Rotranslational state-to-state rates and spectral representation of inelastic collisions in low-temperature molecular hydrogen

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Inelastic collisions in natural H₂ are studied from the experimental and theoretical points of view between 10 and 140 K. Rotational populations and number densities measured by Raman spectroscopy along supersonic expansions of H₂ provide the link between experimental and theoretical rotranslational state-to-state rate coefficients of H₂ in the vibrational ground state. These rates are calculated in the close-scattering approach with the MOLSCAT code employing a recent ab initio H₂–H₂ potential. The calculated rates are assessed by means of a master equation describing the time evolution of the experimental rotational populations. The feasibility for obtaining the rates on the sole basis of the experiment is discussed. The dominant processes \( j_1j_2 \rightarrow j_1'j_2' \) in the investigated thermal range are found to be \( 21 \rightarrow 01 > 30 \rightarrow 12 > 31 \rightarrow 11 \), proving the importance of double processes such as \( 30 \rightarrow 12 \). Good agreement is found between theory and experiment, as well as with earlier ultrasonic measurements of relaxation times. A spectral representation is proposed in order to visualize quantitatively the collisional contributions in any nonequilibrium time evolving process. © 2006 American Institute of Physics.

I. INTRODUCTION

Much of the ongoing research on molecular collisions is aimed at establishing accurate state-to-state collisional cross sections and their associated rate coefficients for a number of small molecular species in a wide range of temperatures. Such data are fundamental for a detailed understanding of energy transfer in elementary collisional processes.

State-to-state cross sections and rates derived from quantum calculations are highly sensitive to the intermolecular potential energy surface (PES) employed in the calculation. Accurate experimental cross sections or rates therefore allow one to assess the accuracy of different PESs. However, the experimental methods developed so far only yield indirect information about state-to-state quantities. Therefore, testing the quality of a PES by comparison with experimental state-to-state quantities is indeed a difficult problem whose solution demands a close cooperation of theoreticians and experimentalists. Perhaps the paradigm of this problem is molecular hydrogen, with much theoretical and experimental material available, which is not of immediate interpretation in most cases. Here we tackle the problem from both sides, experiment and theory, exploring novel experimental techniques with the support of close-coupling scattering calculations based on a recent ab initio PES for H₂–H₂.

Among the environments which depend on nonreactive collisions with H₂ molecules, the interstellar space and the molecular clouds have attracted much attention in recent times. Modeling such rarefied gas media, which are under severe nonequilibrium conditions, is a central problem of present day molecular astrophysics. Other topics such as teledetection based on collisional line broadening, quantification of molecular transport properties, and relaxation phenomena are still open problems in spite of the progress in kinetic theory and computation power during the last three decades.

The range \( 5 \leq T \leq 300 \) K investigated here for molecular hydrogen spans the lower end of the thermal domain of astrophysical interest.

II. METHODOLOGY

The experimental part of this work relies on the master equation (MEQ)

\[
\frac{dP_i}{dt} = n \sum_{j,l=0}^{m} Q_{ij\ell m}(-P_{ij}P_{ij\ell\ell} + P_{ij}P_{m\ell\ell\ell}),
\]

which describes the time evolution of the population \( P_i \) of a rotational level with quantum number \( J = i \) as a consequence of inelastic collisions in a gas of diatomic molecules at instantaneous number density \( n \) and translational or kinetic temperature \( T_r \). The factor

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accounts for the symmetry effects in collisions between indistinguishable molecules.\textsuperscript{4} Indistinguishable molecules will be considered for pH\textsubscript{2}:pH\textsubscript{2} and oH\textsubscript{2}:oH\textsubscript{2} collisions, and distinguishable molecules for oH\textsubscript{2}:pH\textsubscript{2} collisions.

The K’s in MEQ (1) are the so called state-to-state collisional rates (also excitation and deexcitation rates), quantities which account for the elementary collisional process,

\[ H_2(i) + H_2(j) \rightarrow H_2(\ell) + H_2(m), \]

between a H\textsubscript{2} molecule in the rotational state \( J = i \) and another H\textsubscript{2} molecule in the rotational state \( J = j \), in a medium at translational temperature \( T_t \). As a consequence of the collision, the final states of the two molecules become \( J = \ell \) and \( J = m \).

The collisional rates obey the detailed balance

\[ k_{i\ell \rightarrow ij} = k_{i\ell \rightarrow \ell m}, \]

where \( E_i, E_j, E_\ell, \) and \( E_m \) are the energies of the rotational levels involved. For \( E_i + E_j > E_\ell + E_m \), the upward (“up”) and downward (“down”) rates will henceforth be labeled as \( k_{\text{up}}^{ij \rightarrow \ell m} \) and \( k_{\text{down}}^{ij \rightarrow \ell m} \), respectively.

The medium considered here is the gas flowing along the axis of a supersonic expansion with space-time dependent quantities \( n(z,t), P_j(z,t), \) and \( T_j(z,t) \). The particular case studied is natural molecular hydrogen (nH\textsubscript{2}), a 3:1 mixture of orthohydrogen (oH\textsubscript{2}) and parahydrogen (pH\textsubscript{2}) with \( J = \text{odd} \) and \( J = \text{even} \), respectively. These variants of H\textsubscript{2} do not interconvert in the time scale of the supersonic expansion.

