

1 **Title: Valorization of Tunisian secondary date varieties (*Phoenix dactylifera* L.) by**
2 **hydrothermal treatments: new fibre concentrates with antioxidant properties.**

3

4 **Authors: Abdessalem Mrabet^{a,c}, Guillermo Rodríguez-Gutiérrez^a, Rafael Guillén-**
5 **Bejarano^a, Rocío Rodríguez-Arcos^a, Ali Ferchichi^b, Marianne Sindic^c, Ana Jiménez-**
6 **Araujo^{a,*}**

7 ^a Instituto de la Grasa, Consejo Superior de Investigaciones Científicas (CSIC), Avda. Padre
8 García Tejero nº 4, Sevilla-41012, Spain

9 ^b Arid and Oases Cropping Laboratory, Arid Area Institute-Medenine 4119, Tunisia

10 ^c University of Liege - Gembloux Agro-Bio Tech. Department of Food Technol. Passage
11 des Déportés, 2. B-5030 Gembloux, Belgium

12

13 * Corresponding author: A. Jiménez-Araujo

14 E-mail address: araujo@cica.es

15 Instituto de la Grasa, CSIC, Avda. Padre García Tejero nº 4, Sevilla-41012, Spain

16 PHONE +34954692516

17 FAX +34954691262

18

19

20 **ABSTRACT**

21 Two hydrothermal treatments were assayed on secondary date varieties from the Tunisian
22 coastal oasis in order to obtain valuable solid extracts, rich in dietary fibre and antioxidants.

23 The steam explosion treatment (SET) is based on high temperatures and pressures, with an
24 explosive decompression. The steam treatment (ST), without rapid decompression, has lower
25 pressures and temperatures, but longer treatments than SET. The recovery of the fiber

26 concentrates (FCs) was very similar for the different assayed conditions. Only its
27 granulometry showed differences, increasing the fraction with size lower than 4 mm in more
28 intense treatments. Treatment conditions had higher influence than date variety in the FCs
29 chemical composition. It was approximately 0.1-2% soluble sugar, 0.7-2% uronic acids, 4-
30 8% phenols, 5-11% fat, 5-22% non-cellulosic sugars, 9-14% protein, 12-20% cellulose, and
31 44-71% lignin, depending on treatment conditions. They had also very high antiradical
32 activity (230-580 mmols Trolox/Kg FC). The use of ST reactor is highly recommended
33 because it could be easily scaled up to the industry. The FC obtained by hydrothermal
34 treatments from date fruits could be considered as a very interesting ingredient for healthier
35 food, especially for bakery or dairy products due to its pleasant chocolate/coffee flavour.

36

37 **KEYWORDS:** secondary date varieties, hydrothermal treatment, fibre concentrates,
38 chemical composition, antioxidant activity.

39

40 1. INTRODUCTION

41 The fruits of the date palm (*Phoenix dactylifera* L.) are commonly consumed in many parts of
42 the world and a vital component of the diet and a staple food in most of Arabic countries.

43 Dates are one of the main crops in Tunisia and represent a major source of income for the
44 majority of the population in the rural areas. They are also important for the ecology of the
45 country, being the most adaptive crop and tolerant to various environmental stresses (Al-
46 Farsi, 2003). Tunisia is considered to be one of the most important producers for Deglé Nour,
47 cultivar with very good sensory quality and a high commercial value. This production is
48 unfortunately accompanied by a decrease of cultivars characterized by a low commercial
49 quality (secondary cultivars). They are less appreciated and account for approximately 30%
50 of Tunisian dates (Chibane, Benamara, Noui, & Djouad, 2007). Although they are not used as
51 human food, they are rich in many bioactive compounds (Elleuch et al., 2008).

52 Limited data are available regarding the compositional and functional characteristics of
53 secondary dates grown in Tunisia (Elleuch et al., 2008; Mrabet, Rodríguez-Arcos, Guillén-
54 Bejarano, Chaira, Ferchichi, & Jiménez-Araujo, 2012). Attempts should be made to convert
55 these unused varieties into value added products in order to increase the economic feasibility
56 of date industries. Recently, Mrabet et al. (2012) concluded that some of these varieties could
57 be valuable as a source of antioxidant dietary fiber, but pre-treatment could be necessary for
58 the total utilization of the dates. One of the more interesting processes is based on
59 hydrothermal pre-treatments that allow for the extraction of soluble compounds to the liquid
60 phase, leaving a fibrous material as solid fraction (Fernández-Bolaños, Rodríguez, Gómez,
61 Guillén, Jiménez, & Heredia, 2004). Among the existing methods, steam explosion treatment
62 (SET) is one of the most commonly used for fractionation of biomass components. The SET
63 technique is very useful for the separation of cellulose, hemicellulose and lignin from raw
64 materials, as wheat straw (Montane, Farriol, & Salvado, 1998), bagasse (Kar, Gutiérrez, &

65 Kinoshita, 1998), and olive stone (Fernandez-Bolaños et al., 2001). As pressures for SET are
66 very high for usual industrial steam generators (42 Kg/cm²), another pretreatment has been
67 developed (Lama-Muñoz, Rodríguez-Gutiérrez, Rubio-Senent, Gómez-Carretero, &
68 Fernández-Bolaños, 2011). The new one is also based on steam treatment (ST) but without
69 explosive decompression, working at lower pressure (9 Kg/cm²) with longer treatments.
70 The objective of this study was to apply for the first time both thermal pretreatment systems
71 based on steam technology to secondary date varieties from Tunisia, in order to evaluate the
72 chemical composition, functional properties and antioxidant activity of fiber concentrates
73 obtained from the solid fraction. With this information, an industrial treatment could be
74 proposed to valorize these cultivars so important from economical and social points of view.

75

76 **2. MATERIALS AND METHODS**

77 **2.1. Chemicals.**

78 Trifluoroacetic acid (TFA), 3-phenylphenol, anthrone, Folin–Ciocalteu phenol reagent, and
79 2,2-diphenyl-1-picrylhydrazyl (DPPH• free radical) were purchased from Sigma-Aldrich
80 Química (Madrid, Spain). Na₂CO₃, hexane, sodium hydroxide, and acetic acid were from
81 Panreac Química S.A. (Barcelona, Spain). Standards of gallic acid and myoinositol were
82 purchased from Sigma-Aldrich Química. Ethyl acetate and acetonitrile were of HPLC grade
83 purity (Romyl, Teknokroma, Barcelona, Spain). Sulfuric acid and acetone were from Sharlau
84 (Barcelona, Spain). Ethanol was purchased from Alcoholes del Sur (Córdoba, Spain).

