

# PNAS



1

## 2 **Supporting Information for**

### 3 **Self-organized sulfide-driven traveling pulses shape seagrass meadows**

4 **D. Ruiz-Reynés, E. Mayol, T. Sintes, I.E. Hendriks, E. Hernández-García, C.M. Duarte, N. Marbà, and D. Gomila**

5 **Damià Gomila.**

6 **E-mail: [damia@ifisc.uib-csic.es](mailto:damia@ifisc.uib-csic.es)**

#### 7 **This PDF file includes:**

8 Supporting text

9 Figs. S1 to S5

10 Legends for Movies S1 to S4

11 SI References

#### 12 **Other supporting materials for this manuscript include the following:**

13 Movies S1 to S4

## 14 Supporting Information Text

15 **Study site.** The experimental study was conducted at Pollença bay (Mallorca Island, Western Mediterranean), a sheltered bay  
16 exposed to high human pressure (1) with the sea bottom colonized by *Posidonia oceanica*. At the northeast of the bay and at  
17 2-3 m water depth, *P. oceanica* grows forming rings which are delimited by bare sand or *P. oceanica* dead matte colonized by  
18 sparse vegetation of the macroalgae *Caulerpa prolifera* and the seagrass *Cymodocea nodosa*. The measurements for this study  
19 were conducted on six *P. oceanica* rings located at 150-250 meters from the coast, although there is a wave breaker in between  
20 (around 75 and 150 meters of distance to the rings), in September 2021.

21 **Fieldwork sampling.** In each selected *P. oceanica* ring, we laid one transect perpendicular to the vegetation front from one  
22 meter before the inner edge of the front to one meter after the outer edge of it. Along each transect, and every 20 cm, we  
23 measured *P. oceanica* shoot density in  $25 \times 25$  cm squares and collected a sample of the top 10 cm of sediment with 2.6 cm  
24 diameter cores. The sediment was fixed in 1 M zinc acetate (vol:vol). The samples were stored frozen until distillation. Acid  
25 volatile sulfide (AVS) was liberated by the addition of 6 M HCl (in 50% ethanol) and was trapped in zinc acetate, following  
26 Fossing & Jørgensen (1989) (2) and Marbà et al. (2007) (3). The concentrations of AVS pools from the traps were determined  
27 spectrophotometrically according to Cline (1969) (4).

28 The sulfides in the AVS pool can contain sulfides bound to Fe as FeS. However, the carbonate sediment in the coasts of  
29 Mallorca, and in the Pollença bay in particular, are strongly iron-depleted, so the fraction of sulfides bound to Fe as FeS should  
30 be very small in this case (13, 19). Therefore we consider the concentration of AVS pools to be a fair approximation of the  
31 porewater sulfide concentration at this location, which corresponds to the sulfide concentration described by the model.

32 At the inner and outer edges of each ring, we collected three shoots for determination of  $\delta^{34}\text{S}$  in the leaves. In the laboratory,  
33 we scraped the epiphytes from the leaves, soaked the leaves into milliQ water for 5 minutes to remove the salts, and oven-dried  
34 them at  $60^\circ\text{C}$  for 48h. The  $\delta^{34}\text{S}$  in the samples was determined at the Iso Analytical Limited Lab (UK) using a continuous  
35 flow isotope ratio mass spectrometer interfaced with an elemental analyzer (EA-IRMS).

## 36 Model for coupled vegetation density and sulfide concentration dynamics

37 We build on a previous model (5) for the dynamics of the vegetation density  $n \equiv n(x, y, t)$  to include the effects of hydrogen  
38 sulfide concentration  $S \equiv S(x, y, t)$ .

39 The time evolution of the vegetation is described by the following partial differential equation:

$$40 \quad \partial_t n = (\omega_b - \omega_d(n, S))n + d_0 \nabla^2 n + d_1 n \nabla^2 n + d_1 |\nabla n|^2, \quad [1]$$

41 where  $\omega_b$  refers to the branching rate, the terms with spatial derivatives encode clonal growth mechanisms (5), with  $\nabla = (\partial_x, \partial_y)$ ,  
42 and the mortality rate  $\omega_d$  depends not only on the local density but also on the porewater hydrogen sulfide concentration  $S$  in  
43  $\mu\text{M}$ . Direct measurements of demographic rates of *P. oceanica* exposed to different sulfide concentrations indicate a clear effect  
44 of sulfides on mortality, reducing the shoot population growth rate up to 15% when exposed to moderate sulfide concentrations  
45 around  $30\mu\text{M}$  (6). We include this effect in our model increasing the mortality  $\omega_d$  proportionally to the sulfide concentration  $S$ :

$$46 \quad \omega_d(n, S) = \omega_{d0} - \frac{\omega_{d0} a n}{1 + a n} + b n^2 + \gamma S. \quad [2]$$

47 Parameter  $\gamma$  is the sensitivity of the plant to sulfide concentration and measures the increment of the mortality rate for each unit  
48 of sulfide concentration.  $\omega_{d0}$  represents the intrinsic mortality rate of a shoot in the absence of neighboring plants, determining  
49 the typical lifespan of a single shoot, which can depend on external factors such as temperature, salinity, or light availability  
50 (7, 8). Moreover, local density-dependent effects decreasing mortality (facilitation) appear as a result of stress amelioration, for  
51 example reducing wave energy within the meadow or contributing to stabilize or trap sediments (9). Facilitation is included via  
52 the term  $\frac{\mu a n}{1 + a n}$  which grows linearly with  $n$  for low densities and saturates to the value of  $\mu$  which corresponds to the maximum  
53 facilitation. To avoid unrealistic growth, facilitation can not overcome mortality, since it would represent the creation of plants.  
54 Hence,  $\mu \leq \omega_{d0}$  which for simplicity we choose  $\mu = \omega_{d0}$ . The third term in the right hand side of (2) represents negative effects  
55 increasing mortality with density, as a result from self-shading or competition due to resource depletion, determining the  
56 maximum density in the meadow.

