

1 **Monitoring endogenous enzymes during olive fruit ripening and storage:**
2 **Correlation with virgin olive oil phenolic profiles**

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1 **ABSTRACT:**

2 The ability of olive endogenous enzymes activities β -glucosidase, polyphenol oxidase
3 (PPO) and peroxidase (POX), to determine the phenolic profile of virgin olive oil was in-
4 vestigated. Olives used for oil production were stored for one month at 20 °C and 4 °C
5 and their phenolic content and enzymatic activities were compared to those of ripening
6 olive fruits. Phenolic and volatile profiles of the corresponding oils were also analyzed.
7 Oils obtained from fruits stored at 4°C show similar characteristics to that of freshly har-
8 vested fruits. However, the oils obtained from fruits stored at 20 °C presented the lowest
9 phenolic content. Concerning the enzymatic activities, results show that the β -glucosidase
10 enzyme is the key enzyme responsible for the determination of virgin olive oil phenolic
11 profile as the decrease in this enzyme activity after 3 weeks of storage at 20 °C was paral-
12 lel to a dramatic decrease in the phenolic content of the oils.

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16 **KEYWORDS:** Virgin olive oil; olive storage; β -glucosidase; peroxidase; polyphenoloxi-
17 dase; phenolic compounds; volatile compounds; oil stability.

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1 **Highlights**

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- 3 • Similar endogenous enzymatic activities pattern were observed between 4 °C
- 4 stored fruits and fresh harvested ones.
- 5 • Contribution of POX, PPO and β -glucosidase activities in shaping the oil phenol-
- 6 ics profile.
- 7 • β -glucosidase was the key enzyme directly related to the phenolic profile.
- 8
- 9 • Similar phenolic profiles of the oils extracted from fruits stored at 4 °C and fresh
- 10 harvested ones were established.
- 11 • C6 volatile compounds increased during ripening whilst a decrease was noted in
- 12 oils extracted from fruits stored at 20 °C.

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1 **1. Introduction**

2 Virgin olive oil (VOO) is exclusively extracted from olive fruits by means of mechanical
3 techniques (Clodoveo, 2012). Olive constitutes a complex medium in which several en-
4 dogenous enzymes such as pectinases, lipases, lipoxygenases, hydroperoxide lyases, beta-
5 glucosidases, peroxidases, and polyphenol oxidases operate (Clodoveo, Hachicha Hbaieb,
6 Kotti, Mugnozza, & Gargouri, 2014). These enzymes were liberated when fruit tissues
7 were disrupted due to mechanical damage during harvest, pathogenic micro-organism at-
8 tacks occurring during storage or extraction process. VOO sensory properties are essen-
9 tially related to both its content of phenolic compounds (Fregapane & Salvador, 2013) re-
10 sponsible for oil stability and protection against autoxidation, and also its richness in
11 volatile compounds responsible for fruity and green notes (Angerosa, 2002). VOO phe-
12 nolic content depends on endogenous factors, such as phenolic glycosides amount and en-
13 zymatic activities in olive fruit tissues and technological factors during olive oil extrac-
14 tion (Romero-Segura, García -Rodriguez, Sánchez-Ortiz, Sanz, & Pérez, 2012). The most
15 important classes of endogenous enzymes are β -glucosidase which hydrolyze phenolic
16 glycosides and oxidoreductase enzymes which are responsible for phenolic compounds
17 oxidation. Moreover, the level of oxidoreductases activities depends, to a large extent, on
18 the olive variety, maturation degree, location and altitude (Clodoveo et al., 2014; Gutiér-
19 rez, Jiménez, Ruíz, & Albi, 1999).

20 The main phenolic glycosides identified in olive fruits from different cultivars and matu-
21 ration stages are oleuropein, ligstroside, demethyloleuropein, verbascoside, elenolic acid
22 glucoside, luteolin-7-glucoside, apigenin-7-glucoside, rutin and quercetin-3-rutinoside
23 (Gómez-Rico, Fregapane, & Salvador, 2008). These phenolic glycosides are hydrolysed
24 during extraction process by an endogenous β -glucosidase (E.C.3.21.1.21) generating se-
25 coiridoid compounds which constitute the most important phenolic fraction of VOO

1 (Romero-Segura, Sanz, & Pérez, 2009). Secoiridoid compounds are depicted by the dial-
2 dehydic form of decarboxymethyl elenolic acid linked to hydroxytyrosol or tyrosol (3,4-
3 DHPEA-EDA or p-HPEA-EDA), and the aldehydic forms of the oleuropein aglycone
4 (3,4-DHPEA-EA) and the ligstroside aglycone (p-HPEA-EA) (García-Rodríguez,
5 Romero-Segura, Sanz, Sánchez-Ortiz & Pérez, 2011).

6 Moreover, endogenous oxidoreductases in particular, polyphenol oxidase (PPO) and pe-
7 roxidase (POX) promote the oxidation of phenolic compounds during the milling and
8 kneading steps in olive oil extraction process (Servili, Taticchi, Esposto, Urbani, Selvag-
9 gini, & Montedoro, 2008). On the one hand, peroxidases (EC 1.11.1.7) are glycoproteins
10 catalyzing the oxidation of phenolic compounds using either hydrogen peroxide (H₂O₂) or
11 organic peroxides as the oxidizing agent, thus producing free radical, highly reactive and
12 easily polymerizable intermediates (Gajhede, 2001). While the phenol oxidation by the
13 POX activity is limited by the H₂O₂ availability, the auto-oxidation of the phenolic com-
14 pounds in the damaged tissue increases the H₂O₂ concentration which can be used by the
15 POX and thus contribute to phenolic compounds oxidation (Takahama & Oniki, 2000).
16 On the other hand, (PPO) (EC 1.14.18.1) is a copper-containing enzyme broadly distrib-
17 uted in nature, which plays an important role in many plant metabolic processes (Wa-
18 liszewski, Márquez, & Pardo, 2009). Otherwise, it contributes to the enzymatic browning
19 in many plants and vegetables damaged by improper handling, resulting in bruising, com-
20 pression or indentations (Zawitowski, Bilideris, & Eskin, 1991).

21 PPO catalyzes two distinct reactions the o-hydroxylation of monophenols to odiphenols
22 (monophenolase activity) and oxidation of the o-diphenols to o-quinones (diphenolase ac-
23 tivity).