85 **2.2. Plant material.**

86 Three secondary date varieties (Eguwa, Garen Gaze, and Smeti) at the “Tamr stage” (full
87 ripeness) were studied. These varieties were chosen due to their high values of antioxidant
88 activity linked to dietary fiber, as previously reported (Mrabet et al, 2012). They were picked

89 at Gabès littoral oasis (southern Tunisia) during the 2011 harvest season (September-
90 October). All samples were stored at -20 °C until analysis and treatment.

91 **2.3. Thermal treatments.**

92 Prior to both treatments fruits were cut longitudinally to improve the access of steam to the
93 date seed. SET was carried out using a 2-liter capacity reactor (Figure 1, Subfig. A). Date
94 samples (250 g) of each variety were treated with a saturated steam at a maximum operating
95 pressure of 42 Kg/cm². The reactor was equipped with a quick-opening ball valve and an
96 electronic device programmed for the accurate control of steam time and temperature for the
97 final steam explosion. After the treatment, the samples were collected, vacuum filtered
98 through filter paper, and freeze-dried. ST, without explosion, was carried out using a 100-liter
99 capacity reactor (Figure 1, Subfig. B), which can operate at temperatures between 50 and 190
100 °C by direct heating, and at a maximum pressure of 9 Kg/cm². The system allows for the
101 appropriate treatment of dates without explosion or high pressures and temperatures. In this
102 case, date samples consisted of a variety blending. The wet treated material was filtered by
103 centrifugation at 4700 g (Comteifa, S.L., Barcelona, Spain) and freeze-dried.
104 In both treatments, seed pieces higher than 4 mm were removed from the dried solid fractions
105 by sieving. The material under 4 mm was considered the fiber concentrate (FC) to be
106 characterized in the present work.

107 **2.4. Ethanol extraction of date samples and FCs.**

108 One gram of date flesh, seed or FC was extracted twice with 100 ml of 80% ethanol. After
109 filtration, both liquors were collected and made up to 200 ml in a volumetric flask. In these
110 extracts, soluble sugars, total phenols and soluble antiradical activity were determined.

111 **2.5. Proximate composition.**

112 For date flesh, all the determinations were carried out from soluble sugar-free material (80%
113 ethanol extracted). Date seeds and FCs were directly analyzed.

114 Protein content was determined by the Kjeldahl method and applying a factor of 6.25 to
115 convert the total nitrogen into protein content. Soluble sugars were determined using the
116 anthrone method. Klason lignin levels were determined gravimetrically as the acid-insoluble
117 material remaining after a two-stage sulfuric acid hydrolysis. The fat was quantified with
118 hexane using a Soxhlet apparatus. Uronic acids were quantified using the phenyl-phenol
119 method after sulfuric acid hydrolysis. Cellulose was quantified from the TFA-insoluble
120 residue after sulphuric acid hydrolysis by the anthrone method. Non-cellulosic sugar (NCS)
121 composition was determined by hydrolysis with 2N TFA at 121 °C for 1h. The released
122 sugars were quantified as alditol acetates by gas chromatography.

123 The total polyphenol content was quantified for each ethanol extract according to the Folin-
124 Ciocalteu spectrophotometric method, using gallic acid as a reference standard compound.

125 The total phenolic content of the samples was expressed as gallic acid equivalents (g/100 g).

126 **2.6. Determination of the antiradical activity.**

127 Soluble antiradical activity was determined from the ethanol extracts by the DPPH• method
128 (Rodríguez et al., 2005). The efficient concentration EC₅₀, which represents the amount of
129 antioxidant necessary to decrease the initial absorbance by 50%, was calculated from a
130 calibration curve by linear regression for each sample.

131 The antiradical activity of the insoluble residue after ethanol extraction was evaluated as
132 described by Fuentes-Alventosa et al. (2009). As for soluble antiradical activity, EC₅₀ was
133 also calculated. Both activities were expressed as millimoles of Trolox equivalent (TE)
134 antioxidant capacity per kilogram of sample by means of a dose-response curve for Trolox.

135 **2.7. Water-and oil-holding capacity, and solubility.**

136 Water-holding capacity (WHC) and oil-holding capacity (OHC) were determined using the
137 method described by Jiménez, Rodríguez, Fernández-Caro, Guillén, Fernández-Bolaños, &
138 Heredia (2000). Samples (250 mg × 3) were suspended in 15 mL of water or sunflower oil

139 (1.0054 g/mL density), respectively. After 24 h of stirring at room temperature the suspension
140 was centrifuged at 14000g for 1 h. Supernatants were carefully eliminated, and the pellets
141 were weighed. From WHC samples, hydrated pellets were freeze-dried, and their solubility in
142 water (SOL) was determined by the difference in weight before and after the WHC assay,
143 which was expressed as a percentage.

144 **2.8. Statistical analysis.**

145 The results are expressed as the average value of at least three repetitions. To assess the
146 differences in composition, functional characteristics and antiradical activity among samples,
147 a multiple-sample comparison was performed using the Statgraphics Plus program Version
148 2.1. Multivariate analysis of variance (ANOVA), followed by Duncan's multiple comparison
149 test, was performed to differentiate the groups. The level of significance was $P < 0.05$.

150

151 **3. RESULTS AND DISCUSSION.**

152 **3.1. Chemical composition of date fruits.**

153 Prior to hydrothermal treatments, the chemical composition of date fruits from the three
154 varieties was studied. Flesh and seeds were analyzed separately (Table 1). Garen Gaze and
155 Eguwa did not show significant differences in weight but did in flesh/seed ratio. Smeti dates
156 were bigger and with a higher flesh/seed ratio. Only moisture and soluble sugars were present
157 in higher amounts in the flesh than in the seed. Moisture was twice as high in the flesh,
158 varying significantly from 23% in Garen Gaze to 31% in Smeti. Soluble sugars were higher
159 than 60% in the flesh and around 1% in the seed, and they did not show significant
160 differences among varieties. Lignin was the second component in quantity (23-27%) in the
161 flesh and, as was the case for NCS, there were not any statistical differences among varieties.
162 These high percentages could be related to the inedibility of these varieties, as proposed
163 previously (Mrabet et al, 2012). The rest of the components were lower than 10%, with

164 remarkably low amounts of fat (0.3-0.5%) and phenols (0.04-0.07%). Phenols and antiradical
165 activity were in the same range as previously described (Mrabet et al, 2012).