57 The evolution of the sulfide concentration is dominated by three processes, the diffusion of organic matter produced by  
58 the plant, production of sulfides due to decomposition, and sulfide removal. In fact, it is possible to describe the evolution of  
59 both organic matter and sulfides with two equations, similarly to what is done in other models (10), where organic matter  
60 diffuses to the surroundings and generates sulfides, which also are able to diffuse on a slower time scale through the sediment.  
61 However, we prefer to use a simpler description accounting only for sulfide concentration, thus, effectively, sulfides are produced  
62 exogenously at a constant rate and proportionally to death plants, and its diffusion encompasses not only diffusion of sulfides  
63 in the sediment but diffusion of organic matter in the water column due to the flow. The evolution of the concentration of  
64 sulfide  $S \equiv S(\vec{x}, t)$  is then described by

$$65 \quad \partial_t S = c_s \omega_d(n, S)n + P_s - \delta_s S - \delta_o n S + D_s \nabla^2 S, \quad [3]$$

66 where  $c_s$  accounts for the increment of sulfide concentration for each dead shoot,  $\delta_s + \delta_o n$  is the rate of removal of sulfides, and  
67  $D_s$  is the effective diffusion of sulfides. We consider all parameters to take only positive values. Thus, Eqs. (1-3) reproduce the

68 coupled dynamics of vegetation and sulfide concentration and determine the effect of this toxic compound on the growth of the  
69 plant.

70 Concerning the generation of sulfides, both exogenous carbon inputs and endogenous carbon generated by dead plants,  
71 after being buried in the first centimeters of sediment, drive sulphate reduction, resulting in increasing porewater sulfide  
72 concentrations. The sulfate reduction rate (SRR) quantifies the production of sulfide. This flux can not be directly associated to  
73 the net sulfide production, because sulfide is oxidized back again to sulfate if oxygen is present in the upper layer of the sediment.  
74 Furthermore, seagrass roots release oxygen into the sediment, particularly during day time, when the plant photosynthesizes, as  
75 a protection mechanism against sulfide intrusion (11), which increases sulfide oxidation. Hence, to determine the net production  
76 of sulfide it is necessary to consider the balance between these two processes. Seagrass sediment SRR, and sulfide and sulfate  
77 concentrations in the literature (6, 12–16) range between  $100 - 500 \mu\text{M}/\text{d}$ ,  $1 - 300 \mu\text{M}$ , and  $\sim 30 \text{ mM}$  respectively. However,  
78 there are fewer estimations of the sulfide oxidation rates (12). We assume, then, the hydrogen sulfide in the sediment to be  
79 produced proportionally to the density of dead plants,  $c_s \omega_d(n, S)n$ , and introduced also exogenously at a constant rate  $P_s$ .

80 We distinguish two processes regarding sulfide removal. First, sulfide is removed through combination with iron to form  
81 pyrite ( $\text{FeS}_2$ ), which is dependent on iron concentration and independent of shoot density. Second, density-dependent processes  
82 include direct absorption through the roots, resulting in intoxication of the plant (17), and sulfide oxidation to sulfate due to  
83 oxygen released by the plant into the sediment, both of which are assumed to be proportional to shoot density and sulfide  
84 concentration. The contribution of each process to the total removal is unclear and highly dependent on the properties of the  
85 substrate. In particular, carbonate sediments, characterized by low iron availability and slow sulfide removal, are associated  
86 with high seagrass vulnerability to sulfide intrusion (3, 6, 13, 18–20). So, porewater sulfide is removed in Eq. (3) at a constant  
87 rate  $\delta_s$ , accounting for the combination with iron, and proportionally to the shoot density and sulfide concentration,  $\delta_o n S$ ,  
88 accounting for direct absorption and sulfide oxidation.

89 The spatial reach of sulfide impacts on seagrass have been quantified in fish farms, providing insights into the spatial scale  
90 of sulfide effects on *P. oceanica* (21–24). The production of biomass at a given position can affect the growth at important  
91 distances due to either the direct diffusion of sulfides in the substrate or within plants, or the transport of decomposing organic  
92 matter by the water. Nevertheless, the real spatial range of sulfide effects is uncertain yet, ranging from tens of centimeters to  
93 tens of meters. Due to the limited evidence, we describe this process as regular diffusion, exploring the effects of different  
94 diffusion constant values  $D_s$  on the spatiotemporal dynamics.

95 **Parametrization for *Posidonia oceanica*.** The values of the parameters characterizing the clonal growth of several seagrasses,  
96 namely the branching rate and angle, and the elongation velocity of the rhizome are available in the literature. For *P. oceanica*  
97 we take the branching rate  $\omega_b = 0.6 \text{ year}^{-1}$ , the branching angle  $\phi_b = 49^\circ$  ( $\sim 45^\circ$  for simplicity) and the elongation velocity of  
98 the rhizome  $\nu = 6.11 \text{ cm}/\text{year}$  (25). In order to show that rings can form it is useful to explore the parameter  $\omega_b$  in the range  
99  $\sim 0.06 - 0.6$ , which can be justified by measurements indicating a ten fold increment close to the coast. The coefficients of the  
100 terms with spatial derivatives can be approximately determined from these parameters (5). Here we take  $d_0 = 31.1 \text{ cm}^2 \text{ year}^{-1}$ ,  
101 and  $d_1 = 1.04 \times 10^2 \text{ cm}^4 \text{ year}^{-1}$ . The taken value of  $d_0$  leads, for  $\omega_d \sim \omega_b$ , to a vegetation diffusion length of 7 cm, compatible  
102 with the growth of a rhizome over a year for *Posidonia oceanica*.

103 The parameter  $b = 6.67 \text{ cm}^4 \text{ year}^{-1}$  is chosen to have the maximum density of 3000 shoots/ $\text{m}^2$  with no sources of mortality  
104 other than density dependent terms ( $\omega_{d0} = \gamma = 0$ ). Equivalently, to determine the scale for the shoot density we set  
105  $\sqrt{\omega_b/b} = 0.3 \text{ shoots}/\text{cm}^2$ . The range of bistability is mainly controlled by parameter  $a$ . Given the limited evidence, we choose  
106 this parameter such that the change in mortality  $\omega_{d0}$  within the bistable range is comparable to the branching rate  $\sim \omega_b$  for  
107 the other parameters fixed, similarly to what was done in a previous work (26).

