

1 **Could ammonium nutrition increase plant C-sink strength under elevated**
2 **CO₂ conditions?**

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18 **Abstract**

19 Atmospheric carbon dioxide (CO₂) is increasing, and this affects plant photosynthesis and
20 biomass production. Under elevated CO₂ conditions (eCO₂), plants need to cope with an
21 unbalanced carbon-to-nitrogen ratio (C/N) due to a limited C sink strength and/or the reported
22 constrains in leaf N. Here, we present a physiological and metabolic analysis of ammonium
23 (NH₄⁺)-tolerant pea plants (*Pisum sativum* L., cv. snap pea) grown hydroponically with
24 moderate or high NH₄⁺ concentrations (2.5 or 10 mM), and under two atmospheric
25 CO₂ concentrations (400 and 800 ppm). We found that the photosynthetic efficiency of the
26 NH₄⁺ tolerant pea plants remain intact under eCO₂ thanks to the capacity of the plants to
27 maintain the foliar N status (N content and total soluble proteins), and the higher C-skeleton
28 requirements for NH₄⁺ assimilation. The capacity of pea plants grown at 800 ppm to promote
29 the C allocation into mobile pools of sugar (mainly sucrose and glucose) instead of starch
30 contributed to balancing plant C/N. Our results also support previous observations: plants
31 exposed to eCO₂ and NH₄⁺ nutrition can increase of stomatal conductance. Considering the C
32 and N source-sink balance of our plants, we call for exploring a novel trait, combining NH₄⁺
33 tolerant plants with a proper NH₄⁺ nutrition management, as a way for a better exploitation of
34 eCO₂ in C3 crops.

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38 1. Introduction

39 To successfully achieve the goal of sustainable agriculture, it is necessary to maintain equilibrium
40 between the highest possible yield and product quality whilst ensuring minimal environmental impact.
41 Nitrogen (N) fertilisation is the main driver for crop yield, but its intensified application leads to severe
42 environmental pollution, including nitrate (NO_3^-) leaching, emissions of nitrous oxide (the most
43 significant ozone-depleting compound and a powerful greenhouse gas), as well as other forms of N that
44 are toxic for the environment [1]. It is therefore necessary to optimise the application of N fertilizers by
45 better understanding the nutritional requirements of plants. Through nitrification processes, NO_3^- is the
46 most common form in aerated soils; it could therefore be argued that modern breeding should select for
47 efficient NO_3^- assimilation metabolism rather than ammonium (NH_4^+). Nitrification inhibitors (NI),
48 which can stabilise large concentrations of NH_4^+ in the soil, pose new issues: in combination with NI,
49 genotypes that efficiently exploit NH_4^+ nutrition will have an advantage and, thus, this could be
50 considered an advantageous trait in terms of breeding. This capability needs to be evaluated in different
51 environmental scenarios.

52 Actual atmospheric CO_2 concentrations (aCO_2) are increasing despite the urgency of globally reducing
53 the emissions of CO_2 and other greenhouse gases [1]. The effect of increasing aCO_2 on plant
54 responsiveness, particularly photosynthetic performance, has been studied extensively for decades
55 (reviewed by [2]). More specifically, Rubisco (ribulose-1,5-bisphosphate carboxylase–oxygenase)
56 carboxylation in C_3 plants is suboptimal at aCO_2 , and the predicted increase in CO_2 will enhance
57 photosynthesis rates. However, the initial stimulation of photosynthesis is usually reversed in the long
58 term: a phenomenon known as photosynthetic acclimation to elevated CO_2 concentrations (eCO_2) or
59 photosynthetic down-regulation [3]. Among the multiple hypotheses put forward for explaining this
60 biochemical limitation, two are more widely accepted by the scientific community. The first is the C-
61 sink/source hypothesis [4], where the ability of Rubisco to fruitfully exploit an environment with higher
62 substrate for carboxylation is constrained by the limited plant capacity of increasing C sink strength,
63 therefore, plants reduce the photosynthetic capacity to adjust the whole plant C balance. The second one
64 is based on the reduction in the plant N content under eCO_2 because of a C dilution effect [5],
65 phenomenon recently attributed to a decrease in N acquisition across different environments [6],
66 probably in line with NO_3^- assimilation constrains occurring under eCO_2 [7]. In this context, [8] provide
67 a new photosynthetic model for computing the benefit of the C derived from photorespiration on net
68 CO_2 assimilation rates, in plants receiving NO_3^- as their N source and under certain conditions, such as
69 triose phosphate limitations. The constrains of NO_3^- assimilation under eCO_2 has been described in
70 many species and experiments, including bench experiments in *Arabidopsis* [9] and wheat plants in field
71 conditions [10]. Nevertheless, other authors such us [11] indicate that the depleted NO_3^- assimilation
72 capacity under eCO_2 is a source of debate.

73 Increasing NH_4^+ nutrition in crops with respect to NO_3^- has been studied for more than three decades,
74 mainly due to its reduced environmental impact and the fact that less energy is required for plant N
75 assimilation. However, NH_4^+ nutrition may be stressful because high doses of NH_4^+ can be toxic to
76 plants: reduced plant growth with high external NH_4^+ concentrations is a classic effect of NH_4^+ toxicity
77 [12]. More specifically, NH_4^+ nutrition is known to lead to a strong plant C/N imbalance promoted by
78 excessive internal NH_4^+ accumulation with a concomitant high C-skeleton and energy demand
79 (reviewed by Esteban [13]). In fact, by increasing the C-skeleton availability, i.e., higher light intensity,
80 in pea and ammonium-fed wheat plants ([14][15]) or higher CO_2 levels in *Arabidopsis* and ammonium-
81 fed wheat plants [16], the plant C/N balance is improved, and the NH_4^+ toxicity symptoms are alleviated
82 or minimised.