166 The NCS composition was also studied in the flesh and seeds. In Figure 2, Subfig. A, the
167 results for Garen Gaze are presented (Eguwa and Smeti did not show differences with Garen
168 Gaze). In the flesh, glucose was the main sugar (higher than 70%), followed by xylose (12%).
169 The other sugars were lower than 10%. In the seed, mannose was the major sugar (near 60%)
170 and afterwards glucose (around 30%). A glucomannan has been described from the seeds of
171 Libyan dates (Ishrud, Zahid, Ahmad, & Pan, 2001). After hydrolysis, this polysaccharide
172 gave mannose, glucose, arabinose, and galactose in the ratio 55:23:10:10, a very similar
173 composition to that presented in Figure 2, Subfig. A.

174 The percentage of fat (13-17%) and phenols (1-5%) present in date seeds were of great
175 interest. Due to these amounts of phenols, the antiradical activity of seeds was much higher
176 (131-400 mmol TE/Kg dry weight) than that of the flesh (32-47 mmol TE/Kg dry weight), as
177 described by other authors (Nehdi, Oruri, Khalil, & Al-Resayes, 2010). Although the seed is
178 the minor and inedible portion of date fruits, its composition is very important in this study
179 because during hydrothermal treatments, seeds would be partially disorganized and their
180 components could modify the composition of final products to a great extent.

181 **3.2. Hydrothermal treatments and overall recovery.**

182 As explained in Materials and Methods, two different hydrothermal treatments were applied
183 to date samples. SET (Figure 1, Subfig. 1) has a rapid decompression (explosion) which helps
184 in lignocellulosic material disorganization and solubilization. ST (Figure 1, Subfig. 2)
185 operates at lower pressures, with longer time of treatment and without explosion. The latter
186 could be more easily applied to the industry due to its lower work pressure, having similar
187 results to the first one when working with non pure lignocellulosic materials.

188 With SET, the varieties were studied separately in 250g (x2) fresh weight samples (Table 2).
189 The overall recovery was similar between duplicates and among the different treatments and
190 varieties. Only Smeti showed little difference in duplicates at 180 °C. The average recovery
191 of dry solid residue was 25.06%. This percentage seemed to be low when compared with
192 other products, such as olive stones, treated in the same reactor where the recovery average
193 was 59.12% (Fernández-Bolaños et al, 2001). It is important to take into account that soluble
194 sugars accounted for more than 50% of date flesh dry weight (Table 1). If the percentages of
195 recovery were calculated on the dry and soluble sugar-free basis, they would increase to 50-
196 60%, similar to those reported previously by Fernández-Bolaños et al. (2001) for olive stones
197 and by Garrote, Cruz, Domínguez & Parajó (2008) for barley husks.

198 Although treatments at different temperatures did not lead to different recoveries in SET, this
199 factor influenced the granulometry of recovered solid. In Garen Gaze, the increase in
200 temperature from 180 °C to 200 °C brought about a rise in the percentage of material <4 mm
201 from 37% to 58% (average values of duplicate experiences). The same results were found on
202 Smeti, changing from 48% to 58%. Although the duplicates had high variability in Eguwa,
203 the average values showed the same behavior as the other varieties (41% for 180 °C and 49%
204 for 200 °C).

205 In the ST reactor the same results were found: higher treatment temperature did not lead to a
206 different recovery, but affected the granulometry. The treatment at 180 °C almost doubled the
207 percentage of solid <4 mm. Comparing both reactors, seed fragments higher than 4 mm from
208 the SET had a hardness similar to those of the fresh seed, but those from the ST could be
209 broken by finger press. This could be a technological advantage because a milder milling
210 process would be needed for improving their granulometry for other industrial uses.

211 **3.3. Chemical composition of FCs.**

212 In Table 3, the chemical composition of FCs is shown. For SET, the presented data for each
213 treatment was the average value of the duplicates. After treatments, lignin was the major
214 component, followed by cellulose, NCS and protein (in some cases, protein was more
215 abundant than NCS). Fat and phenols were in similar amounts, and uronic acids and soluble
216 sugars were the lowest, and not present in some samples. FCs obtained by ST had similar
217 composition to those from SET, showing significant increases in lignin and uronic acids, and
218 decreases in NCS and phenols. In order to compare the composition of FCs with that of the
219 original date fruit, comparisons should be made on a soluble sugar-free basis. So, from the
220 insoluble part of the flesh, lignin represented around 60.3-72.4%, NCS 18.5-19.6%, cellulose
221 4.2-4.5%, protein 4.3-5.9%, uronic acids 3.6-5.8%, fat 0.9-1.4%, and phenols 0.1-0.2%,
222 depending on date variety (columns in italics in Table 1). Compared with these percentages,
223 the average composition of FCs (Table 3) from both reactors showed enrichment in fat,
224 protein, cellulose and phenols in all treatments.

225 The date variety and the treatment lead to significant differences in some components. The
226 SET reactor was especially designed to increase accessibility and separate the main
227 components of lignocellulosic biomass (cellulose, hemicellulose, and lignin). In FCs obtained
228 by SET, the content in lignin was lower than that of original fruits (if compared with data
229 corrected without soluble sugars – columns in italic in Table 1), showing a partial
230 solubilization of this component into the liquid fraction. In fact, some derivatives from lignin
231 alcohols were found in treatment liquids (unpublished results). There were significant
232 increases in lignin at 200 °C. This fact could be probably due to a higher degree of seed
233 hydrolysis at this temperature. As it was commented above, a decrease about 10-20% was
234 quantified at 200 °C in the fraction of seed pieces >4 mm (Table 2). Seeds were also very rich
235 in lignin (20-25% in dry weight basis, Table 1); therefore, an increase of seed percentages in
236 FCs could imply higher amounts of lignin in FC composition. Besides, repolymerization of

237 lignin and/or its reaction with other compounds (probably derived from sugar-decomposition
238 reactions) could also play a significant role (Garrote et al., 2008). In the ST reactor, for both
239 treatments, lignin was similar to the initial samples, and it increased significantly with
240 temperature (around 60% at 165 °C and 71% at 180 °C). In this case there was no explosion
241 step, so the lignin solubilization was very limited. As it was commented for SET, the increase
242 in lignin with temperature must be related to the higher degree of seed hydrolysis at 180 °C
243 than at 165 °C (Table 2) and to the repolymerization and cross-reactions of lignin.