24 VOO from Arbequina is highly appreciated by the consumers thanks to its excellent sen-
25 sorial quality (Yousfi, Weiland, & García, 2012). However, this quality depends on the

1 maturity of the fruit. In fact, Morello, Romero, and Motiva (2004) showed a clear reduc-
2 tion of sensory positive attributes and VOO oxidative stability due to the reduction of
3 contents on photosynthetic pigments (chlorophylls and carotenoids) and phenolic com-
4 pounds when the fruits are overripe. Thus, it is recommended that Arbequina fruit should
5 be harvested at an early maturation stage and for a short period (Romero, Diaz, &Tous,
6 2002). However, this situation will make the milling industries face an extreme concen-
7 tration of production, which might exceed their processing capacity, and oblige them to
8 store the fruit for longer periods varying from weeks to months.

9 Several studies have been carried out to elucidate the influence of fruits storage on oil
10 quality. Storage of 'Picual' olives at 5 °C kept the value of free acidity, peroxide value,
11 ultraviolet absorbance, and sensory quality of the resulting oil within the limit admitted
12 for extra quality until a 45-day-fruit storage (García, Gutiérrez, Barrera, & Albi, 1996a).
13 Moreover, García, Gutiérrez, Castellano, Perdiguero, Morilla, and Albi (1996b) and
14 Canet and García (1999) demonstrated that olive refrigeration (*Olea europaea* cv. Vil-
15 lalonga and Blanqueta) at 5 °C on an industrial scale delayed deterioration of the physical,
16 chemical, and sensorial parameters of oil quality, without changes in the initial oil quality
17 during the 30 day storage. Clodoveo, Debora, Tommaso and Giancarlo (2007) studied the
18 effect of different temperatures and storage atmospheres on Coratina olive oil quality. Re-
19 sults showed that oils retained their initial chemical qualities during 30 days of olives
20 storage at 5 °C under a flow of humidified air or a flow of 3% O₂ + 5% CO₂. However,
21 the oils lose their qualities after 15 days of olive storage at room temperature.

22 Given that phenolic content is a key parameter to guarantee the VOO nutritional and or-
23 ganoleptic properties, so as to obtain a product with a high economic value, suitable for
24 commercialization in international markets (Lazzez et al., 2011), it is important to under-

1 stand the evolution of the enzymatic activities related to phenolic metabolism during olive
2 fruits storage.

3 Nevertheless, according to our knowledge no studies have been carried out to investigate
4 those endogenous enzymatic activities during fruits storage and their impact on extracted
5 VOO quality.

6 The main purpose of the present study is to follow the evolution of olive (β glucosidase,
7 POX and PPO) during fruit storage at two different temperatures and to assess their pos-
8 sible role in shaping the phenolic profile of VOO.

9 **2. Materials and methods**

10 *2.1. Chemicals*

11 Reagents for enzymatic activity extraction and measurements were supplied by Sigma-
12 Aldrich (St. Louis, MO) except for phenolic compounds (oleuropein, and verbascoside)
13 purchased from Extrasynthese (Genay, France).

14 *2.2. Plant material*

15 Olives (*Olea europaea*) of the Arbequina variety, cultivated at the experimental fields of
16 Institute de la Grasa (Seville, Spain), were harvested (10 kg) in plastic containers at the
17 green mature stage October 2012, 24 weeks after flowering (24WAF). Two different stor-
18 age conditions were tested: refrigeration at 4 °C and ambient temperature (20 ± 2 °C) dur-
19 ing one month (24 to 28 WAF). Every 7 days, samples of 1 kg were taken from each stor-
20 age condition and used for olive oil extraction while about 250g were used for the enzy-
21 matic activities determination. The same experimental procedure was carried out with
22 fresh harvested fruits.

23 *2.3. Olive oil extraction*

24 Olive oil was extracted using an Abencor analyzer (Comercial Abengoa, S.A., Seville,
25 Spain) that simulates the industrial process of VOO production at laboratory scale (Mar-

1 tínez, Muñoz, Alba, & Lanzón, 1975). Milling of olive fruits (1 kg) was performed using
2 a stainless steel hammer mill operating at 3000 rpm provided with a 5 mm sieve. Malaxa-
3 tion was carried out for 30 min with the Abencor thermo beater operated at 30 °C. Cen-
4 trifugation of the kneaded paste was performed in a basket centrifuge at 3500 rpm for 1
5 min. After centrifugation, the oils were decanted and paper filtered. Then, the oils were
6 transferred into dark glass bottles, and stored in the dark at -20 °C until analysis.

7 *2.4. Enzyme extraction*

8 Acetone powders were prepared from mesocarp of fresh harvested olive fruits, following
9 the method described by García-Rodríguez et al. (2011).

10 *2.4.1. POX extraction*

11 POX enzyme extracts were prepared as described previously by García-Rodríguez et al.
12 (2011). 0.5 g of olive seeds was homogenized in 5 mL of buffer by means of an Ultra-
13 turrax homogenizer (5×1min) and then centrifuged at 15,000 g for 20 min at 4 °C. The
14 clear supernatant was used as crude extract.

15 *2.4.2. PPO extraction*

16 PPO enzyme extracts were prepared from acetone powder, following the method de-
17 scribed by García-Rodríguez et al. (2011).

18 *2.4.3. β -glucosidase extraction*

19 β -glucosidase enzyme extracts were prepared also from acetone powder as reported by
20 Romero-Segura et al. (2009).

21 *2.5. Enzyme activity assay*

22 *2.5.1. POX activity assay*

23 Peroxidase activity in the extracts was determined spectrophotometrically at 470 nm ($\epsilon =$
24 $26600 \text{ M}^{-1} \text{ cm}^{-1}$), using the method described by García-Rodríguez et al. (2011). One unit

1 of POX activity corresponds to the amount of enzyme oxidizing 1 μmol of guaiacol per
2 min.

3 2.5.2. *PPO activity assay*

4 PPO activity was determined by constantly monitoring the increase in absorbance at 400
5 nm related to the oxidation of tert-butylcatechol (TBC) (García-Rodríguez et al., 2011).
6 One unit of PPO activity was defined as the amount of enzyme forming 1 μmol of TBC-
7 quinone per min.

8 2.5.3. *β -glucosidase activity assay*

9 β -Glucosidase activity was determined spectrophotometrically by following the hydroly-
10 sis of the synthetic glucoside *p*-nitrophenyl- β ,D-glucopyranoside (pNPG) at 405 nm ($\epsilon =$
11 $552.8 \text{ M}^{-1} \text{ cm}^{-1}$), according to Romero-Segura et al. (2012). One unit of β -glucosidase ac-
12 tivity was defined as the amount of enzyme able to liberate 1 μmol of *p*-nitrophenol per
13 min.

