

1 **Fermentation of texturized pea protein in combination with proteases for**
2 **aroma development in meat analogues**

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22 **ABSTRACT**

23 The potential use of texturized pea protein in meat analogues was investigated by comparing
24 the effects of fermentation on pea and myofibrillar pork proteins in a model system including
25 additives, microbial starters and proteases. Model fermentation was controlled for 15 days by
26 pH decrease, microbial counts and free amino acids increase. Besides, volatile production and
27 sensory properties were evaluated at the end of fermentation. Protein type affected free amino
28 acid generation and volatile profile. Models supplemented with proteases showed an increase
29 in amino acid derived compounds (branched aldehydes and alcohols) and fruity odor notes.
30 During fermentation, protease addition significantly reduced the production of linear aldehydes
31 (pentanal, hexanal and octanal) in vegetal models, while pyrazine compounds were not affected.
32 This changes in the volatile profile reduced the legume-beany odor, although increased the
33 perception of toasted cereal-like notes generated by the texturization process.

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35 **KEYWORDS:** meat analogue, plant based, pea protein isolate, fermentation, off-flavor

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37 1. INTRODUCTION

38 Flavor is an essential issue in the development of meat- and processed-meat analogues¹.
39 Changes in the ingredients or processing affect greatly the flavor of these products and,
40 consequently, consumer preference, which is highly influenced by cultural habits and
41 experience². The main components in the formulation of meat analogues are plant protein-rich
42 ingredients, such as plant protein isolates and soy or wheat concentrates, as well as legumes
43 like pea and lupine, rice or potato³. Peas belong to the *Fabaceae* family and are popular for
44 their low cost, and high protein content⁴. The protein ingredients are the most important
45 component for differentiation of meat analogues, because of their ability to provide meat-like
46 structure and nutritional health⁵.

47 Numerous studies have demonstrated that the inclusion of plant protein in food products
48 formulation is the origin of undesirable volatile flavors⁶. The removal or covering of the plant
49 protein off-flavors, as well as those originated from flavor interactions with the plant proteins,
50 are key in the study of meat analogues flavor^{7,8}. Indeed, many studies have focused on the flavor
51 of cooked meat analogues, and the effect of the addition of flavorings and aroma precursors
52 during their manufacture⁹. In case of fermented and dry-cured products, aroma seems to be the
53 biggest challenge for the formulation of meat analogues.

54 The fermentation of plant proteins has the potential to produce pleasant aroma compounds of
55 interest in the design of fermented dry sausage analogues. The fermentation of plant-based
56 foods to generate different flavor profiles is widely known in Asia since ancient times. Several
57 of these fermented foods have been described as having a taste profile with umami
58 characteristics. Moreover, many of these foods have been characterized in terms of their aroma
59 profile and taste, as in case of Chinese fermented soybean curd or white sufu and the Japanese
60 fermented soybean paste miso¹⁰.

61 The protein sources most widely used in meat analogues are soy and pea isolates ¹¹. However
62 the application of a fermentative process of these protein sources for production of fermented
63 meat analogues has been scarcely investigated. Recent studies have proposed the fermentation
64 of pea protein with a combination of lactic acid bacteria (LAB) and yeast starter cultures to
65 reduce the off-flavors produced by the presence of hexanal and other oxidation products like 2-
66 pentylfuran, (E, E)-2,4-decadienal, hexanal, nonanal, (E, E)-2,4-nonadienal, octanal, (E)-2-
67 nonenal and (E)-2-octenal^{12,13}. Furthermore, the application of microbial consortia (LAB and
68 molds) in the fermentative process in combination with enzymatic hydrolysis has been
69 proposed as a way to improve the taste of soy protein isolates ¹⁴. This improvement was
70 observed in both the taste and functionality (emulsifying and foaming properties) of the protein
71 isolate and, in addition, the fermented protein isolates showed a reduced beany flavor.

72 In traditional fermented dry-cured products, flavor generation depends on precursors produced
73 during the fermentative process and the activity of microbial starters selected to ferment animal
74 proteins. The ability of these starters to generate precursors and aromas has not been tested on
75 plant proteins. Moreover, their activity may be hindered by their ability to hydrolyze vegetal
76 proteins, which could be improved applying exogenous proteolytic enzymes. In summary, the
77 aim of this study was to determine the functionality of microbial starters, combined with
78 proteases, in the fermentation of texturized pea proteins. The fermentation process and its
79 outcome were compared with that of an identical model system formulated with extracted pork
80 meat proteins undergoing the same treatment. The results of this study could provide
81 information about the potential use of texturized pea proteins in dry-cured meat analogues
82 manufacturing.

83 **2. MATERIAL AND METHODS**

84 **2.1 Isolation of myofibrillar proteins from pork meat.** The isolation of myofibrillar
85 proteins was performed following the method of Molina & Toldrá¹⁵ using the muscle
86 *Longissimus thoracis et lumborum*. The process consisted on the homogenization of the meat
87 with 0.03 N phosphate buffer pH 7.4 using a stomacher (IUL masticator, Barcelona, Spain)
88 followed by a centrifugation process at 10.000 g during 20 min. The pellet was collected and
89 the process was repeated three times for the removal of sarcoplasmic proteins. The final pellet
90 was resuspended in a solution containing 0.1 N buffer phosphate, 0.7 M potassium iodide and
91 0.02 % sodium azide at pH 7.4, and then filtered through glass wool and diluted again in a
92 solution containing 0.1 N buffer phosphate and 0.02% sodium azide at pH 7.4. Finally, the
93 suspension was removed by centrifugation and the pellet containing myofibrillar proteins was
94 collected and used in the formulation of the models.

95 **2.2. Preparation of vegetal and animal fermentation models.** The fermentation model
96 systems included animal or vegetal proteins together with common additives used in the
97 fermentation of meat products (salt, glucose and nitrifying agents), previously dissolved in
98 distilled water and filter (0.22 µm) sterilized (Grynia, Labbox, Barcelona, Spain), and
99 microbial starters. A commercial protease (Flavourzyme >500 U/g, Sigma, Merck, Germany)
100 was applied as flavouring enzyme in some of the models as described in Table S1. Two
101 models, animal (A) and vegetal (V), were prepared. The animal model (A) was formulated
102 with the extracted myofibrillar proteins (8 % w/v), while the vegetal model (V) was prepared
103 with texturized pea protein (Manufacturas Ceylan, Valencia, Spain) (8 % w/v) previously
104 homogenized in a blender. Two additional models containing the flavouring enzyme (0.02 %)
105 were prepared from the animal (AE) and vegetal (VE) models. The ingredients (3 % NaCl, 2
106 % Glucose, 0.015 % NaNO₂, 0.015 % KNO₃) of the four models were homogenized in a
107 blender and inoculated with the commercial starter TRADI-302 (0.0125 %), containing,
108 *Lactobacillus sakei*, *Staphylococcus carnosus*, and *Staphylococcus xylosum*, (Chr. Hansen,

109 Hoersholm, Denmark) and yeast *D. hansenii* (L5, 10^6 cells/ml)¹⁶ as indicated in Table S1. The
110 fermentation experiments of the four models (A, AE, V and VE) were prepared in triplicate.
111 The models were incubated at 25 °C in a heater (Incuberm Digit, Raypa, Barcelona, Spain)
112 and samples were taken at days 0, 3, 8 and 15. The evolution of the fermentation was followed
113 by the decrease in pH, microbial counts, and free amino acids production. The sample for
114 microbial analysis was homogenized with saline solution in a sampling bag with a side filter
115 (Scharlab, Barcelona, Spain) using a Pulsifier II (Microgen Bioproducts, Camberley, UK) (3
116 pulses of 30 s). The sample for physic-chemical analysis was centrifuged at 10.000 g for 20
117 min and the supernatant filtered through a 0.2 µm filter (Minisart NML, Sartorius, Göttingen,
118 Germany) and used for pH measurement with a portable pH-meter (HI 99163, Hanna
119 Instruments Inc., Hoonsocket, USA). The supernatant was further used for free amino acids
120 and volatile compound analysis. The samples for volatiles analysis were acidified using 200
121 µL of trichloroacetic acid to inactivate protease activity, then neutralized with 1 N NaOH and
122 kept at -20 °C until further analysis. Additionally, at the end of the fermentation (15 days) the
123 remaining fermented model was kept for sensory analysis.

124 **2.3. Microbiological analysis.** The analysis was performed as described by Belloch et al.¹⁷, .
125 In summary, the homogenized samples were used to prepare decimal dilutions which were
126 subsequently spread in triplicates on the appropriate media plates for microbial counts as
127 follows: total mesophilic bacteria (TMB) on Plate Count Agar (PCA) (Condalab, Madrid,
128 Spain) at 30 °C for 2 days, LAB on MRS Agar (Scharlau, Barcelona, Spain) at 30 °C for 2
129 days, Gram positive cocci (GC+) on Mannitol Salt Agar (MSA) (Scharlau, Barcelona, Spain)
130 at 30 °C for 2 days, enterobacteria (E) on Violet Red Bile Glucose Agar (VRBGA) at 37 °C
131 for 24 h, and yeasts and moulds (YM) on Rose Bengal Agar Chloramphenicol (Scharlau,
132 Barcelona, Spain) at 30 °C for 3 days. Results from the microbial counts were expressed as
133 log CFU/g.

134 **2.4. Volatile compounds analysis.** Volatile compounds present in the headspace of the liquid
135 sample were analyzed as described in Perea-Sanz et al.,¹⁶ by extracting the compounds with a
136 solid phase microextraction (SPME) device (Supelco, Bellefonte, PA, USA). Samples
137 consisting of 4 mL of supernatant previously defrosted, were placed in a headspace vial (20
138 ml, Gerstel, Germany) containing 1.88 g NaCl and equilibrated at 37°C during 30 min. Then,
139 the volatile compounds were extracted for 1 h at 37°C under shaking at 250 rpm using the
140 SPME fibre (85 µm, CAR/PDMS StableFlex fibre, 1cm). The extracted volatile compounds
141 were analyzed in an Agilent HP 7890 series II GC with an HP 5975C mass selective detector
142 (Hewlett-Packard Palo Alto, CA, USA) and a Gerstel MPS2 multipurpose sampler (Gerstel,
143 Germany). The fiber was desorbed in the GC injection port at 240°C for 5 min in splitless
144 mode. Volatile compounds were separated using a DB-624 capillary column (30 m x 0,25
145 mm, 1,40 µm Agilent Technologies, USA) and analyzed using the MS detector in SCAN
146 mode. Volatile compounds were identified by comparison with mass spectra from the library
147 database (Nist¹⁷), by linear retention indices calculated using the series of n-alkanes C8-C22
148 (Aldrich, Merck, Germany)¹⁸, and by comparison with authentic standards (Table S5).
149 Quantification was performed in SCAN mode using either total or extracted ion area (TIC or
150 EIC) on an arbitrary scale. Each model supernatant was analyzed in triplicate and the results
151 were expressed as abundance units (AU) 10⁻⁵ per g of protein in the media, and the differences
152 in volatiles produced depending on the protein source, animal or vegetal, were determined.

