

EFFECT OF BRIQUETTE COMPOSITION AND SIZE ON THE QUALITY OF THE RESULTING COKE

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ABSTRACT

Five briquettes were prepared using sawdust, a non-coking coal and a binder. Industrial coal blends were used to study the influence of the type of sawdust (pine and chestnut), the binder (coal tar and coal-tar sludge) and the size of the briquettes on the quality of the cokes produced from mixtures containing up to 15 wt.% of the five briquettes. The effect of the briquettes and briquette components on the fluidity of the industrial coal blends was investigated. It was found that biomass and non-coking coal produced a decrease in fluidity, whereas the binders increased it. The combined effect of both types of additive had the global effect of decreasing fluidity. Mixtures of the briquettes with the industrial coal blends were carbonized in a 17 kg movable wall oven in order to assess their influence on the quality of the cokes produced. Their cold mechanical strength (JIS DI150/15 index), reactivity to CO₂ (CRI index) and post-reaction strength (CSR index) were also tested. The composition of the ash of the sawdusts and the reactivity of the briquette components were used as an indication of the effect on coke reactivity. The effects on cold mechanical strength and post-reaction strength were different in some cases.

Keywords: coke; biomass; briquettes; mechanical strength; reactivity

1. Introduction

The steel industry is an energy and carbon-based intensive process and therefore a major contributor to global anthropogenic CO₂ emissions [1-4]. At the same time, cokemaking is a process where the recycling of wastes is possible, especially high carbon and low ash wastes like sawdust and charcoal [5-9], plastics [10-12], and bituminous wastes [13-15]. Thus recycling in the cokemaking industry could provide a way to reduce the environmental impact of CO₂ emissions, reduce costs and widen the raw materials spectrum to include non-fossil fuels.

The inclusion of sawdust in coal blends for cokemaking has clear advantages such as its low sulphur and ash content and its zero contribution to CO₂ emissions but it also has a number of disadvantages including its low char yield, deleterious effect on coal fluidity [16,17] and low bulk density [18]. A possible way to increase the bulk density of the biomass is to prepare briquettes. Partial briquetting of coal charges was introduced by the Japanese industry in the 70s, as this technique enabled the amount of expensive prime coking coal in the blend to be reduced and a cheap non-coking coal to be used instead without any deterioration of the quality of the resultant coke [19]. In a previous paper [7] a comparison of the direct addition of sawdust and addition via briquetting was carried out. However, other factors such as size and composition still need to be evaluated in order to know whether it is possible to apply sawdust briquetting to cokemaking.

Various binders can be used for the preparation of briquettes. However both sawdust and non-coking coal have a deleterious effect on the development of coal fluidity [16,17] making pitch and coal tar preferable binders considering that both of these produce an increase in coal fluidity [20,21]. Coal-tar pitch has already been successfully used as a binder [22]. The role of pitch in briquettes comprising high-rank and coking coals is to interact with them and modify their carbonization behaviour so that the system is sufficiently fluid to wet the surface of non-fusing coals. In addition it

needs to be able to form a binder coke with a mosaic optical texture that connects coal-derived coke with inerts. The drawback with coal-tar pitch is its high carcinogenic compounds content [23]. An alternative option is to use coal tar which does not cause as great an increase in fluidity as coal tar pitch but is nevertheless a liquid and has fewer carcinogenic polyaromatics.

The aim of the present research work is to determine the influence of the type of sawdust and the size of the briquettes and binder used for their preparation on the quality of the coke produced from the co-carbonization of the briquettes with industrial coal blends.

2. Experimental

2.1. Materials characterization

Three industrial coal blends (CB2, CB3, CB4) were used together with the briquettes in the carbonization tests. The briquettes were prepared using a low volatile non-coking coal (C) and two sawdusts, one from chestnut (SC) and the other from pine (SP). As binder for the preparation of the briquettes coal tar (T) and coal tar sludge (M) from the coking plant were used. The raw gas evolving from the coke ovens is treated in order to separate the permanent gases, ammonia, benzol, and tar. The coal tar sludge (M) collected at the bottom part of the tar decanter, apart from tar also contains some coal and coke that is drawn away and deposited on the bottom of the decanter. Five briquettes with different compositions were prepared using a roll press briquetting machine (Table 1). The briquettes were ellipsoid shaped with axes 46 and 42 mm in length and weighting around 23 g. On the basis of their different compositions it is possible to study: 1. the influence of including SC (B1 vs. B4 and BM1 vs. BM4); 2. the influence the type of binder (B4 vs. BM4 and B1 vs. BM1); 3. the effect of the two sawdusts, chestnut vs. pine (BM1 vs. BM1_{sp}). To study the effect of the size, samples of briquette of weight between 4 and 6 g, with the same composition as BM1_{sp} and BM4 were used for the co-carbonization tests and labelled BM1_{sp-F} and BM4_F.

