{A/Y}	1.090 ± 0.048 ^{C/Y}
L-PG	1.02 ± 0.01 ^{C/Z}	7.23 ± 0.05 ^{C/Y}	7.37 ± 0.05 ^{C/Y}	0.763 ± 0.085 ^{B/Y}
L-SL	0.87 ± 0.02 ^{B/Y}	6.81 ± 0.13 ^{B/X}	5.92 ± 0.13 ^{B/Y}	0.848 ± 0.052 ^{B/Y}

Different letters (A, B, C,) indicate significant differences (p≤0.05) between samples for the same time of analysis.

Different letters (X, Y, Z) indicate significance differences (p<0.05) between the same sample for different times of analysis and the same parameter.

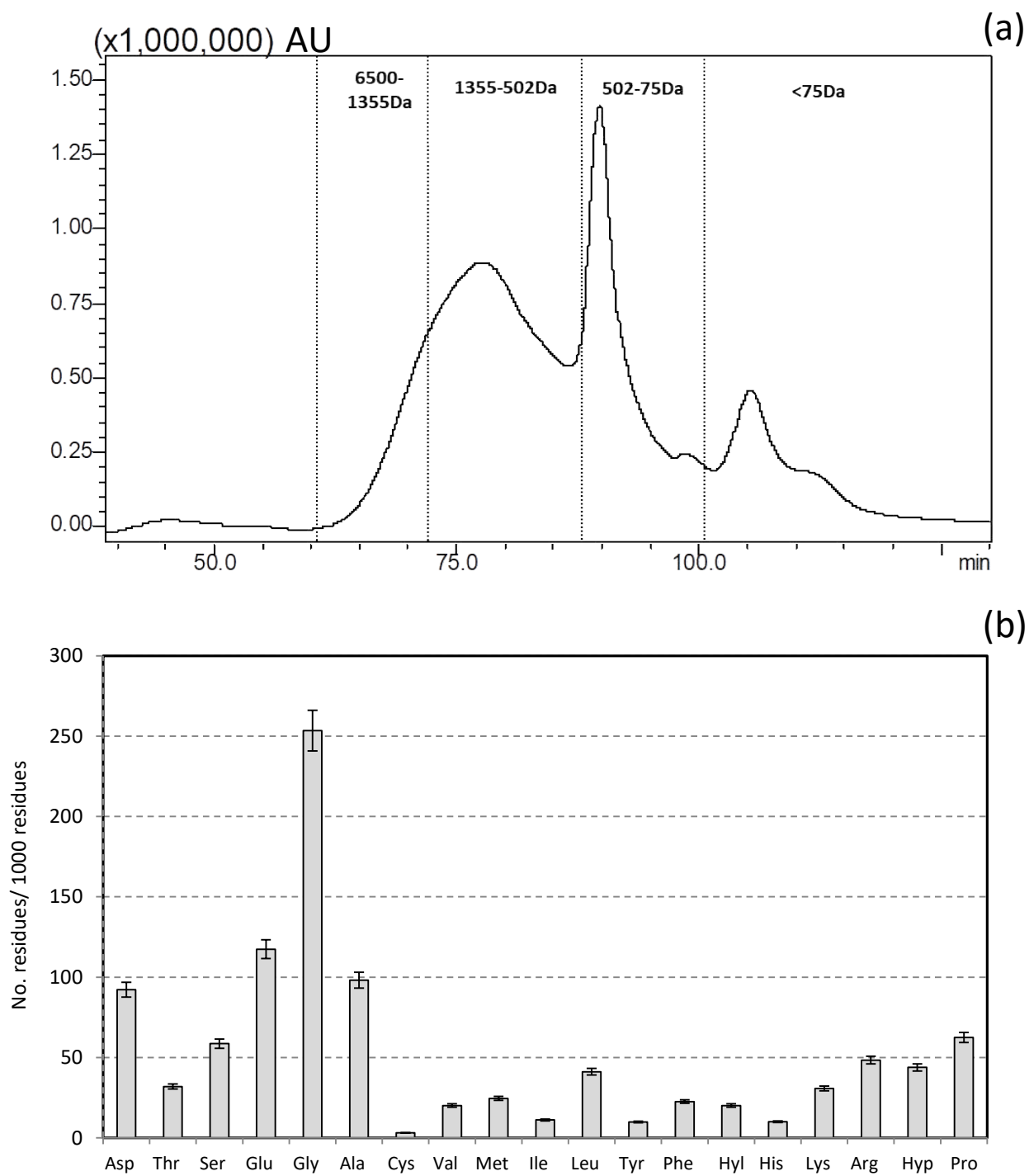


Figure 1

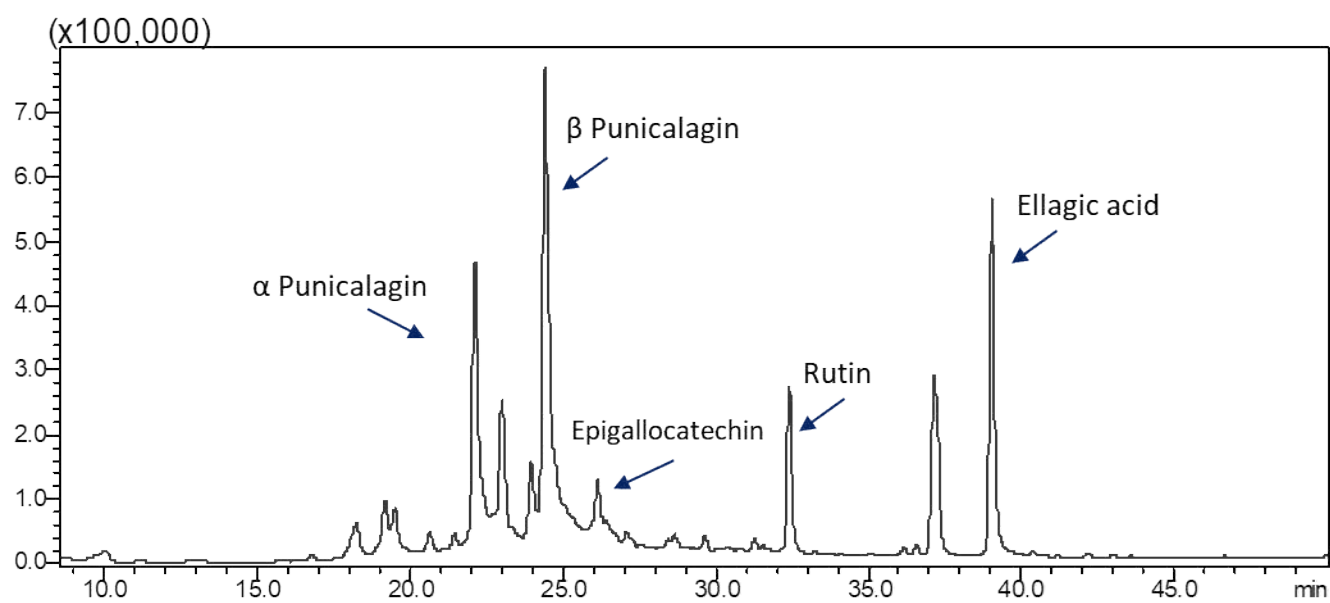


Figure 2

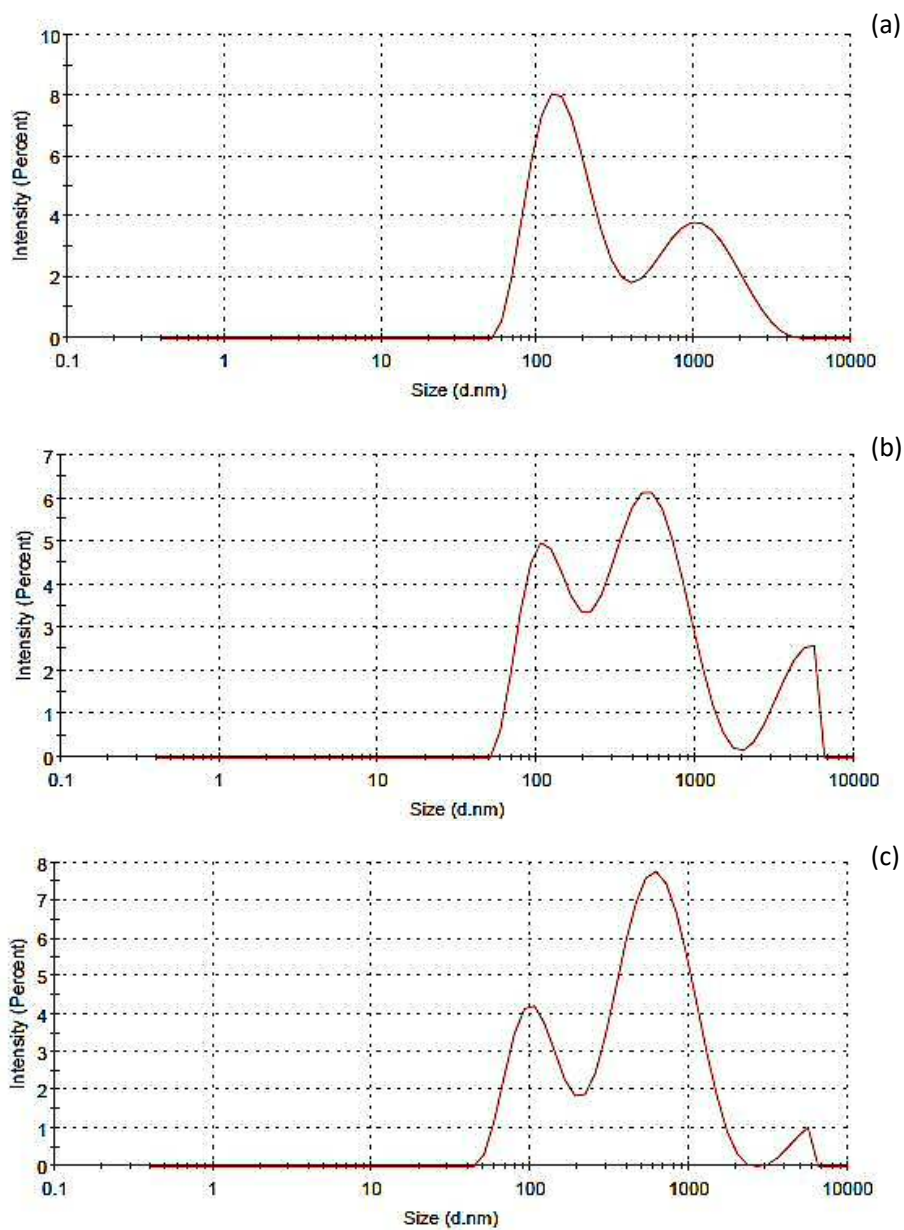