The rotational populations in Eq. (1) are then normalized as

\[ \sum_{i=\text{odd}} P_i = 3/4, \quad \sum_{i=\text{even}} P_i = 1/4. \]

Since up and down rate coefficients at temperature \( T_t \) may differ by many orders of magnitude, it is convenient to reformulate the MEQ (1) in terms of only down rate coefficients, which show a far smoother dependence on \( T_t \). With aid of Eq. (4), the MEQ (1) is then formally rewritten as

\[ dP_j/dt = \sum_{\tau \omega \rho} \pm a_{\tau \omega \rho} P_{\tau \omega \rho}^{\downarrow}, \]

with explicit form and \( a_{\tau \omega \rho} \) coefficients defined in the Appendix in terms of the instantaneous experimental quantities \( n, P_j, \) and \( T_j \). The indices \( \tau, \omega, \rho, \) and \( \sigma \) run over the values of the rotational quantum number \( J \) allowed by ortho-para nonconversion, with at least one of the indices being equal to \( i \). Only \( \tau = 0, \pm 2 \) and \( \omega = 0, \pm 2 \) processes are relevant for nH\textsubscript{2} at \( T_t = 300 \) K. Each \( a_{\tau \omega \rho}P_{\tau \omega \rho}^{\downarrow} \) term in Eq. (6) accounts in a factorized form (instantaneous local properties \( \times \) intermolecular properties) for the net contribution of the \( \sigma \rho \rightarrow \tau \omega \) and \( \tau \omega \rightarrow \sigma \rho \) time-symmetric processes to the population rate \( dP_j/dt \).

The rotational population rates \( dP_j/dt \) as well as the \( a_{\tau \omega \rho} \) coefficients are determined directly from the experiment as discussed below. In turn, the \( k_{\text{up}}^{\tau \omega \rho} \)’s may be obtained from first principle calculations or, with some limitations, from the experiment as shown in Sec. V.

### III. Experiment

The experimental quantities involved in the \( a_{\tau \omega \rho} \) coefficients of Eq. (6) and in its left-hand term \( dP_j/dt \) were determined from the intensities of the \( Q(0), Q(1), Q(2), \) and \( Q(3) \) Raman lines of H\textsubscript{2} at 4161.2, 4155.3, 4143.5, and 4125.9 cm\textsuperscript{-1}, respectively. These intensities were measured at a number of points located along the axis of supersonic expansions of nH\textsubscript{2}, at distances \( z \) from the nozzle orifice. In this region (zone of silence) the flow is laminar, and the distance \( z \) can be related with the expansion time \( t \) by means of the flow velocity \( V(z) = dz/dt \). The population rates are then determined from

\[ \frac{dP_j}{dt} = \frac{dP_j}{dz} V(z). \]

The quantities \( P_j, dP_j/dz, \) and \( n \) needed in Eq. (6) were measured as described in earlier works,\textsuperscript{18,22} while \( V(z) \) was determined assuming the conservation of enthalpy along the expansion axis.

The translational temperature \( T_t \) at each position \( z \) of the expansion axis has been deduced in terms of the corresponding local number density and rotational populations by means of

\[ T_t(z) = T_0 \left[ \left( \frac{n(z)}{n_0} \right) \prod_j \left( \frac{P_j(z)}{g_j(2J+1)} \right)^{P_j(z)} \right]^{2/3}, \]

with

\[ C = \prod_j \left( \frac{P_j^0}{g_j(2J+1)} \right)^{-P_j^0}, \]

where \( g_j = 1 \) for \( J = \text{even} \) and \( g_j = 3 \) for \( J = \text{odd} \) are the nuclear spin weights of pH\textsubscript{2} and oH\textsubscript{2}, respectively; \( n_0 \) is the stagnation number density, and \( P_j^0 \) are the rotational populations at the stagnation temperature \( T_0 \). Equation (8) has been derived from the statistical definition of entropy, assuming entropy invariance along the axis of the supersonic expansion.

Two independent data sets were measured from two jets of 99.9999% pure nH\textsubscript{2} generated at stagnation pressures \( p_0 = 1.28 \) and 2.55 bars at \( T_0 = 295 \) K, expanding the gas through a circular nozzle of diameter \( D = 313 \) \( \mu \)m. The remaining experimental conditions are similar to those of Ref. 18. The experimental quantities \( n(z), V(z), \) and \( P_j(z) \) are shown in Figs. 1 and 2. The rotational temperatures of oH\textsubscript{2} and pH\textsubscript{2} have been measured from the ratio of populations \( \nu_{1}\nu_{1}/\nu_{2}\nu_{2} \) and \( \nu_{2}\nu_{2}/\nu_{1}\nu_{1} \), with the constraints of Eq. (A1) (Appendix). The rotational and translational temperatures are shown in Fig. 3. The translational temperatures resulting from Eq. (8) are very similar for both expansions. On the contrary, the rotational temperatures differ by up to \( \approx 15 \) K, always with \( T_r(\text{pH}_2) < T_r(\text{oH}_2) \).

\[ \sum_{\text{odd}} P_i = 3/4, \quad \sum_{\text{even}} P_i = 1/4. \]
The population rates \( dP_i/dt \) were determined by means of Eq. (7), employing the flow velocities of Fig. 1. The slopes \( dP_i/dz \) were obtained numerically from the populations \( P_i(z) \) of Fig. 2. For optimal accuracy, smoothing functions of the form

\[
P_i(z) = a_i(z + z_0) + b_i + c_i(z + z_0)^2
\]

were fitted to consecutive seven-point sets. The coefficients \( a_i, b_i, \) and \( c_i \) are different for each set while \( z_0 \) accounts for the true origin of the expansion, the so-called sonic surface, where the Mach number reaches the value \( M = 1 \). In the present expansions it appears to be located at \( =100 \) and \( =200 \) \( \mu \)m inside the nozzle for the expansions at 1.28 and 2.55 bars, respectively. The experimental population rates \( dP_0/dt \) and \( dP_1/dt \) for the ground rotational states of pH\(_2\) and oH\(_2\) along the \( p_0=1.28 \) bar expansion are shown in Fig. 4. The uncertainties in Fig. 4 are two-standard deviations from statistical errors plus an additional 10% for possible systematic errors not accounted for by the statistics.