244 NCS also varied with treatments because the hydrothermal treatments lead to a high
245 solubilization of hemicelluloses (Fernandez-Bolaños et al., 2001). In the SET reactor, the FCs
246 were significantly richer in NCS when treated at 180 °C than at 200 °C, and, except for Smeti
247 at 180 °C, there was no difference due to variety. When both reactors were compared, ST at
248 165 °C led to a similar NCS content (9.99%) than that of SET at 200° C, and FC obtained
249 after the 180 °C 30' process had even less NCS (4.86%). Their composition was studied by
250 gas chromatography and the results are presented in Figure 2, Subfigures B and C (for SET
251 only Garen Gaze has been shown because no difference was found among varieties). When
252 comparing these graphs with that of fresh flesh (Subfigure A) an important decrease in
253 glucose was found. This sugar could be solubilized into the liquid fraction obtained during
254 the treatments, but also transformed to hydroxymethylfurfural due to the thermal process.

255 This aromatic compound was identified in large amounts in all liquid samples (unpublished
256 data). The FCs were richer in xylose, as it was the main sugar quantified in date dietary fiber
257 (Mrabet et al., 2012). Besides this decrease in glucose, a decrease in xylose and an increase in
258 mannose were observed between treatments at 180 °C and 200 °C (Subfigure B). As
259 mentioned for glucose, xylose could also be released into treatment liquors as a soluble sugar
260 or decomposed to furfural, which was also identified in liquids and in the volatile fraction
261 (unpublished data). The increase in mannose could be related to a higher degree of hydrolysis

262 of the seeds in the 200° C samples than in the 180° C ones (Table 2), since mannose was the
263 major sugar in the seeds (Subfigure A). Comparing reactors (Subfigures B and C), SET at
264 200 °C was very similar to ST at 165 °C, where glucose, mannose and xylose were in the
265 same proportion. In the ST treatment at 180 °C, glucose accounted for more than 80%,
266 showing an important solubilization and/or degradation of sugars.

267 Cellulose did not change significantly because of treatment or variety in the SET reactor, its
268 content varied from 14-19%. In ST, the increase in temperature led to a loss in cellulose
269 content. However, in all cases its content was higher in solids after treatment than in the fresh
270 date (about 4% when calculated on dry and soluble sugar-free basis). Hydrothermal
271 treatments lead to enrichment in cellulose (among other components) in the recovered solid
272 (Fernández-Bolaños et al, 2004; Garrote et al, 2008). Besides, the partial hydrolysis of seeds,
273 whose content in cellulose was higher than that of flesh (Table 1), could also contribute to the
274 increase in the percentage of cellulose.

275 Fat varied from 5-11% in SET samples, only Smeti at 200 °C being significantly higher than
276 others. In both ST assays, fat was around 6% and there was no difference between them
277 either. When comparing with fresh flesh, this enrichment in fat must be due to the partial
278 disorganization of date seeds, much richer in fat than the flesh (Table 1). The soluble sugar
279 content was very low (0.07-2%), neither variety nor treatment having significant influence.

280 The protein content was similar in all the samples (9-13%), only FC from Garen Gaze at 200
281 °C was significantly different. Uronic acids decreased the most with treatments and even
282 disappeared in some samples. In ST solids, they were more abundant than in the others.

283 Phenols increased in all samples when compared with date fruits. In SET, increasing
284 treatment temperatures led to a significant raise in this component. The FC from the
285 treatment at 165 °C in the ST reactor had a phenol percentage similar to that of 180 °C in the
286 SET reactor. A temperature increase in the ST did not imply significant changes in phenols.

287 For SET, the factorial ANOVA analysis concluded that the factor variety had influence only
288 on fat and uronic acids, and the factor treatment affected the cited components plus lignin,
289 NCS and phenols. There was no interaction between both factors in any case. Therefore,
290 treatment was the most influential factor, and this is the reason why it was decided to study
291 different ST treatment conditions with samples consisting of a variety blending. Besides, this
292 could increase the efficiency from an industrial point of view.

293 **3.4. Functional properties and antiradical activity.**

294 WHC, OHC and SOL were measured in all the FCs (Table 4). In the SET reactor, these three
295 parameters showed few differences. WHC was around 6 mL/g, OHC higher than 8 mL/g, and
296 SOL about 20%. After factorial ANOVA, it could be concluded that variety or temperature
297 did not have any effect. Only in SOL, lower treatment temperature led to significantly higher
298 SOL values. The same effect of temperature was observed in the ST reactor, but in this case
299 the three properties were affected. WHC, SOL and OHC decreased by increasing the
300 temperature from 165 °C to 180 °C. Significantly higher values of WHC were obtained with
301 the ST than with the SET one. The opposite results were found for OHC.

302 Comparing these results with those obtained from fresh date fiber (Mrabet et al, 2012),
303 hydrothermal treatments led to a decrease in WHC and an increase in OHC (12-13 mL
304 water/g fresh date fibre and 3-4 mL oil/g fresh date fiber). Despite the decrease in WHC, the
305 obtained values are in the range of other agricultural by-products proposed as valuable
306 dietary fiber sources, such as pear pomace, 5 mL water/g (Mckee & Latner, 2000) and
307 grapefruit peel, 8.5 mL water/g (Larrauri, Rupérez, Borroto, & Saura-Calixto, 1997). Of great
308 interest was the high value of OHC which was similar to the highest values found in the
309 bibliography, 5-8 mL/g for asparagus byproduct (Fuentes-Alventosa et al, 2009). These
310 results for WHC and OHC make date FC valuable as texture or viscosity modifier. Its use
311 would be also appropriate in food products that require emulsifying properties.

312 Nowadays, the antioxidant activity of a food ingredient is a very interesting property for the
313 formulation of healthier foods. The secondary date varieties studied in this work were chosen
314 due to their high antioxidant activity linked to dietary fiber (Mrabet et al, 2012). In fact, the
315 antiradical activity presented in Table 1 was constituted in a percentage higher than 90% by
316 the ethanol-insoluble fraction for the three varieties (data not shown). Both hydrothermal
317 treatments increased the total antiradical activity (Table 4), being even higher than that
318 quantified in date seeds (Table 1) in most cases. A percentage between 60-80% of the FC
319 antiradical capacity was due to the soluble fraction, unlike quantified for fresh flesh and
320 seeds. In FCs, this activity could be due to soluble antioxidant compounds as alcohols,
321 aldehydes, ketones, benzoic and cinnamic acids (Conde, Moure, Domínguez, & Parajó, 2011)
322 derived from non-cellulosic structural components of date fruits, including lignin and
323 hemicelluloses. In SET, activity increased significantly with temperature in both soluble and
324 insoluble fractions. The opposite behavior was found in ST, where activity decreased with
325 temperature. The influence of hydrothermal treatment parameters were also noticed by
326 González, Cruz, Domínguez, & Parajó (2004) and Garrote et al. (2008) working with
327 eucalyptus wood and barley husks respectively. These authors concluded that treatments at
328 higher severities led to extracts with decreased antioxidant activity, suggesting a different
329 nature and/or structure of the active compounds. The factorial ANOVA analysis concluded
330 that the variety was also a significant factor for SET soluble activity, but not for the insoluble
331 one. FCs from Eguwa were the samples with the highest soluble antiradical activity (360 and
332 447 mmol TE/Kg for 180 °C and 200 °C, respectively). This variety had also the highest
333 activity in its fresh seeds (Table 1), so their partial disorganization due to treatments could
334 probably lead to FCs with higher antioxidant activity. When both reactors were compared,
335 there were significant differences only in the soluble antiradical activity, being lower in FCs
336 obtained by ST. The total activity for date FC varied from 240-580 mmol TE/Kg. It could be