14 2.6. *Extraction of fruit phenolic compounds*

15 Fruit phenolics were extracted according to a previously developed protocol (García-
16 Rodríguez et al., 2011). Representative fruits samples (1.65 g) were kept at 4 °C during
17 24h with dimethyl sulphoxide (10 mL). The extracts obtained were filtered through 0.22
18 μm before HPLC analysis.

19 2.7. *Extraction of VOO phenolic compounds*

20 VOO phenolics were isolated by SPE on a diol-bonded phase cartridge (Supelco, Belle-
21 fonte, PA) following a previously described procedure (Mateos et al., 2001). A solution of
22 *p*-hydroxyphenyl-acetic acid ($4.64 \times 10^{-2} \text{ mg/mL}$) and *o*-coumaric acid (9.6×10^{-3}
23 mg/mL) in methanol was used as internal standard in this extraction procedure. 0.5 mL of
24 standard solution was added to each oil sample (2.5 g).

25 2.8. *Analysis of fruit and VOO phenolic compounds*

1 Phenolic extracts were then analyzed by HPLC in a Beckman Coulter liquid chromato-
2 graphic system equipped with a System Gold 168 detector, a solvent module 126 and a
3 Mediterranea Sea 18 column (4.0 mm i.d. x 250 mm, particle size 5 μ m) (Teknokroma,
4 Barcelona, Spain) following a previously described methodology (Luaces et al., 2007).
5 Quantification and identification of phenolic compounds was performed as described by
6 García-Rodríguez et al. (2011).

7 *2.9. Stability of VOO*

8 Oxidative stability was expressed as the oxidation induction time (h), measured with the
9 Rancimat apparatus Model CH 8970 (Metrohm AG, Herisau, Switzerland) using an oil
10 sample of 2.5 g, warmed to 100°C, and a purified air flow rate of 20 l/h. Three determina-
11 tions per oil were carried out.

12 *2.10. Extraction and analysis of VOO volatile compounds*

13 Olive oil samples were conditioned to room temperature and then placed in a vial heater
14 at 40 °C. After 10 min of equilibrium time, volatile compounds from headspace were ad-
15 sorbed on a solid-phase microextraction (SPME) fiber DVB/Carboxen/PDMS 50/30 μ m
16 (Supelco Co., Bellefonte, PA). Sampling time was 50 min at 40 °C. Desorption of volatile
17 compounds trapped in the SPME fiber was performed directly into the gas chromatograph
18 (GC) injector.

19 Volatiles were analyzed using a HP-6890 GC equipped with a DB-Wax capillary column
20 (60 m \times 0.25 mm inner diameter; film thickness, 0.25 μ m; J&W Scientific, Folsom, CA).

21 Quantification and identification of volatile compounds was carried out following the
22 method described by Sánchez-Ortiz, Pérez, and Sanz. (2013). Volatile compounds were
23 clustered into different classes according to the polyunsaturated fatty acid and the LOX
24 pathway branch origin. Quantitative data for every volatile class was done as previously
25 detailed by Sánchez-Ortiz et al. (2013).

1 *2.11 Statistical analysis*

2 Data were statistically evaluated using Statgraphics Plus 5.1 (Manugistic Inc., Rockville,
3 MD). Analysis of variance (ANOVA) was applied, and comparison of means was done
4 by the Student-Newman-Keuls/Duncan test at a significance level of 0.05.

5 Moreover, the correlation between the amounts of total phenolics, o-diphenolics and se-
6 coiridoid compounds and the oxidative stability of the oils was assessed by simple regres-
7 sion applying the best fitting model (linear). The degree of correlation was expressed by
8 regression coefficient (r).

9 **3. Results and discussion**

10 POX, PPO and β -glucosidase activities were extracted and assayed in crude protein ex-
11 tracts obtained from green mature Arbequina fruits stored during 4 weeks at two different
12 temperatures (4 and 20 °C \pm 2 °C) in order to determine if endogenous enzymatic activi-
13 ties levels could be correlated to the oils phenolic profiles.

14 *3.1. Monitoring the olive endogenous enzymatic activities (POX, PPO, and β -* 15 *glucosidase) during fruit ripening and storage*

16 The evolution of POX activity in Arbequina olive seeds during ripening and fruit storage
17 at 4 and 20 °C is illustrated in (Figure 1). Concerning the POX activity during fruit ripen-
18 ing from 24 to 28 WAF, the initial value was found to be low at early fruit maturation
19 stages (9.4 \pm 0.2 U/g FW at 24 WAF). Then results demonstrate an increment on POX ac-
20 tivity during the ripening with a maximum activity level equal to 13.93 \pm 0.4 U/g FW at
21 28 WAF. These activities confirmed those reported by García-Rodríguez et al. (2011).

22 A gradual increase in seeds POX activities was also observed during storage, except in
23 the second week of fruits storage at 4 °C when the POX activity passed from 9.4 U/g FW
24 to 8.77 \pm 0.3 U/g FW. Moreover, during the first week of storage, POX activities were
25 almost constant for both storage temperatures and were respectively. 11.36 \pm 0.3 and 10.5

1 ± 0.2 U/g FW at 4 and 20 °C. On the contrary, during the second and third week of stor-
2 age, POX activities were higher in olives stored at 20 °C compared to those stored at 4
3 °C. Maximum activity level was observed on the last week of storage, reaching 15 ± 0.6
4 U/g FW and 14 ± 0.5 U/g FW for Arbequina fruits stored at 20 and 4 °C, respectively.

5 Results show that POX activities in olives seeds during storage and ripening have almost
6 the same behavior with slightly higher activities in fruits stored at ambient temperature.

7 The evolution of PPO activity in olive fruit mesocarp during ripening and storage at two
8 different temperatures was shown in (Figure 1). Results show that PPO activities decrease
9 during fruit ripening at both storage temperatures which was more marked when storage
10 was carried out at 20 °C. In fact, after one month of storage at 20 °C, PPO activity passed
11 from 287.1 ± 10.4 to 147.7 ± 24 U/g FW corresponding to 49% reduction in respect to the
12 initial value. However, when fruits were stored at 4°C, the diminution was only 24%
13 (218.4 ± 15.7 U/g FW) and followed similar pattern to that found in fresh harvested fruits.

14 In this sense, the PPO activity values found along Arbequina fruit ripening, were in good
15 agreement with previous studies on PPO activity in the fruits of Gordal and Manzanilla
16 olive varieties which had reported that olive PPO decreases sharply during the first month
17 of olive fruit development and then at a slower rate during fruit maturation and ripening
18 (Hornero- Méndez, Gallardo-Guerrero, Jarén-Galán, & Mínguez-Mosquera, 2002).