153 **2.5. Free amino acid analysis.** The abundance of free amino acids released from the
154 proteolytic activity in the liquid sample was measured following the methodology described
155 by Aristoy and Toldrá¹⁹, which includes the deproteinization and derivatization of the sample.
156 Norleucine (10 mM in 0.01 M HCl) was used as internal standard. The separation of free
157 amino acids was done by reversed-phase HPLC chromatography in an Agilent Series 1100
158 equipment (Agilent, CA, USA) equipped with a Waters Nova Pack® C18 column (3.9 × 300

159 mm; Waters Corporation, MA, USA) at 52°C using a photodiode array detector²⁰. The
160 separated amino acids were detected at 254 nm. Each medium supernatant was analyzed in
161 triplicate. Identification of amino acids was achieved by comparison against a solution of
162 mixed standards (Sigma, Merck, Germany), and quantification was based on the calculated
163 response factors. They were calculated using five amino acid standard levels in the presence
164 of the added internal standard (norleucine). The final results were expressed as mg of free
165 amino acid per g of protein in the model, and the differences in released free amino acids
166 depending on the protein source, animal or vegetal, were determined.

167 **2.6. Sensory analysis.** The sensory analysis was done from model samples at the end of the
168 fermentation process (15 days) using the detection frequency method²¹ to reveal the aroma
169 impact of volatile compounds in the models. Odors were evaluated by six trained panellists,
170 4 females and 2 males with an average of 40 years old, who evaluated the odors by smelling
171 the model samples as reported in Belloch et al.²² The aroma descriptors were recorded and the
172 results were expressed as the number of times a descriptor was detected by the panellists^{21,23}.

173 **2.7. Statistical analysis.** Data were analyzed using the Generalised Linear Model (GML)
174 procedure in the statistical software XLSTAT 2018 (Addinsoft, Barcelona, Spain). Data
175 analysis, using the linear mixed model, included two factors: protein source (vegetal or
176 animal) and enzyme as fixed effects, and replicates as random effects. Differences between
177 sample means were analysed according to Tukey's test, when a significant effect of the
178 treatment group was detected ($P < 0.05$). Principal component analysis (PCA) was performed
179 to evaluate the relationships between variables (pH, microbial counts, free amino acids and
180 volatiles) and models at the four sampling times. Heatmaps plotted using XLSTAT 2018 were
181 based on the relative abundance of identified volatile compounds in the models at the four
182 sampling times.

183

184 3. RESULTS

185 The evolution of the fermentation, free amino acids content, changes in the volatile profile
186 and sensory analysis of the fermentative process in the vegetal and animal models,
187 supplemented or not with a protease, was monitored. The analyses were done at the beginning
188 of the fermentation (day 0), at the middle (day 3 and 8), and at the end of the process (day 15).

189 **3.1. Evolution of the fermentation: pH and microbial counts.** The evolution of pH and
190 microbial counts in the fermented models is shown in Figure 1 and Tables S2 and S3. Values
191 of pH decreased significantly during the fermentation of the animal (Fig 1A and 1B) and
192 vegetal models (Fig 1C and 1D). Moreover, the addition of the proteolytic enzyme accelerated
193 significantly the pH decrease (Table S2). Fermentation time increased significantly microbial
194 counts, usually at days 3 or 8 of fermentation. Microbial counts were lower in the animal (Fig
195 1A) than in the vegetal model (Fig 1C). In contrast, the addition of the protease decreased
196 bacterial counts in both models (Fig 1B and 1D; Table S3). This decrease was significant in
197 case of GC⁺ counts in the animal model (AE), while in the vegetal model (VE) the effect was
198 observed in both GC⁺ and in LAB counts. In contrast, the differences in PCA and YM counts
199 between models with or without enzyme were not significant. No enterobacteria were detected
200 along the fermentation process.

201 **3.2. Determination of free amino acids in the models along the fermentation.** The total
202 content of free amino acids along the fermentation is reported in Figure 2, while the values
203 for individual amino acids are in Tables S4 and S5. In general, free amino acid content
204 significantly increased in all models along the fermentation time (Fig 2), except for few amino
205 acids (glu, his, thr, met, phe, and trp) in the animal models (Table S4). The addition of enzyme
206 also increased significantly the amino acid content in both, vegetal and animal, models (Tables
207 S4 and S5). This increase was about 100 times higher in the vegetal than in the animal models
208 (Fig 2). In the animal model the addition of enzyme significantly increased the production of

209 amino acids ala, pro, val, ile, leu, orn and lys, but the amount produced was only 2 to 3 times
210 higher than at the initial time. In contrast, the amount of free amino acids produced by enzyme
211 addition in the vegetal model was higher, around 8 fold in case of try, ala, thr and glu and 12
212 fold in case of phe, ile, leu and val.

213 **3.3. Differences in the volatiles profile between the models along the fermentation.** The
214 volatile organic compounds (VOCs) profile was very different in the vegetal and animal
215 models (Fig 3 and 4; Tables S6-S8). Sixty-two VOCs were identified in the model's
216 headspace, but the chemical structure was confirmed in only 54 of them (Table S6). Eight
217 VOCs, including 3 pyrazines, were tentatively identified by mass spectrometry. The main
218 difference between the vegetal and animal models was the presence of pyrazines in the vegetal
219 models, which were absent in the animal models. Also, four additional compounds, 3-methyl-
220 3-buten-1-ol, 3-methyl-1-butanol acetate, ethylbenzene and 3-pentanone were only detected
221 in the vegetal models (Table S6).

222 The evolution of the VOCs profile classified by chemical group (Fig 3) along the fermentation
223 of all models indicates that alcohols constituted the most abundant group, followed by
224 aldehydes and ketones. The evolution of the fermentation can be recognized by the significant
225 increase of alcohols with time in all models (Fig. 3), being this increase higher in the vegetal
226 (Fig 3C) than in the animal model (Fig 3A). Besides, the addition of enzyme impacted
227 differently the vegetal and animal models. In the vegetal models (Table S8), alcohols such as
228 ethanol and 2-ethyl-1-hexanol were the most abundant compounds found in the V model,
229 while in the VE model increased several methyl branched alcohols (2- and 3-methyl-1-
230 butanol) and phenylethyl alcohol. Similarly, branched aldehydes 2-methyl and 3-methyl
231 butanal, were in higher abundance in the VE than in the V model. The abundance of ketone
232 compounds generally increased in the VE model respect to the V model (Figure 3D). Few
233 changes were observed in pyrazines abundance along the fermentation and the addition of

234 enzyme did not produce a clear trend. In contrast, the addition of enzyme in the animal model
235 did not cause many significant differences in the volatile profile (Table S7). The main
236 differences were the increase in branched aldehydes (2-methyl and 3-methyl butanal) in the
237 AE model, as happened in the vegetal model VE (Figure 3A and 3B, Table S7).

238 A more comprehensive comparison of the compounds constituting the volatile profile of the
239 models was plotted in a heatmap with hierarchical clustering (Fig 4). The dendrogram at the
240 top shows that the models are divided in two groups by the type of protein employed, animal
241 (right) vs. vegetal (left). Moreover, differences within each group can also be observed. In the
242 vegetal model, the effect of the enzyme had larger impact than the fermentation time, as
243 samples VE3, VE8 and VE15 appear separated from the rest of the samples. In the animal
244 model the main impact was caused by the fermentation time as samples A15, AE8, AE15 were
245 separated (left) from the rest. The dendrogram on the left shows which compounds support
246 the differences between and within the models. The presence of pyrazines and few ketones
247 constitute the core of cluster E, which separates between the vegetal and animal models. The
248 remaining clusters of compounds account for the main differences within the models. Cluster
249 D composed by several alcohols and ketones separates samples A8, A15 and AE15 from the
250 other samples in the animal model, as well as the VE from the V samples in the vegetal model.
251 The separation of A8, A15 and AE15 samples is also supported by compounds in cluster C,
252 composed by several ketones, branched aldehydes and alcohols. Finally, cluster B constituted
253 mainly by linear aldehydes separates initial samples 0 and 3 from later samples 8 and 15 in
254 the V model.

255 A further analysis of the data was applied to study the effect of time and enzyme addition on
256 the fermentation of pea protein vs. pork myofibrillar protein, and the results were plotted in a
257 principal components analysis (Fig 5). The PCA explained 62.8 % of the variability. The first
258 factor (42.5%) separated the animal samples from the vegetal samples, whereas the second

259 factor (20.3 %) separated the samples containing enzyme by fermentation time. Microbial
260 counts and free amino acids were clearly related to the V model samples. Moreover, it is worth
261 to note that all free amino acids are closely related to the VE model samples. Regarding the
262 volatile compounds, pyrazines seem to be the main variable separating V from A models,
263 while alcohols and aldehydes separate VE from the V model.

264 **3.4. Sensory properties of the fermented models.** The odor profile of the models was
265 evaluated at the end of the fermentation time (15 days) (Figure 6), and significant ($P < 0.05$)
266 differences were found among all models. The animal models were defined by descriptors
267 fruity, sour, and cooked vegetal, while the vegetal models were described by toasted cereal,
268 legume, cocoa and cheesy odor notes in addition to fruity and sour. The addition of enzyme
269 had a significant impact on the odor profile of the vegetal model. The legume and cocoa notes
270 in the V model were replaced by toasted cereal, cheesy and fruity notes in the VE model.
271 Furthermore, the addition of enzyme significantly decreased the sour odors. In case of the
272 animal models, the addition of enzyme (AE model) only increased the fruity and cooked
273 vegetal odors already present in the A model.

274

275 **4. DISCUSSION**

276 In order to develop attractive plant-based fermented meat analogs, we have evaluated the
277 potential of pea protein isolates fermentation in combination with enzymatic proteolysis, to
278 improve flavor. Moreover, we have compared these findings with those obtained applying a
279 similar fermentative process using extracted meat proteins. The results from our study show
280 (Fig 1) that the fermentation process progressed in a similar way using texturized pea protein
281 or myofibrillar pork proteins, although the presence of the enzyme (protease) accelerated the
282 process. This may be due to the increase in free amino acids produced by the proteolytic

283 activity (Fig 2), which would increase the metabolic activity of LAB and, consequently, the
284 decrease in pH. Enzyme addition (VE and AE models) caused a slight negative effect on LAB
285 counts in both models; however, this decrease did not seem to have a large impact neither in
286 the pH decrease nor in the fermentation progress. On the contrary, yeast growth was not
287 affected by the presence of enzyme in the models, which could have important consequences
288 for aroma generation²⁴. The most important difference between the models, animal (A) and
289 vegetal (V), was the higher microbial counts in the vegetal model, which might indicate that
290 the texturized pea protein is more accessible to the microorganisms, thus facilitating
291 proteolysis activity. This agrees with the slight increase in free amino acid abundance in the V
292 model respect to the A model (Figure 2, Tables S4 and S5). Previous studies have demonstrated
293 that hydrolysis of myofibrillar proteins using *Staphylococcus carnosus* exoproteases highly
294 increases the concentration of free amino acids Glu and Gly, and moderately in case of His,
295 Thr, Val, Leu, Phe and Lys^{25,26}. In contrast, the addition of the starter culture (A model), which
296 also includes *S. carnosus*, did not produce a significant increase of protein hydrolysis, and only
297 the addition of the commercial protease (AE model) produced a significant increase of the
298 proteolytic activity against myofibrillar pork proteins. In agreement with previous studies,
299 some of the most abundant free amino acids produced in the AE model (animal model with
300 enzyme) (Table S4) were the same as those produced by hydrolysis of myofibrillar proteins
301 using *S. carnosus* exoproteases²⁶. The texturized pea protein (VE model) underwent a similar
302 proteolysis and fermentative process than the animal model (AE) but, in comparison, the free
303 amino acid yield in the former was significantly higher than in the latter. This result would
304 indicate that the pea protein is more accessible to enzymatic activity than the myofibrillar pork
305 proteins. Moreover, the large proteolysis yield of the vegetal model (VE) would suggest that
306 not only exopeptidase activities are present. Furthermore, the amino acid composition of plant
307 proteins can be very different from the one found in meat proteins²⁶ and, in case of pea proteins,

308 the most abundant amino acids are Glu, Arg, Leu, Lys, whereas the less abundant are Met and
309 Cys, in agreement with previous studies²⁷.