108 Estimates of sulfide removal  $\delta_s + \delta_o n \sim 1 \text{ year}^{-1}$  were determined using time series (3). The variability in estimating  
109 such parameters is very large and some estimates are bigger than the values of the parameters  $\delta_s$  and  $\delta_o$  compatible with  
110 the existence of excitable pulses. According to Eq. (8), the parameter  $\delta_s$ , which is the most relevant for having excitability,  
111 must be smaller than  $\omega_b$ , what means that the evolution of sulfides is slower than the time scale of vegetation. Given the  
112 limited precision of the data available in the literature, we take  $\delta_s = 3.6 \times 10^{-2} \text{ year}^{-1}$  and  $\delta_o = 6 \times 10^{-2} \text{ year}^{-1} \text{ cm}^2$  to set  
113 the system in the excitable region. This value is smaller than a previous estimation (3) but reasonable within the variability  
114 of sulfide concentration measures. The production of sulfides  $P_s$  is included in the model to allow to reproduce a constant  
115 background of sulfides. In our measurement this background concentration, determined using the values of sulfide concentration  
116 outside of the ring, is  $S_0 = P_s/\delta_s \sim 31.4 \mu\text{M}$ , and we set  $P_s = 1.13 \mu\text{M} \text{ year}^{-1}$ . Parameter  $c_s \sim 30.6 \mu\text{M} \text{ cm}^2$  is set to fit  
117 the scale of the measured sediment sulfide concentrations, ensuring the order of magnitude, while being in agreement with  
118 other previous experimental measurements (6). To determine the values of  $c_s$  we have used the relation between the sulfide  
119 concentration and vegetation density for the stationary homogeneous solution  $S^* = \frac{c_s \omega_b n^* + P_s}{\delta_s + \delta_o n^*}$ , which linearized allows to  
120 estimate an order of magnitude for  $c_s$  from the relation between sulfide concentration and density determined experimentally  
121  $S_{exp} \sim P_s/\delta_s + (c_s \omega_b \delta_s - P_s \delta_o)/\delta_s^2 n_{exp}$  with the other parameters fixed. Regarding sulfide diffusion, there are no precise  
122 estimates of the diffusion coefficient of sulfides, which we have changed significantly to investigate its role in the dynamics. The  
123 precise values used are given in the figure captions. The value used in Fig. 3 to generate traveling pulses is  $D_s = 3.6 \text{ cm}^2$   
124  $\text{year}^{-1}$  which correspond to a diffusion length  $\sqrt{D_s/\delta_s} = 0.1 \text{ m}$ , compatible with diffusion through the sediment.

125 Finally, the sensitivity to sulfides is one of our main control parameters and we vary it significantly to explore the different  
126 dynamical regimes. The precise values are indicated in the figures. The value used in the numerical simulations of the rings  
127  $\gamma = 7.19 \times 10^{-3} \mu \text{ M}^{-1} \text{ year}^{-1}$  ( $\gamma \frac{c_s}{\sqrt{\omega_b b}} = 1.1 \times 10^{-1}$ ), is less than twice (1.8 fold) the experimental value of  $\gamma = 4.1 \times 10^{-3}$

128  $\mu \text{ M}^{-1} \text{ year}^{-1}$  ( $\gamma \frac{c_s}{\sqrt{\omega_b b}} = 6.27 \times 10^{-2}$ ) estimated in the literature (6). Moreover, it is worth mentioning the experimental  
 129 value would also exhibit excitability although in a smaller range of mortality. We also point out that the relevant parameter  
 130 to observe a given dynamical regime is not  $\gamma$  alone, but the product  $\gamma c_s$  (see Fig. 2), and  $c_s$  determines the scale of the  
 131 sediment sulfide concentration. We have chosen  $c_s$  to fit the measured concentrations, but sediment sulfide concentration is  
 132 highly variable over time compared to the changes in vegetation density. Therefore the values of  $\gamma$  used are referred to the  
 133 concentrations measured in September 2021, which may not account for previous sulfide exposure. Should the concentration be  
 134 higher in other periods, the effective values of  $\gamma$  could be lower.

135 **Dimensionless parameters.** For simplicity it is convenient to work with dimensionless units, such that time, space and density  
 136 of shoots in the new units are given by:  $t' = \omega_b t$ ,  $\vec{r}' = \sqrt{\frac{\omega_b}{d_0}} \vec{r}$ ,  $n' = \sqrt{\frac{b}{\omega_b}} n$  and  $S' = \frac{1}{c_s} \sqrt{\frac{b}{\omega_b}} S$ . We note that the branching  
 137 rate fixes the temporal scale, the spatial scale is determined by the vegetation diffusion, the scale of the density of shoots is  
 138 determined by the saturation parameter  $b$ , and, finally, the scale of sulfide concentration is determined by  $c_s$ . In the following  
 139 we drop the primes from the variables and parameters expressed in the new units.

140 **Dynamical regimes and linear stability analysis.** The model given by Eqs. (1-3) describes the coupled dynamics of seagrass  
 141 shoot density and porewater sulfide concentration, allowing to study the effects of instabilities on the spatial distribution  
 142 of vegetation and the emerging spatiotemporal regimes. The homogeneous steady states  $n^*$  of Eqs. (1-3) are given by the  
 143 solutions of the nonlinear equation  $(\omega_b - \omega_d(n^*, S^*))n^* = 0$ , where  $S^* = \frac{c_s \omega_b n^* + P_s}{\delta_s + \delta_o n^*}$ . The solutions can be multiple depending  
 144 on the parameters.

145 Under favorable conditions (i.e. when the density-independent mortality is smaller than the branching rate  $\omega_{d0}/\omega_b < 1$ , light  
 146 blue shaded region in Fig. 2), bare soil ( $n^* = 0$ ), which is always a solution of the equations independently of the parameters,  
 147 is unstable and vegetation grows and forms a homogeneous meadow with density  $n^* = n_p^*$ . Given a fixed branching rate,  
 148 increasing mortality  $\omega_{d0}$  leads to a reduction in the stationary value of shoot density, until large values of  $\omega_{d0}$  lead to bare soil  
 149 as the only possible stable configuration (white region). As a result of facilitative effects, the systems exhibits bistability at  
 150 values of  $\omega_{d0}$  above, but close to, the critical value ( $\omega_{d0}/\omega_b = 1$ ) provided moderate sensitivity to sulfide concentration ( $\gamma$ ),  
 151 leading to the populated and unpopulated solutions coexisting for a range of  $\omega_{d0}$  values above  $\omega_b$  (dark blue shaded region  
 152 in Fig. 2). In this regime an additional homogeneously populated unstable solution  $n^* = n_u^*$  with an intermediate density  
 153 between 0 and  $n_p^*$  exists too.