83 In this context, the hypothesis that “strict NH_4^+ nutrition under eCO_2 , could enhance C sink for satisfying
84 the high C skeletons requirement for NH_4^+ assimilation” arises. Furthermore, these conditions would
85 help to maintaining a balanced plant C/N ratio with higher Rubisco carboxylation rates. For this purpose,
86 we used NH_4^+ tolerant pea plants (*Pisum sativum* cv. snap pea; [17]), which were grown at a sufficient
87 N (2.5 mM NH_4^+) and high NH_4^+ concentration (10 mM NH_4^+) and at two levels of CO_2 . Here we suggest
88 that NH_4^+ nutrition could be considered as a promising N source alternative to face eCO_2 conditions
89 improving plant responsiveness by strengthening plant C sink. This article expects to give light about
90 the C/N metabolites management by a NH_4^+ -tolerant plant could be an strategy to cope with future CO_2
91 scenarios using NH_4^+ nutrition as alternative N source.

92

93 2. Material and Methods

94 Plant material, growth conditions and biomass determination

95 Seeds of an NH_4^+ -tolerant pea variety (*Pisum sativum* L., cv. sugar snap, [17]) were surfaced sterilised,
96 germinated at 26°C in for 96 h in the dark, in a perlite:vermiculite (1:2) substrate and grown
97 hydroponically. This genotype was chosen to avoid the masking effect of possible NH_4^+ toxicity
98 symptoms that other genotypes could show, therefore making it possible to study only the effect of
99 contrasting CO_2 concentrations under ample N fertilisation conditions. Modified nitrogen-free ‘Rigaud
100 Puppo’ solution was used [18]: 1.15 mM K_2HPO_4 ; 2.68 mM KCl; 0.7 mM CaSO_4 ; 0.07 mM Na_2Fe -
101 EDTA; 0.85 mM MgSO_4 ; 16.5 μM Na_2MoO_4 ; 3.7 μM FeCl_3 ; 3.4 μM ZnSO_4 ; 16 μM H_3BO_3 ; 0.5 μM
102 MnSO_4 ; 0.1 μM CuSO_4 ; 0.2 μM AlCl_3 ; 0.1 μM NiCl_2 ; 0.06 μM KI. The solution was adjusted to pH 6.5
103 by adding H_3PO_4 and was then buffered with CaCO_3 (0.5 mM). NH_4^+ was supplied as $(\text{NH}_4)_2\text{SO}_4$ at two
104 concentrations: 1.25 and 5 mM (i.e., 2.5 and 10 mM of NH_4^+ in nutrient solution). The one element that
105 remains unbalanced between the treatments is sulphur which has been reported to have minor effects on
106 the absorption of other mineral elements and, therefore, should not induce significant changes in the
107 nutritional status of the plants [19]. The nutrient solution was replaced twice a week to maintain the N
108 level, and the pH was kept at 6.5. The plants were cultured in two modified controlled-environment
109 chambers (Heraeus-Votsch HPS-500, Norrköping, Sweden) at two different CO_2 concentrations:
110 ambient CO_2 (a CO_2 ; 400 ppm) and elevated CO_2 (e CO_2 ; 800 ppm) \pm 5%. The growth chamber conditions
111 were 22/18°C (day/night), 65 % relative humidity and with a photoperiod of 16 hours and 300 $\mu\text{mol m}^{-2}$
112 s^{-1} photosynthetic photon flux density. The plants were grown under these conditions for 4 weeks. At
113 the end of this period, the plants were collected for determinations. Three independent experiments were
114 performed. The data presented is a combination of the three experiments.

115

116 At harvest, the plants were separated into roots and shoots. Samples were collected, frozen in liquid N_2
117 and stored at -80°C . Dry weights (DW) were determined by drying the plant material in an oven at 80°C
118 for 48 h until stabilisation.

119 Gas exchange determinations

120 Gas exchange in leaves was recorded in the last fully expanded leaf between 3 h and 8 h after the start
121 of the photoperiod using a portable GFS-3000 gas exchange system (Walz, Effeltrich, Germany). The
122 order of the measurement of the plants were complete randomized using a sample random sample
123 function in excel . The measurements were taken at an air flow rate of 300 ml min^{-1} , 25°C , 1000 μmol
124 $\text{m}^{-2}\text{s}^{-1}$ irradiance. Maximum photosynthesis (A_{max}) was recorded at 400 and 800 ppm CO_2 , depending
125 on the growing conditions. The A/Ci curves were constructed at 1000 $\mu\text{mol m}^{-2}\text{s}^{-1}$ irradiance and at the
126 following CO_2 levels: 400, 250, 125, 250, 400, 600, 800, 1000, 1200 ppm CO_2 . The A/Ci curves were

127 modelled according to [20]. Leaf steady-state respiration measurements were recorded 2 h after the
128 sunset in the chamber. Inlet CO₂ was adjusted to the treatment; with an air flow rate of 300 ml min⁻¹, 0
129 μmol m⁻²s⁻¹ irradiance.

130 **Respiratory capacity of the roots**

131 Root respiration measurements were taken using 0.05 g⁻¹ fresh weight (FW) and 0.5 to 1 cm-long root
132 cuttings using a Rank Oxygen electrode (LD2, Hansatech, UK) at 25°C in a total volume of 4 mL of
133 nutrient solution. The capacity of the cytochrome respiratory pathway was studied following the
134 application of 20 mM SHAM to the electrode chamber. The residual pathway capacity was measured
135 following the addition of 0.1mmol/L KCN. The alternative respiration capacity was calculated from the
136 difference between total respiration minus cytochromic and residual respiration.