337 hardly found in bibliography agricultural by-products with so high antioxidant activity. Only
338 citrus by-products (Marín, Soler-Rivas, Benavente-García, Castillo, & Pérez-Álvarez, 2007)
339 and red grape pomace (Llobera & Cañellas, 2007) were reported to have similar antioxidant
340 activity to date FC, 70-240 and 427 mmol TE/Kg respectively. This fact is of great interest
341 for a possible application of date FC as food ingredient.

342 The changes observed in this property were very similar to those above commented for
343 phenols (Table 1). In fact, a regression analysis between phenols and soluble antiradical
344 activity showed r^2 value of 0.8617 for a linear model, showing in this way that both
345 parameters were highly related.

346 **3.5. Comparative study between reactors.**

347 The SET reactor has the technological advantage of the disruption of lignocellulosic material
348 due to the explosive decompression. However, this treatment is hardly applied by the industry
349 due to its high working pressure (up to 42 Kg/cm²). The ST reactor works at lower pressure
350 (up to 9 Kg/cm²) which is easily reached by industrial steam generators. In this other reactor,
351 longer treatment periods have similar effects to explosive decompression when working with
352 materials which are not very lignified. When date fruits were processed, the recovery of the
353 solid residue was similar (Table 2), but important differences were found in granulometry. As
354 was expected, the percentage of solid with a size lower than 4 mm was higher in SET than in
355 ST, and in treatments with higher temperatures. However, the hardness of pieces with size
356 higher than 4 mm was lower in the ST reactor which implies an advantage in the milling of
357 the recovered solid and, therefore, promotes its use as active carbon, animal feeding, abrasive
358 or in cosmetic formulations as an exfoliating agent (Rodríguez, Lama, Rodríguez, Jiménez,
359 Guillén, & Fernández-Bolaños, 2008; Al-Muhtaseb, 2010).

360 FCs from both reactors showed significant differences in composition (Table 3). Those from
361 the ST were richer in lignin and uronic acids, but poorer in NCS and phenols. Their soluble

362 antiradical activity was lower too. The treatment at 165° C for 30 min led to an FC with a
363 composition which was very similar to that of the FCs obtained with the SET reactor.
364 Smoother treatments (at lower temperature or shorter than 30 min) could improve the
365 composition and antiradical activity. The functional properties were affected by the
366 treatment: WHC increased in the ST reactor, but OHC decreased. Just as it happened for
367 composition, the FC from the treatment at 165° C had better values than that from 180° C.
368 The ST at optimized conditions of secondary varieties of Tunisian date fruits could be an
369 interesting alternative for date growers, because this reactor could be easily scaled up. The
370 obtained FCs have similar chemical composition and functional characteristic to those
371 obtained by SET, and they could be a valuable ingredient for the formulation of healthier
372 foods (fiber or antioxidant enriched). They have a balanced nutritional composition (around
373 6% fat, 10% protein, and 70-80% dietary fiber). Date seed oil has been studied by other
374 authors, and its composition in vitamins, minerals and fatty acids made it valuable for food
375 formulation (Nehdi et al, 2010; Habib & Kamal, 2013). Besbes, Blecker, Deroanne, Lognay,
376 Drira, & Attia (2005) studied the effects of heating on date seed oil and they concluded that
377 this oil resisted thermal treatment during a long period of time (30-40 h). So, it is probable
378 that the oil from FCs had good quality parameters. Besides, these FCs have very high
379 antioxidant activity, similar to the highest antioxidant agricultural by-products (Llobera &
380 Cañellas, 2007; Marín et al, 2007), citrus by-products and Manto Negro red grape pomace.
381 The pleasant chocolate/coffee flavor of the solids is another positive characteristic for their
382 use in food formulation, especially in dairy or bakery products. It would be necessary to
383 adjust the ST conditions to optimize the composition, balance of components, and antioxidant
384 activity, but this is a promising process for valorization of Tunisian secondary date varieties
385 which will help in the industry of the production areas. The use of the secondary date
386 varieties in the food industry as healthy ingredients could also help in the fight against the

387 reduction in vegetable genetic variability. In this way, the growth of these native cultivars
388 from the Tunisian coastal oasis for technological purposes may play an important role in the
389 economical, social and ecological level of the people from this developing region.

390

391 **4. ABBREVIATIONS USED**

392 SET: steam explosion treatment; ST: steam treatment; DPPH: 2,2-diphenyl-1-picrylhydrazyl
393 free radical; FC: fiber concentrate; NCS: non-cellulosic sugars; TFA: trifluoroacetic acid;
394 WHC: water holding capacity; SOL: solubility; OHC: oil holding capacity.

395

396 **5. ACKNOWLEDGEMENTS**

397 This research was supported by the Ministerio de Asuntos Exteriores y Cooperación –
398 Agencia Española de Cooperación Internacional para el Desarrollo MAEC-AECID (Spain),
399 and by the Banq Islamique de Développement BID (Saudi Arabia).

400

401 **6. REFERENCES**

402 Al-Farsi, M.A. (2003). Clarification of date juice. *International Journal of Food Science and*
403 *Technology*, 38, 241-245.

404 Al-Muhtaseb, S.A. (2010). Adsorption and desorption equilibria of nitrogen, methane,
405 ethane, and ethylene on date-pit activated carbon. *Journal of Chemical and Engineering*
406 *Data*, 55, 313-319.

407 Besbes, S., Blecker, C., Deroanne, C., Lognay, G., Drira, N., & Attia, H. (2005). Heating
408 effects on some quality characteristics of date seed oil. *Food Chemistry*, 91, 469-476.