154 The homogeneous solutions are affected by different instabilities that determine the spatiotemporal evolution of the vegetation  
 155 density. We use linear analysis to calculate the stability of the stationary homogeneous solutions by considering perturbations  
 156 of the form  $n = n^* + n_p$ ,  $S = S^* + S_p$ . The linearized systems reads:

$$157 \quad \partial_t n_p = (\omega_b - \omega_d(n^*, S^*) + \beta(n^*))n_p - \gamma n^* S_p + (d_0 + d_1 n^*)\nabla^2 n_p, \quad [4]$$

$$158 \quad \partial_t S_p = c_s(\omega_d(n^*, S^*) - \beta(n^*))n_p + (c_s \gamma n^* - \delta_s - \delta_o n^*)S_p$$

$$159 \quad -\delta_o S^* n_p + D_s \nabla^2 S_p, \quad [5]$$

160 where  $\beta(n^*) = -2bn^* + \frac{\omega_{d0} \alpha n^*}{(1 + \alpha n^*)^2}$ . Considering perturbations of the form  $\sim e^{-i\vec{q} \cdot \vec{x}}$  the following eigenvalue problem results,

$$161 \quad \begin{vmatrix} \omega_b - \omega_d^* + \beta^* - D_n^* q^2 - \lambda & -\gamma n^* \\ c_s(\omega_d^* - \beta^*) - \delta_o S^* & c_s \gamma n^* - \delta_s - \delta_o n^* - D_s q^2 - \lambda \end{vmatrix} = 0 \quad [6]$$

162 where  $q^2 = \vec{q} \cdot \vec{q} = q_x^2 + q_y^2$  and  $\beta^* = \beta(n^*)$ ,  $\omega_d^* = \omega_d(n^*, S^*)$  and  $D_n^* = d_0 + d_1 n^*$ . Thus, the eigenvalues for the unpopulated  
 163 solution ( $n^* = 0$ ) are  $\lambda = \omega_b - \omega_{d0} - \gamma P_s / \delta_s - d_0 q^2$  and  $\lambda = -\delta_s - D_s q^2$ . The first eigenvalue is negative for all values of  $q$   
 164 when  $\omega_b > \omega_{d0} + \gamma P_s / \delta_s$ , which corresponds to a transcritical bifurcation determining the stability of the unpopulated solution.  
 165 The second eigenvalue, which is always negative, determines the time scale of sulfides close to the unpopulated solution being  
 166 determined by the removal rate  $\delta_s$ . Both eigenvalues have high-wavenumber damping given by diffusion.

167 The eigenvalues of the populated solutions are given by roots  $\lambda$  of the following characteristic polynomial:

$$168 \quad \begin{aligned} & \lambda^2 - \lambda(\beta^* - (D_n^* + D_s)q^2 + c_s \gamma n^* - \delta_s - \delta_o n^*) \\ & - \beta^*(\delta_s + \delta_o n^*) + c_s \omega_b \gamma n^* - \gamma n^* \delta_o S^* \\ & - (\beta^* D_s + D_n^*(c_s \gamma n^* - \delta_s - \delta_o n^*))q^2 \\ & + D_n^* D_s q^4 = 0. \end{aligned} \quad [7]$$

169 Finding the roots of (7), one obtains that the homogeneous solution  $n^* = n_u^*$  is always unstable, while  $n^* = n_p^*$  is always stable  
 170 for low enough  $\omega_{d0}$ . Instead for values of  $\omega_{d0}$  comparable to  $\omega_b$ , and large enough sensitivity to sulfide  $\gamma$ , the total mortality  
 171  $\omega_d$  increases significantly with increasing sulfide concentration, destabilizing the homogeneous vegetation distribution in two  
 172 different ways.

173 On one hand, it can experience an oscillatory instability of the homogeneous meadow, a Hopf bifurcation, where vegetation  
 174 experiences a periodic cycle (pink region in Fig. 2). The Hopf bifurcation can be identified by imposing  $\lambda(q=0) = i\omega$  with  
 175  $\omega \in \mathbb{R}$  in Eq. (7) and determining this way the parameter regimes leading to oscillatory behavior. The presence of oscillations



176 can be easily understood considering the negative feedback loop between vegetation and sulfides. The growth of vegetation  
 177 leads to sulfide production with a certain time lag, which beyond a threshold concentration causes vegetation density to decline.  
 178 The production of sulfides, then, decreases as well until its concentration becomes low enough to allow the vegetation to regrow,  
 179 generating a spatially homogeneous periodic oscillation of the vegetation density and sulfide concentration.

180 On the other hand, the homogeneous solution can also experience a Turing instability, which is calculated imposing the  
 181 conditions  $Re[\lambda(q)] = Re[\frac{\partial \lambda(q)}{\partial q}] = 0$  on Eq. (7) and it is indicated in Fig. 2 with dashed lines for different values of the sulfide  
 182 diffusion constant  $D_s$ . As a result of this instability spatial modulations grow leading to the formation of regular patterns. The  
 183 spatiotemporal dynamics resulting from these two combined instabilities leads to a very complex Turing-Hopf behavior (27). In  
 184 this work we focus on the case in which the dynamics is dominated by the Hopf bifurcation, which corresponds to low values of  
 185  $D_s$  (Fig. 2), in particular, values of the interaction length of vegetation driven by sulfides  $\sqrt{D_s/\delta_s}$  smaller than half meter.  
 186 In this case, starting from a populated homogeneous solution and increasing  $\omega_{d0}$ , the Hopf bifurcation occurs first and the  
 187 oscillatory dynamics dominates over pattern formation.