IV. THEORETICAL CROSS SECTIONS AND RATES

Comparing theory with experiment is a useful exercise for understanding in detail the elementary inelastic collisional processes in hydrogen. To this purpose a number of scattering calculations have been carried out within the framework of the quantum mechanical coupled channels method.\(^5\)\(^{19}\) In these calculations a rigid rotor potential energy surface (Diep and Johnson)\(^20\) has been employed on a grid of total energies from 354.35 to 5000 cm\(^{-1}\). At least two closed \((i, j)\) “well-ordered” rotational levels (i.e., with \( i \approx j \))
The relevant down rate coefficients of H$_2$ calculated from the PES of Diep and Johnson$^{20}$ are shown in Table I, where $\beta=hc/k_B=1.4388$ K/cm$^{-1}$ and B is the rotational constant of H$_2$. The homologous rates reported by Monchick and Schaefer$^5$ for 60$^{\text{a}} \lesssim T_e \lesssim 150$ K are within 10% of those reported in Table I, while several rates at 100 and 300 K calculated by Le Bourlout et al.$^{25}$ using Schwenke’s PES (Ref. 30) agree within 25%.

As can be seen in Table I, from 5 to 300 K the $k_{\text{down}}$ rates only vary by one order of magnitude, facilitating an overview of the different contributions to the collisional energy transfer for various thermal conditions such as, for instance, along supersonic expansions.

V. DISCUSSION

The experimental and the theoretical results described above will be compared below according to two different approaches: the direct problem and the inverse problem.

In the direct problem a set of preexisting rate coefficients obtained, for instance, from theoretical calculations, or from an hypothetical independent experiment, will be assessed by means of the spectroscopic results from a supersonic expansion, according to the methodology of Sec. II.

In the inverse problem we will discuss how to obtain information about the rate coefficients from the supersonic expansion with as little aid as possible from preexisting theoretical values. In the ideal case of a perfect set of rate coefficients and an error-free experiment, the quantitative conclusions of the direct and inverse problems should be identical. In a real case, however, the various sources of uncertainty in the experiment and in the theoretical calculation impair such an ideal convergence. Nonetheless, much knowledge can be gained about the elementary collisional processes by comparing the conclusions from the direct and

\begin{table}
\centering
\caption{Calculated $k_{\text{down}}$ rate coefficients for H$_2$:H$_2$ collisions (units of $10^{-20}$ m$^3$/s).}
\begin{tabular}{|c|c|c|c|c|c|c|c|c|c|c|c|c|c|c|c|}
\hline
$\Delta E/k_B$ & 4$\beta$B & 4$\beta$B & 6$\beta$B & 6$\beta$B & 6$\beta$B & 8$\beta$B & 10$\beta$B & 10$\beta$B & 10$\beta$B & 10$\beta$B & 12$\beta$B & 16$\beta$B \\
\hline
\hline
 300 & 864 & 631 & 167 & 283 & 281 & 291 & 258 & 135 & 164 & 164 & 177 & 4 & 5 \\
 200 & 681 & 502 & 112 & 189 & 190 & 195 & 164 & 83 & 96 & 97 & 105 & 2 & 3 \\
 150 & 574 & 428 & 86 & 145 & 146 & 151 & 122 & 61 & 68 & 70 & 75 & 1 & 2 \\
 125 & 514 & 389 & 74 & 124 & 125 & 129 & 103 & 51 & 56 & 58 & 62 & 1 & 1 \\
 100 & 452 & 350 & 63 & 104 & 106 & 109 & 86 & 42 & 45 & 47 & 51 & 1 & 1 \\
 90 & 428 & 333 & 58 & 96 & 98 & 102 & 81 & 39 & 41 & 43 & 47 & 1 & 1 \\
 80 & 403 & 319 & 54 & 89 & 91 & 94 & 75 & 36 & 39 & 43 & 47 & 1 & 1 \\
 70 & 379 & 303 & 50 & 82 & 84 & 88 & 69 & 33 & 34 & 36 & 39 & 1 & 1 \\
 60 & 356 & 292 & 47 & 76 & 78 & 81 & 64 & 30 & 31 & 33 & 36 & 1 & 1 \\
 50 & 334 & 281 & 43 & 70 & 73 & 76 & 61 & 28 & 28 & 31 & 33 & 1 & 1 \\
 40 & 319 & 279 & 40 & 65 & 68 & 71 & 61 & 27 & 26 & 29 & 32 & 0 & 1 \\
 30 & 311 & 290 & 38 & 61 & 64 & 65 & 67 & 26 & 25 & 28 & 31 & 0 & 1 \\
 20 & 330 & 359 & 39 & 60 & 64 & 60 & 86 & 28 & 27 & 30 & 33 & 0 & 1 \\
 10 & 469 & 761 & 46 & 67 & 73 & 69 & 246 & 41 & 49 & 49 & 1 & 1 \\
 5 & 979 & 3601 & 63 & 83 & 94 & 92 & 2035 & 89 & 93 & 95 & 107 & 1 & 2 \\
\hline
\end{tabular}
\begin{flushright}
$\beta B=84.98$ K.
\end{flushright}
\end{table}
inverse problems. This is particularly relevant considering how difficult it is to retrieve significant information about the individual state-to-state rates for H₂: H₂ collisions from the experiments reported so far.