310 The generation of free amino acids is closely related to the formation of volatile compounds
311 affecting aroma. For example, in fermented meat products the generation of sulfur amino acids
312 promotes formation of sulfur compounds which contribute to savory properties of the meat
313 product²⁸. An important result from our study was that the fermented models, animal and
314 vegetal, generated different volatile profiles which were derived from the different amino acid
315 composition of the proteins present in the models. The volatile profile of hydrolyzed
316 myofibrillar proteins using *S. carnosus* exoproteases²⁶ has been reported to include VOCs such
317 as linear aldehydes, alcohols and ester compounds after only 2 h of hydrolysis. Among these
318 compounds, two were found derived from phenylalanine, benzenacetaldehyde and phenylethyl
319 alcohol. The generation of these two compounds was also observed in the animal models (A
320 and AE) (Table S7). However, benzeneacetaldehyde was absent or scarcely produced in the
321 vegetal models (V and VE), whereas phenylethyl alcohol was found abundantly in the VE
322 model (Table S8). The main differences between a purely enzymatic hydrolysis²⁶ and our study
323 are the addition of microbial starters, and the longer incubation times (up to 15 days). These
324 differences were responsible in the VE model for the generation of compounds derived from
325 phenylalanine (benzenacetaldehyde and phenylethyl alcohol), as well as those derived from
326 isoleucine and leucine, like branched aldehydes (2-methyl- and 3-methyl butanal) and their
327 respective alcohols (Figure 4).

328 The effect of the long fermentation time, applied in our models, on the volatile profile is not
329 easy to analyze since from the beginning of fermentation (day 0) both models, animal and
330 vegetal, had a very dissimilar volatile profile. The largest difference was the presence in the
331 pea protein models of odor-active carbonyl compounds (linear aldehydes and 2-pentyl furan,

332 Figure 4) responsible for the beany flavor²⁹, and pyrazine compounds derived from the
333 degradation of fatty acids and amino acids, respectively³⁰. The presence of different aldehydes,
334 ketones and pyrazines responsible for the beany flavor in the vegetal models largely depends
335 on the initial pea protein composition³¹ and texturization process²⁹, but also on the volatile
336 extraction technique employed during analysis, which affects the VOCs profile qualitative and
337 quantitatively³². The large influence of these factors on VOCs profile limits comparisons of
338 results between studies using the same extraction conditions. Nevertheless, odor compounds
339 responsible for the pea protein isolate flavor such as hexanal, benzaldehyde, heptanal and 1-
340 octen-3-ol, derived from lipid oxidation processes^{33,34}, were also present in the vegetal models
341 (Figure 4). Regarding the pyrazines, the ones present in vegetal models may be derived from
342 Maillard reactions during the texturization process as 2,5-dimethyl-pyrazine³⁰. Other pyrazines
343 are inherent constituents of the pea protein as methoxypyrazines³⁵, while 2-isobutyl-3-
344 hydroxypyrazine vary during the isolation process of pea proteins and affect the aroma
345 profile.³⁴

346 The contribution of microbial starters to food aroma has been widely explored³⁶. Moreover,
347 their application in fermented meat products for their ability to transform free amino acids,
348 generated by the endogenous proteolytic system, into volatile compounds has been amply
349 proven³⁷. In case of vegetal proteins, most efforts have focused on the removal of beany off
350 flavors, especially on the transformation of aldehydes and ketones into alcohols or carboxylic
351 acids by the activity of alcohol dehydrogenases (ADH) and aldehyde dehydrogenases (ALDH)
352 present in microorganisms³¹. Among the most studied microorganisms for this application are
353 LAB (*Lactobacillus acidophilus*, *Limosilactobacillus fermentum*, *Lactiplantibacillus*
354 *plantarum* and *Streptococcus thermophilus*) and *Saccharomyces cerevisiae*. In case of the
355 animal and vegetal models used in our study, both the formulation of the models and the

356 microbial starter were selected to imitate a fermented meat product, therefore bacterial
357 (TRADI-302, Chr. Hansen, Denmark) and fungal starters¹⁶ used for that purpose were applied.

358 Since the beginning of the fermentation, ketones and aldehydes were detected in high
359 abundance in the vegetal models (V and VE) (Figure 4), as already observed in previous
360 studies¹³. Fermentation reduced aldehydes such as pentanal and nonanal in the V model, and
361 hexanal in VE model. Similar reductions of hexanal and nonanal have been attributed to *S.*
362 *cerevisiae* and *L. plantarum* fermentations of pea protein for 6 and 8 h, respectively¹³. However,
363 fermentation was not able to reduce ketones and pyrazines, specially 2,5-dimethylpyrazine,
364 which contributes to the nutty and cereal-like odor in fermented pea³⁰. The alcohols increase
365 observed during fermentation of the vegetal models is in accordance with previous studies¹³.
366 The presence in the VE model of methyl branched alcohols (2-methyl and 3-methyl butanol)
367 (Figure 4, Table S8) could be a direct consequence of the large amounts of free amino acids,
368 which made possible the generation of methyl branched alcohols by microbial activity.

369 The impact of the VOCs on the aroma of the fermented models can not be solely determined
370 by the calculation of the odor activity values (OAV). Besides, the extraction method employed
371 (SPME) only allows the comparison of the volatile profile among models and fermentation
372 time, and it requires the application of accurate quantitation methodologies³⁸. These limitations
373 were overcome applying a sensory analysis of the models. This analysis revealed that fruitiness
374 and cooked vegetal odors detected in the AE model could be explained by the presence of D-
375 limonene, benzene, acetaldehyde, branched aldehydes and terpinen-4-ol, respectively (Figure
376 5). In the vegetal model (VE), the reduction of the legume and cocoa odor notes, as well as the
377 increase of toasted cereal notes was related to the reduction of aldehydes^{29,31}. On the contrary,
378 pyrazines abundance was not affected by fermentation time or enzyme addition, and probably
379 increased the perception of the nutty and cereal-like odor in the vegetal models³⁰. In this regard,

380 recent studies have revealed the potential of plant hydrolysates to simulate the meaty aroma by
381 producing volatile compounds through Maillard reactions^{39, 40}. The combination of Maillard
382 reactions and protein hydrolysis, using the same enzyme applied in our study, on wheat and
383 rice⁴⁰ and soy³⁹ revealed similar nutty and toasted aroma notes. These odors were attributed to
384 alkyl pyrazines resulting from the Maillard reaction and derived from the free amino acids
385 generated thorough hydrolysis. Similarly, in our study the presence of alkyl pyrazines was
386 detected at the beginning of the fermentation, therefore their origin can be attributed mainly to
387 the texturization process of pea proteins which employed high pressure and temperatures²⁹.

388 In conclusion, the potential of the fermented vegetal models to simulate the meaty aroma
389 should be focused on the elimination not only of the beany compounds but also of the pyrazines
390 producing toasted-cereal like odors. Moreover, the generation of volatiles which could reduce
391 or mask these off-aromas in the vegetal model is largely affected by the level of proteolysis
392 and generation of free amino acids, which are used as volatile precursors by the microbial
393 starters. Finally, the whole food matrix composition and not only the proteins is a source of
394 flavor compounds, therefore the interaction mechanisms between proteins, fat, and volatile
395 compounds will affect flavor perception in plant-based foods³⁵. In summary, these models are
396 far from a real food system and the elucidation of the aroma impact of the compounds
397 generated through fermentation of plant proteins should be done through proper quantitation
398 on future developed plant-based foods.

399 **ABBREVIATIONS AND NOMENCLATURE**

400 Solid phase microextraction (SPME), total mesophilic bacteria (TMB), lactic acid bacteria
401 (LAB), Gram positive cocci (GC+), enterobacteria (E), yeasts and molds (YM), volatile
402 organic compounds (VOCs), odor activity values (OAV).

403 **DECLARATION OF COMPETING INTEREST**

404 The authors declare no competing financial interest.

405 **SUPPORTING INFORMATION.**

406 Supplementary tables are included with the composition of animal and vegetal models,
407 identification of volatile compounds, and the data of microbial counts, free amino acid content,
408 and volatile compounds content in the models.

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410 The authors are thankful to Javier Calvo for his technical assistance and to the sensory panel.

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538

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543 **FIGURE CAPTIONS**

544 **Figure 1.** Effect of fermentation time and addition of enzyme on pH and microbial counts (log
545 cfu/g) of the vegetal and animal models. The results from the animal models are in Figures A
546 (without enzyme, A model) and B (with enzyme, AE model). The results from the vegetal
547 models are in Figures C (without enzyme, V) and D (with enzyme, VE model). Symbols
548 represent pH (□), TMB (△), LAB (●), GC+ (○) and Y&M (▼). Details about the individual
549 variables and ANOVA results of the fermentation time and enzyme effects on the models are
550 reported in Tables S2 and S3.

551 **Figure 2.** Evolution of total free amino acids content (mg/g protein) in animal and vegetal
552 models. Figures: A, animal model without enzyme (A, ●) and animal model with enzyme (AE,
553 ○), and B, vegetal model without enzyme (V, ●) and vegetal model with enzyme (VE, ○).

554 **Figure 3.** Abundance (AU x 10⁵/g protein) of volatile compounds summarized by chemical
555 group detected in the headspace of the animal and vegetal models along the fermentation.
556 Figures: A (animal model without enzyme, A), B (animal model with enzyme, AE), C (vegetal
557 model without enzyme, V) and D (vegetal model with enzyme, VE). Compounds: aldehydes
558 (●), alcohols (○), esters (▼), alkanes (△), ketones (■), pyrazines (□), other (◇).