188 In the bistable regime the homogeneous steady state  $n_p^*$  coexists with the unpopulated state  $n^* = 0$  and the unstable  
 189 homogeneous steady state  $n_u^*$ . The intermediate density  $n_u^*$  acts as a threshold for the facilitation to operate and sustain an  
 190 homogeneous meadow. On top of this, the presence of a cycle bifurcating from the Hopf instability of the upper homogeneous  
 191 solution  $n_p^*$ , and whose amplitude grows as mortality is increased, can lead to a situation in which, during the cycle, the  
 192 density reaches the value of the unstable solution  $n_u^*$ . At this point the cycle is destroyed, since facilitative interactions are  
 193 not strong enough to support vegetation for densities below  $n_u^*$ . This transition, known as homoclinic bifurcation, is a global  
 194 bifurcation that changes the dynamics of the system drastically. Close to the onset of the homoclinic bifurcation, the dynamics  
 195 of the periodic oscillation become very slow for densities close to  $n_u^*$ , increasing the period of the oscillations, which diverges at  
 196 threshold (28). When the homoclinic transition is crossed and there is no limit cycle (orange region in Fig. 2), the dynamics  
 197 becomes excitable. Since,  $n_u^*$  acts as a threshold for the vegetation to grow, vegetation will decrease exponentially to zero  
 198 below this threshold, while densities larger than  $n_u^*$  will allow the vegetation to grow, increasing also the sulfide concentration  
 199 until a certain point in which mortality overshoots leading to vegetation density to decrease again below  $n_u^*$ , and then to zero.  
 200 This excitable excursion reproduces a single cycle of the former oscillatory state. In this excitable regime the final state is  
 201 always bare soil independently of the initial density, but if the initial vegetation is dense enough to overcome the threshold,  
 202 then the system produces a pulse of vegetation before ending up in bare soil.

203 The curve delimiting the homoclinic bifurcation in parameter space (orange region in Fig. 2) is born at two Takens-Bogdanov  
 204 (TB<sub>1,2</sub>) points, where the Hopf coincides with the saddle node bifurcation. Writing the stationary condition as  $\omega_{d0} \equiv \omega_{d0}(n^*)$   
 205 one can impose the saddle node condition  $\frac{\partial \omega_{d0}(n^*)}{\partial n^*} = 0$  and introduce it in Eq. (7). This forces one of the eigenvalues to be  
 206 zero. Imposing the second eigenvalue to be zero allows to determine the TB points. The three conditions give involved relations  
 207 which are solved numerically.

208 Under the right conditions, an analytical criteria for the existence of a TB can be derived. In general the maximum  
 209 vegetation density is determined by the term  $bn^2$  present in the total mortality. However the term  $\gamma S$  also prevents an unlimited  
 210 growth determining the maximum value of the density. Under these circumstances a good approximation is to neglect the term  
 211  $bn^2$ . For  $\delta_o = P_s = 0$ , this allows to find an approximated value of the mortality at which the TB<sub>1</sub> takes place:

$$212 \quad \frac{\omega_{d0, TB_1}}{\omega_b} \approx \frac{1}{1 - \frac{\delta_s^2}{\omega_b^2}} \quad [8]$$

213 Since the TB<sub>1</sub> point is the origin of the homoclinic bifurcation, the existence of a TB point is good indication of the existence  
 214 of excitable dynamics in a nearby parameter region. This requires the right hand side of Eq. (8) to be positive, as the mortality  
 215 and branching rates take positive values only. Thus, excitability is more easily observed for  $\delta_s < \omega_b$ . This implies that the  
 216 formation of rings are most likely to occur if the removal rate of sulfides  $\delta_s$  is smaller than the branching rate of the plant  $\omega_b$ .

217 **Numerical simulations.** The model described by (1), (2) and (3) is a system of two coupled partial differential equations (PDEs)  
 218 which is integrated in time with a pseudo-spectral method (29). We consider a squared grid with  $N_x$  and  $N_y$  grid points  
 219 with periodic boundary conditions and we integrate the linear terms in Fourier space exactly, while the nonlinear terms are  
 220 integrated using a second-order in time approximation.

221 The excitable region is calculated using numerical simulations for the temporal systems where the diffusion in space has not  
 222 been considered. Starting from the stationary populated solution  $n^*$ ,  $S^*$  plus a small perturbation as initial condition we let  
 223 the system time evolve for long times and compute the amplitude of the limit cycle. Following this procedure for different  
 224 values of  $\omega_{d0}$  in the bistable range one can identify when the minimum density of the limit cycle coincides with the value of the  
 225 unstable branch  $n_u^*$ . This process is repeated for different values of the parameter  $\gamma$  to identify the homoclinic bifurcation in  
 226 the phase diagram in Fig. 2.

227 Numerical simulations of excitable pulses or rings are performed using a grid of size  $N_x = N_y = 512$  starting with an initial  
 228 condition with  $S(\vec{x}) = S_0$  and with spots of vegetation located at different positions where the density  $n$  follows a Gaussian  
 229 function in two dimensions where the maximum density has a value of the density equal or greater than  $n_u^*$  for each parameter  
 230 set. Neither the width of the Gaussian nor the initial position have effect on the final ring. A different initial spatial distribution  
 231 has been used capturing the details of vegetation from ortophotos. The images have been processed in gray scale filtering  
 232 regions with vegetation over a certain threshold and converted to a binary array  $N_x = N_y = 1024$  with value equal to one for  
 233 the position of vegetation. A diffusive filter has been applied to obtain soft transitions at the interfaces by multiplying by

234  $e^{-k^2 t}$  the Fourier amplitudes  $A_k$  of the array, where  $k$  is the wavenumber and  $t$  controls the softness of the interfaces. Finally,  
235 multiplying the values of the array by the stationary value of the density  $n^*$  for each parameter set, the array has been scaled  
236 to obtain the desired values of the maximum of the vegetation density for the initial condition.