A. The direct problem

The rate coefficients of H₂ calculated in Sec. IV are assessed in the range of kinetic temperatures 10 < T ≤ 140 K by means of the two supersonic expansions of nH₂ described in Sec. III. The rate coefficients k_{\text{down}}^{\text{corr}} to be tested are summarized in Table I, while the corresponding a_{\text{corr}}’s are given in the Appendix in terms of the experimental quantities n, P, and T along the expansions.

The population rates dP_{J}/dt and dP_{J}^{\text{d}}/dt measured at the corresponding points of the expansions yield the left-hand term (LHT) of the MEQ (6). On the other hand, the sum of products a_{\text{corr}}k_{\text{corr}} are given in Eqs. (A3) and (A4) of the Appendix yields the right-hand term (RHT) of the MEQ (6).

Comparing the experimental LHTs with the theoretical experimental RHTs for the J=0 and J=1 rotational levels of 1.28 and 2.55 bar expansions provides a consistency test of the calculated rates with the experiment. This comparison is shown in Fig. 4 for the 1.28 bar expansion. Similar results are obtained for the 2.55 bar expansion.

A summary of the statistics of the experiment is given in Table II. There the ratio LHT/RHT is employed as a figure of merit, whose ideal value should be LHT/RHT=1; N_{\text{data}} are the number of experimental data points from both expansions, scrambling information about the J=0 and J=1 rotational levels in the referred thermal range.

The statistics of Table II suggest that the global accuracy of the k_{\text{corr}} rates of Table I is likely to be better than 10% for 90 < T < 140 K, better than 20% for 40 < T < 90 K and better than 25% for 10 < T < 40 K. Since the uncertainties of the experimental population rates dP_{J}/dt, coefficients a_{\text{corr}} and calculated k_{\text{corr}}’s are highly uncorrelated, the empirical ratio 0.99 < LHT/RHT < 1.06 from Table II for a set of N_{\text{data}}=66 data points also suggests that the systematic overall uncertainty of the experiment is probably less than 10%.

No information about the relative weight and accuracy of the individual collisional contributions to the population rates can be inferred from Table II. However, such information can be deduced primarily from the numerical values of the a_{\text{corr}} coefficients defined in Eqs. (A5) (Appendix), in combination with Table I. This shows that 21=01 and 30=12 are the dominant collisional induced transitions contributing to dP_{J}/dt, with approximate relative weights of 60% and 40%. The remaining collisions contribute just marginally, their overall weight decreasing rapidly with the energy gap involved in the collision and with the precollisional rotational populations. For the dP_{J}/dt rate, the dominant processes are 30=12 and 31=11, with relative weights of 70% and 30% and marginal contribution from other collisions. Thus the figure of merit LHT/RHT of Table II largely refers to the 21=01, 30=12, and 31=11 contributions, in this order of importance. The corresponding rates on Table I are thus expected to be accurate to ≈10% for 90 < T < 140 K, ≈15% for 40 < T < 90 K, and ≈20% for 10 < T < 40 K. Little can be said about the remaining rates, apart from their marginal importance in the thermal range of the experiment, 10 < T < 140 K and 140 < T < 210 K.

The above information may be represented in terms of collisional spectra showing in detail the contribution of the various collisions to the population rates dP_{J}/dt of pH₂ and dP_{J}/dt of OH₂ at any point along the expansion of nH₂. As an example the collisional spectra at a particular point of the 1.28 bar expansion are shown in Fig. 5.

In the collisional spectra the abscissa (energy) represents the global energy ΔE=k_Bβ[B(σ(σ+1)+ρ(ρ+1)−1 τ+1)−ω(ω+1)] involved in the τω→σp collisionally induced

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TABLE II. Statistics of the experiment.

<table>
<thead>
<tr>
<th>Range (K)</th>
<th>LHT/RHT</th>
<th>N_{\text{data}}</th>
</tr>
</thead>
<tbody>
<tr>
<td>90 &lt; T ≤ 140</td>
<td>1.03(6)</td>
<td>16</td>
</tr>
<tr>
<td>40 &lt; T ≤ 90</td>
<td>1.06(14)</td>
<td>16</td>
</tr>
<tr>
<td>20 &lt; T ≤ 40</td>
<td>0.99(20)</td>
<td>16</td>
</tr>
<tr>
<td>10 &lt; T ≤ 20</td>
<td>1.02(20)</td>
<td>18</td>
</tr>
</tbody>
</table>

*One standard deviation in parentheses.*
transition. Positive $\Delta E$’s stand for globally up (or excitation) processes and negative $\Delta E$’s stand for globally down (or deexcitation) processes. Note that some globally up or down processes are mixed “up-down” or “down-up” processes with simultaneous excitation of one molecule and deexcitation of the other. Symmetric $\pm \Delta E$’s correspond to the time-symmetric $aP = -\tau a$ processes.