19 As previously observed for POX activity, PPO activity in olive fruits stored at 4 °C fol-
20 lowed the same trend that in fresh fruit, and was significantly different from those stored
21 at 20 °C.

22 The olive β -glucosidase activity during ripening and storage was determined and results
23 were shown in (Figure 1). The activities found for both storage temperatures (4 and 20
24 °C) were almost equal on the first week of storage (31.2 ± 1.8 and 33 ± 2.1 U/g FW re-
25 spectively). However, their behavior was very different during the last three weeks. In

1 fact, at 20 °C, the β -glucosidase activity fell considerably from 33 to 0.6 ± 0.2 U/g FW.
2 While, at 4 °C, it continued to increase until the third week of storage (57.2 ± 6.3 U/g
3 FW) and then decreased slightly at the last week of storage (50.6 ± 6.2 U/g FW).
4 Regarding, the β -glucosidase ripening evolution, the maximum activity was attained at
5 the 27 WAF (66.2 ± 4.5 U/g FW) and then decreased slowly to reach 54.3 ± 3.5 U/g FW
6 at the 28 WAF corresponding to the green-brown stage. The activity was high in green
7 fruit, when the oleuropein amount was highest and low in last stage when the oleuropein
8 concentration declined. These results were in accordance with those reported by Bitonti,
9 Chiappetta, Innocenti, Muzzalupo, and Uccella (2000) concerning the evolution of
10 oleuropein concentration at different ripening stages. In fact, in green fruits the debitter-
11 ing, due to the hydrolysis of oleuropein, does not occur because enzyme and substrate are
12 in the separate cell compartments although the respective concentrations in tissues are
13 very high. In green-brown fruits the debittering is provoked by the damage of cell struc-
14 tures, due to the senescence of tissues or pathogens injury or mechanical damage that
15 brings in contact β -glucosidase and oleuropein, despite lower enzyme activity. Moreover,
16 Mazzuca, Spadafora, and Innocenti (2006) showed that the β -glucosidase activity trend
17 during ripening appeared to be linked to oleuropein degradation and the release of glucose
18 and aglycones molecules, with the consequent physiological debittering of fruit tissues.
19 For these reasons, the clear decline on β -glucosidase activity from the first week of olive
20 storage at ambient temperature could be related to the accelerated ripening and to the
21 damage of cell structures happening during olives storage at high temperature (García et
22 al., 1996b). However, at 4 °C, the decrease on β -glucosidase activity was lagged by 2
23 weeks compared to that determined in the ambient temperature storage. This may be re-
24 lated to the fact that olive refrigeration delayed olive ripening and tissue softening (García
25 et al., 1996a; Yousfi et al., 2012).

1 In conclusion, monitoring the olive endogenous enzymatic activities (POX, PPO, and β -
2 glucosidase) during fruit ripening and storage showed a similar pattern between 4 °C
3 stored fruits and fresh harvested ones.

4 *3.2. Phenolic compounds*

5 Given the importance of the phenolic fraction, the change in the phenolic compounds
6 over time in fruits and oils could be an essential quality control parameter of VOO.

7 To investigate whether the observed enzymatic activities changes correlate with fruits and
8 oils phenolic profile, the monitoring of fruits and oils phenolic compounds in different
9 experimental conditions (fresh harvested fruits and stored ones) was studied.

10 *3.2.1. Behavior of phenolic compounds in fruits*

11
12 The phenolic composition of fresh harvested and stored olive fruits is shown in Table 1.

13 As illustrated, significant differences on phenolic composition were noted between sam-
14 ples. Moreover, the most representative complex phenols identified in olive fruits were
15 oleuropein, dimethyloleuropein, hydroxytyrosol and ligstroside. The content of oleuropein
16 decreased significantly during ripening and vanished after 4 weeks of storage at 20 °C. In
17 fact, the levels of oleuropein decreased from 7382.3 $\mu\text{g/g}$ to 1840.5 $\mu\text{g/g}$ (25% of its ini-
18 tial content) during ripening. However, in fruits stored at 4 °C, oleuropein content in-
19 creased during the two first weeks up to 10768.3 $\mu\text{g/g}$

20 Dimethyloleuropein, is a phenolic glycoside characteristic of the Arbequina variety, that
21 is considered by some authors as a degradation product of the oleuropein (Gomez-Rico et
22 al., 2008). Dimethyloleuropein doubled its content during ripening. In fact, it passes from
23 6027.7 $\mu\text{g/g}$ to 11399.8 $\mu\text{g/g}$ from 24 to 28 WAF. In the same way, the content of this
24 compound also increased significantly during olive storage at 20 °C and the same pattern
25 was observed for hydroxytyrosol.

1 Oleuropein degradation could explain hydroxytyrosol and dimethyloleuropein concentra-
2 tion increment.

3 The concentration of ligstoside decreases along ripening and storage particularly at 20 °C.

4 *3.2.2. Phenolic compounds in the oils*

5 Changes in oil phenolic profiles obtained from fresh harvested fruits and from olive fruits
6 stored at different temperatures are shown in Table 2. Olive oils samples did not show
7 significant qualitative differences in their phenolic fraction profiles. However, significant
8 quantitative differences were observed for many of the phenolic compounds.

9 The most representative phenolic compounds were 3,4-DHPEA-EDA and p-HPEA-EDA.

10 In fact, the concentration of these two compounds in the VOO obtained from fruits har-
11 vested at 24 WAF was 1.0951 ± 0.0545 and 0.452 ± 0.0258 $\mu\text{mol/g}$ olive oil, respectively.

12 The secoiridoid compound p-HPEA-EA was not detected in this oil and other simple phe-
13 nols, such as vanillic, cinnamic acid and *p*-coumaric acid, were detected in very low
14 amounts. Considerable changes were observed in oil phenolic composition obtained
15 through fruit ripening. The content of total phenolics, o-diphenolics and secoiridoids de-
16 rivatives in the oils decreased about 20-25% from fruits harvested at 24 WAF to those of
17 28 WAF.

18 The total content of phenolic compounds, orthodiphenols and secoiridoids derivatives in
19 the oils extracted also decreased as the fruit storage period progressed. Moreover, the
20 negative effect of storage time on oil phenolic compounds was more noticeable when
21 storage was carried out at 20 °C than at 4 °C. In fact, when fruits were kept at 20 °C, the
22 reduction rate of these compounds was 90.31, 95.78 and 91.73% respectively whereas, at
23 4 °C, it was 22.5, 25.93 and 26.17 % respectively.