559 **Figure 4.** Heatmap representing the volatile profile of the animal and vegetal models during
560 the fermentation. Samples: animal models without (A) and with enzyme (AE) and vegetal
561 models without (V) and with enzyme (VE). Numbers in the samples represent fermentation time
562 in days. Colors in the heatmap indicate relative abundance of each volatile compound: blue,
563 relatively high abundance; red, relatively low abundance.

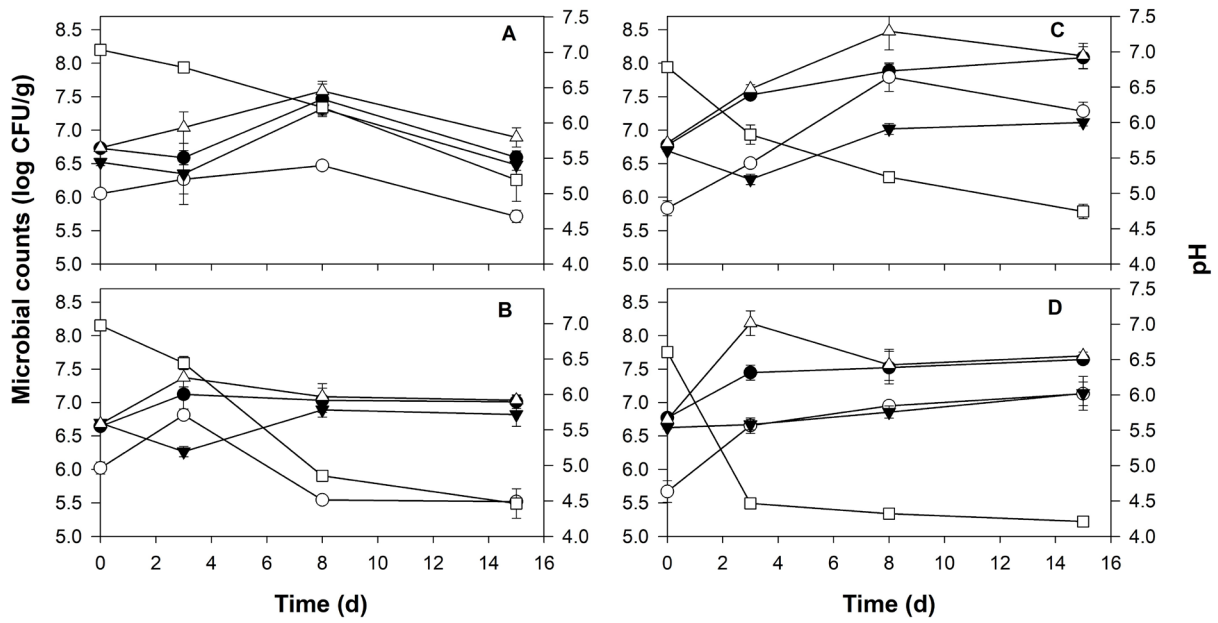
564 **Figure 5.** Principal component analysis showing the relationship among variables, microbial
565 counts, pH, free amino acids and volatile compounds, and the animal and vegetal models along
566 the fermentation. Animal models are represented by samples A (without enzyme) and AE (with

567 enzyme, whereas vegetal models are represented by samples V (without enzyme) and V (with
568 enzyme). The numbers in the models represent the fermentation time in days of the samples.

569 **Figure 6.** Odor profile of the animal and vegetal models after 15 d of incubation. Animal
570 models are represented by A (blue) and AE (orange) lines, whereas vegetal models are
571 represented by V (gray) and VE (yellow) lines.

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573 **Figure 1.**



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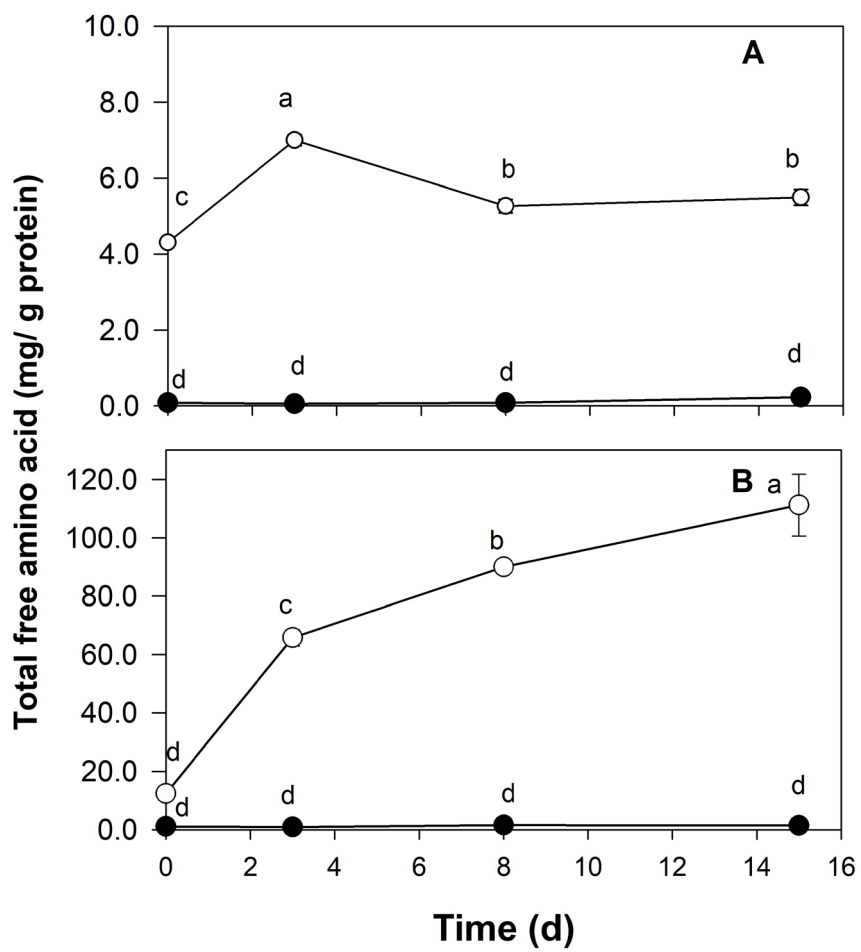
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587 **Figure 2.**



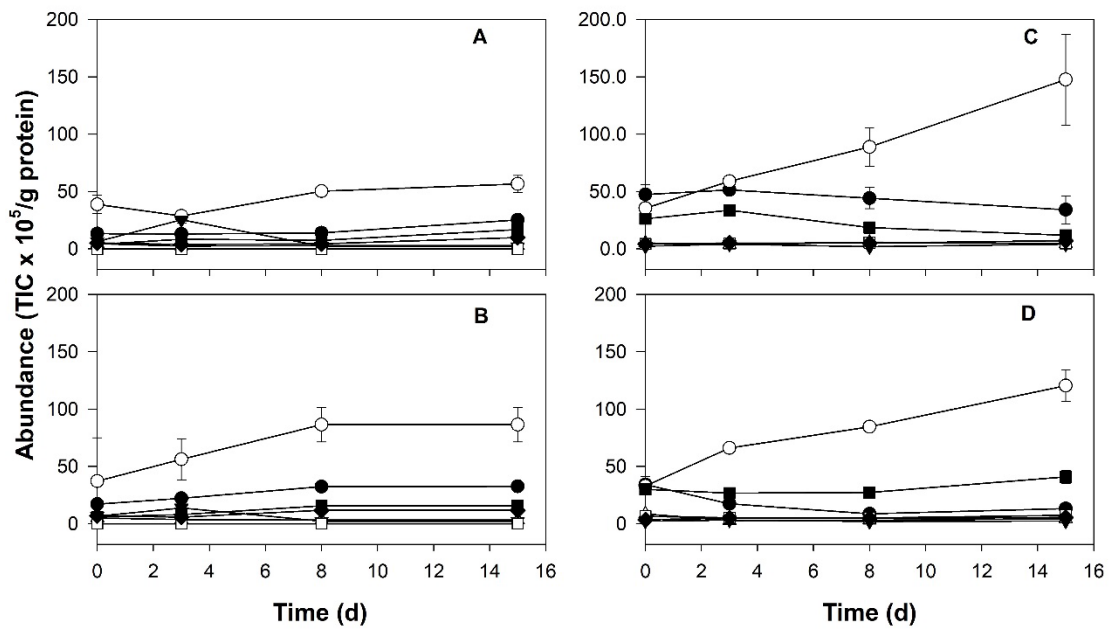
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591 **Figure 3.**