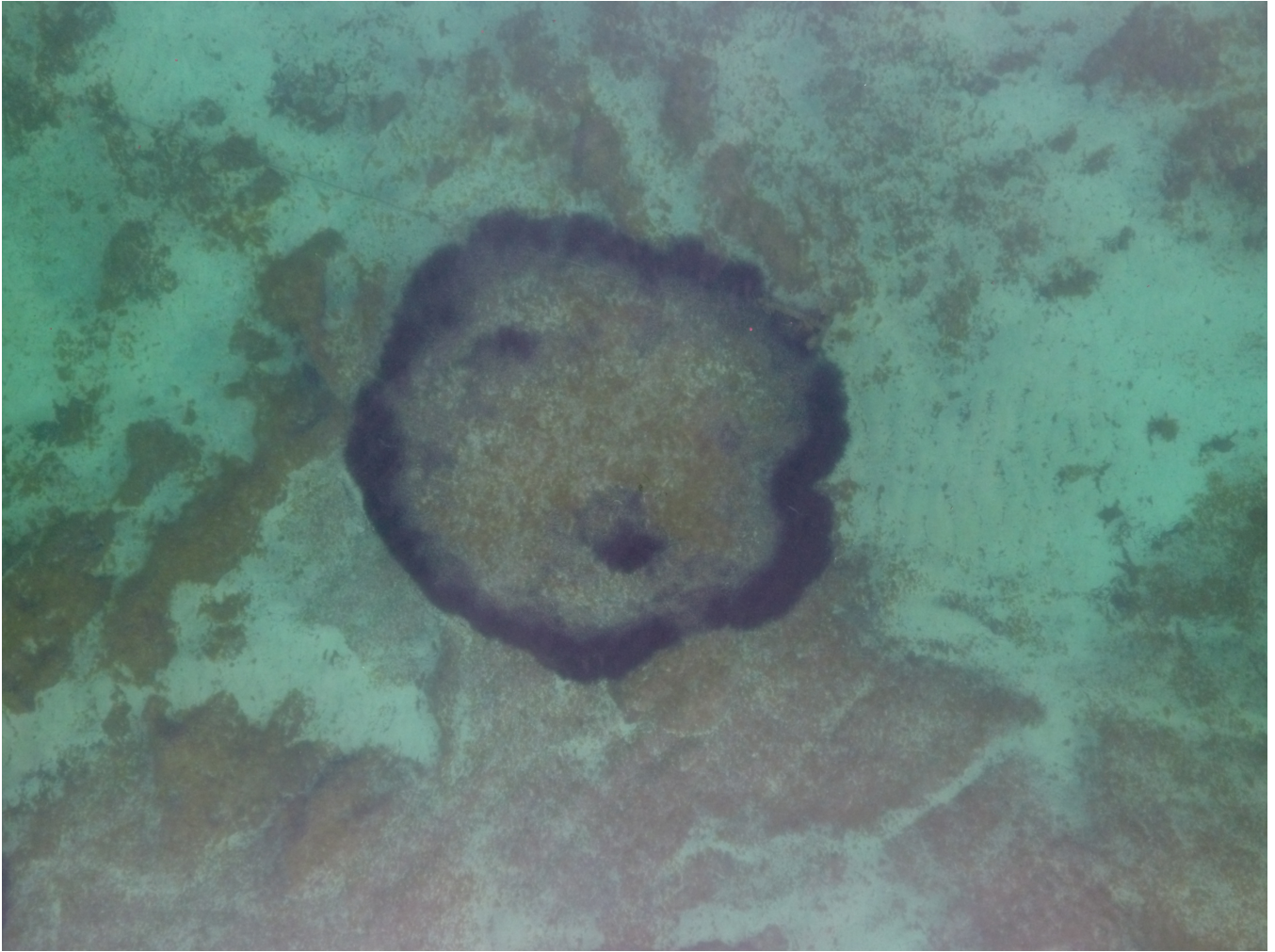


**Fig. S1.** Aerial image of the study site. Drone image of the study site in Pollença bay showing the environment where the vegetation patterns of *Posidonia oceanica* develop.



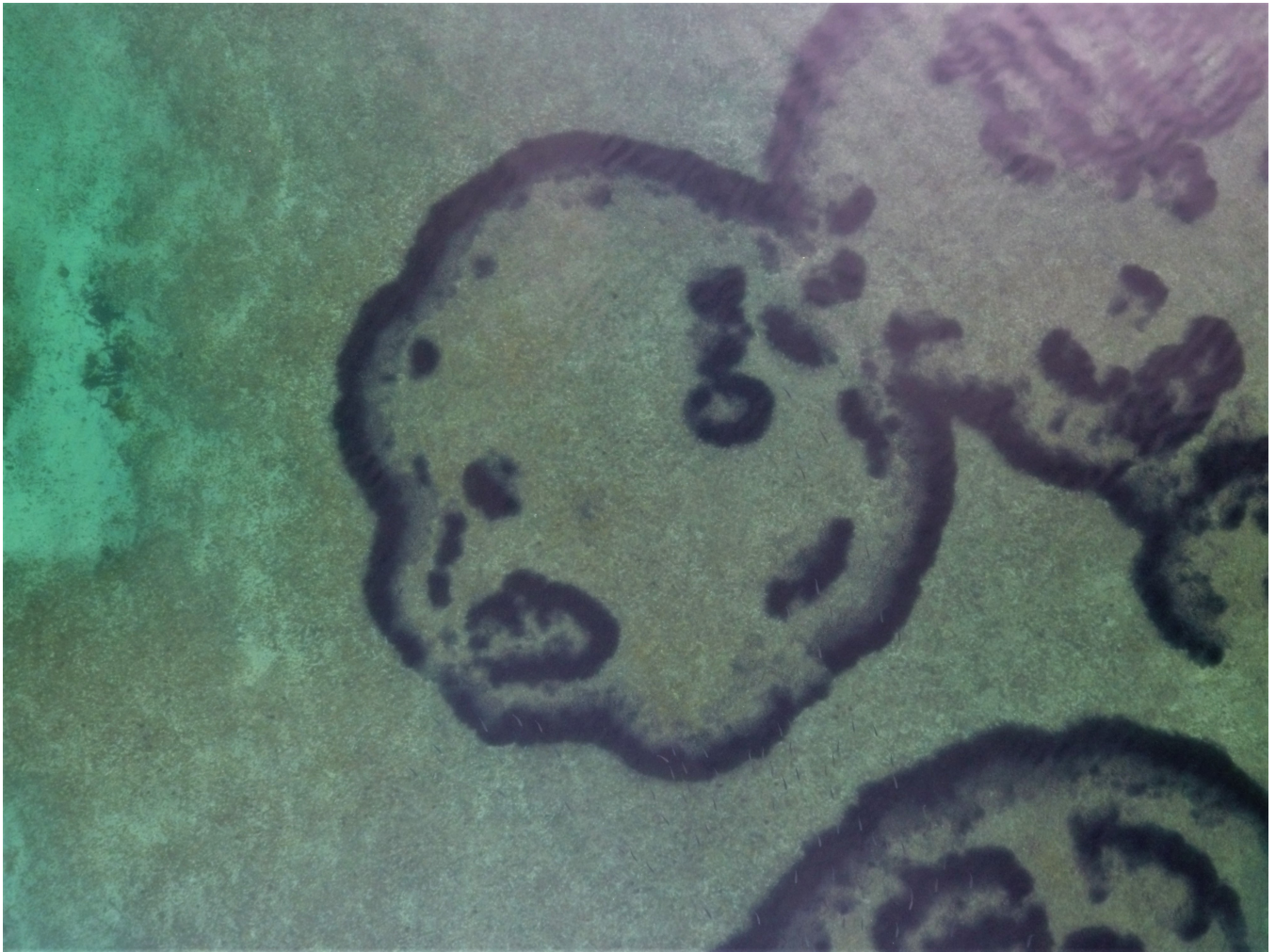


**Fig. S2.** General aerial view of the patterns observed in the *Posidonia oceanica* meadows present in the Pollença bay.