24 In addition, secoiridoids compounds, 3,4-DHPEA-EDA and 3,4-DHPEA-EA, were the
25 most affected phenolic compounds along the total storage period. Indeed, oil loses 28.81

1 and 97.07% of these compounds when fruits were stored at 4 °C and 20 °C respectively.
2 However, the loss of p-HPEA-EDA was 18.85 and 77.54 % at 4 and 20 °C. In the same
3 way, the reduction rate of 3,4-DHPEA-EDA, and p-HPEA-EDA was respectively 27.41,
4 and 13.59% during olive ripening. However, only 3.36% of 3,4-DHPEA-EA was reduced
5 during fruit ripening.

6 Concerning the phenolic alcohols, the tyrosol concentration varied between the analyzed
7 samples. In fact, it maintained its concentration during ripening and storage at 4°C and
8 decreased only after 3 weeks when fruits were kept at 20 °C. Tyrosol concentration may
9 be associated to the ligstroside aglycone and derivatives hydrolysis which contains tyrosol
10 as an aromatic alcoholic fragment.

11 However, hydroxytyrosol showed a decrease at the last ripening stage and after 4 weeks
12 of storage at 20 °C. It was noted that tyrosol concentration was higher than hydroxytyro-
13 sol. This result was similar to those reported by several authors (Tsimidou, Papadopoulos,
14 & Boskou, 1992; Caponio, Allogio, & Gomes, 1999).

15 Regarding flavonoids, for all samples the concentration of luteolin was higher than that of
16 apigenin. Moreover, the flavonoids contents of the oils extracted from olives stored at 4
17 °C were higher than those at 20 °C.

18 In oils extracted from stored olives at 4 °C, both luteolin and apigenin contents registered
19 a clear rise during all the storage period. However, in those extracted from olives kept at
20 20 °C, flavonoids contents increased until 2 weeks of storage and then decreased sharply
21 during the last two weeks of fruit storage.

22 The significant reduction of phenolic compounds found in the oils obtained from olives
23 stored at 20 °C can be related to the fact that high storage temperatures affect the physio-
24 logical status of the olive fruits and accelerate the processes of fruit ripening. Moreover,
25 the olives suffer softening and become very sensitive to mechanical damage and to the ac-

1 tion of pathogenic microorganisms which are able to metabolize a wide variety of aro-
2 matic compounds, such as phenol and its derivatives (Watanabe, Hino, Onodera, Kajie, &
3 Takahashi, 1996). However, olive refrigeration at 4 °C delayed deterioration of the fruit
4 in terms of physical, chemical, and sensorial quality so that VOOs obtained from cold
5 stored fruits retain good organoleptic and nutritional properties. In addition, low tempera-
6 tures may have a bacteriostatic action by controlling microbial development (Clodoveo et
7 al., 2007).

8 In conclusion, our results revealed a very similar phenolic profile of the oils extracted
9 from fruits stored at 4 °C and fresh harvested ones. Whereas, the oil extracted from fruits
10 kept at 20 °C has a very poor quality marked by its low phenolic content.

11 Moreover, our data showed that the β -glucosidase was the key enzyme directly related to
12 the oil phenolic profile since the sharp decrease in β -glucosidase activity after 3 weeks of
13 storage at 20 °C was parallel to a dramatic decrease of phenolic compounds in the corre-
14 sponding oils even if the fruit was rich in phenolic compounds.

15 *3.3. Oil stability*

16 The oil stability evaluates the time (hours) of their resistance to oxidation and plays a key
17 role in the assessment of the olive oil quality.

18 Oxidative stability, measured by Rancimat method, of the oils extracted during fruit rip-
19 ening and storage was shown in Table 3. Results showed that the oil stability was clearly
20 affected by fruit ripening and storage conditions. In fact the oil stability decreased with
21 olive maturation degree and storage. Moreover, this reduction was more important when
22 fruits were stored at 20 °C. Indeed, 60.62% of oil stability was lost when olives were kept
23 at 20 °C during 4 weeks, whereas only 20 and 15.93% was lost respectively during refrig-
24 eration and ripening. García and Yousfi (2006) demonstrated that storage at low tempera-
25 ture delayed olive fruit ripening and that VOO stability decreased with olive maturation.

1 The results obtained were in good agreement with those found in a previous study on Pic-
2 ual fruits stored at 5 °C, 8 °C and ambient temperature during 60 days, in which the oil
3 stability decreased 35, 70 and 93% respectively.

4 Numerous studies have reported a correlation between phenolic compounds content and
5 oil oxidative stability. As described elsewhere, the contribution of phenolic compounds to
6 oil stability was estimated at approximately 30%, of fatty acids at 27% and of carotenoids
7 at 6% (Apparicio, Roda, Albi, & Gutiérrez, 1999).

8 Figure 2 showed the relation between the oil phenolics, o-diphenolics and secoridoids
9 content, with their corresponding oxidative stabilities. Data shown by this figure evi-
10 denced a positive correlation between the oxidative stability measured by Rancimat and
11 total phenolics, o-diphenolics and secoridoid compounds analysed by HPLC. As illustrat-
12 ed, the maximum correlation ($r=0.985$) was found between total phenols in oils extracted
13 from fruits stored at higher temperature and the resistance to oxidation. This result was
14 due to the lowest amounts of phenolics compounds in oils obtained from stored olive
15 fruits at high temperature and confirms the strict dependence between phenolic concentra-
16 tion and oil stability.

17 *3.4. Volatile compounds*

18 Given that the aroma is a key quality parameter for VOO, the volatile profiles of the dif-
19 ferent oils obtained were studied (Table 4). C6 aldehydes and alcohols and their corre-
20 sponding esters, are the most important compounds in the VOO aroma, from either a
21 quantitative or a qualitative point of view (Kotti, Gargouri, Chiavaro, & Bendini, 2011).

22 Moreover, Kalua, Allen, Bedgood Jr, Bishop, Prenzler, & Robards (2007) reported that
23 storage of the fruit decreased the aldehyde and ester content that is responsible for the
24 positive aroma.

1 During olive fruit ripening an increase of C6 compounds was observed, from 22645 to
2 29770 ng/g oil. The contents of C6 compounds, in oils extracted from fruits stored at 20
3 °C decreased through storage. Whereas oils extracted from stored fruits at 4 °C showed a
4 progressive increase in these compounds until 2 weeks. These compounds are responsible
5 for the positive aroma perceptions in olive oils.