137

138 **NH₄⁺, N, C and mineral content in plant tissues**

139 The soluble NH₄⁺ content in the cells was obtained by centrifuging (20,000×g, 30 min) tissue (approx.
140 0.2 g FW) incubated in 1mL of milli-Q water at 80°C in a bath for 5 min. The supernatants (leaves and
141 roots) were stored at -20°C until they were analysed using ion chromatography. The soluble cation
142 content was determined using an isocratic method with a 20 mM methanesulphonic acid solution as the
143 eluent in a Dionex-DX500 ion chromatograph (Dionex, CA, USA) with Ion Pac CG12A and Ion Pac
144 CS12A columns. Detection was performed by conductivity as above. The NH₄⁺ content was expressed
145 as mg g⁻¹ DW.

146 The content of N and C (%; w/w), as well as the other minerals, was calculated from the dry material.
147 Leaves and roots were ground in a mixer mill (MM200, Retsch, Haan, Germany). 2–3 mg of DW were
148 placed into tin capsules and analysed through Dumas combustion in an elemental analyser CNS 2500
149 (CE Instruments, Milan, Italy). The N₂ and CO₂ produced were detected by thermal conductivity.
150 Acetanilide was used as a standard in the total N content parameter. The C/N ratio was calculated by
151 dividing the percentage of C by the percentage of N. The mineral content was determined after acid
152 digestion using inductively coupled plasma/optical emission spectrometry (iCAP 6500 Duo, Thermo
153 Fisher Scientific, Cambridge, UK). The N-use efficiency (i.e., internal NUE) was calculated as the ratio
154 between the plant biomass (g) and the amount of N absorbed per plant (g).

155

156 **Compounds related to C and N metabolism**

157 **i) Soluble sugar, starch, and organic acid content**

158 Soluble carbohydrates (fructose, glucose, and sucrose) were extracted from roots and leaves (0.2 g of
159 FW) in boiling ethanol (80%, volume/volume). The ethanol-insoluble residue was dried and the starch
160 extracted, and the glucose produced by the amyloglucosidase enzyme was analysed as for soluble
161 carbohydrates [21]. Soluble sugars were expressed as $\mu\text{mol g}^{-1}$ of DW, and starch was expressed as
162 μmol of glucose g^{-1} DW. Fucose 0.5 mM was used as the internal standard in the extracts.

163 For organic acid determination, frozen (-80°C) pea leaf or root samples (0.2 g) were homogenised to a
164 fine powder in liquid N using a mortar and pestle. A 1.5 mL aliquot of 5% (weight/volume)
165 trichloroacetic (TCA) acid in water was added. The extracts were kept frozen at -20°C until use.
166 Succinate, malate, α -ketoglutarate, oxaloacetate, and citrate contents were determined via ion
167 chromatography in a DX-500 system (Dionex Corporation) by gradient separation using a
168 DionexIonPac AS11 (4 mm \times 250 mm) column and a Dionex ASRS Ultra II (4 mm) suppressor column
169 with the Dionex Ion-Pac ATC-3 (9 mm \times 24 mm) ion trap, and a pre-column Dionex Ion-Pac AG11 (4
170 mm \times 50 mm). The samples were injected with an AS40 autosampler (Dionex) at a 1:20 dilution in
171 milli-Q water. A 2 mL min. flow of solvent (methanol 18% NaOH 0.2 mM) was applied, and organic
172 acid separation was performed using a NaOH gradient (from 0.2 mM to 35 mM) for 16 min. Detection
173 involved a conductivity method in the electrochemical detector ED 40 (Dionex). Organic acid content
174 was expressed as mg g^{-1} DW.

175 **ii) Amino acid content**

176 Frozen plant tissue (0.1 g) was ground with liquid N_2 and homogenised with 1 ml HCl 1M. The extract
177 was centrifuged at $16000\times\text{g}$ and 4°C for 10 min. The supernatant was then pH adjusted to 7 with NaOH
178 and stored at -20°C . The amino acids were derivatised at room temperature for between 12 and 16 h with
179 FITC dissolved in 20 mM acetone/borate, at pH 10. Single amino acids were determined by high-
180 performance capillarity electrophoresis using a Beckman Coulter PA-800 apparatus (Beckman Coulter
181 Inc., Brea, CA, USA). The potential applied was -20 kV. The background buffer was 80 mM borax, 45
182 mM α -cyclodextrin, at pH 9.2. Because of the analytical method used, the asparagine and proline content
183 pooled in the same pick. The units were expressed as $\mu\text{mol g}^{-1}$ DW.

184

185 **Proteins and enzymatic activity determination**

186 **i) Total soluble protein and RuBisCo protein contents**

187 Frozen leaves (0.2 g) were homogenised in a mortar with liquid N_2 in a 0.1 M phosphate buffer, at pH
188 7. The samples were centrifuged at $20,000\times\text{g}$ and 4°C for 20 min, then 5 μg of total soluble protein from
189 the supernatants was loaded onto SDS-PAGE gel (12.5% acrylamide), and stained with Gel-Code Blue
190 Stain reagent (Pierce Biotechnology, Inc., Rockford, USA). To estimate the large subunit of RuBisCo

191 content, densitometry analysis was conducted using the program Quant 1 in Gel Doc 2000 (Bio-Rad,
192 USA).

193 The total soluble protein (TSP) content was calculated using the Bradford method. The units were
194 expressed as mg protein g⁻¹ DW.

195 **ii) Activity and protein gel blot analysis of glutamine synthetase enzyme**

196 Glutamine synthetase (GS; EC 6.3.1.2) activity was determined using a glutamyl hydroxamate (GHM)
197 synthesis-based biosynthetic assay (following [22]) and expressed as μmol GHM g⁻¹ dry weight (DW)
198 min⁻¹. The total soluble protein was calculated according to the Bradford method and expressed as mg
199 protein g⁻¹ DW.