409 Chibane, H., Benamara, S., Noui, Y., & Djouad, A . (2007). Some physicochemical and
410 morphological characterizations of three varieties of Algerian common date. *European*
411 *Journal of Scientific Research*, 18, 134–140.

412 Conde, E., Moure, A., Domínguez, H., Parajó, J.C. (2011). Production of antioxidants by
413 non-isothermal autohydrolysis of lignocellulosic wastes. *LWT-Food Science and Technology*,
414 *44*, 436-442.

415 * Elleuch, M., Besbes, S., Roiseux, O., Blecker, C., Deroanne, C., Drira, N.E., & Attia, H.
416 (2008). Date flesh: chemical composition and characteristics of the dietary fibre. *Food*
417 *Chemistry*, *111*, 676-682. This reference is a complete description of two edible date
418 varieties, and it is important as a comparison with the inedible varieties studied in this work.

419 * Fernandez-Bolaños, J., Felizón, B., Heredia, A., Rodríguez, R., Guillén, R., & Jiménez, A.
420 (2001). Steam-explosion of olive stones: hemicelluloses solubilization and enhancement of
421 enzymatic hydrolysis of cellulose. *Bioresource Technology*, *79*, 53-61. In this key reference
422 there is a complete description of the steam explosion treatment applied in the present work.

423 Fernández-Bolaños, J., Rodríguez, G., Gómez, E., Guillén, R., Jiménez, A., Heredia, A.
424 (2004). Total recovery of the waste of two-phase olive oil processing: isolation of added-
425 value compounds. *Journal of Agricultural and Food Chemistry*, *52*, 5849–5855.

426 Fuentes-Alventosa, J.M., Rodríguez-Gutiérrez, G., Jaramillo-Carmona, S., Espejo-Calvo, J.A.,
427 Rodríguez-Arcos, R., Fernández-Bolaños, J., Guillén-Bejarano, R., & Jiménez-Araujo, A.
428 (2009). Extraction method on phytochemical composition and antioxidant activity of high
429 dietary fibre powders obtained from asparagus byproducts. *Food Chemistry*, *116*, 484–490.

430 Garrote, G., Cruz, J.M., Domínguez, H., & Parajó, J.C. (2008) Non-isothermal autohydrolysis
431 of barley husks: product distribution and antioxidant activity of ethyl acetate soluble
432 fractions. *Journal of Food Engineering*, *84*, 544-552.

433 González, J., Cruz, J.M., Domínguez, H., & Parajó, J.C. (2004). Production of antioxidants
434 from *Eucalyptus globulus* wood by solvent extraction of hemicelluloses hydrolysates. *Food*
435 *Chemistry*, *84*, 243-251.

- 436 Habib, H.M., & Kamal, H. (2013). Carotenoids, fat soluble vitamins and fatty acid profiles of
437 18 varieties of date seed oil. *Industrial Crops and Products*, *42*, 567-572.
- 438 Ishrud, O., Zahid, M., Ahmad, V.U., & Pan, Y. (2001). Isolation and structure analysis of a
439 glucomannan from the seeds of Lybian dates. *Journal of Agricultural and Food Chemistry*,
440 *49*, 3772-3774.
- 441 Jiménez, A., Rodríguez, R., Fernández-Caro, I., Guillén, R., Fernández-Bolaños, J., &
442 Heredia, A. (2000). Dietary fibre content of tables olives processed under different European
443 styles: study of physicochemical characteristics. *Journal of the Science of Food and*
444 *Agriculture*, *87*, 1–6.
- 445 Kar, W. E., Gutiérrez, C.V., & Kinoshita, C.M. (1998). Steam explosion of sugarcane
446 bagasse as a pretreatment for conversion to ethanol. *Biomass and Bioenergy*, *14*, 1-9.
- 447 * Lama-Muñoz, A., Rodriguez-Gutierrez, G., Rubio-Senent, F., Gómez-Carretero, A., &
448 Fernández-Bolaños, J. (2011). New hydrothermal treatment of alperujo enhances the content
449 of bioactive minor components in crude pomace olive oil. *Journal of Agricultural and Food*
450 *Chemistry*, *59*, 1115–1123. In this key reference there is a complete description of the steam
451 treatment (ST) applied in the present manuscript.
- 452 Larrauri, J.A., Rupérez, P., Borroto, B., & Saura-Calixto, F. (1997). Seasonal changes in the
453 composition and properties of a high dietary fibre powder from grapefruit peel. *Journal of the*
454 *Science of Food and Agricultural*, *74*, 308-312.
- 455 Llobera, A., & Cañellas. J. (2007). Dietary fibre content and antioxidant activity of Manto
456 Negro red grape (*Vitis vinifera*): pomace and stem. *Food Chemistry*, *101*, 659-666.
- 457 Marín, F.R., Soler-Rivas, C., Benavente-García, O., Castillo, J., & Pérez-Álvarez, J.A.
458 (2007). By-products from different citrus processes as a source of customized functional
459 fibres. *Food Chemistry*, *100*, 736-741.

460 Mckee, L.H., & Latner, T.A. (2000). Underutilized sources of dietary fiber: a review. *Plant*
461 *Food for Human Nutrition*, 55, 285-304.

462 Montane, D., Farriol, X., & Salvado, J. (1998). Fractionation of wheat straw by steam
463 explosion pretreatment and alkali delignification, cellulose pulp and byproducts from
464 hemicellulose and lignin. *Journal of Wood Chemistry and Technology*, 18, 171-191.

465 * Mrabet, A., Rodríguez-Arcos, R., Guillén-Bejarano, R., Chaira, N., Ferchichi, A., &
466 Jiménez-Araujo, A. (2012). Dietary fiber from Tunisian common date cultivars (*Phoenix*
467 *dactylifera* L.): chemical composition, functional properties, and antioxidant capacity.
468 *Journal of Agricultural and Food Chemistry*, 60, 3658-3664. This is a comparative study of
469 ten secondary date varieties. Authors identify some varieties worthy to be valorised as source
470 of healthy food ingredients. These varieties were studied in the present work.

471 * Nehdi, I., Omri, S., Khalil, M.I., & Al-Resayes, S.I. (2010). Characteristics and chemical
472 composition of date palm (*Phoenix canariensis*) seeds and seed oil. *Industrial Crops and*
473 *Products*, 32, 360-365. This reference is a very complete work about date seeds and date seed
474 oil. It is important in the discussion about composition and possible uses of date seeds.