6 The trend of C5 compounds including also aldehydes and alcohols compounds was simi-
7 lar to that of C6 compounds with a difference in the case of cold storage consisting in a
8 decrease of C5 compounds content after 1 week of storage at 4 °C.

9 The esters content was also significantly affected by temperature and storage period. In
10 fact, the content of these compounds was higher when fruits were stored at 20 °C than at
11 4 °C. Moreover, their concentrations reached highest level after 2 weeks of storage at
12 both temperatures. However, these values were lower than the concentration measured in
13 fresh fruits at the same period.

14 **4. Conclusion**

15 This study demonstrated that storage conditions affects VOO flavor quality. However, ol-
16 ive refrigeration at 4 °C retarded deterioration in the oil quality parameters by delaying
17 alteration of the olive fruits. In this sense, for most quality parameters, a very similar pat-
18 tern was observed between fruits stored at 4 °C and freshly harvested fruits. Thus, the
19 prolonged storage of fruits caused a decrease in corresponding oil phenolic content. This
20 negative effect was more noticeable when fruits were kept at ambient temperature (20 ± 2
21 °C) than at 4 °C.

22 A positive correlation was obtained between the decrease of total phenolics, *o*-diphenolics
23 and secoiridoid compounds and the loss of oxidative stability measured by Rancimat
24 method.

1 Results evidenced the contribution of POX, PPO and β -glucosidase activities in determin-
2 ing the oil phenolics profile. However, the experimental data obtained point to olive β -
3 glucosidase as the key enzyme to determine the phenolic profile of the VOO.

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1 **Appendix:**

2 ABBREVIATIONS USED

3 VOO = Virgin Olive Oil

4 POX = Peroxidase

5 PPO = Polyphenoloxidase

6 *p*NPG = *p*-nitrophenyl- β ,D-glucopyranoside

7 TBC = Tert-butylcatechol

8 SDS =Sodium Dodecyl Sulfate

9 PMSF= Phenyl Methyl Sulfonyl Fluoride

10 WAF = Weeks After Flowering

11 EDTA = Ethylen Diamine Tetraacetic Acid

12 DTT = Dithiothreitol

13 3,4-DHPEA-EDA = di-aldehydic form of the oleuropein aglycone

14 *p*-HPEA-EDA = di-aldehydic form of the ligstroside aglycone

15 3,4-DHPEA-EA = aldehydic form of the oleuropein aglycone

16 *p*-HPEA-EA = aldehydic form of the ligstroside aglycone

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5 of Higher Education and Scientific Research.

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48 **FIGURE CAPTIONS**

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1 Figure 1. Evolution of olive endogenous enzymatic activities (U/g FW, means of three
2 replicates) during fruit ripening (a) and along the storage of Arbequina fruits at 4°C (b)
3 and at 20°C (c)

4
5 Figure 2. Correlation between the amounts of total phenolics, o-diphenolics and
6 secoridoids compounds with the oxidative stability of the oils extracted from fresh har-
7 vested Arbequina fruits (a) and fruits stored at 4°C (b) and at 20°C (c). The correlation
8 was assessed by simple regression applying the best fitting model (linear). The degree of
9 correlation was expressed by regression coefficient (r).

10

11 Table 1. Concentration of phenolic compounds in fresh harvested Arbequina fruits and
12 fruits stored at different temperatures (4 and 20°C) during 4 weeks.

13 Table 2. Concentration of phenolic compounds in VOOs obtained from fresh harvested
14 Arbequina fruits and fruits stored at different temperatures (4 and 20°C) during 4 weeks.

15

16 Table 3. Oxidative stability index (expressed as hours) of oil samples extracted from fresh
17 harvested Arbequina fruits and fruits stored at different temperatures (4 and 20°C) during
18 4 weeks.

19

20 Table 4. Concentration of Volatile compounds in VOOs obtained from fresh harvested
21 Arbequina fruits and fruits stored at different temperatures (4 and 20°C) during 4 weeks.

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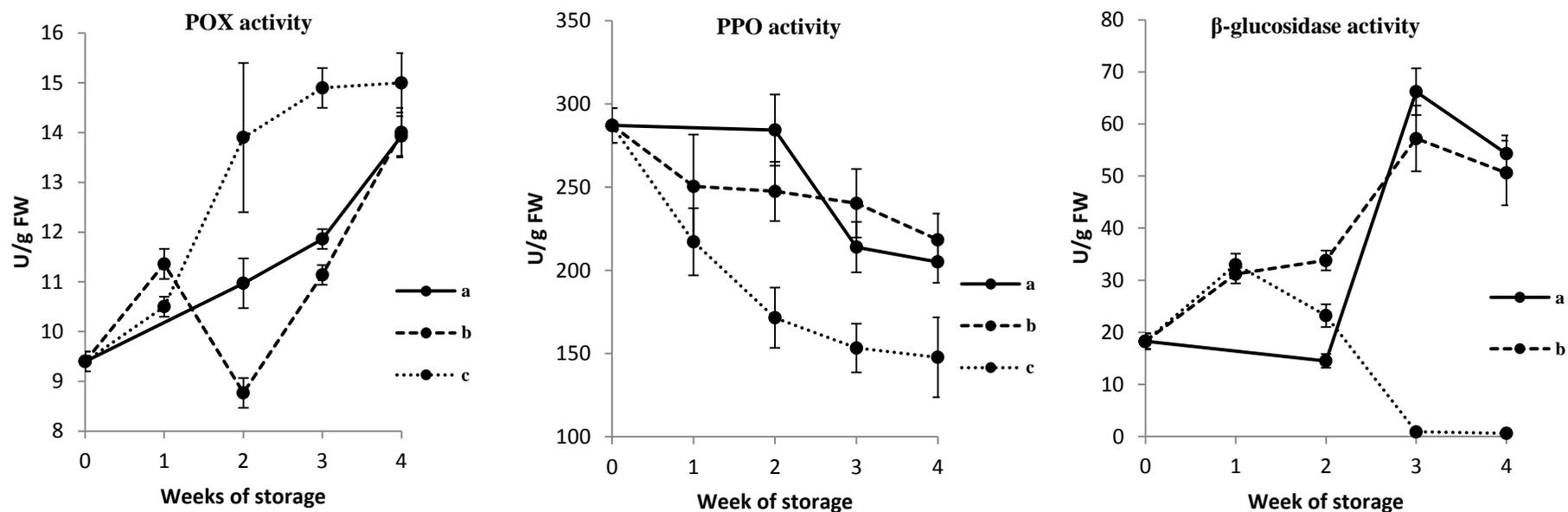
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4 Figure 1. Evolution of olive endogenous enzymatic activities (U/g FW, means of three replicates) during fruit ripening (a) and along the storage
5 of Arbequina fruits at 4°C (b) and at 20°C (c)