200 Plant samples were ground with liquid N₂ and then homogenised with 1.5:1 (volume/weight) extraction
201 buffer (50 mM Tris-HCl pH 8; 1 mM EDTA; 10 mM 2-mercaptoethanol; 5 mM dithiothreitol; 10 mM
202 MgSO₄; 1 mM cysteine; 0.6% polyvinylpyrrolidone). Phosphatase inhibitor cocktails 1 and 2
203 (Sigma-Aldrich, St. Louis, MO, USA) were added to a final concentration of 2.5 μL mL⁻¹ each. The
204 extracts were centrifuged at 20,000×g and 4 °C for 30 min.

205 SDS-PAGE was run with the following antibody: anti-GS IgG, which was raised in rabbit against a
206 specific peptide from pea GS (Acc. # CAJ87510.1; [17]) and used at a 1:2000 dilution overnight at 4°C.
207 A peroxidase conjugated goat anti-rabbit IgG, followed by luminescence detection with the ECLTM
208 Plus kit (AmershamBiosciences, Buckinghamshire, UK), was used in foliar tissues, and analkaline
209 phosphatase labelled goat-anti-rabbit IgG was visualised with NBT-BCIP (Sigma-Aldrich) in root
210 samples.

211

212 **Statistical analysis**

213 All statistical analyses were performed using the Statistical Product and Service Solutions software
214 package (SPSS, USA) version 15.0 for Windows. The data was analysed using a one- and two-way
215 analysis of variance (ANOVA) for all parameters. In the one-way analysis of variance (one-way
216 ANOVA), the Duncan post-hoc test was used as a method to separate treatment means; in the post-hoc
217 tests displayed in the figures and tables the letters represent the significant differences between
218 treatments. Moreover, a two-way analysis of variance (two-way ANOVA) test was used in order to
219 study the effect of two factors, NH₄⁺ concentration, and CO₂ level, on the parameters analysed and the
220 interaction between these factors (CO₂ x NH₄⁺); significant differences and interaction between the
221 factors are indicated with asterisks in the figures and tables. All the statistical analyses were conducted
222 at a significance level of 5% (p ≤ 0.05).

223

224 3. Results

225 **Plant growth and C-N content of NH₄⁺ fed pea plants and eCO₂**

226 Under eCO₂, the shoot and root biomass increased for both NH₄⁺ concentrations relative to aCO₂ (Figure
227 1). Nevertheless, pea plant biomass accumulation was independent of the external NH₄⁺ concentration
228 supplied at each CO₂ level (Figure 1). In fact, there was no plant growth reduction at the higher NH₄⁺
229 concentration (Figure 1). The plants showed no symptoms of stress: there was no appreciable browning
230 of the roots or necrotic tips; no lesions on the stem or leaves; and no chlorosis or vascular browning in
231 any treatment (Supplemental Figure S1). Plants grown under eCO₂ increased their shoot biomass by
232 86% at 2.5mM NH₄⁺ and 32% at 10 mM NH₄⁺; meanwhile the increase observed in root biomass was
233 28% and 50% at 2.5mM and 10 mM NH₄⁺, respectively, in comparison with aCO₂. Although 10 mM
234 NH₄⁺ enhanced shoot biomass at day 14, the shoot biomass accumulation was equal at the endpoint,
235 after 21 days (Supplemental Table 1).

236 The C content was the same between the treatments in both tissues. Remarkably, the N content was
237 higher at 10 mM NH₄⁺ at both CO₂ levels. Our treatments did not influence the leaf NH₄⁺ content.
238 Contrastingly, in the roots, more NH₄⁺ was found at 10 mM NH₄⁺ at both CO₂ levels (Table 2).

239 **Increased C source under eCO₂ and ammonium nutrition: photosynthetic performance, 240 carbohydrate availability and leaf and root respiration rates**

241 The maximum photosynthetic rates (A_{max}) were higher under eCO₂ (Table 1). Remarkably, the
242 maximum velocity of the Rubisco carboxylation rate (V_{c_{max}}) and maximum electron transport rate (J_{max})
243 did not differ significantly between the treatments. Curiously, an increase in stomatal conductance (g_s)
244 was observed under eCO₂ and 10 mM NH₄⁺, concomitant to the increase in dark respiration (R_D) of the
245 leaf and A_{max} (Table 1). Note that this significant increase in R_D coincided with a higher sucrose content
246 in leaves, which could be offering elevated respiratory substrate in this organ.

247 The contrasting CO₂ levels had a profound influence on the carbohydrate pools (i.e., soluble sugars and
248 starch). Plants grown under eCO₂ showed a significantly decreased leaf starch pool with respect to aCO₂,
249 which was accompanied by a concomitant release of soluble sugars, fructose, glucose and sucrose in the
250 root, which was especially important at 10 mM NH₄⁺ (Figure 2). This suggests a shoot-to-root
251 translocation of C driven by the greater availability of C and the higher external NH₄⁺ concentration.
252 This provides a clue to the important C and N sink that the roots can represent under these growth
253 conditions, to help maintain the C/N ratios independently from the CO₂ concentrations. Moreover, of
254 the leaf soluble sugars, only sucrose presented a significant increase under eCO₂ and at 10 mM NH₄⁺.

255 The CO₂ treatment did not impact the tricarboxylic acid content, while the N treatment did. In leaves,
256 malate and citrate levels were reduced under high NH₄⁺ conditions at both CO₂ concentrations, while
257 the α-ketoglutarate content presented the contrary response. (Supplemental Figure S2). In roots, higher
258 levels of succinate and α-ketoglutarate were found under high NH₄⁺ conditions at both CO₂
259 concentrations. Meanwhile, the malate and citrate content in the root only increased concomitantly with
260 the NH₄⁺ concentration under aCO₂. As for root respiration, the cytochrome respiration was not modified
261 by the treatments while the alternative pathway decreased at high NH₄⁺ conditions under eCO₂
262 (Supplemental Figure S3); this could be induced as a way to dissipate excess energy. Residual
263 respiration was not affected in pea plants.