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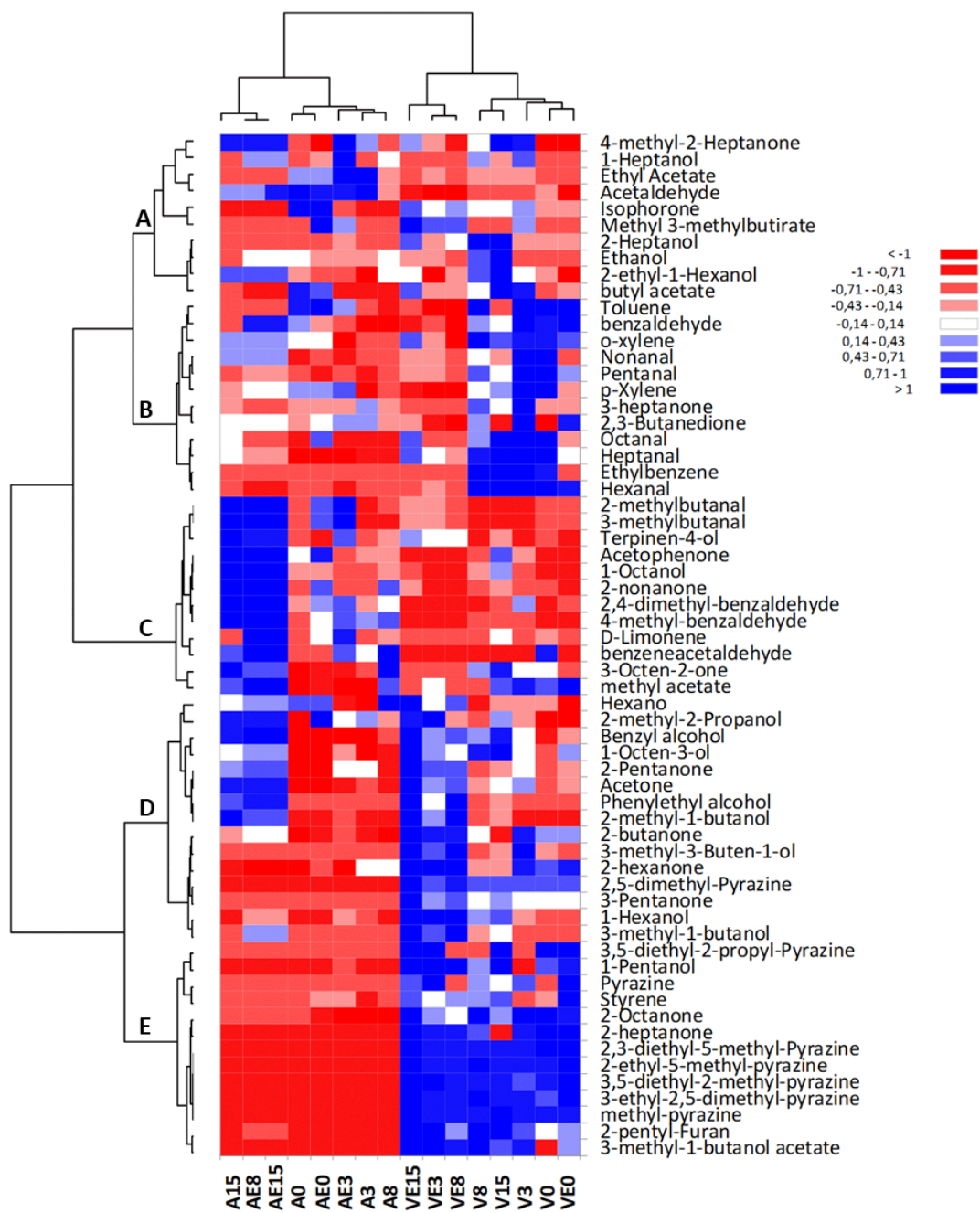
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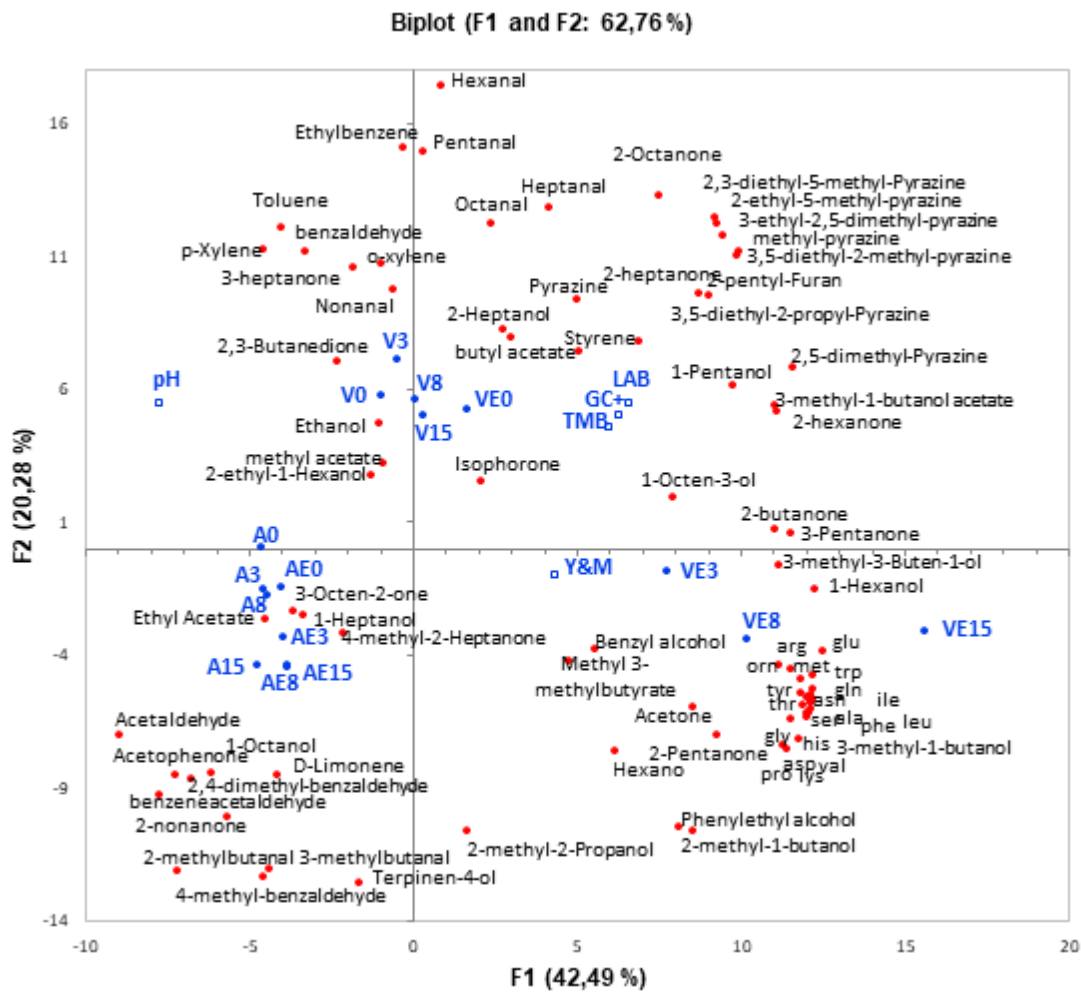
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600 **Figure 5.**



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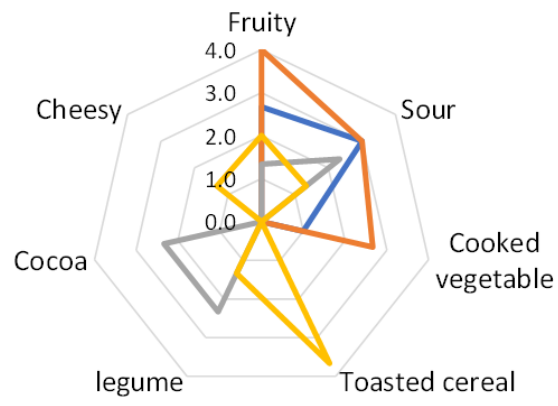
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606 **Figure 6.**

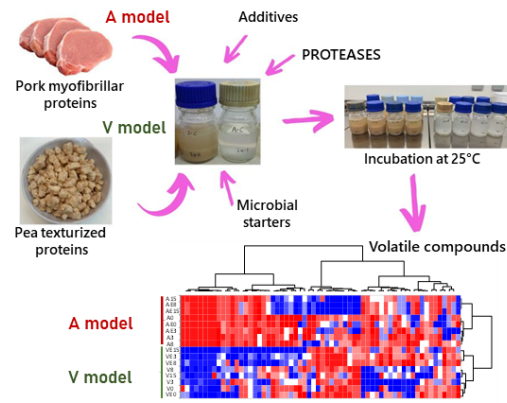


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610 **Graphic for table of contents (TOC)**



611

Supplementary Tables

Manuscript title: Fermentation of texturized pea protein in combination with proteases for aroma development in meat analogues

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Table S1. Composition of animal (myofibrillar pork protein) and vegetal (texturized pea protein) models.

Composition (g/100ml)	Models			
	Animal A	Animal + Enzyme AE	Vegetal V	Vegetal + Enzyme VE
Myofibrillar pork protein	8	8	-	-
Texturized pea protein			8	8
NaCl	3	3	3	3
Glucose	2	2	2	2
NaNO ₂	0.015	0.015	0.015	0.015
KNO ₃	0.015	0.015	0.015	0.015
Starter TRADI-302	0.0125	0.0125	0.0125	0.0125
D. hansenii (L5) (cells/ml)	10 ⁶	10 ⁶	10 ⁶	10 ⁶
Protease (Flavourzyme) (g/g protein)		0.02		0.02

Table S2. Microbial counts (log cfu/g) and pH from the fermentative process of the animal models without and with proteolytic enzyme. Samples were taken at 0, 3, 8 and 15 days of incubation.

	Animal model without enzyme				Animal model with enzyme				RMSE ²	P _t ³	P _E	P _{txE}
	A0 ¹	A3	A8	A15	AE0	AE3	AE8	AE15				
PCA	6.74 bc	7.04 abc	7.59 a	6.89 abc	6.68 c	7.4 ab	7.09 abc	7.04 abc	0.24	**	ns	ns
LAB	6.73 b	6.59 b	7.46 a	6.60 b	6.64 b	7.12 ab	7.03 ab	7.01 ab	0.24	**	ns	**
GC+	6.05 bcd	6.27 bc	6.47 ab	5.71 de	6.02 cd	6.81 a	5.54 e	5.52 e	0.15	***	*	***
YM	6.52 ab	6.35 c	7.32 a	6.48 ab	6.69 ab	6.27 b	6.89 ab	6.82 ab	0.31	**	ns	*
pH	7.04 a	6.79 ab	6.22 b	5.19 c	6.97 a	6.44 ab	4.85 cd	4.46 d	0.23	***	***	***

¹Animal models containing myofibrillar proteins without (A) and with (AE) enzyme at 0, 3, 8 and 15 d of incubation. ²RMSE: root mean square of the errors.

³P_f: P value of the time effect, P_E: P value of enzyme effect, P_{txE}: P value of interaction between time and enzyme effects. ***: P < 0.001; **: P < 0.01; *: P < 0.5; ns: P > 0.05. ⁴Different letters in the same row indicate significant differences among models and sampling times.

Table S3. Microbial counts (log cfu/g) and pH from the fermentative process of the vegetal models without and with proteolytic enzyme. Samples were taken at 0, 3, 8 and 15 days of incubation.

	Vegetal model without enzyme				Vegetal model with enzyme				RMSE ²	P _t ³	P _E	P _{txE}
	V0 ¹	V3	V8	V15	VE0	VE3	VE8	VE15				
PCA	6.81 cd	7.62 bc	8.48 a	8.11 ab	6.8 d	8.2 ab	7.6 bcd	7.7 ab	0.29	***	ns	**
LAB	6.78 c	7.53 ab	7.88 ab	8.08 a	6.77 c	7.45 b	7.52 ab	7.64 ab	0.21	***	*	ns
GC+	5.84 de	6.51 cd	7.79 a	7.28 ab	5.67 e	6.65 bc	6.95 bc	7.13 abc	0.24	***	*	*
YM	6.69 ab	6.27 b	7.02 a	7.12 a	6.63 ab	6.67 ab	6.86 a	7.14 a	0.19	***	ns	ns
pH	6.78 a	5.83 b	5.23 c	4.74 d	6.61 a	4.47 de	4.32 e	4.21 e	0.12	***	***	***

¹Vegetal models containing texturized pea protein without (V) and with (VE) enzyme at 0, 3, 8 and 15 d of incubation. ²RMSE: root mean square of the errors. ³P_f: P value of the time effect, P_E: P value of enzyme effect, P_{txE}: P value of interaction between time and enzyme effects. ***: P < 0.001; **: P < 0.01; *: P < 0.5; ns: P > 0.05. ⁴Different letters in the same row indicate significant differences among models and sampling times.

Table S4. Free amino acids content (mg/g protein) in the animal model without and with proteolytic enzyme. Samples were taken at 0, 3, 8 and 15 days of incubation.