475 Rodríguez, G., Lama, A., Rodríguez, R., Jiménez, A., Guillén, R., & Fernández-Bolaños, J.
476 (2008). Olive stone an attractive source of bioactive and valuable compounds. *Bioresource*.
477 *Technology*, 99, 5261-5269.

478 Rodríguez, R., Jaramillo, S., Rodríguez, G., Espejo, J.A., Guillén, R., Fernández-Bolaños, J.,
479 Heredia, A., & Jiménez, A. (2005). Antioxidant activity of ethanolic extracts from several
480 asparagus cultivars. *Journal of Agricultural and Food Chemistry*, 53, 5212-5217.

481 Rodríguez-Gutiérrez, G., Lama-Muñoz, A., Ruiz-Méndez, M.V., Rubio-Senent, F., &
482 Fernández-Bolaños, J. (2012). New olive-pomace oil improved by hydrothermal pre-
483 treatments. In: *Olive Oil-Constituents, Quality, Health Properties and Bioconversions*.
484 Dimitrios, B. (ed.), In Tech, pp: 249-266. ISBN: 978-953-307-921-9.

485

486 **FIGURE CAPTIONS**

487

488 Figure 1.- Diagrams of hydrothermal treatment reactors (Rodríguez-Gutiérrez, Lama-Muñoz,
489 Ruiz-Méndez, Rubio-Senent, & Fernández-Bolaños, 2012). Subfigure A: Steam explosion
490 reactor (SET). Subfigure B: Steam reactor (ST).

491 1.- Steam generator. 2.- Steam accumulator. 3.- Reactor chamber (2 L capacity). 4.-

492 Expansion chamber. 5.- Reactor chamber (100 L capacity). 6.- Cooler. 7.- Steam

493 accumulator.

494

495 Figure 2.- Glycosyl composition, expressed as relative percentages, of fresh Garen Gaze

496 dates (flesh and seed) and FCs after hydrothermal treatments. Subfigure A: Fresh Garen Gaze

497 dates. Subfigure B: Garen Gaze variety, SET reactor. Subfigure C: mixed varieties, ST

498 reactor.

499 GG.- Garen Gaze variety. Rha.- Rhamnose. Fuc.- Fucose. Ara.- Arabinose. Xyl.- Xylose.

500 Man.- Mannose. Gal.- Galactose. Glu.- Glucose.

Table 1.- Chemical composition (%) and antiradical capacity (mmol TE/Kg) of date varieties.

	GG		EG		SM				
	Flesh	Seed	Flesh	Seed	Flesh	Seed			
Weight/fruit (g)	6.99±0.97 a		7.74±1.05 a		11.49±0.74 b				
Flesh/seed ratio	5.93±1.02 b		5.02±0.74 a		8.17±1.28 c				
Moisture	22.98±2.64 b	11.63±1.17 a	26.67±0.99 c	13.34±0.67 a	31.47±1.25 d	12.93±0.46 a			
Soluble sugars ^a	61.61±4.92 b	0.26±0.00 a	62.01±5.05 b	1.50±0.01 a	60.64±3.86 b	0.58±0.01 a			
Lignin ^a	25.15±0.92 c	<i>66.98^b</i>	25.44±0.06 c	27.5±0.71 c	<i>72.39</i>	19.60±0.85 a	23.75±2.54 bc	<i>60.34</i>	23.20±0.56 bc
NCS ^a	6.96±0.54 a	<i>18.53</i>	20.55±2.06 d	7.24±0.26 a	<i>19.06</i>	19.73±0.89 c	7.73±0.63 a	<i>19.64</i>	10.84±0.69 b
Cellulose ^a	1.58±0.21 a	<i>4.21</i>	20.28±1.37 c	1.72±0.15 a	<i>4.53</i>	17.04±0.94 b	1.74±0.23 a	<i>4.42</i>	17.79±1.28 b
Protein ^a	1.70±0.10 a	<i>4.53</i>	5.88±0.04 d	1.61±0.05 a	<i>4.27</i>	4.64±0.20 c	2.33±0.07 b	<i>5.92</i>	5.79±0.36 d
Uronic acids ^a	1.60±0.21 a	<i>4.26</i>	4.03±0.33 d	1.36±0.11 a	<i>3.58</i>	4.57±0.43 e	2.28±0.21 b	<i>5.79</i>	2.60±0.22 c
Fat ^a	0.52±0.01 a	<i>1.38</i>	17.53±0.71 c	0.35±0.01 a	<i>0.92</i>	17.75±1.41 c	0.44±0.02 a	<i>1.12</i>	13.54±0.71 b b
Phenols ^a	0.04±0.00 a	<i>0.11</i>	1.05±0.03 b	0.07±0.00 a	<i>0.18</i>	5.72±0.27 d	0.04±0.00 a	<i>0.10</i>	4.24±0.13 c
Antiradical activity ^{a, c}	47.75±0.09 b	131.30±6.56 c	38.58±0.06 ab	400.40±36.03 e	32.67±0.12 a	320.20±28.82 d			

^a Expressed as dry weight basis.

^b Results in italics are expressed as dry and soluble sugar-free weight basis

^c Total antiradical activity, as a sum of soluble and insoluble activities.

TE.- Trolox equivalent. GG.- Garen Gaze. EG.- Eguwa. SM.- Smeti. NCS.- non cellulosic sugars

Table 2.- Conditions of hydrothermal treatments and recovery of the dry solid residue and its granulometry.

			Fruit dry weight (g)	% recovery	% < 4mm ^a (FC) ^b	% > 4mm ^a
GG	180° C, 5' SET	1	198.49	23.44	36.07	63.93
		2	200.47	25.03	38.54	61.46
	200° C, 5' SET	1	199.19	25.43	58.39	41.61
		2	199.94	25.62	59.18	40.82
EG	180° C, 5' SET	1	189.16	28.04	50.89	49.11
		2	189.18	26.77	32.03	67.97
	200° C, 5' SET	1	189.16	25.41	53.24	46.76
		2	189.21	24.08	45.84	54.16
SM	180° C, 5' SET	1	176.37	26.84	48.63	51.37
		2	176.77	19.93	47.95	52.05
	200° C, 5' SET	1	176.67	24.83	59.71	40.29
		2	176.80	25.38	58.06	41.94
	165° C, 30' ST		2959.34	26.35	29.65	70.34
	180° C, 30' ST		4682.50	23.24	55.00	45.00

^a This percentage is expressed on total recovered solid residue basis.

^b These fractions constituted the analyzed fiber concentrates (FC).

GG.- Garen Gaze. EG.- Eguwa. SM.- Smeti. SET.- Steam explosion treatment. ST.- Steam treatment.