The ordinate (intensity) shows the contribution of a particular collision to the $J=0$ and $J=1$ rotational population rates. According to the scheme

$$P_J(t + \Delta t) = P_J(t) + \frac{dP_J}{dt} \Delta t,$$  \hspace{1cm} (15)

positive intensities imply an increase of population in the rotational ground levels in time, i.e., an elementary cooling down process. Negative intensities are associated with a decrease of population of the ground levels, i.e., an elementary warming up process. Note that in each spectrum the sum of all positive and negative intensities yields a positive balance, accounting for the global rotational cooling of both species, $\text{pH}_2$ and $\text{oH}_2$, in the mixture of expanding $\text{nH}_2$.

In the collisional spectra associated with the zone of silence of supersonic expansions the unbalance between positive and negative intensities increases at larger distances along the supersonic expansion and tends to vanish asymptotically at negative short distances, i.e., inside the nozzle, in the preexpansion chamber. At this point, positive and negative intensities become equal, the spectra showing inversion symmetry with respect to the origin. This reflects the thermodynamical equilibrium at the stagnation temperature and density, with a pictorial representation of the collisional status in detail.

The corresponding spectra at the same point of the 2.55 bar expansion are qualitatively similar to but somewhat more intense than those of the 1.28 bar expansion shown in Fig. 5. In general the global intensity of the collisional spectra decreases with distance $z$ by about the same factor of Fig. 4, accompanied by moderate variations in the relative intensities associated with the various collisional transitions. This smooth trend may be expected to show a dramatic change at the normal shock wave associated with the end of the zone of silence of the expansions, where the flowing gas becomes suddenly rethermalized.

The spectra prove useful to represent graphically the collisional evolution of any nonequilibrium state regardless of how complex it may be, as for instance, within a shock wave. Note that the spectra are not universal, even under thermodynamical equilibrium, since they depend on the particular colliding species through the state-to-state rates, which are specific to the intermolecular potential.

1. para-$\text{H}_2$ rotational cooling in expanding $\text{nH}_2$

The spectrum in Fig. 5(a) shows that the dominant collisions contributing to the $dP_0/dt$ rate are the $21 \rightarrow 01$ ones, with a positive contribution. The time-symmetric collisions $01 \rightarrow 21$ have a far smaller negative contribution. The net balance is positive, implying that the net effect of the $21 \rightarrow 01$ collisions strongly increases the population rate of the ground level along the expansion, i.e., the efficiency of the rotational cooling of $\text{pH}_2$.

It is somewhat surprising that the next most important contribution to $dP_0/dt$ is due to the $30 \rightarrow 12$ and $12 \rightarrow 30$ collisions, with a net negative contribution amounting to $\approx -0.25$ to $\approx -0.50$ of the main one described above. Since all $k_{\tau \rightarrow \tau'}$’s are positive by definition, this happens because the coefficient $a_{3012}$ is negative. Thus the $30 \rightarrow 12$ and $12 \rightarrow 30$ collisions tend to produce a net rotational heating of the $\text{pH}_2$ fraction of expanding $\text{nH}_2$. Such a peculiar behavior is to be expected for up-down or down-up collisions where one of the colliding partners becomes excited and the other one becomes simultaneously deexcited by a different amount of energy.

The $20 \rightarrow 00$ collisions in $\text{nH}_2$ contribute to the cooling rate with only 15%–20% of the main contribution $21 \rightarrow 01$, while $22 \rightarrow 02$ with only 6%–10% and $23 \rightarrow 03$ with less than 4% of the main contribution, as shown in Fig. 5(a).

Note that the reversal collisions, at symmetric negative values of the energy gap $\Delta E$, also reverse the cooling down, or the heating up effects, but attenuated, reflecting the intrinsic nonequilibrium character of the supersonic expansion. The remaining collisions contributing to $dP_0/dt$ [see Eq. (A3) (Appendix)] only have a marginal effect on changes in the $J=0$ rotational populations of expanding $\text{nH}_2$.

2. ortho-$\text{H}_2$ rotational cooling in expanding $\text{nH}_2$

As shown in Fig. 5(b) the main contributions to $dP_1/dt$ are positive. The dominant ones are the down-up $30 \rightarrow 12$ and the up-down $12 \rightarrow 30$ collisions. In contrast to pH$_2$, the net contribution of these collisions in oH$_2$ is the rotational cooling of the oH$_2$ fraction of expanding nH$_2$. The next contribution, amounting to about 50% of the former, arises from the down $31 \rightarrow 11$ and the up $11 \rightarrow 31$ collisions. The down $30 \rightarrow 10$ and the up $10 \rightarrow 30$ collisions only contribute with $\approx 10\%$ to $\approx 15\%$ of the dominant ones, while the remaining collisions have a marginal contribution in the investigated thermal range. The down-up $41 \rightarrow 23$ and the up-down $23 \rightarrow 41$ collisions have a marginal contribution in spite of the large $k_{41-23}$ rate and the small total energy gap of $4BR$ (see Table I) involved in these collisions, which are compensated by the very small value of the accompanying $a_{4123}$ coefficient.

On the whole, in a sample of expanding nH$_2$ the global $dP_0/dt$ rate is positive and somewhat larger than the $dP_1/dt$ rate, as shown in Figs. 4 and 5. This implies that the rotational cooling efficiency of pH$_2$ in nH$_2$ is higher than that of oH$_2$, leading to the $T_r(pH_2) < T_r(oH_2)$ breakdown of rotational temperatures shown in Fig. 3.