Elemental analysis was carried out following the standard ISO 562 and ISO 1171 procedures for humidity, ash and volatile matter respectively. For the elemental analysis the following standard procedures were used: ASTM D 5016-98 and ASTM D 5373-02 for C, H and N using a LECO CHN-2000 and a LECO S-144 DR instrument for the analysis of S.

The apparent density of the briquettes was determined by means of water displacement by immersing 6 briquettes in a 500 ml container. The density of briquettes BM1_{SP-F} and BM4_F could not be measured because they disintegrate in water.

2.2. Textural characterization

The particle size used to determine the porous structure of the materials was between 1.18 and 0.8 mm. The true density (ρ_{He}) of the sawdusts was measured by means of helium pycnometry on a Micromeritics Accupyc 1330 Pycnometer. Their apparent density (ρ_{Hg}) was determined using mercury at 0.1 MPa on a Micromeritics autopore IV 9500 mercury porosimeter. From the true and apparent densities the open porosity corresponding to pore sizes of less than 12 μm was calculated by means of the following equation :

$$\varepsilon (\%) = \left(1 - \frac{\rho_{Hg}}{\rho_{He}} \right) \cdot 100 \quad (1)$$

The total pore volume (V_T) was obtained from the equation:

$$V_T (cm^3 / g) = \left(\frac{1}{\rho_{Hg} (g / cm^3)} - \frac{1}{\rho_{He} (g / cm^3)} \right) \quad (2)$$

The pore size distribution was calculated by applying increasing pressure to the sample from 0.1 to 227 MPa. This resulted in pore sizes in a range of 12 μm to 5.5 nm according to the Washburn equation.

Pore size was classified into two categories: macropores (12 μm > d_p > 50 nm) and mesopores (50 nm > d_p > 5.5 nm).

2.3. Variation of coal blend fluidity due to briquette addition

The thermoplastic properties of mixtures of the coal blends with 5, 10 and 15 wt.% of briquettes B1, BM1, BM1sp, B4 and BM4 were measured. Also the effect of the two binders, the sawdusts and the non-coking coal on the fluidity of the coal blends was assessed by means of the Gieseler test (ASTM D2639-74), this test having been successfully used previously to determine the modification of coal fluidity due to the use of additives [17,20,24]. The sample was heated at 3 $^{\circ}\text{C}/\text{min}$ up to a final temperature of 550 $^{\circ}\text{C}$, while a constant torque was applied to the stirrer inside the crucible containing the sample. The spin rate of the stirrer was measured continuously until it stopped. The parameters derived from this test were: (i) softening temperature, T_s ; (ii) the temperature of maximum fluidity, T_f ; (iii) resolidification temperature, T_r ; (iv) plastic range, $T_r - T_s$, which is defined as the difference between the resolidification and softening temperatures; and (v) maximum fluidity, MF, expressed as dial divisions per minute (ddpm).

2.4. Thermogravimetric analysis (TG)

Gasification was studied on a TA Instruments SDT 2960 thermobalance. Samples of weight 3-5 mg with a particle size of < 0.212 mm were heated in N_2 up to 1100 $^{\circ}\text{C}$ and once the temperature was stabilized they were treated with CO_2 using a flow of 100 ml/min until a conversion degree higher than 50 % was reached. The cokes/chars employed for the gasification in the thermobalance were prepared in a horizontal oven using a heating rate of 5 $^{\circ}\text{C}/\text{min}$. The carbon conversion (x) and gasification rate or reactivity (r) were calculated by means of the following equations:

$$x = \frac{m_0 - m_t}{m_0 - m_{ash}} \times 100 \quad (\text{Eq. 1})$$

$$r = \frac{dx}{dt} \quad (\text{Eq. 2})$$

where m_0 represents the initial mass of char and m_t , the mass at time t .