264 **Increased N assimilation derived from eCO₂ and high NH₄⁺ levels is accompanied by** 265 **decreased total amino acid accumulation and increased total soluble protein**

266 The glutamine synthetase (GS) activity in leaves was about twice as high as in roots. In leaves, no
267 differences were found between treatments (Table 3). In roots, the GS activity was significantly higher
268 under eCO₂. When both NH₄⁺ and CO₂ concentrations were high, the root GS polypeptide content was
269 increased under eCO₂ (Table 3; Supplemental Figure S4). This increase in the activity and content of
270 GS in the roots under eCO₂ was not accompanied by an increase in the total amino acid content (Table
271 3). Plants exposed to eCO₂ showed a reduced total amino acid content in both leaves and roots,
272 independently of the external NH₄⁺ concentration (Table 3).

273 In terms of the amino acid profile, the major amino acids (serine, alanine, cysteine and
274 asparagine/proline) represented 78% in leaves and 87% in roots. In leaves, a decrease in the serine and
275 asparagine/proline content was observed at both ammonium and glutamine levels at 10 mM NH₄⁺ under
276 eCO₂ (Supplemental Figure S5). The serine content, the precursor of which is 3-phosphoglycerate, is
277 strongly diminished under eCO₂. This decrease in serine coincides with an increase in the levels of
278 alanine, an amino acid that is synthesized from pyruvate; this is synthesised in the glycolytic pathway
279 in a process subsequent to 3-phosphoglycerate. With regard to the major amino acids in the root, a
280 decrease in the content of serine and glutamine was observed at both levels of ammonium and cysteine
281 at 2.5 mM NH₄⁺, this decrease coinciding with an increase in asparagine in the roots under eCO₂.

282 Finally, the total soluble protein (TSP) content in leaves increased under eCO₂. Interestingly, this was
283 not due to the variation in the levels of the large subunit of Rubisco (RbLs), which in turn was lower
284 under eCO₂ (Table 1). The N treatment did not influence either the leaf TSP or RbLs. Nevertheless, the
285 root TSP content was greater under the high NH₄⁺ concentration, while no differences were found
286 between CO₂ levels (Table 3).

287

288 4. Discussion

289 Elevated CO₂ leads to increased photosynthetic activity and plant biomass regardless of external 290 and internal NH₄⁺ concentrations

291 A plant's capacity to tolerate NH₄⁺ nutrition has been attributed to several factors, including their genetic
292 background, i.e., the inter- and intraspecific variation, among other things. [17] described intraspecific
293 variation for pea plant (*Pisum sativum* L.) sensitivity to NH₄⁺ nutrition, concluding that *Snap pea* plants
294 could be considered a reference cultivar for NH₄⁺ tolerance, since the biomass production of these plants
295 is not affected by NH₄⁺ concentrations. In our study, Snap pea plant growth was also independent of the
296 NH₄⁺ concentration supplied, and the plant biomass variation observed was driven by the availability of
297 C (i.e., eCO₂). Similarly to [17], we found a large accumulation of NH₄⁺ in the roots, although the root
298 GS was similar (without concomitant changes in the leaves). Interestingly, in contrast with [17], who
299 claim that the Snap pea's tolerance is based on its greater root respiration, in our case the root respiration
300 remained unvaried between the different NH₄⁺ levels under aCO₂. For that reason, our plants could have
301 been promoting the 'sequestration' of NH₄⁺ in the root cell vacuoles, preventing its transport to the
302 leaves, which are more sensitive to NH₄⁺ accumulation [23,24], as occurs in NH₄⁺-tolerant reference
303 species such as rice [25]. We believe that the discrepancies between the results of [17] and our study are
304 related to a very tight control of the pH during our experiment.

305 Although in plants with C3 photosynthetic metabolism exposure to eCO₂ increases photosynthesis rates,
306 often such stimulation is partially reversed in an adaptation process known as 'photosynthetic
307 acclimation' [4]. Photosynthetic acclimation is accompanied by alterations in the gas exchange
308 characteristics that are indicative of a decreased carboxylation capacity [26]. In our study, the A_{max} of
309 plants exposed to eCO₂ increased regardless of the NH₄⁺ dose and presented greater GS activity.
310 Furthermore, these plants maintained unaltered V_{cmax} and J_{max} rates. While A_{max} increased and TSP
311 (together with leaf N) were not altered by growth at eCO₂, our study showed that the Rubisco content
312 decreased significantly under eCO₂. Due to the fact that plants tend to maximise resource-use efficiency,
313 the reduction in the Rubisco content would imply a reallocation of the N away from the CO₂ fixation
314 machinery into more limiting processes, such as carbohydrate synthesis and non-photochemical
315 processes[26]. Such processes may contribute to an increase in sink activity. Our study confirmed a
316 reduction in N allocation to Rubisco (a major leaf N storage form), leading to more NUE for
317 photosynthesis and biomass accumulation. This is a remarkable observation, as our previous work on
318 pea plants in eCO₂ with NO₃⁻ as their N source registered a large leaf N and TSP reduction, precipitating
319 V_{cmax} [27]. Furthermore, the decline in leaf N and TSP reduction is a phenomenon overwhelmingly
320 observed under eCO₂ in multiple plant species, including wheat, *Arabidopsis*, and many others (Leakey
321 et al., 2009; Jauregui et al., 2016; Rubio Asensio & Bloom, 2017). In contrast, newer publications have
322 reported the capacity of the plants to maintain leaf N status under eCO₂ when NH₄⁺ is the N source

323 received by the plants (extensively reviewed in [16][9][28][29]). Our results therefore fuel the debate
324 over the impact different sources of N have in eCO₂ [11], including a novel nuance: plants that are able
325 to tolerate high levels of NH₄⁺ under aCO₂ successfully exploit high levels of NH₄⁺ under eCO₂, evading
326 leaf N, TSP and RbLs depletion and thus, maintaining an advantageous photosynthetic response to
327 eCO₂. In this way, we can suggest that exploiting NH₄⁺-tolerant plants in combination with NH₄⁺-based
328 nutrition could be a strategy for dealing with the reported reduction in the nutritional content of crops
329 [30] in higher CO₂ concentrations than at present: we detected no N or other mineral depletion, as P or
330 S could not be detected during the vegetative stage, and those minerals could potentially be allocated
331 into the harvested tissues. Although an NH₄⁺-rich environment is complex to address in croplands due
332 to nitrification –even when current formulation of nitrification inhibitors are used– these conditions can
333 be achieved in greenhouses through fertigation.