	Animal model without enzyme				Animal model with enzyme				RMSE	Pt	PE	P txE
	A0 ¹	A3	A8	A15	AE0	AE3	AE8	AE15				
Asp	0.01 d	0.01 d	0.00 d	0.01 d	0.12 c	0.22 a	0.17 b	0.11 c	0.02	***	***	***
Glu	0.00	0.00	0.00	0.02	0.08	0.16	0.08	0.08	0.067	ns	**	ns
Ser	0.00 e	0.00 e	0.00 e	0.00 e	0.16 b	0.24 a	0.11 c	0.04 d	0.013	***	***	***
Asn	0.00 e	0.00 e	0.00 e	0.00 e	0.18 b	0.28 a	0.13 c	0.05 d	0.013	***	***	***
Gly	0.00 d	0.00 d	0.00 d	0.00 d	0.07 c	0.15 a	0.10 b	0.11 b	0.012	***	***	***
Gln	0.00 d	0.00 d	0.00 d	0.00 d	0.24 b	0.36 a	0.13 c	0.08 c	0.02	***	***	***
His	0.00	0.00	0.00	0.00	0.05	0.11	0.08	0.11	0.040	ns	***	ns
Thr	0.07 b	0.05 b	0.07 b	0.16 ab	0.24 ab	0.33 a	0.23 ab	0.15 ab	0.068	ns	***	**
Ala	0.00 c	0.00 c	0.00 c	0.02 c	0.22 b	0.58 a	0.41 a	0.48 a	0.063	***	***	**
Arg	0.00 b	0.00 b	0.00 b	0.00 b	0.36 a	0.34 a	0.00 b	0.15 ab	0.112	*	***	*
Pro	0.00 c	0.00 c	0.00 c	0.00 c	0.05 c	0.17 b	0.34 a	0.36 a	0.032	***	***	***
Tyr	0.00 c	0.00 c	0.00 c	0.00 c	0.26 b	0.38 a	0.00 c	0.00 c	0.022	***	***	***
Val	0.00 d	0.00 d	0.00 d	0.00 d	0.33 c	0.57 b	0.60 b	0.66 a	0.014	***	***	***
Met	0.00 c	0.00 c	0.00 c	0.03 bc	0.16 a	0.16 a	0.08 abc	0.09 ab	0.031	ns	***	*
Ile	0.00 d	0.00 c	0.00 d	0.00 d	0.37 c	0.65 b	0.71 a	0.72 a	0.018	***	***	***
Leu	0.00 d	0.00 c	0.00 d	0.00 d	0.69 c	1.13 a	1.04 b	1.07 b	0.019	***	***	***
Phe	0.00 b	0.00 b	0.00 b	0.00 b	0.49 a	0.61 a	0.51 a	0.41 a	0.071	ns	***	ns
Trp	0.00 b	0.00 b	0.00 b	0.00 b	0.03 ab	0.01 ab	0.00 b	0.27 a	0.090	ns	*	ns
Orn	0.00 b	0.00 b	0.00 b	0.00 b	0.00 b	0.06 ab	0.11 a	0.13 a	0.031	*	***	*
Lys	0.00 c	0.00 c	0.00 c	0.00 c	0.22 b	0.51 a	0.45 a	0.44 a	0.027	***	***	***
Total aac	0.08 d	0.06 d	0.08 d	0.23 d	4.31 c	7.00 a	5.27 b	5.50 b	0.189	***	***	***

¹Animal models containing myofibrillar proteins without (A) and with (AE) enzyme at 0, 3, 8 and 15 d of incubation. ²RMSE: root mean square of the errors.

³Pf: P value of the time effect, P_E: P value of enzyme effect, P_{txE}: P value of interaction between time and enzyme effects. ***: P < 0.001; **: P < 0.01; *: P < 0.05; ns: P > 0.05. ⁴Different letters in the same row indicate significant differences among models and times.

Table S5. Free amino acids content (mg/g protein) in the vegetal model without and with proteolytic enzyme. Samples were taken at 0, 3, 8 and 15 days of incubation.

	Vegetal model without enzyme				Vegetal model with enzyme				RMSE	Pt	PE	PtxE
	V0	V3	V8	V15	VE0	VE3	VE8	VE15				
Asp	0.03 b	0.06 b	0.06 b	0.04 b	0.06 b	0.36 a	0.53 a	0.60 a	0.086	***	***	***
Glu	0.45 c	0.36 c	0.47 c	0.45 c	0.60 c	3.17 b	4.28 a	5.07 a	0.320	***	***	***
Ser	0.00 d	0.00 d	0.01 d	0.01 d	0.36 d	1.79 c	2.38 b	2.84 a	0.147	***	***	***
Asn	0.18 e	0.03 e	0.03 e	0.01 e	0.59 d	1.25 c	1.61 b	2.00 a	0.086	***	***	***
Gly	0.01 d	0.03 d	0.05 d	0.03 d	0.10 d	0.68 c	1.08 b	1.59 a	0.058	***	***	***
Gln	0.00 c	0.02 c	0.04 c	0.01 c	0.80 b	1.85 a	2.03 a	2.05 a	0.081	***	***	***
His	0.00 b	0.01 b	0.02 b	0.07 b	0.14 b	0.51 a	0.55 a	0.59 a	0.077	***	***	**
Thr	0.14 d	0.13 d	0.28 d	0.14 d	0.36 d	1.38 c	2.29 b	3.11 a	0.140	***	***	***
Ala	0.03 d	0.03 d	0.08 d	0.08 d	0.56 d	3.09 c	3.96 b	4.73 a	0.254	***	***	***
Arg	0.00 d	0.00 d	0.00 d	0.00 d	0.92 c	2.19 a	2.05 ab	1.66 b	0.151	***	***	***
Pro	0.02 b	0.01 b	0.10 b	0.07 b	0.18 b	0.66 a	0.89 a	1.08 a	0.155	***	***	**
Tyr	0.01 b	0.00 b	0.02 b	0.00 b	0.62 b	4.55 a	5.52 a	5.04 a	0.356	***	***	***
Val	0.02 c	0.00 c	0.01 c	0.04 c	0.78 c	5.63 b	7.31 ab	9.21 a	0.857	***	***	***
Met	0.13 d	0.10 d	0.02 d	0.03 d	0.12 d	0.56 c	0.92 b	1.22 a	0.088	***	***	***
Ile	0.00 c	0.00 c	0.06 c	0.05 c	1.09 c	7.08 b	9.63 ab	12.51 a	1.024	***	***	***
Leu	0.02 d	0.01 d	0.02 d	0.06 d	2.72 d	17.00 c	24.06 b	30.61 a	1.773	***	***	***
Phe	0.03 d	0.01 d	0.09 d	0.04 d	1.84 d	11.17 c	16.91 b	22.11 a	1.493	***	***	***
Trp	0.03 c	0.02 c	0.06 c	0.08 c	0.22 c	0.71 b	1.05 b	1.53 a	0.119	***	***	***
Orn	0.00 c	0.07 bc	0.16 bc	0.18 bc	0.00 c	0.18 bc	0.49 ab	0.79 a	0.153	***	***	*
Lys	0.01 e	0.00 e	0.01 e	0.04 de	0.29 d	1.97 c	2.45 b	2.92 a	0.097	***	***	***
Total aac	1.11 d	0.88 d	1.57 d	1.45 d	12.35 d	65.78 c	89.98 b	111.25 a	6.553	***	***	***

¹Vegetal models containing texturized pea protein without (V) and with (VE) enzyme at 0, 3, 8 and 15 d of incubation. ²RMSE: root mean square of the errors. ³Pf: P value of the time effect, P_E: P value of enzyme effect, P_{txE}: P value of interaction between time and enzyme effects. ***: P < 0.001; **: P < 0.01; *: P < 0.05; ns: P > 0.05. ⁴Different letters in the same row indicate significant differences among models and times.

Table S6. Volatile compounds identified in the headspace of the animal and vegetal models. Animal models contain myofibrillar proteins without (A) and with (AE) enzyme. Vegetal models contain texturized pea protein without (V) and with (VE) enzyme.

Compound	tr ¹ (min)	LRI DB 624 ²	LRI std DB624 ²	RI ³	Models	
					A - AE	V - VE
Aldehydes						
1 Acetaldehyde	2.13	469	466	a	s ⁴	s
2 3-methylbutanal	11.02	690	687	a	s	s
3 2-methylbutanal	11.72	701	698	a	s	s
4 Pentanal	15.35	738	736	a	s	s
5 Hexanal	24.53	841	839	a	s	s
6 Heptanal	31.92	942	939	a	s	s
7 Octanal	39.41	1047	1044	a	s	s
8 Nonanal	46.27	1150	1148	a	s	s
9 Benzaldehyde	37.17	1017	1013	a	s	s
10 benzeneacetaldehyde	43.99	1109	1104	a	s	s
11 4-methyl-benzaldehyde	46.12	1148	-	b	s	s
12 2,4-dimethyl-benzaldehyde	52.79	1292	-	b	s	s
Alcohols						
13 Ethanol	3.11	508	507	a	s	s
14 2-methyl-2-Propanol	4.62	569	-	b	s	s
15 3-methyl-3-Buten-1-ol	20.40	790	787	a	n	s
16 3-methyl-1-butanol	20.89	795	793	a	s	s
17 2-methyl-1-butanol	21.11	798	795	a	s	s
18 1-Pentanol	23.47	828	823	a	s	s
19 1-Hexanol	30.77	925	921	a	s	s
20 1-Heptanol	37.63	1024	1021	a	s	s
21 1-Octen-3-ol	38.12	1030	1028	a	s	s
22 2-Heptanol	32.40	949	947	a	s	s
23 Benzyl alcohol	44.70	1122	1120	a	s	s
24 1-Octanol	44.89	1126	1123	a	s	s
25 Phenylethyl alcohol	48.71	1195	1191	a	s	s
26 2-ethyl-1-Hexanol	42.21	1083	1083	a	s	s
Ester compounds						
27 Methyl acetate	4.18	551	549	a	s	s
28 Ethyl Acetate	7.58	635	635	a	s	s
29 Methyl 3-methylbutirate	21.79	806	804	a	s	s
30 Butyl acetate	25.05	848	846	a	s	s
31 3-methyl-1-butanol acetate	29.56	907	905	a	n	s
Alkanes						
32 Hexano	5.40	600	600	a	s	s
33 Toluene	20.18	788	790	a	s	s
34 Ethylbenzene	27.78	883	881	a	n	s
35 p-Xylene	28.42	891	893	a	s	s
36 o-xylene	30.24	917	915	a	s	s
37 Styrene	30.39	919	921	a	s	s

Ketones							
38	Acetone	3.65	530	527	a	s	s
39	2,3-Butanedione	7.08	627	624	a	s	s
40	2-butanone	7.35	631	629	a	s	s
41	2-Pentanone	14.89	733	731	a	s	s
42	3-Pentanone	15.63	741	740	a	n	s
43	2-hexanone	24.12	836	835	a	s	s
44	3-heptanone	30.98	928	-	b	s	s
45	2-heptanone	31.49	936	933	a	s	s
46	4-methyl-2-Heptanone	34.56	981	-	b	s	s
47	2-Octanone	38.77	1038	1034	a	s	s
48	2-nonanone	45.79	1142	1139	a	s	s
49	3-Octen-2-one	43.13	1095	1094	a	s	s
50	2-pentyl-Furan	36.52	1009	1009	a	s	s
Other compounds							
51	D-Limonene	39.15	1043	1046	a	s	s
52	Acetophenone	45.30	1133	1134	a	s	s
53	Isophorone	49.13	1203	1207	a	s	s
54	Terpinen-4-ol	50.18	1229	1228	a	s	s
Pyrazines							
55	Pyrazine	18.63	772	772	a	n	s
56	Methyl-pyrazine (94) ⁵	26.00	860	860	a	n	s
57	2,5-dimethyl-Pyrazine (108)	32.06	944	943	a	n	s
58	2-ethyl-5-methyl-pyrazine (121)	38.46	1034	1033	a	n	s
59	3-ethyl-2,5-dimethyl-pyrazine (135)	44.00	1109	1109	a	n	s
60	2,3-diethyl-5-methyl-Pyrazine (150)	48.08	1183	-	b	n	s
61	3,5-diethyl-2-methyl-pyrazine (150)	48.35	1188	-	b	n	s
62	3,5-diethyl-2-propyl-Pyrazine (122)	48.56	1192	-	b	n	s

¹Tr: retention time, ²LRI: Linear retention indices of the compounds (LRI DB624) or standards (LRI-std) eluted from GC-MS using a DB-624 capillary column. ³Reliability of identification: a, identification by mass spectrum and by coincidence with the LRI of an authentic standard; b, tentatively identification by mass spectrum. ⁴(s) present in model, (n) absent in model. ⁵Target ion (m/z in parenthesis) used to quantify the compound when the peak was not completely resolved.