It may be concluded from this analysis that some down-up and up-down collisions are as relevant as or even more relevant for the rotational transfer of energy than the single down and up collisions involving a passive molecule. This effect is expected to be increasingly important at higher temperatures.
B. The inverse problem

Equations (A3) and (A4) of the Appendix tell us that at a point $z$ of a particular $n\text{H}_2$ expansion characterized by $n(z)$, $P_\zeta(z)$, and $T_\zeta(z)$, the number of unknowns $k_{\tau\alpha\beta\gamma}$ is by far larger than the just two independent equations available, for $dP_\eta/dt$ and $dP_\zeta/dt$. This a priori unfavorable situation can, in principle, be overcome by juxtaposing data from other expansions at different stagnation conditions, up to completing a linear system with more equations than $k_{\tau\alpha\beta\gamma}$ unknowns. This strategy is impaired in practice, however, due to the strong correlation between homologous equations from different expansions and by the uncertainties of the experimental $dP_\eta/dt$’s and $a_{\alpha\beta\gamma}$ coefficients.

As inferred from the direct problem, Fig. 5 shows that only three families of collisions contribute significantly to $dP_\eta/dt$ and $dP_\zeta/dt$: $30 = 12$, $2m = 0m$, and $3m = 1m$, with $m = 0, 1, 2, 3$. Other contributions to the $dP_\eta/dt$’s are negligible since the corresponding $k_{\tau\alpha\beta\gamma}$ (given in Table I) and/or the accompanying $a_{\alpha\beta\gamma}$ coefficients are small.

At the present level of experimental accuracy the only way to retrieve meaningful information about the significant $k_{\tau\alpha\beta\gamma}$ rate coefficients from the supersonic expansion experiment is to impose some qualitative constraints on the basis of the above given arguments. Two sorts of constraints will be considered here: (i) over the $k_{\tau\alpha\beta\gamma}$ unknowns, taking into account the theoretical results of Table I, and (ii) over the $a_{\alpha\beta\gamma}$ coefficients, according with their experimental values.

On this basis the following effective unknowns are proposed for $n\text{H}_2$: $n\text{H}_2$ collisions at $10 \leq T_i \leq 140$ K and $140 \leq T_i \leq 210$ K:

$$k_{30-12},$$

$$k_{21-01} = k_{22-02} = k_{23-03} = 3/2k_{20-00},$$

$$k_{31-11} = k_{32-12} = k_{33-13} = k_{30-10},$$

neglecting all other $a_{\alpha\beta\gamma}k_{\tau\alpha\beta\gamma}$ terms.

In addition to the former constraints, one may observe from Table I that the $k_{\tau\alpha\beta\gamma}$’s are smooth functions of the translational temperature. This suggests partitioning the range of temperature $10 \leq T_i \leq 140$ K into several subranges, augmenting the number of equations. Each subrange is then characterized by an average value of the three representative unknowns $\langle k_{21-01} \rangle$, $\langle k_{30-12} \rangle$, and $\langle k_{31-11} \rangle$, with one equation per measured temperature point within the thermal subrange. The analysis of the system for each thermal subrange shows the strong correlation between $k_{30-12}$ and $k_{31-11}$, which cannot be broken down due to the relatively high experimental uncertainty. Only the combination $k_{30-12} + bk_{31-11}$, with $b = 5$, is well defined under the present experimental conditions, the most relevant of which is the 3:1 fixed ratio of $\text{O}_2$ to $\text{H}_2$ nuclear spin variants in $n\text{H}_2$.

The mean-square solution of the corresponding overdetermined systems for the four thermal subranges covering from $T_i = 10$ up to $140$ K is shown in Table III. The quoted experimental uncertainties include an estimate of the experimental error as well as the variation of $\langle k_{21-01} \rangle$ or $\langle k_{30-12} \rangle$ + $5\langle k_{31-11} \rangle$ within the corresponding thermal subrange; the uncertainty of the theoretical values only considers the latter variation. The results of Table III, although limited in practical terms, confirm again the good agreement of theory with experiment for the dominant rates $k_{21-01}$, $k_{30-12}$, and $k_{31-11}$.

C. Comparison with other experiments

Due to the contribution from resonant rotation-rotation, rotation-translation, and elastic vibrational dephasing processes, the interpretation of line broadening coefficients of Raman lines of $\text{H}_2$ in terms of state-to-state rate coefficients rests upon models.11 Therefore a comparison with present experiments will not be attempted here. On the other hand, relaxation times from ultrasonic absorption/dispersion experiments provide a simpler material to compare with. For two-level systems such as $\text{O}_2$ and $\text{H}_2$ at temperature $T_i \leq 300$ K, the relaxation time may be expressed as

$$\tau_{\alpha\beta\gamma} = \frac{1.363 \times 10^{-28}T_i}{k_{\alpha\beta\gamma} + k_{\beta\alpha\gamma}},$$

where $(\rho r)$ is in atm s for $T_i$ in Kelvin and $k$’s in $10^{-20}$ m$^3$ s$^{-1}$; $k_{\alpha\beta\gamma}$ refer to the $20 = 00$ rates of $\text{H}_2$; the $\text{H}_2$ collisions, $31 = 11$ rates of $\text{O}_2$; $\text{H}_2$ collisions, and $21 = 01$ rates of $\text{H}_2$; $\text{O}_2$ collisions interchanging energy between the rotational levels of the vibrational ground state. The relaxation times calculated from the rates of Table I are shown in Table IV jointly with experimental results from ultrasonic measurements. This comparison confirms the good quality of the PES (Ref. 20) employed in the present close-coupling calculation. Tables III and IV also confirm indirectly the consistence of present experiment with the ultrasonic relaxation time measurements.