2.5. Carbonization tests and coke quality determination

Carbonization tests were carried out in a movable wall oven of approximately 17 kg capacity (MWO17) [25]. The dimensions of the oven are 250 mm L x 165 mm W x 790 mm H. The samples were charged when the oven had reached 1100 °C. The temperature of the wall was kept constant throughout the test. The duration of the coking was approximately 3.5 h. The industrial coal blends were carbonized with a bulk density of $782 \pm 5 \text{ kg/m}^3$

The cold mechanical strength of the cokes produced was assessed by the JIS test (JIS K2151 standard procedure). After the test the coke was sieved and the D1150/15 index was calculated from the amount of coke with a particle size greater than 15 mm. Coke reactivity and mechanical strength after reaction were assessed by means of the NSC test (ASTM D5341 standard procedure) [26]. Two indices were derived from this test i.e. the CRI index which represents the loss of weight of a 200 g sample of coke with size between 19 and 22.4 mm after reaction with CO_2 at 1100 °C for two hours and the CSR index which represents the percentage of partially reacted coke that remains on a 9.5 mm sieve after 600 revolutions in a standardized drum. The relationship between the CSR values obtained in a MWO of 17 kg capacity and those obtained in a MWO of 300 kg capacity has been reported previously [27].

3. Results and discussion

3.1. Characteristics of the material

The three industrial coal blends used have similar characteristics (Table 2) sulphur and ash contents around 0.56 and 8.4 wt.% respectively and a maximum Gieseler fluidity between 600 and 1000 ddpm. The highest oxygen and lowest carbon contents correspond to the sawdust.

It is well known that the bulk density of the charge has a strong influence on the quality of the coke, especially on the mechanical strength and strength after the reaction with CO₂ (CSR index) [28]. In a previous research paper it was shown that the direct addition of sawdust to an industrial coal blend produced a decrease in the bulk density of the charge of 17 kg/m³ while the addition of sawdust in the form of briquettes increased it by 11 kg/m³ [7].

The apparent density of the briquettes is shown in Table 3. The highest apparent density corresponds to B4 which was prepared using a non-coking coal and tar as binder. For briquettes prepared with the same binder those containing biomass have a lower density ($\rho_{B1} < \rho_{B4}$ and $\rho_{BM1_{SP}} < \rho_{BM4}$). The effect of SC with tar as binder (B1) compared to B4 is a reduction in density of 8 %, while in the case of briquettes prepared with coal tar sludge (M) as binder and SP (BM1_{SP}) compared to BM4, the reduction is around 24 %. The type of sawdust is also important, in the case of briquettes prepared with coal tar sludge and SC ($\rho_{BM1} = 1179 \text{ kg/m}^3$), while the density of the briquettes prepared with coal tar sludge and pine sawdust is lower ($\rho_{BM1_{SP}} = 873 \text{ kg/m}^3$). In order to explain these results two factors may play a role: the particle size distribution of the sawdusts and their porous structure. While SC has a 4 wt.% proportion of particles lower than 0.5 mm, SP has 20 wt.% of this size [6]. The finer particle size distribution implies that there is a greater contact surface available for coverage by the binder. The structure of the two sawdusts was studied by SEM in order to find differences between soft and hardwood [29] that might corroborate differences in their ability to produce dense briquettes. Figure 1 shows images which clearly indicate that the porous structure in both types of sawdust is different. While SC

has pores of two sizes (see [Figure 1a and 1b](#)) corresponding to vessels (greater than 100 μm) and tracheids (approx. 15-25 μm) which are common in hardwood, SP (see [Figure 1c and 1d](#)) presents a more regular porous structure with only one size corresponding to small tracheids for the transport of fluids. In the case of SC the larger size of the pores will facilitate the wetting of particles and the agglomeration of the sawdust since the binder will penetrate more easily.

In order to confirm the different porous structures of the two sawdusts used i.e. SP (softwood) and SC (hardwood), the textural characteristics of both sawdusts have been assessed by means of He and Hg picnometry and Hg porosimetry. [Table 4](#) displays the textural characteristics of the sawdusts. The total pore volume (V_{total}) and porosity (ϵ) are higher in the case of SC but the most significant difference is to be found in their different macropore volumes (V_{macro}). The macropore volume of SC is more than double that of SP, facilitating the entrance of the binder to produce denser briquettes.