334 **Maintaining the leaf N status and favouring C allocation into mobile sugars over starch** 335 **accumulation is an efficient strategy for overcoming photosynthetic acclimation under eCO₂**

336 The responsiveness of the photosynthetic machinery to eCO₂ has been previously associated with
337 increases in leaf carbohydrate that induce photosynthetic protein repression, leading to a down-
338 regulation of photosynthetic capacity [31]. Within this context, an efficient whole-plant partition-
339 allocation of C between mobile carbohydrates and end products, and the development of new sink tissue
340 determines plant responsiveness under eCO₂ [32][33]. Indeed, the capacity of our plants to avoid
341 photosynthetic acclimation may be linked to the efficient C-allocation strategy adopted, prioritising
342 short-term and mobile C-storage pools (soluble sugars and tricarboxylic acids) over the long-term
343 storage (starch). The mobile C pools released from starch under eCO₂ were especially relevant at high
344 NH₄⁺ concentrations, these being principally long-term transport sugars, such as sucrose (in leaves and
345 roots), as well as immediately consumable/usable/available sugars, such as glucose and fructose (mainly
346 in the roots). Furthermore, starch accumulation under eCO₂ is a recurrent observation with NO₃⁻
347 nutrition [30][5], including in our previous experiments with the same genotype of pea plants [27].
348 Contrastingly, with NH₄⁺ nutrition, in our experiment we found a significant reduction of starch content
349 under eCO₂ relative to aCO₂. This decreased starch content was undoubtedly related to the reallocation
350 of C into the mobile pools of sugars and the higher growth rates observed under eCO₂.

351 **Might the stomatal conductance be enhanced with NH₄⁺ nutrition under eCO₂?**

352 Stomatal closure is the first physiological response of plants exposed to increasing CO₂ concentrations
353 (Flexas et al., 2007), and in multiple species and experiments the g_s parameter is seen to typically drop
354 under eCO₂ [35][36]. In fact, this response is often associated with photosynthetic acclimation as CO₂
355 levels rise. However, our experiment shows that pea plants exposed to eCO₂ present greater g_s activity
356 than those grown under aCO₂. Despite strict NH₄⁺ nutrition being usually underrepresented in the

357 published literature, this atypical increase in g_s under eCO_2 has also been reported when plants receive
358 NH_4^+ nutrition without noticeable symptoms of NH_4^+ toxicity [37][28]. Additionally, [29] reported
359 increased g_s under high NH_4^+ concentrations compared to low NH_4^+ levels, arguing that N depletion was
360 the cause of this drop. Interestingly, other leaf conductance, such as mesophyll conductance, is also
361 enhanced with NH_4^+ nutrition compared to NO_3^- [29]. Despite the relevance of this observation for whole
362 plant C management, neither the molecular mechanisms nor the signalling cascade underlying the
363 differential effect of the N forms on plant conductance has been elucidated. For instance, general plant
364 leaf responses to eCO_2 include an increase in epidermal and guard cell size, an increase in stomatal area,
365 a decrease in epidermal and stomatal density, and decreased stoma opening [38]. In this regard, [39]
366 demonstrated that the interaction of eCO_2 and the N source can influence the stomatal and epidermal
367 anatomy in wheat plants, not only due to the increased CO_2 concentration, but also to the N source.
368 These authors showed that wheat plants growing at 600 ppm CO_2 and with NH_4^+ as the sole N source
369 exhibited a smaller stomatal opening area and lower stomatal density than those grown with NO_3^- .
370 Furthermore, they also observed that NH_4^+ toxicity notably affects the morphological traits of wheat
371 leaves, including their size and shape [36]. Thus, the interesting question arises of whether NH_4^+ tolerant
372 cultivars, which are not affected by toxicity symptoms of NH_4^+ , such as this snap pea variety [17], could
373 avoid stomatal anatomy alterations under eCO_2 . Further investigation is needed to address this question
374 in depth.

375 **Root NH_4^+ assimilation is an important C sink at eCO_2 and high NH_4^+ concentrations**

376 Our data supports the idea that there is a strengthening of the root C sink in eCO_2 under high NH_4^+
377 concentrations. At 2.5 mM NH_4^+ , the relative increase in biomass is mainly due to shoot growth, while
378 at 10 mM it is mainly due to root growth. Besides, NH_4^+ transamination from glutamine to asparagine
379 can be a means of transporting the root-assimilated ammonia to shoot. The increased ability of leaves to
380 deliver photoassimilates under eCO_2 , helps root for NH_4^+ assimilation, as accordingly shows its
381 increased GS activity, strengthening its C sink. This is especially evident in plants grown with 10 mM
382 NH_4^+ under eCO_2 , with higher leaf sucrose and glucose levels than with 2.5 mM NH_4^+ , and higher C
383 skeletons (sugars and carboxylic acids) in roots, leading root C allocation to support primary NH_4^+
384 assimilation. In the literature, increased C allocation in roots of plants grown under NH_4^+ nutrition has
385 been usually observed [13]. This observation has been attributed to the high requirement of C skeletons
386 to obtain an extra energy input as well as to incorporate NH_4^+ into organic compounds, avoiding its
387 accumulation in plant cells. Up to now, this strategy has only been considered as a mechanism of NH_4^+
388 tolerance in some plant species. Here it can be considered as a strategy for overcome to eCO_2
389 maintaining biomass production and N status of crop plants.