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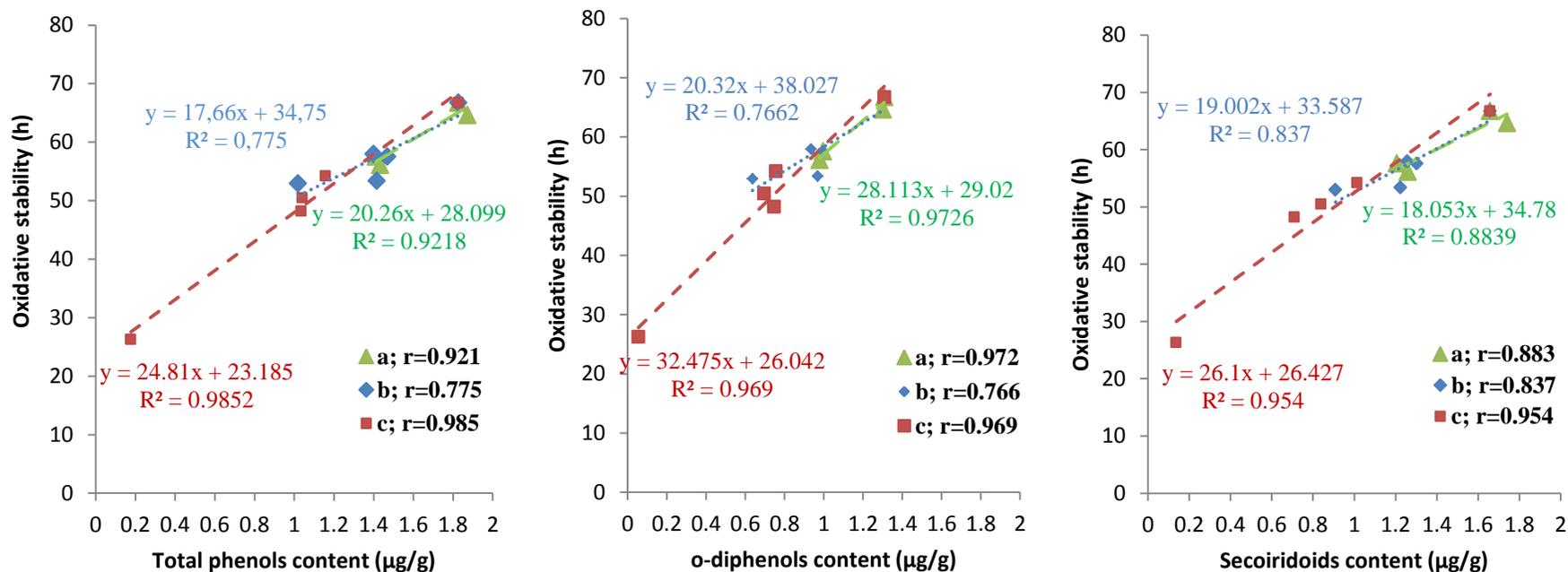
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(R. Hachicha Hbaieb).

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8 Figure 2. Correlation between the amounts of total phenolics, o-diphenolics and secoridoids compounds with the oxidative stability of the oils
9 extracted from fresh harvested Arbequina fruits (a) and fruits stored at 4°C (b) and at 20°C (c). The correlation was assessed by simple regres-
10 sion applying the best fitting model (linear). The degree of correlation was expressed by regression coefficient (r).

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Table 1. Phenolic contents ($\mu\text{mol/g}$ olive pulp) in Arbequina fruits ripen on the tree and stored at different temperatures (4 and 20°C) during 4 weeks.

Fruit Phenolics ($\mu\text{mol/g}$ FW)	<u>On-tree ripening</u>				<u>Storage at 4°C</u>				<u>Storage at 20°C</u>			
	0 (24 WAF)	2 week	3 weeks	4 weeks	1 week	2 weeks	3 weeks	4 weeks	1 week	2 weeks	3 weeks	4 weeks
Hydroxytyrosol	2.5	1.9a	1.5a	2.2a	2.1	2.1a	2.0a	2.3a	3.1	4.4b	5.1b	6.7b
Demethyloleuropein	11.4	16.4b	14.8b	21.7b	9.3a	9.1a	12.4a	17.0a	17.7b	23.8c	34.9c	36.0c
Verbascoside	1.7	1.7	0.8a	1.4a	1.7	1.7	1.4b	1.5a	1.6	1.6	2.1c	2.4b
Luteolin-7-glucoside	0.5	0.7	0.5	0.8	0.7	0.9	0.6	0.8	0.8	0.6	0.8	0.6
Oleuropein	13.7	6.2a	2.7a	3.4b	16.6b	19.9c	15.3b	17.1c	8.4a	2.3b	1.6a	0a
Ligstroside	12.7	0.7b	0.5a	0a	1.1b	1.3c	0.8b	1.0c	0.7a	0.5a	0.6a	0.7b

*Mean values from three determinations.

**For each compound and week values with different letters are significantly different ($P \leq 0.05$).

1 Table 2: Concentration of phenolic compounds in VOOs obtained from fresh harvested Arbequina fruits and fruits stored at different temperatures (4 and 20°C) during 4
2 weeks