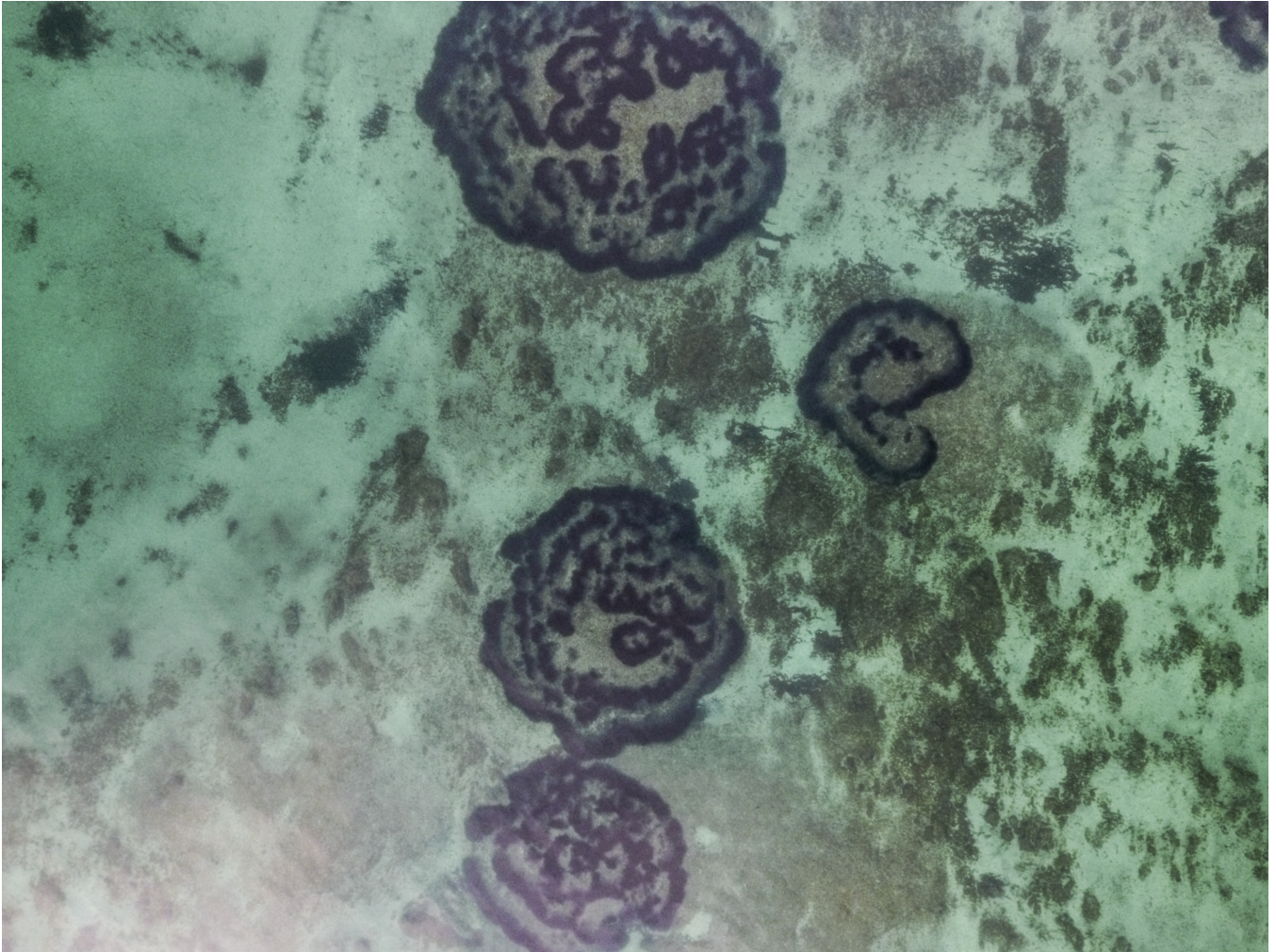
**Fig. S3.** Excitable ring. High-resolution 2021 drone image of an excitable ring of *Posidonia oceanica* in the Pollença bay.





**Fig. S4.** Collision of two rings. High-resolution 2021 drone image of the region where two excitable rings collided.





**Fig. S5.** Spiral of *Posidonia oceanica*. High-resolution 2021 drone image showing a spiral of *Posidonia oceanica* and other disordered patterns as a result of the excitable dynamics.

237 **Movie S1. Time evolution of excitable rings under spatially homogeneous mortality: Movie of the simulation**  
238 **shown in Figs. 3 (a-e) for homogeneous mortality conditions starting from initial conditions resembling aerial**  
239 **pictures of the vegetation from 1973. A collision of two rings is observed during the evolution.**

240 **Movie S2. Time evolution of excitable rings under spatially heterogeneous mortality conditions: Movie of the**  
241 **simulations shown in Figs. 3 (f-j) for spatially heterogeneous mortality conditions starting from vegetation**  
242 **initial conditions resembling aerial pictures from 1973.**

243 **Movie S3. Animation of aerial images of the study site in Pollença bay from 1973 to 2018 showing the actual**  
244 **evolution of the spatial distribution of vegetation patterns.**

245 **Movie S4. Animation showing a zoom of the collision of two excitable rings from historic aerial images from**  
246 **1973 to 2018.**