390

391 **5. Conclusion**

392 We have described the physiological mechanisms underlying the response of an NH_4^+ tolerant pea plant
393 grown hydroponically under solely NH_4^+ nutrition and elevated CO_2 conditions ($e\text{CO}_2$). The NH_4^+
394 tolerant pea plants overcome photosynthetic acclimation to $e\text{CO}_2$ and increased their biomass,
395 maintaining leaf N status by reallocating surplus C for the primary assimilation instead of starch
396 formation. In terms of root plasticity, NH_4^+ nutrition could be additionally considered as a strong N
397 source, able to increase C sink strength in conditions of $e\text{CO}_2$. In this way, improved C source / sink
398 balance, maintained photosynthetic capacity and kept plant N status. A proper NH_4^+ nutrition
399 management could motivate plants to strenghten the C demand from their sinks organs and metabolic
400 process, which in turn constitute a plant trait to avoid photosynthetic acclimation in C3 crops under
401 $e\text{CO}_2$ conditions.

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409

410 **7. CRediT author statement**

411 Ivan Jauregui: Conceptualization, Investigation and Writing- Original draft preparation, Writing -
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416 and Writing - Review & Editing

417

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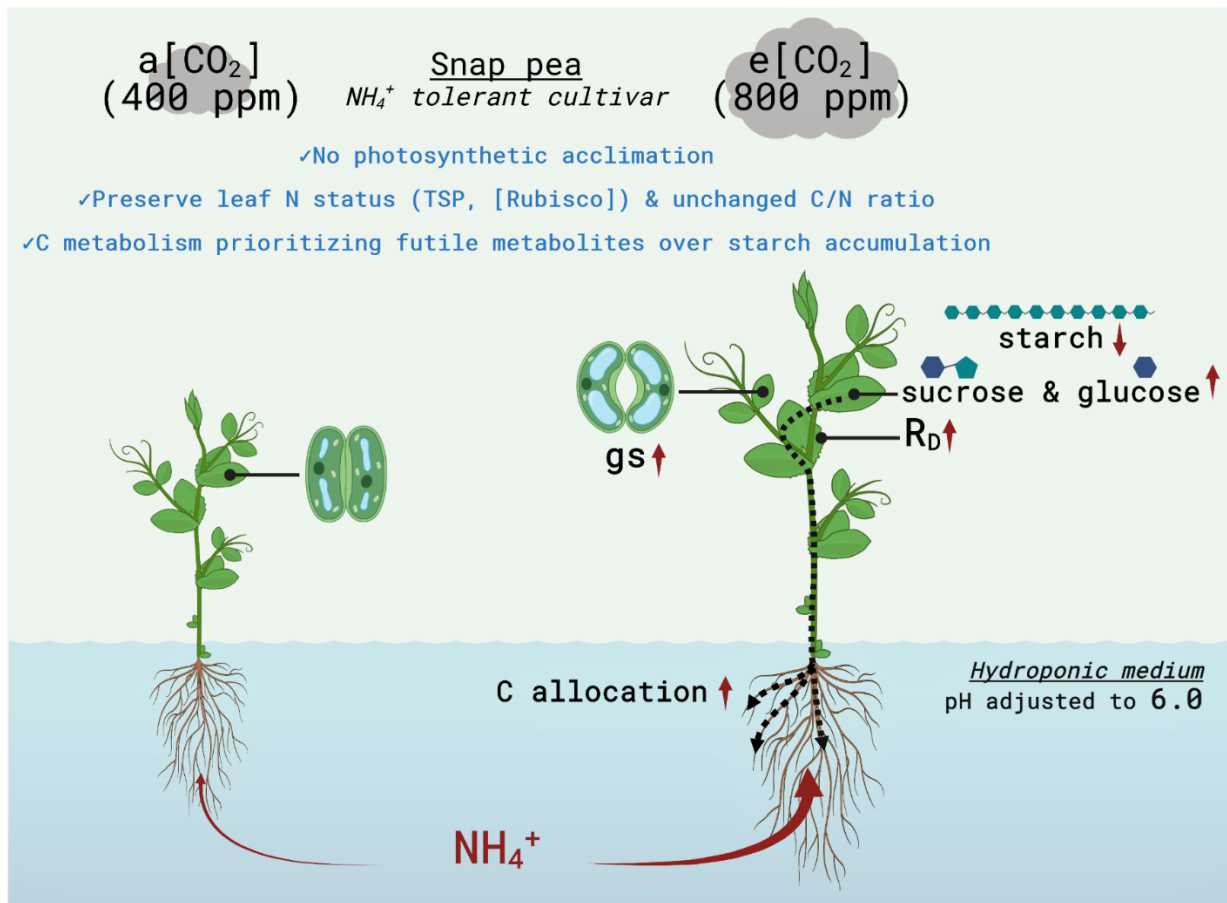
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549

550 Graphical Abstract (created with *biorender.com*)

551

Highlights

552

- We explore a novel strategy to cope with the widely reported reductions of leaf nitrogen content in plants exposed to atmospheric elevated CO₂ (eCO₂) using ammonium nutrition and ammonium-tolerant pea plants.

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555

- Our Pea-ammonium tolerant avoids acclimation to eCO₂.

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- Pea-ammonium tolerant plants favour C allocation into futile C pools over starch accumulation, and, preserves leaves nitrogen status.

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- We bring light into an underrepresented physiological response of plants exposed to eCO₂: stomatal conductance can increase in eCO₂ if the source of nitrogen is ammonium.

559

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- Besides, we found that we can consider the roots tissues as a source of C because of the energy cost of NH₄⁺ assimilation.