Figure 3

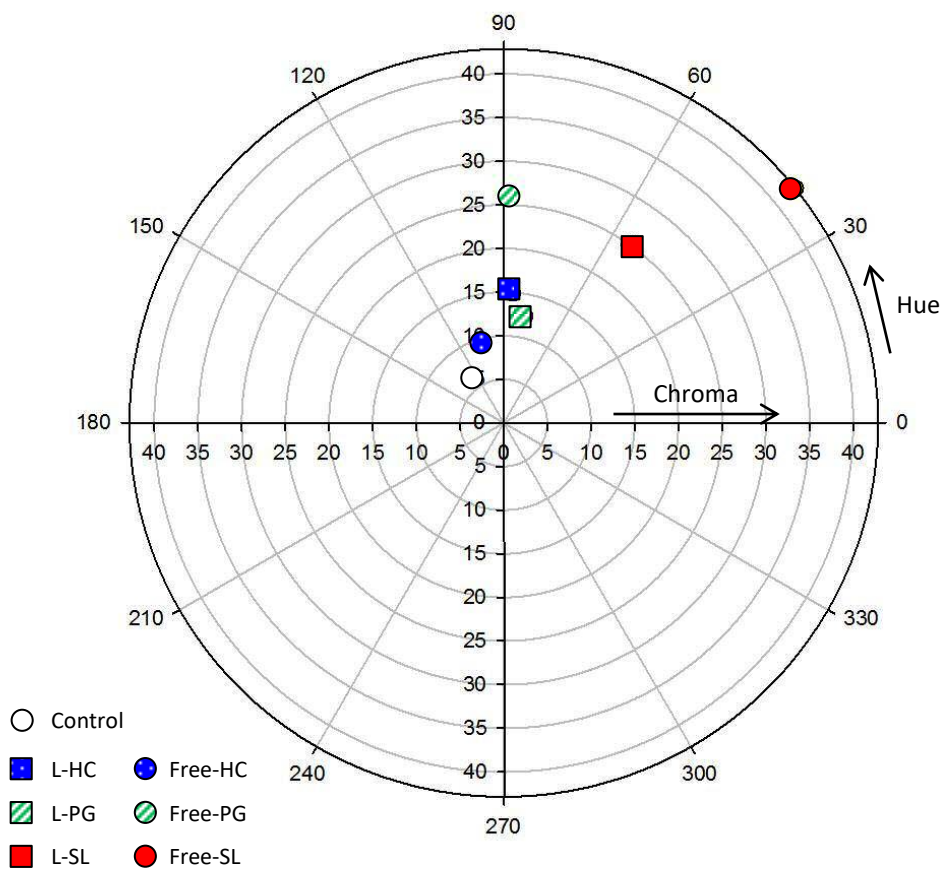
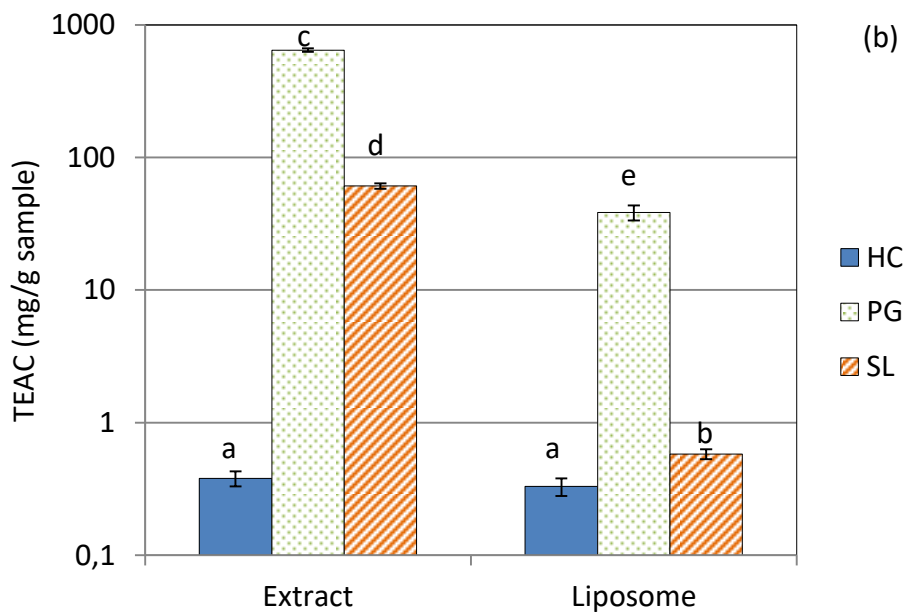
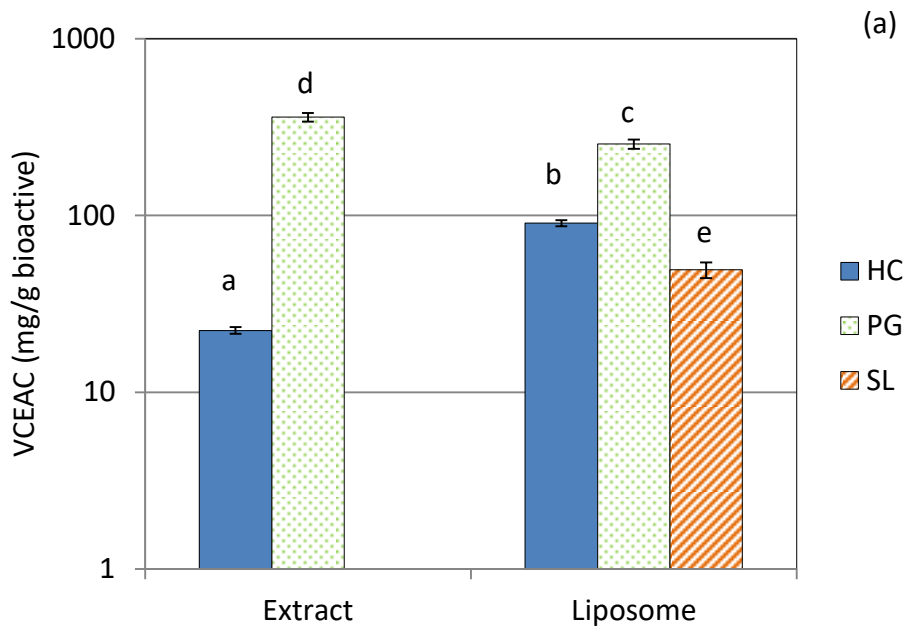


Figure 4



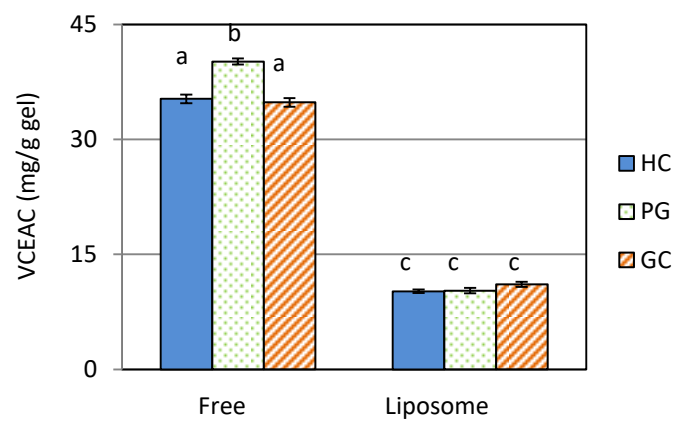


Figure 6