VI. SUMMARY AND CONCLUSIONS

Some aspects of the complexity of the collisional problem of $\text{H}_2$ are shown in this work. The experimental methodology employed is based on a fundamental kinetic equation, the so called “master equation” [Eq. (1)], which appears to be free from interpretative ambiguities for binary collisions. In it the state-to-state rate coefficients appear explicitly

<table>
<thead>
<tr>
<th>Subrange (K)</th>
<th>$\langle k_{21-01} \rangle$</th>
<th>$\langle k_{30-12} \rangle + 5\langle k_{31-11} \rangle$</th>
</tr>
</thead>
<tbody>
<tr>
<td>90 &lt; $T_i$ &lt; 140</td>
<td>115 ± 31</td>
<td>776 ± 159</td>
</tr>
<tr>
<td>117 ± 22</td>
<td>750 ± 114</td>
<td></td>
</tr>
<tr>
<td>133 ± 24</td>
<td>703 ± 129</td>
<td></td>
</tr>
<tr>
<td>40 &lt; $T_i$ &lt; 90</td>
<td>75 ± 19</td>
<td>555 ± 32</td>
</tr>
<tr>
<td>81 ± 15</td>
<td>544 ± 86</td>
<td></td>
</tr>
<tr>
<td>97 ± 13</td>
<td>513 ± 63</td>
<td></td>
</tr>
<tr>
<td>20 &lt; $T_i$ &lt; 40</td>
<td>62 ± 15</td>
<td>429 ± 36</td>
</tr>
<tr>
<td>63 ± 3</td>
<td>501 ± 45</td>
<td></td>
</tr>
<tr>
<td>10 &lt; $T_i$ &lt; 20</td>
<td>59 ± 15</td>
<td>429 ± 49</td>
</tr>
<tr>
<td>64 ± 5</td>
<td>613 ± 150</td>
<td></td>
</tr>
</tbody>
</table>

*Experimental, this work.

{Theory, this work.

{Theory, adapted from Ref. 2.

$10^{-20}$ m$^3$ s$^{-1}$}
TABLE IV. Calculated and experimental relaxation times in H₂ collisions (pτ) in units of 10⁻¹⁰ atm s.

<table>
<thead>
<tr>
<th>Tₜ (K)</th>
<th>Transitions</th>
<th>Expt. (pτ)</th>
<th>Calc. (pτ)</th>
</tr>
</thead>
<tbody>
<tr>
<td>300</td>
<td>20=00</td>
<td>130⁶</td>
<td>128</td>
</tr>
<tr>
<td>170</td>
<td>187±2⁵</td>
<td>192</td>
<td></td>
</tr>
<tr>
<td>111.5</td>
<td>223±4⁵</td>
<td>212</td>
<td></td>
</tr>
<tr>
<td>90.5</td>
<td>211±2⁵</td>
<td>207</td>
<td></td>
</tr>
<tr>
<td>77</td>
<td>220±8⁵</td>
<td>197</td>
<td></td>
</tr>
<tr>
<td>300</td>
<td>31=11</td>
<td>226⁶</td>
<td></td>
</tr>
<tr>
<td>30=10</td>
<td>231</td>
<td></td>
<td></td>
</tr>
<tr>
<td>77</td>
<td>21=01</td>
<td>121±5⁶</td>
<td>120</td>
</tr>
</tbody>
</table>

Notes:
1. This work, from Table I.
2. Reference 15.
5. Weighted over 31=11 and 30=10 in nH₂.

as four-index rates which are free from averages over rotational states. Each index refers to a clearly identified rotational quantum state of both colliding molecules before and after the collision. This allows for unambiguous comparison with the results from high level quantum calculations, as well as with other experiments.

The following conclusions can be emphasized.

1. The supersonic expansion spectroscopic experiment spanning the range 10<Tₜ<140 K confirms to ±10% the accuracy of the k₃₀₋₁₂, k₂₁₋₀₁, and k₃₁₋₁₁ rate coefficients for H₂:H₂ collisions derived from close-coupled calculations based on the PES of Diep and Johnson.⁶

2. The rates k₃₀₋₀₀, k₂₁₋₀₁, k₃₀₋₁₀, and k₃₁₋₁₁ calculated here reproduce the relaxation time measurements from ultrasonic absorption experiments at 77<Tₜ<300 K (Refs. 13–15) to better than 5%.

3. Appropriate reorganization of the master equation in the form of Eq. (6) allows for a detailed pictorial representation of the collisional status of any medium of instantaneous number density n(t), kinetic temperature Tₜ(t), and rotational distribution of populations Pₖ(t) in the form of spectral diagrams as in Fig. 5. They are useful for visualizing the intricate internal energy transfer patterns due to collisions, regardless of equilibrium or nonequilibrium conditions.

4. It is shown by experiment and calculations that the 30=12 double process, with active participation of both molecules, one getting deexcited and the other getting excited in the collision by a different amount of energy, is as important as the single 21=01 and 31=11 processes where one of the molecules remains passive.

5. The net balance of 30=12 collisions in expanding nH₂ is a heating up of the pH₂ fraction and a simultaneous cooling down of the oH₂ fraction, at the same speed, as shown in Fig. 5. Paradoxically, this does not lead to a violation of the 3:1 abundance ratio of ortho-para species in nH₂. This ratio is always maintained along the expansion, as imposed in the master equation through the constraints given in the Appendix [Eq. (A1)].

6. A behavior different from 30=12 occurs for the 21=01 and 31=11 processes. Their net balance cools down, respectively, the pH₂ and oH₂ fractions of supersonically expanding nH₂.