Table S7. Volatile compounds content (AU 10⁻⁵/g protein) in the headspace of the animal models (containing myofibrillar proteins) without and with proteolytic enzyme. Samples were taken at 0, 3, 8 and 15 days of incubation.

	Animal model without enzyme				Animal model with enzyme				RMSE ²	P _t ³	P _E	P _{txE}
	A0 ¹	A3	A8	A15	AE0	AE3	AE8	AE15				
Aldehydes												
Acetaldehyde	1.39	1.47	0.72	0.71	0.85	1.01	0.77	0.97	0.33	ns	ns	ns
3-methylbutanal	0.06 bc ⁴	0.00 c	0.00 c	3.96 a	2.04 abc	3.79 a	3.05 abc	3.22 ab	0.93	*	***	*
2-methylbutanal	0.00 bc	0.00 b	0.43 ab	3.46 a	1.54 ab	3.25 a	2.79 a	3.00 a	0.70	*	***	ns
Pentanal	0.05 ab	0.00 b	0.23 ab	0.22 ab	0.00 b	0.24 ab	0.30 a	0.26 ab	0.06	*	ns	ns
Hexanal	1.30	2.49	2.25	1.70	1.29	1.03	1.10	1.10	0.65	ns	*	ns
Heptanal	0.23 b	0.29 ab	0.26 ab	0.59 a	0.16 b	0.22 b	0.47 ab	0.50 ab	0.10	**	ns	ns
Octanal	0.14	0.21	0.29	0.74	0.75	0.29	0.38	0.36	0.16	ns	ns	ns
Nonanal	0.58 b	0.82 b	1.34 b	2.64 a	1.02 b	0.45 b	2.66 a	2.66 a	0.76	**	ns	ns
Benzaldehyde	7.03 ab	3.61 b	3.68 b	5.16 ab	6.00 ab	5.19 ab	8.33 a	8.33 a	1.31	ns	*	*
Benzeneacetaldehyde	0.91	2.46	3.83	2.05	0.00	2.09	4.29	4.51	1.44	ns	ns	ns
4-methyl-benzaldehyde	0.90 c	2.53 abc	3.05 abc	5.10 ab	2.08 bc	2.98 abc	5.97 a	5.97 a	1.15	***	*	ns
2,4-dimethyl-benzaldehyde	1.20 b	1.03 b	1.32 ab	4.10 a	1.80 ab	2.17 ab	3.19 a	3.19 ab	0.85	**	ns	ns
Alcohols												
Ethanol	6.78	5.12	7.61	4.28	6.54	5.78	10.46	9.56	6,15	ns	ns	ns
2-methyl-2-Propanol	0.00 e	0.23 c	0.17 d	0.29 b	0.31 a	0.19 d	0.30 ab	0.30 ab	0.00	***	***	***
3-methyl-3-Buten-1-ol												
3-methyl-1-butanol	0.00 b	0.29 b	0.50 b	0.23 b	0.00 b	0.91 b	5.50 a	5.76 a	0.77	***	***	***
2-methyl-1-butanol	0.00 b	0.00 b	0.00 b	1.57 a	0.00 b	0.72 ab	1.56 a	1.63 a	0.26	***	***	**
1-Pentanol	0.00 b	0.00 b	0.00 b	0.01 b	0.00 b	0.32 a	0.00 b	0.00 b	0.02	***	***	***
1-Hexanol	1.50	0.64	0.50	0.78	0.39	3.93	3.61	3.52	1,36	ns	*	ns
1-Heptanol	0.00	0.00	6.99	3.55	5.27	29.25	12.18	9.85	10,48	ns	*	ns
1-Octen-3-ol	1.19 de	1.38 de	1.79 cd	3.07 ab	0.76 e	2.49 bc	3.38 a	3.38 a	0.26	***	***	***
2-Heptanol	0.44 ab	0.24 c	0.21 c	0.25 bc	0.27 bc	0.48 a	0.33 abc	0.33 abc	0.04	ns	*	**
Benzyl alcohol	0.00 b	0.48 b	0.81 b	1.96 a	0.00 b	0.00 b	2.28 a	2.40 a	0.22	***	**	**
1-Octanol	3.46 b	2.53 b	3.49 b	8.83 a	3.43 b	2.68 b	8.30 a	8.33 a	1.05	***	ns	**
Phenylethyl alcohol	0.00 b	0.00 b	0.21 b	4.59 a	0.00 b	0.00 b	6.48 a	6.57 a	0.53	***	***	***
2-ethyl-1-Hexanol	26.55	18.32	28.81	33.35	22.70	24.42	33.61	33.61	6.45	*	ns	ns
Esthers comp												

Methyl acetate	0.33 b	0.49 ab	1.11 ab	1.10 ab	0.56 ab	0.35 b	1.42 a	1.42 a	0.30 ***	ns	ns
Ethyl Acetate	5.55 c	24.84 a	1.10 c	0.65 c	5.18 c	13.16 b	0.57 c	0.57 c	1.71 ***	***	***
Methyl 3-methylbutyrate	0.00 b	0.00 b	0.00 b	0.04 b	0.72 a	0.12 b	0.00 b	0.00 b	0.07 ***	***	***
Butyl acetate	0.44 a	0.20 ab	0.17 b	0.24 ab	0.40 ab	0.18 ab	0.17 b	0.17 b	0.07 **	ns	ns
3-methyl-1-butanol acetate											
Alkanes											
Hexane	0.37	0.41	0.51	0.32	0.27	0.18	0.33	0.37	0.24 ns	ns	ns
Toluene	3.65 a	2.22 b	2.03 b	2.32 ab	4.09 a	3.11 ab	2.38 b	2.38 b	0.67 **	ns	ns
Ethylbenzene											
p-Xylene	0.28	0.19	0.23	0.24	0.27	0.30	0.26	0.26	0.07 ns	ns	ns
o-xylene	0.15 a	0.09 b	0.13 a	0.17 a	0.14 a	0.00 b	0.18 a	0.17 a	0.02 ***	ns	**
Styrene	0.25 b	0.00 d	0.11 c	0.30 b	0.57 a	0.49 a	0.21 bc	0.21 bc	0.01 ***	***	***
Ketones											
Acetone	0.23 b	1.65 b	0.76 b	4.85 a	0.25 b	1.24 b	4.45 a	4.45 a	0.49 ***	**	***
2,3-Butanedione	0.63	0.80	0.56	0.67	0.64	0.79	0.72	0.72	0.23 ns	ns	ns
2-butanone	0.70 d	1.21 cd	0.80 d	2.03 ab	1.07 cd	1.62 bc	2.20 a	2.20 a	0.17 ***	***	***
2-Pentanone	0.00 c	0.84 ab	0.42 bc	1.16 a	0.00 c	0.90 ab	1.22 a	1.23 a	0.15 ***	**	**
3-Pentanone											
2-hexanone	0.27	1.17	1.03	0.39	0.62	0.22	0.30	0.29	0.40 ns	ns	ns
3-heptanone	0.89	1.03	0.51	0.00	0.37	0.75	0.12	0.17	0.33 ns	ns	ns
2-heptanone	0.46 ab	0.41 b	0.40 b	0.75 a	0.51 ab	0.40 b	0.55 ab	0.55 ab	0.10 *	ns	ns
4-methyl-2-Heptanone	0.43 b	0.69 ab	0.45 b	0.83 a	0.40 b	0.97 a	0.87 ab	0.93 a	0.19 *	ns	ns
2-Octanone	0.42 ab	0.27 b	0.27 b	0.46 a	0.35 ab	0.28 b	0.45 a	0.46 a	0.05 **	ns	*
2-nonanone	0.81 b	0.83 b	2.43 ab	4.73 a	2.51 ab	0.83 b	3.68 a	3.68 a	1.40 *	ns	ns
3-Octen-2-one	0.00 b	0.75 ab	1.42 ab	1.81 a	0.70 ab	0.16 b	1.28 ab	1.39 ab	0.47 *	ns	ns
Other compounds											
2-pentyl-Furan	0.00 b	0.00 b	0.00 b	0.00 b	0.00 b	0.00 b	0.16 a	0.14 a	0.03 **	***	**
D-Limonene	0.26 b	0.61 b	1.07 b	0.00 b	0.76 b	2.11 ab	4.47 a	4.63 a	0.87 *	***	ns
Acetophenone	4.16 ab	3.47 b	3.82 b	6.70 a	5.54 ab	3.26 b	6.53 a	6.53 a	1.42 *	ns	ns
Isophorone	0.84 a	0.00 c	0.00 c	0.00 c	0.85 a	0.28 b	0.00 c	0.00 c	0.07 ***	ns	ns
Terpinen-4-ol	0.00	0.52	0.76	3.68	0.00	0.95	1.47	1.32	1.00 ns	ns	ns

¹Animal models containing myofibrillar proteins without (A) and with (AE) enzyme at 0, 3, 8 and 15 d of incubation. ²RMSE: root mean square of the errors. ³Pf: P value of the time effect, P_E: P value of enzyme effect, P_{txE}: P value of interaction between time and enzyme effects. ***: P < 0.001; **: P < 0.01; *: P < 0.5; ns: P > 0.05. ⁴Different letters in the same row indicate significant differences among models and times.

Table S8. Evolution of the content of volatile compounds (AU 10⁻⁵/g protein) in the headspace of vegetal models (containing pea protein) without and with proteolytic enzyme. Samples were taken at 0, 3, 8 and 15 days of incubation.