To study the effectiveness of the two binders, a comparison of the density of B4 with that of BM4 and of B1 with that of BM1 was carried out. The values in [Table 3](#) indicate that BM4 has a density 19 % lower than that of B4, while BM1 has a density which is only 10 % lower than that of B1. The different effectiveness of the two binders may be related to the presence of inert materials in the case of M [\[30\]](#). BM1_{SP} has the lowest density (38 % lower than B4) because it combines the lesser effectiveness of the tar decanter sludge (M) as binder and the presence of SP which produces briquettes with a lower density.

3.2. Modification of coal thermoplastic properties

As a first step for studying the modification by the briquettes of coal fluidity, the effect of the briquette components was tested. [Figure 2a and 2b](#) show the effect of increasing amounts of binder, biomass and non-coking coal respectively. Both binders produce an increase in fluidity but it is greater in the case of the tar, probably due to the

amount of inerts present in the tar decanter sludge. Such bituminous additives favor coal fluidity because they provide molecular species that have the effect of plasticizing the entire system. On the other hand sawdust and non-coking coal cause a decrease in fluidity (Figure 2b). It can be seen that an addition of 2 wt.% of sawdust produces a reduction of between 40-45 % in the fluidity of the coal blend, which is slightly higher in the case of pine sawdust. The drastic reduction in coal fluidity due to biomass has already been reported in the case of individual coals and blends [6,16,17]. The reduction observed when adding to a high volatile and highly fluid coal was slightly greater [16] than for the coal blends [6]. The effect is lower for coal blends probably due to a compensatory effect coming from the number of coals of different rank and origin. Clearly more work is needed in order to determine the effect of sawdust as a function of coal rank and in coal blends.

Figure 2b shows that the non-coking coal (C) produced a reduction in fluidity but to a lower extent than the two sawdusts. Fitting the points corresponding to each additive in Figure 2b produces lines with slopes of -17.8, -16.2 and -3.4 for SP, SC and the non-coking coal respectively.

The composition of the briquettes includes a binder that increases fluidity but also inerts that produces a decrease in fluidity. In order to confirm the combined effect, mixtures of the coal blend with increasing amounts of the briquette were prepared and their fluidity tested, as shown in Figure 3. All the briquettes tested produced a reduction in the fluidity of the coal blends. The deleterious effect in decreasing order was as follows: B4<BM4<BM1sp<B1<BM1. According to the results in Figure 2, the briquettes with no sawdust produce the least deleterious effect followed by those with sawdust. Although the binders increase fluidity they are not able to compensate for the decrease in fluidity caused by the non-coking coal and the sawdust. The fluidity of the mixture was in all cases greater than 300 ddpm which is inside the range required for a coking coal i.e. between 200 and 1000 ddpm.

3.3. Modification of coke quality due to briquette addition

The cokemaking industry uses standardized methods to evaluate the quality of the cokes produced. In the case of mechanical strength the tests usually provide two indices, one of which represents the cohesiveness of the material and the other resistance to abrasion. It is usually the case that the higher the cohesiveness is, the lower the resistance to abrasion. Another quality parameter normally measured is its reactivity to CO₂ and its mechanical strength after reaction which is shown by means of the CRI/CSR indices. These two parameters give an indication of the behavior of a coke in the blast furnace. The raw materials used to prepare the briquettes have a different origin, composition, structure etc. which will affect the coke properties (i.e. the CRI and CSR indices will be different). To clarify the effect on coke reactivity, the briquette components were gasified in CO₂ at 1100 °C. This temperature was chosen because it is the temperature used to determine the CRI index used in the cokemaking industry to characterize coke quality. Figure 4 shows the variation of conversion (x) and reactivity (r) expressed as the derivative of the conversion with time for all the briquette components. From the TG/DTG data the following parameters were derived (see Table 5): time and reactivity for a conversion of 50 % ($t_{x=50\%}$ and $r_{x=50\%}$), time for maximum reactivity (t_{rmax}), maximum reactivity (r_{max}) and conversion at maximum reactivity (x_{rmax}). The two sawdusts required the least time to achieve a conversion of 50 %, around 4 min while in the case of the rest of the materials more than 8 min was needed. In addition the maximum gasification rate was more than three times that of the other materials (Table 5). The high reactivity of the biomass chars is due to the presence of K and Ca salts which catalyze the gasification reaction [31]. In the present research work the composition of the ashes from the two sawdusts was studied and the results are presented in Table 6. The basicity index has been calculated taking into account the ash content and the ratio of basic to acidic oxides [32]. This index is higher in the case of SC due to the higher ash content but if the ash content is ignored the ratio of

basic to acidic oxides is very similar for both sawdusts (1.49 for SC and 1.57 for SP). The sawdust from pine has a very low ash content (0.3 wt.% db) but the K content of the ashes is much higher than in SC (14 vs. 4 wt.%) and some authors have linked the K content to the catalytic effect of the biomass ashes [33]. The two sawdusts have the highest reactivity followed by the non-coking coal and the coal tar sludge and the tar. This reactivity order may serve as an indicator of the modification of the reactivity of the cokes prepared with the briquettes.