7. The population or depletion of the J=0 level in nH₂ is mainly due to the 21=01 collisions in competition with the 30=12 ones. The role of the 20=00 processes is secondary, as shown in Fig. 5(a).

Concerning the experimental method the following conclusions can be drawn.

1. Raman intensity measurements in supersonic jets were proven suited for investigating the rotation-translation energy transfer by collisions in a thermal range hardly accessible to any other experimental method. Jets provide collisions under controlled nonequilibrium conditions, depending on stagnation conditions, nozzle, and position, while Raman spectroscopy allows for unprecedented high spatial resolution and accuracy at any point of the jet. The wide range of temperatures, 10=ₜ≤140 K, has been investigated here in a single experiment.

2. Nonetheless, the accuracy of the experimental results is still far from optimization. Translational temperatures in the experiment rest upon the hypothesis that the H₂ jets are isentropic. Nonisentropic behavior might bias the determined rates to an extent that is not possible to quantify at the present state of the research. The fact that the kinetic temperatures of two different expansions are nearly identical is interpreted as a proof in favor of isentropic behavior.

3. The main source of uncertainty in the upper end of the investigated thermal range, close to Tₜ=140 K in this case, arises from the difficulty in determining accurately the position of the sonic surface of the expansion (Mach=1), apparently located inside the nozzle by about 100–200 μm. This might cause some uncertainty in the initial flow velocity and in the population rates dP₁/dt at the beginning of the expansion.

4. The correlation between k₃₀₋₁₂ and k₃₁₋₁₁ could be broken down experimentally by expanding prepared H₂ samples with different oH₂:pH₂ ratios. This can be inferred from Eqs. (A5) where the relative values of the a₃₀₁₂ and a₃₁₁₁ coefficients are shown to depend on the oH₂:pH₂ ratio through the P₁, P₃, P₆, and P₂ rotational populations.

5. Encouragingly, the present experimental rates of H₂, the close-coupling calculations, and the ultrasonic relaxation time measurements are in good agreement, in spite of the former limitations and uncertainties

Much experimental work remains to be done in order to improve our knowledge about H₂:H₂ collisions, in particular, to narrow the margin of uncertainty, to widen the thermal range, and to break the correlations between some observable rates by expanding oH₂:pH₂ mixtures of proportion different from the 3:1 in nH₂.
ACKNOWLEDGMENTS

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APPENDIX: MASTER EQUATION COEFFICIENTS

In the thermal range investigated experimentally here, approximately $10 \leq T_r \leq 140 \, K$ and $140 \leq T_r \leq 210 \, K$, for the rotational levels of $H_2$ of $J=0, 1, 2, 3$, whose population can be measured with sufficient accuracy, the following constraints hold to a good approximation:

$$P_0 + P_2 = 1/4, \quad P_1 + P_3 = 3/4,$$  \hspace{1cm} (A1)

$$dP_2/dt = -dP_0/dt, \quad dP_3/dt = -dP_1/dt.$$  \hspace{1cm} (A2)

In the referred thermal range the relevant terms of the master equation (6) are expressed as a function of the $k_{\text{rot}}$ rates by

$$dP_0/dt = a_{3012}k_{3012} + a_{2002}k_{2000} + a_{2101}k_{2101} + a_{2202}k_{2202} + a_{3023}k_{3023} + a_{0422}k_{0422} + a_{4103}k_{4103} + a_{2000}k_{2000} + a_{3011}k_{3011},$$  \hspace{1cm} (A3)

$$dP_1/dt = a_{3223}k_{3223} + a_{3101}k_{3101} + a_{3111}k_{3111} + a_{3212}k_{3212} + a_{3313}k_{3313} - a_{4103}k_{4103} + a_{3012}k_{3012} - a_{3102}k_{3102} + a_{3112}k_{3112} + a_{0422}k_{0422} + a_{4103}k_{4103} + a_{2000}k_{2000} + a_{3011}k_{3011},$$  \hspace{1cm} (A4)

where the arrows are omitted for simplicity. The $a_{\text{rot}}$ coefficients are given in the rigid rotor approximation by

$$a_{3012} = n((7/15)P_1P_2 e^{-6BB/T} - P_3P_0),$$

$$a_{4123} = n((27/35)P_1P_2 e^{-4BB/T} - P_4P_1),$$

$$a_{2000} = n(-5P_0P_2 e^{-6BB/T} + P_2P_0),$$

$$a_{2101} = n(-5P_0P_2 e^{-6BB/T} - P_2P_1),$$

$$a_{2202} = n(-5P_0P_2 e^{-6BB/T} + P_2P_2),$$

$$a_{2303} = n(-5P_0P_3 e^{-6BB/T} + P_3P_3),$$

$$a_{0422} = n((9/25)P_2P_2 e^{-8BB/T} - P_4P_0),$$

$$a_{3111} = n(-(7/3)P_1P_0 e^{-10BB/T} + P_3P_0),$$

$$a_{3212} = n(-(7/3)P_3P_2 e^{-10BB/T} + P_3P_3),$$

$$a_{3313} = n(-(7/3)P_1P_3 e^{-10BB/T} + P_3P_3),$$

$$a_{4103} = n(-(27/7)P_0P_3 e^{-10BB/T} + P_1P_3),$$

$$a_{2200} = n(-(5P_0P_2 e^{-12BB/T} + 2P_3P_2),$$

$$a_{2301} = n(-(35/3)P_0P_3 e^{-16BB/T} + P_3P_3).$$  \hspace{1cm} (A5)