## 247 **References**

- 248 1. I Mazarrasa, N Marbà, J Garcia-Orellana, P Masqué, CM Arias-Ortiz, Ariane nad Duarte, Dynamics of carbon sources  
249 supporting burial in seagrass sediments under increasing anthropogenic pressure. *Limnol. Oceanogr.* **62**, 1451–1465 (2017).
- 250 2. H Fossing, BB Jørgensen, Measurement of bacterial sulfate reduction in sediments: evaluation of a single-step chromium  
251 reduction method. *Biogeochemistry* **8**, 205–222 (1989).
- 252 3. N Marbà, et al., Iron additions reduce sulfide intrusion and reverse seagrass (*Posidonia oceanica*) decline in carbonate  
253 sediments. *Ecosystems* **10**, 745–756 (2007).
- 254 4. JD Cline, Spectrophotometric determination of hydrogen sulfide in natural waters 1. *Limnol. Oceanogr.* **14**, 454–458  
255 (1969).
- 256 5. D Ruiz-Reynés, F Schönsberg, E Hernández-García, D Gomila, General model for vegetation patterns including rhizome  
257 growth. *Phys. Rev. Res.* **2**, 023402 (2020).
- 258 6. ML Calleja, N Marbà, CM Duarte, The relationship between seagrass (*Posidonia oceanica*) decline and sulfide porewater  
259 concentration in carbonate sediments. *Estuarine, Coast. Shelf Sci.* **73**, 583–588 (2007).
- 260 7. N Marbà, CM Duarte, Mediterranean warming triggers seagrass (*Posidonia oceanica*) shoot mortality. *Glob. Chang. Biol.*  
261 **16**, 2366–2375 (2010).
- 262 8. CM Duarte, J Kalff, Latitudinal influences on the depths of maximum colonization and maximum biomass of submerged  
263 angiosperms in lakes. *Can. J. Fish. Aquatic Sci.* **44**, 1759–1764 (1987).
- 264 9. JL Gutiérrez, et al., *Physical ecosystem engineers and the functioning of estuaries and coasts*. (Elsevier Inc.) Vol. 7, pp.  
265 53–81 (2011).
- 266 10. J de Fouw, et al., A facultative mutualistic feedback enhances the stability of tropical intertidal seagrass beds. *Sci. Reports*  
267 **8**, 12988 (2018).
- 268 11. O Pedersen, J Borum, C Duarte, M Fortes, Oxygen dynamics in the rhizosphere of *Cymodocea rotundata*. *Mar. Ecol.*  
269 *Prog. Ser.* **169**, 283–288 (1998).
- 270 12. BB Jørgensen, The sulfur cycle of a coastal marine sediment (Limfjorden, Denmark). *Limnol. Oceanogr.* **22**, 814–832  
271 (1977).
- 272 13. M Holmer, CM Duarte, N Marbà, Sulfur cycling and seagrass (*Posidonia oceanica*) status in carbonate sediments.  
273 *Biogeochemistry* **66**, 223–239 (2003).
- 274 14. MS Frederiksen, M Holmer, J Borum, H Kennedy, Temporal and spatial variation of sulfide invasion in eelgrass (*Zostera*  
275 *marina*) as reflected by its sulfur isotopic composition. *Limnol. Oceanogr.* **51**, 2308–2318 (2006).
- 276 15. BB Jørgensen, Mineralization of organic matter in the sea bed—the role of sulphate reduction. *Nature* **296**, 643–645  
277 (1982).
- 278 16. J Borum, et al., Eelgrass fairy rings: sulfide as inhibiting agent. *Mar. Biol.* **161**, 351–358 (2014).
- 279 17. O Pedersen, T Binzer, J Borum, Sulphide intrusion in eelgrass (*Zostera marina* L.). *Plant, Cell Environ.* **27**, 595–602  
280 (2004).
- 281 18. M Holmer, P Storkholm, Sulphate reduction and sulphur cycling in lake sediments: a review. *Freshw. Biol.* **46**, 431–451  
282 (2001).
- 283 19. M Holmer, CM Duarte, N Marbà, Iron additions reduce sulfate reduction rates and improve seagrass growth on organic-  
284 enriched carbonate sediments. *Ecosystems* **8**, 721–730 (2005).
- 285 20. N Marbà, et al., Sedimentary iron inputs stimulate seagrass (*Posidonia oceanica*) population growth in carbonate sediments.  
286 *Estuarine, Coast. Shelf Sci.* **76**, 710–713 (2008).
- 287 21. MS Frederiksen, M Holmer, E Díaz-Almela, N Marbà, CM Duarte, Sulfide invasion in the seagrass *Posidonia oceanica* at  
288 Mediterranean fish farms: assessment using stable sulfur isotopes. *Mar. Ecol. Prog. Ser.* **345**, 93–104 (2007).
- 289 22. M Holmer, MS Frederiksen, Stimulation of sulfate reduction rates in Mediterranean fish farm sediments inhabited by the  
290 seagrass *Posidonia oceanica*. *Biogeochemistry* **85**, 169–184 (2007).
- 291 23. M Holmer, et al., Sedimentation of organic matter from fish farms in oligotrophic Mediterranean assessed through bulk  
292 and stable isotope ( $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$ ) analyses. *Aquaculture* **262**, 268–280 (2007).

- 293 24. M Holmer, et al., Effects of fish farm waste on *Posidonia oceanica* meadows: synthesis and provision of monitoring and  
294 management tools. *Mar. Pollut. Bull.* **56**, 1618–1629 (2008).
- 295 25. T Sintès, N Marbà, CM Duarte, Modeling nonlinear seagrass clonal growth: Assessing the efficiency of space occupation  
296 across the seagrass flora. *Estuaries Coasts* **29**, 72–80 (2006).
- 297 26. D Ruiz-Reynés, et al., Fairy circle landscapes under the sea. *Sci. Adv.* **3**, e1603262 (2017).
- 298 27. D Walgraef, *Spatio-temporal pattern formation: with examples from physics, chemistry, and materials science*. (Springer-  
299 Verlag, New York), p. 309 (1997).
- 300 28. A Arinyo-i Prats, P Moreno-Spiegelberg, M Matías, D Gomila, Traveling pulses in Class-I excitable media. *Phys. Rev. E*  
301 **104**, L052203 (2021).
- 302 29. R Montagne, E Hernández-García, A Amengual, M San Miguel, Wound-up phase turbulence in the complex Ginzburg-  
303 Landau equation. *Phys. Rev. E* **56**, 151–167 (1997).