562

Table 1. Effect of CO₂ level (Ambient, 400 *versus* Elevated, 800 μmol mol⁻¹) and NH₄⁺ concentration (2.5 *versus* 10 mM) in *Pisum Sativum* plants (cv. *snap pea*) on Photosynthesis (A, μmol CO₂ m⁻²s⁻¹), Rubisco maximum carboxylation rate (V_{cmax}, μmol CO₂ m⁻²s⁻¹), maximum electron transport rate contributing to RuBP regeneration (J_{max}, μmol CO₂ m⁻²s⁻¹), stomatal conductance (g_s, mol CO₂ m⁻²s⁻¹), dark respiration (R_D, μmol CO₂ m⁻²s⁻¹), Rubisco Large Subunit content (RbLs, optical density units). Each value represents the mean of biological replicates ± SD, n=3 and n=11 for Rubisco content. Statistical analysis was made by a two factors Analysis of the Variance (ANOVA). The asterisk (*) represent significant differences and *n.s.*, no significant differences (P ≤ 0.05).

		A	V _{cmax}	J _{max}	g _s	R _D	RbLs
aCO ₂	2,5 mM NH ₄ ⁺	18 ± 1 b	77 ± 9 a	150 ± 10 a	183 ± 6 b	-0.49 ± 0.02 a	68 ± 9 a
	10 mM NH ₄ ⁺	17 ± 1 b	78 ± 4 a	150 ± 6 a	186 ± 10 b	-0.60 ± 0.13 a	58 ± 4 a
eCO ₂	2,5 mM NH ₄ ⁺	27 ± 1 a	72 ± 5 a	127 ± 6 a	203 ± 20 ab	-1.12 ± 0.07 b	48 ± 6 b
	10 mM NH ₄ ⁺	28 ± 2 a	70 ± 4 a	136 ± 7 a	240 ± 6 a	-1.56 ± 0.03 c	46 ± 3 b
CO ₂		*	n.s.	n.s.	*	*	*
NH ₄ ⁺		n.s.	n.s.	n.s.	n.s.	n.s.	n.s.
CO ₂ x NH ₄ ⁺		n.s.	n.s.	n.s.	n.s.	n.s.	n.s.

Table 2. Effect of CO₂ level (Ambient, 400 *versus* Elevated, 800 μmol mol⁻¹) and NH₄⁺ concentration (2.5 *versus* 10 mM) in *Pisum Sativum* plants (cv. *snap pea*) on ammonium content (mg g⁻¹ DW) nitrogen (%), C/N ratio in shoot and root. Each value represents the mean of biological replicates ± SD, n=3 and n=6 for NH₄⁺ content.

Statistical analysis was made by a two-way Analysis of the Variance (ANOVA). The asterisk (*) represent significant differences and *n.s.*, no significant differences ($P \leq 0.05$).

		LEAF			ROOT		
		NH ₄ ⁺	N	C/N	NH ₄ ⁺	N	C/N
aCO ₂	2,5 mM NH ₄ ⁺	0.66 ± 0.1 a	4.6 ± 0.4 a	9.4 ± 0.8 a	4.2 ± 0.5 b	4.8 ± 0.0 b	7.6 ± 0.3 a
	10 mM NH ₄ ⁺	0.80 ± 0.2 a	5.4 ± 0.3 a	7.9 ± 0.5 a	9.0 ± 1.1 a	6.2 ± 0.3 a	6.3 ± 0.4 b
eCO ₂	2,5 mM NH ₄ ⁺	0.72 ± 0.1 a	4.4 ± 0.6 a	10.0 ± 1.5 a	4.1 ± 0.6 b	4.4 ± 0.2 b	8.7 ± 0.5 a
	10 mM NH ₄ ⁺	0.85 ± 0.1 a	5.5 ± 0.3 a	7.7 ± 0.4 a	7.0 ± 0.4 a	6.3 ± 0.2 a	6.2 ± 0.3 b
CO ₂		n.s.	n.s.	n.s.	n.s.	n.s.	n.s.
NH ₄ ⁺		n.s.	*	n.s.	*	*	*
CO ₂ x NH ₄ ⁺		n.s.	n.s.	n.s.	n.s.	n.s.	n.s.

Table 3. Effect of CO₂ level (Ambient, 400 versus Elevated, 800 μmol mol⁻¹) and NH₄⁺ concentration (2.5 versus 10 mM) in *Pisum Sativum* plants (cv. *snap pea*) on free amino acids content (μmol g⁻¹ DW), total soluble protein (TSP) (mg g⁻¹ DW) and GS activity (μmol GHM g⁻¹ DW min⁻¹) in leaf and root. Each value represents the mean of biological replicates ± SD, n=6. Statistical analysis was made by a two factors Analysis of the Variance (ANOVA). The asterisk (*) represent significant differences and *n.s.*, no significant differences (P ≤ 0.05).

		LEAF			ROOT		
		Total free amino acids content	TSP	GS activity	Total free amino acids content	TSP	GS activity
aCO ₂	2,5 mM NH ₄ ⁺	1428 ± 87 a	178 ± 20 c	83 ± 5 a	1050 ± 104 a	76 ± 5 b	35 ± 3 b
	10 mM NH ₄ ⁺	1567 ± 187 a	195 ± 25 bc	85 ± 6 a	729 ± 38 b	105 ± 7 a	41 ± 4 ab
eCO ₂	2,5 mM NH ₄ ⁺	998 ± 74 ab	255 ± 15 ab	90 ± 3 a	851 ± 24 b	81 ± 3 b	46 ± 3 a
	10 mM NH ₄ ⁺	1273 ± 80 b	272 ± 21 a	93 ± 3 a	740 ± 9 b	104 ± 7 a	51 ± 3 a
CO ₂		*	*	n.s.	n.s.	n.s.	*
NH ₄ ⁺		n.s.	n.s.	n.s.	*	*	n.s.
CO ₂ x NH ₄ ⁺		n.s.	n.s.	n.s.	n.s.	n.s.	n.s.