	0 (24 WAF)	<u>On-tree ripening</u>			<u>Storage at 4°C</u>				<u>Storage at 20°C</u>			
		2 weeks	3 weeks	4 weeks	1 week	2 weeks	3 weeks	4 weeks	1 week	2 weeks	3 weeks	4 weeks
Phenolics (µmol/g olive oil)												
Hydroxytyrosol	0.0036*	0.0033	0.0033	0.0028	0.0033	0.00273	0.0045	0.0031	0.0026	0.0034	0.0030	0.0021
Tyrosol	0.0052	0.0054	0.0061	0.0046	0.0064	0.0074	0.0065	0.0073	0.0073	0.0079	0.0044	0.0032
Vanillic acid	0.0027	0.0016	0.0018	0.0008	0.0026	0.0023	0.0017	0.0015	0.0014	0.0008	0.0003	0.0002
<i>p</i> -coumaric acid	0.0029	0.0025	0.0013	0.0007	0.0034	0.0027	0.0018	0.0016	0.0013	0.0021	0.0014	0.0065
Hydroxytyrosol Ac.	0.0902	0.0529	0.1230	0.0943	0.0585	0.0317	0.0722	0.0972	0.0652	0.1177	0.2728	0.0154
3,4-DHPEA-EDA	1.0951	1.1381	0.8004	0.7949	0.7798	0.5195	0.8081	0.7470	0.6126	0.5112	0.4185	0.0311
OA-isomers	0.0400	0.0252	0.0038	0.0058	0.0270	0.0255	0.0365	0.0272	0.0165	0.0050	0.0051	0.0012
<i>p</i> -HPEA-EDA	0.4527	0.5057	0.3461	0.3912	0.3931	0.3157	0.3941	0.3674	0.3360	0.2771	0.2480	0.1017
Pinoresinol	0.0077	0.0084	0.0083	0.0084	0.0075	0.0036	0.0074	0.0084	0.0069	0.0075	0.0047	0.0020
Cinnamic acid	0.0012	0.0010	0.0009	0.0011	0.0004	0.0007	0.0011	0.0067	0.0006	0.0005	0.0007	0.0003
Acetoxipinoresinol	0.0427	0.0436	0.0475	0.0420	0.0495	0.0478	0.0577	0.0551	0.0462	0.0452	0.0269	0.0068
3,4-DHPEA-EA	0.0695	0.0711	0.0565	0.0671	0.0563	0.0476	0.0632	0.0820	0.0487	0.0449	0.0386	0.0003
<i>p</i> -HPEA-EA	0	0	0	0	0	0	0	0	0	0	0	0
Luteolin	0.0097	0.0103	0.0094	0.0140	0.0093	0.0094	0.0106	0.0123	0.0096	0.0131	0.0083	0.0023
Apigenin	0.0025	0.0031	0.0029	0.0049	0.0027	0.0026	0.0031	0.0038	0.0027	0.0032	0.0023	0.0007
Total phenolics**	1.8264	1.8726b	1.4119b	1.4332b	1.4005b	1.0197a	1.4691b	1.4153b	1.8264	1.1581a	1.0403a	1.0355a
Total <i>o</i>-phenolics	1.3084	1.3011b	0.9965b	0.9792b	0.9345b	0.6366a	0.9952b	0.9690b	1.3084	0.7554a	0.6955a	0.7466a
Secoiridoids	1.6575	1.7402b	1.2070b	1.2592b	1.2563b	0.9084a	1.3020b	1.2237b	1.6575	1.0139a	0.8384a	0.7103a

3 *) Average coefficient of variance was lower than 9%. (**) Total values with different letters within each week are significantly different ($P \leq 0.05$).

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Table 3. Oxidative stability index (expressed as hours) of oil samples extracted from fresh harvested Arbequina fruits and fruits stored at different temperatures (4 and 20°C) during 4 weeks.

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	Weeks				
	0 (24 WAF)	1	2	3	4
Fresh harvested fruits	66.7±1.5	-	64.6±2.3	57.5±0.3	56.1±0.7
Stored fruits at 4°C	66.7±1.5	57.9±0.3	52.9±0.6	57.5±0.3	53.4±0.6
Stored fruits at 20°C	66.7±1.5	54.2±0.3	50.4±0.6	48.2±0.4	26.2±0.3

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1 Table 4. Concentration of Volatile compounds in VOOs obtained from fresh harvested Arbequina fruits and fruits stored at different temperatures (4 and 20°C) during 4
2 weeks.

Volatile class (ng/g olive oil)	<u>On-tree ripening</u>				<u>Storage at 4°C</u>				<u>Storage at 20°C</u>			
	0 (24 WAF)	2 weeks	3 weeks	4 weeks	1 week	2 weeks	3 weeks	4 weeks	1 week	2 weeks	3 weeks	4 weeks
C6/LnA aldehydes	21553±1422	25322±424	27208±222	26247±11	23813±1225	27687±195	23399±1298	23212±192	15865±441	15970±255	11557±160	9418±262
C6/LnA alcohols	723±79	1441±202	1701±42	1372±178	1228±140	1052±7	922±83	941±55	887±118	708±16	1284±247	874±12
Σ C6/LnA	22277±1342	26763±221	28910±179	27619±190	25042±1365	28739±187	24322±1381	24153±248	16752±323	16678±272	12842±408	10292±275
C6/LA aldehydes	353±121	499±132	629±24	903±13	561±149	807±36	494±28	682±10	714±11	987±56	763±38	1709±42
C6/LA alcohols	14.34±2	218±274	231±10	51±2	46±44	20±0	21±6	30±2	14±1	19±1	167±32	268±5
Σ C6/LA	367±124	717±407	860±34	954±10	608±104	827±37	515±22	713±13	729±10	1007±57	931±71	1978±36
C5/LnA carbonyls	497±54	584±48	582±49	448±15	351±32	259±59	246±16	259±20	412±97	414±24	334±48	214±5
C5/LnA alcohols	745±2	1166±288	1276±178	825±26	847±232	494±1	357±4	401±49	717±102	614±84	492±26	435±14
Σ C5/LnA	1242±51	1751±336	1859±228	1274±10	1199±264	754±57	603±12	661±70	1130±4	1029±109	826±74	650±20
C5/LA carbonyls	32±13	109±6	132±18	35±2	47±19	32±3	37±6	48±9	60±25	72±49	41±34	26±1
C5/LA alcohols	9±3	92±81	157±8	25±17	42±40	32±8	37±34	24±7	28±16	59±35	42±23	7±1
Σ C5/LA	42±10	201±87	289±10	60±15	89±58	65±5.	75±40	72±1	89±9	131±13	84±11	33±2
LOX esters	83±42	422±250	338±35	137±5	246±124	221±186	80±29	99±33	104±8	268±13	161±106	87±43
Non-LOX esters	70±28	339±12	425±1	183±4	50±11	176±103	144±10	133±56	176±58	329±17	197±68	159±35
Σ Esters	154±14	761±263	764±34	320±9	296±113	397±290	224±39.	232±89	280±49	597±31	358±174	246±7
Σ Volatile AA	120±43	236±54.6	340±57	158±77	132±36	79±8	77±14	72±37	98±8	183±17	113.85±26.07	131±81
Σ Terpenes	8989±353	8241±698	8190±269	5698±621	6636±205	4022±563	3814±252	3794±199	5293±298	6017±755	5281±465	4153±351
Total volatiles*	33194	38674b	41215c	36087c	34005b	34888c	29633b	29701b	24375a	25646a	20439a	17487a

3 * Total values with different letters within each week are significantly different ($P \leq 0.05$).

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