Table 3.- Chemical Composition (%) of FCs after Hydrothermal Treatments, Expressed as % on Dry Weight Basis.

		Lignin	Cellulose	NCS	Protein	Fat	Phenols	Soluble sugars	Uronic acids
GG	180° C, 5' SET	50.38±2.41 ab	14.36±1.22 a	16.59±1.62b	13.60±0.46 b	7.19±0.69 b	4.38±0.01 a	1.77±0.46 b	0.84±0.00 a
	200° C, 5' SET	60.80±1.00 cd	16.34±5.03 a	10.53±0.42 a	9.31±1.31 a	7.74±0.20 b	7.43±0.17 b	1.21±0.39 b	tr
EG	180° C, 5' SET	53.57±0.07 bc	17.45±1.61 a	13.75±0.20 b	12.84±0.48 b	4.93±0.11 a	5.09±0.74 a	0.07±0.01 a	1.64±0.16 b
	200° C, 5' SET	63.07±2.22 d	15.36±1.77 a	8.92±0.07 a	11.45±0.31 ab	5.61±0.24 ab	7.82±0.26 b	1.89±0.17 b	0.76±0.02 a
SM	180° C, 5' SET	44.02±4.29 a	17.84±1.96 a	21.77±3.22 c	13.06±1.71 b	6.80±0.39 ab	4.24±0.27 a	2.01±0.37 b	0.70±0.22 a
	200° C, 5' SET	58.25±1.50 bcd	19.77±1.18 a	9.66±0.12 a	13.28±0.39 b	10.92±1.35 c	7.40±0.08 b	0.08±0.01 a	tr
Average		55.02 *	16.86 *	13.54 *	12.26 *	7.20 *	6.06 *	1.17 *	0.68 *
165° C, 30' ST		59.72±0.31 A	18.08±1.97 A	9.99±0.09 A	11.41±0.85 A	6.77±0.56 A	4.24±0.18 A	1.25±0.22 A	2.05±0.20 A
180° C, 30' ST		70.98±0.92 B	12.36±0.79 B	4.86±0.04 B	9.63±0.37 A	6.09±0.60 A	3.91±0.42 A	0.89±0.08 A	1.18±0.35 B
Average		65.25 **	15.22 *	7.42 **	10.52 *	6.43 *	4.08 **	1.07 *	1.55 **

Values are the means of at least triplicate assays. Means bearing the same symbol (lower case letter for SET reactor, capital letter for ST one, and * for comparison between reactors) are not significantly different at the 5% level as determined by the Duncan multiple-range test.

FC.- Fiber concentrate. GG.- Garen Gaze. EG.- Eguwa. SM.- Smeti. tr.- traces. SET.- Steam explosion treatment. ST.- Steam treatment. NCS.- non cellulosic sugars

Table 4.- Functional Properties and Antiradical Activity (mmol TE/Kg dry weight) of FCs after Hydrothermal Treatments.

		Antiradical activity				
		WHC mL/g	% SOL	OHC mL/g	Soluble	Insoluble
GG	180° C, 5' SET	5.37±0.65 a	22.71±0.92 a	8.70±0.31 a	206.84±8.33 a	93.06±1.14 b
	200° C, 5' SET	5.88±0.17 ab	23.28±2.46 a	8.68±0.15 a	264.35±4.47 b	131.96±3.19 c
EG	180° C, 5' SET	6.28±0.26 b	22.55±2.46 a	8.32±0.60 a	360.24±43.64 c	65.30±6.03 a
	200° C, 5' SET	5.97±0.28 ab	19.88±2.45 a	8.74±0.89 a	446.77±29.52 d	138.25±18.93 c
SM	180° C, 5' SET	5.68±0.76 ab	32.32±2.79 b	8.09±0.97 a	198.24±25.70 a	79.46±5.90 ab
	200° C, 5' SET	6.27±0.35 b	19.67±2.77 a	7.62±0.46 a	277.34±6.56 b	101.25±6.98 b
Average		5.91 *	23.40 *	8.35 *	292.30 *	101.55 *
165° C, 30' ST		8.50±0.43 B	24.75±0.71 B	7.06±0.67 B	184.90±3.57 B	127.29±11.88 B
180° C, 30' ST		6.01±0.52 A	20.14±1.45 A	6.03±0.25 A	150.07±13.79 A	90.39±10.31 A
Average		7.25 **	22.44 *	6.55 **	167.48 **	108.84 *

Values are the means of at least triplicate assays. Means bearing the same letter (lower case for SET reactor, capital for ST, and * for comparison between reactors) are not significantly different at the 5% level as determined by the Duncan multiple-range test.

TE.- Trolox equivalent. FC.- Fiber concentrate. GG.- Garen Gaze. EG.- Eguwa. SM.- Smeti. WHC.- Water holding capacity. SOL.- Solubility. OHC.- Oil holding capacity. SET.- Steam explosion treatment. ST.- Steam treatment.

Figure 1

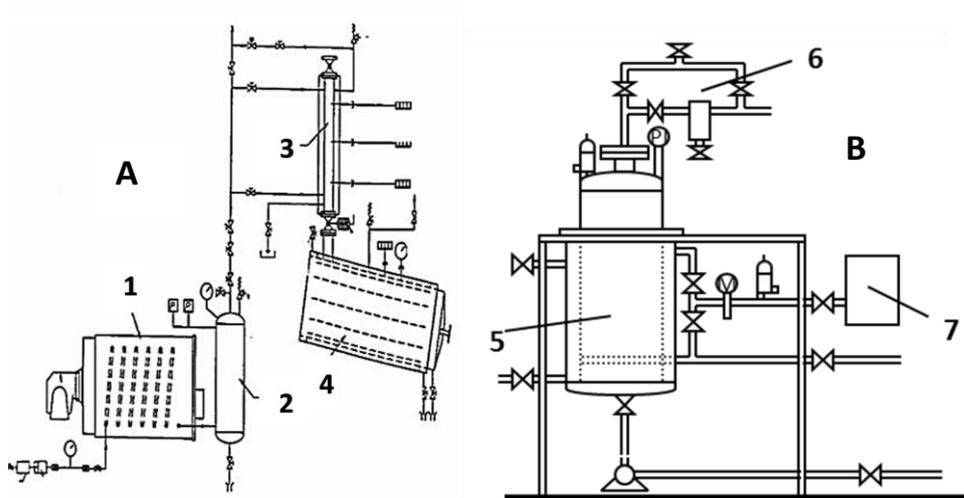


Figure 2