	Vegetal model without enzyme				Vegetal model with enzyme				RMSE ²	Pt ³	PE	P txE
	VO ¹	V3	V8	V15	VE0	VE3	VE8	VE15				
Aldehydes												
Acetaldehyde	0.47	0.45	0.44	0.37	0.26	0.24	0.21	0.27	0.13	ns	**	ns
3-methylbutanal	0.22 c ⁴	0.00 d	0.00 d	0.00 d	0.37 c	0.86 a	0.40 c	0.59 b	0.06	***	***	***
2-methylbutanal	0.29 c	0.00 d	0.00 d	0.00 d	0.40 c	0.69 a	0.37 c	0.53 b	0.04	***	***	***
Pentanal	3.18 a	3.37 a	1.57 b	0.73 bc	1.10 bc	0.29 c	0.21 c	0.58 bc	0.38	***	***	***
Hexanal	24.53 a	25.01 a	29.98 a	29.20 a	19.43 ab	6.91 bc	1.06 c	1.72 c	5.32	ns	***	*
Heptanal	0.93 abc	1.16 a	0.88 abc	1.11 ab	0.55 bc	0.56 abc	0.44 c	0.77 abc	0.21	ns	***	ns
Octanal	1.19 abc	1.38 ab	0.86 abc	1.57 a	0.52 bc	0.41 c	0.43 bc	0.96 abc	0.33	*	***	ns
Nonanal	7.69 a	7.30 a	2.32 b	1.83 b	1.02 b	1.39 b	1.21 b	1.76 b	1.53	*	***	**
Benzaldehyde	8.25 abc	10.78 a	7.22 bcd	6.74 bcde	9.69 ab	4.97 cde	3.43 e	4.35 ed	1.19	***	***	***
Benzeneacetaldehyde	1.32	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.44	ns	ns	ns
4-methyl-benzaldehyde	0.35 c	0.89 a	0.87 a	0.71 ab	0.30 c	0.43 bc	0.31 c	0.61 abc	0.11	**	***	**
2,4-dimethyl-benzaldehyde	0.37 b	1.85 a	0.79 b	0.80 b	0.64 b	0.30 b	0.31 b	0.32 b	0.33	*	**	**
Alcohols												
Ethanol	0.60	1.67	15.21	55.14	0.70	4.91	2.53	3.17	21.82	ns	ns	ns
2-methyl-2-Propanol	0.00 b	0.16 ab	0.16 ab	0.27 a	0.00 b	0.28 a	0.18 ab	0.26 a	0.06	***	ns	ns
3-methyl-3-Buten-1-ol	0.59 cd	2.23 b	0.00 d	0.27 d	0.00 d	1.93 bc	3.88 a	5.17 a	0.39	***	***	***
3-methyl-1-butanol	0.00 c	0.98 c	1.70 c	4.43 bc	0.36 c	8.70 bc	14.94 ab	24.06 a	4.56	**	***	*
2-methyl-1-butanol	0.00 c	0.00 c	0.17 c	0.46 bc	0.00 c	1.05 bc	2.40 ab	3.84 a	0.70	**	***	**
1-Pentanol	2.08	0.00	1.17	3.12	2.13	2.71	2.32	2.41	1.12	ns	*	ns
1-Hexanol	0.98 f	5.17 e	9.25 d	11.45 c	2.58 f	16.55 b	17.64 b	27.36 a	0.69	***	***	***
1-Heptanol	2.84 bc	14.70 a	12.86 a	4.69 b	3.27 bc	2.89 bc	1.58 c	2.03 bc	0.85	***	***	***
1-Octen-3-ol	2.04 d	2.84 cd	4.22 abc	4.76 ab	3.42 bcd	3.16 bcd	2.91 cd	5.74 a	0.58	***	ns	**
2-Heptanol	0.34 b	0.57 b	4.95 ab	6.77 a	0.36 b	0.68 b	1.08 b	2.21 ab	1.60	*	*	ns
Benzyl alcohol	0.62 e	1.12 cde	1.50 bcd	1.81 ab	1.01 de	1.45 bcd	1.59 abc	2.13 a	0.19	***	**	ns
1-Octanol	2.14 bc	2.81 bc	3.21 ab	4.46 a	2.07 bc	2.19 bc	1.73 c	2.30 bc	0.46	**	***	**
Phenylethyl alcohol	0.00 f	0.00 f	0.02 ef	0.86 d	0.33 e	2.46 c	6.63 b	12.93 a	0.13	***	***	***
2-ethyl-1-Hexanol	25.69 bc	28.77 bc	34.78 b	54.60 a	17.75 c	18.65 c	25.73 bc	28.42 bc	4.06	***	***	**
Esthers comp	2.31	4.41	2.71	4.33	2.61	3.30	2.11	2.42				

Methyl acetate	1.13	ab	1.27	a	0.69	b	1.07	ab	1.32	a	0.89	ab	0.71	b	0.72	b	0.19	**	ns	*
Ethyl Acetate	0.99		2.12		1.25		2.26		0.81		1.31		0.42		0.00		0.76	ns	*	ns
Methyl 3-methylbutyrate	0.00	c	0.27	b	0.00	c	0.00	c	0.00	c	0.35	b	0.37	b	0.66	a	0.06	***	***	***
Butyl acetate	0.20	d	0.41	b	0.30	bcd	0.73	a	0.25	d	0.26	cd	0.27	cd	0.39	bc	0.05	***	***	***
3-methyl-1-butanol acetate	0.00	e	0.33	bcd	0.48	abc	0.27	cd	0.24	d	0.50	ab	0.35	bcd	0.65	a	0.07	***	***	***
Alkanes																				
Hexane	0.19		0.33		0.10		0.19		0.00		0.27		0.41		0.49		0.19	ns	ns	*
Toluene	3.94		3.52		4.08		2.44		4.14		1.41		1.41		2.39		0.85	*	**	*
Ethylbenzene	0.10	ab	0.22	a	0.17	ab	0.14	ab	0.00	b	0.00	b	0.00	b	0.00	b	0.06	ns	***	ns
p-Xylene	0.39	a	0.41	a	0.28	ab	0.29	ab	0.24	ab	0.19	b	0.24	ab	0.22	b	0.05	ns	***	ns
o-xylene	0.20	a	0.21	a	0.23	a	0.18	a	0.18	a	0.14	a	0.00	b	0.17	a	0.04	*	***	***
Styrene	0.89	b	0.51	b	1.06	b	1.27	b	4.12	a	0.86	b	1.03	b	1.31	b	0.64	**	**	**
Ketones																				
Acetone	1.82	b	3.22	b	2.15	b	2.88	b	2.00	b	3.32	b	4.32	b	8.58	a	1.17	***	***	***
2,3-Butanedione	0.46		2.14		0.95		0.27		1.10		0.35		0.36		0.70		0.70	ns	ns	*
2-butanone	2.95	ab	3.79	ab	2.33	ab	1.15	b	2.69	ab	3.55	ab	3.63	ab	6.18	a	1.46	ns	*	*
2-Pentanone	0.41	b	0.91	b	0.51	b	0.71	b	0.59	b	1.21	b	1.37	b	2.93	a	0.50	***	***	**
3-Pentanone	0.24	b	0.24	b	0.26	b	0.32	b	0.28	b	0.36	b	0.45	b	1.61	a	0.26	***	***	***
2-hexanone	1.72	bc	1.91	abc	1.03	c	0.93	c	1.89	abc	2.14	ab	2.39	ab	2.87	a	0.42	ns	***	**
3-heptanone	0.34	b	5.58	a	1.76	ab	1.17	b	0.46	b	0.00	b	0.00	b	0.00	b	1.45	*	**	*
2-heptanone	15.95	a	13.37	ab	10.64	ab	4.93	b	18.92	a	13.33	ab	12.73	ab	15.27	ab	3.37	*	*	ns
4-methyl-2-Heptanone	0.29	e	0.89	a	0.60	bc	1.05	a	0.29	e	0.50	cd	0.36	de	0.70	b	0.08	***	***	***
2-Octanone	1.09		1.14		0.94		0.79		0.94		0.75		0.67		1.05		0.19	ns	ns	*
2-nonanone	0.96	b	0.90	b	0.96	b	1.14	a	0.70	b	0.70	b	0.67	b	1.10	a	0.17	*	*	ns
3-Octen-2-one	0.76	b	0.77	b	0.96	b	1.28	a	0.52	c	0.52	c	0.41	c	0.47	c	0.10	***	***	***
Other compounds																				
2-pentyl-Furan	0.40	c	0.64	bc	1.08	a	0.84	ab	0.46	bc	1.05	a	0.50	bc	0.78	abc	0.13	***	ns	***
D-Limonene	0.52	ab	0.28	b	0.31	b	0.91	a	0.23	b	0.37	b	0.29	b	0.44	ab	0.17	*	*	ns
Acetophenone	2.63	b	3.49	b	3.03	b	4.79	a	2.62	b	2.71	b	2.63	b	2.73	b	0.41	***	***	***
Isophorone	0.26	ab	0.38	ab	0.31	ab	0.34	ab	0.17	b	0.32	ab	0.42	ab	0.54	a	0.12	*	ns	ns
Terpinen-4-ol	0.27	d	0.24	d	0.19	d	0.48	c	0.30	d	0.74	b	0.88	ab	0.99	a	0.11	***	***	***
Pyrazines																				
Pyrazine	0.00	b	0.31	ab	0.23	ab	0.11	b	0.91	a	0.40	ab	0.00	b	0.26	ab	0.23	ns	*	*
methyl-pyrazine (94)	0.29		0.29		0.30		0.33		0.30		0.32		0.32		0.36		0.04	ns	ns	ns

2,5-dimethyl-Pyrazine (108)	1.28	1.41	1.51	1.30	1.38	1.37	1.61	3.31	0.78	ns	ns	ns								
2-ethyl-5-methyl-pyrazine (121)	0.58	0.60	0.67	0.56	0.67	0.58	0.55	0.64	0.08	ns	ns	ns								
3-ethyl-2,5-dimethyl-pyrazine (135)	1.68	2.12	2.58	2.22	2.76	2.28	2.14	2.45	0.40	ns	ns	*								
2,3-diethyl-5-methyl-Pyrazine (150)	0.12	ab	0.10	b	0.11	ab	0.11	b	0.10	b	0.12	ab	0.01	*	ns	ns				
3,5-diethyl-2-methyl-pyrazine (150)	0.21	abc	0.17	c	0.21	abc	0.20	bc	0.27	a	0.23	abc	0.22	abc	0.25	ab	0.02	ns	***	ns
3,5-diethyl-2-propyl-Pyrazine (122)	0.10	ab	0.00	c	0.08	b	0.09	b	0.12	a	0.08	b	0.00	c	0.10	ab	0.02	***	*	***

¹Vegetal models containing pea protein without (V) and with (VE) enzyme at 0, 3, 8 and 15 d of incubation. ²RMSE: root mean square of the errors. ³Pf: P value of the time effect, P_E: P value of enzyme effect, P_{txE}: P value of interaction between time and enzyme effects. ***: P < 0.001; **: P < 0.01; *: P < 0.5; ns: P > 0.05. ⁴Different letters in the same row indicate significant differences among models and times.