The variation of the coke quality indices is expressed as the difference between the value obtained for the coke produced from the industrial coal blend and that corresponding to the coke prepared from the mixture of the coal blend and the briquette. The quality of the cokes produced from the base blends is presented in Table 7. As can be observed, the CSR value which is the parameter generally accepted by cokemaking industry to evaluate coke quality is the same for all three base blends. Figure 5 shows the variation of the mechanical strength index (DI150/15) and the CRI and CSR indices with the amount of briquettes B1, B4, BM1 and BM4. In the graphs the difference in the quality index corresponding to the base blend and that corresponding to its mixtures with briquettes is expressed as Δ JIS D150/15, Δ CRI and Δ CSR. It is possible by means of this comparison to evaluate the effect of including sawdust (B1 vs. B4 and BM1 vs. BM4) and the effect of the type of binder (B1 vs BM1 and B4 vs.

BM4). The addition of these four types of briquettes produces a decrease in the JIS index. The lowest values correspond to the briquettes that contain sawdust. In the case of 15 wt.% addition the results are very similar both with and without biomass (BM1 and BM4 respectively) when coal tar sludge (M) is used as binder. It appears that, when sawdust is used, the presence of filler material in the binder has a beneficial effect on the integration of the biomass within the briquette and as a result in the coke

matrix, by reducing the risk of fissures or weak points where rupture is likely to occur. When sawdust is included, M is better than T as binder.

The addition of briquettes with no sawdust (B4 and BM4) does not produce any significant variation of the CRI index (less than 2 points for 15 wt.% addition). On the other hand the inclusion of biomass in the briquettes causes an increase in the CRI index, especially when coal tar sludge is used as binder ($\Delta\text{CRI}=8$ for 15 wt.% BM1). In this case the binder makes a difference and coal tar produces briquettes that result in a lower increase in coke reactivity. As is well-known, the CRI and CSR indices are linearly related and low CRI values are associated to high CSR values [26,28]. The addition of briquettes with no sawdust produces cokes with similar CSR values as those of the base blend. The worst results were obtained for briquette BM1 with 15 wt.% addition: $\Delta\text{CSR}=-14$. These poor results may be due to the lower density of the briquettes that include sawdust, as can be seen in Table 3. In order to explain the differences observed between using tar (T) or coal tar sludge (M) as binder, mixtures of the coal blend CB4 with direct addition of T and M were carried out. The results corresponding to the quality of the cokes produced are presented in the Supplementary Information (Figure 1S). The addition of both binders produced cokes with a lower mechanical strength than that of the base blend. The greatest difference in the D150/15 index was observed for the addition of 2.25 wt.% of binder corresponding to a 15 wt.% briquette content in the charge, with M producing a coke with a better mechanical strength than T. This might explain why the mechanical strength of the coke obtained in the case of 15 wt.% addition of B1 and BM1 was higher when M was used than with T (Figure 5). In the case of the high temperature properties of the cokes (CRI and CSR indices), T produced cokes with slightly better characteristics than M ($\Delta\text{CRI}=2$ and $\Delta\text{CSR}=2$) up to 1.5 wt.% addition which corresponds to a briquette addition of 10 wt.%. The impairment of coke quality observed in the case 15 wt.% addition of BM1 can be attributed to the effect of the sawdust.

The effect of the two types of sawdust in the preparation of the briquettes was also studied by adding briquettes BM1 and BM1_{SP}. Figure 6 displays the variation in the coke quality indices with additions of these two briquettes up to 15 wt.%. It can be seen (Figure 6) that cold mechanical strength decreases with the addition of both types of briquettes. The greatest difference was observed for 15 wt.% addition which in the case of briquettes prepared with SP implies a difference in the JIS index of 7 points, while for briquettes with SC the drop is 5. This is explained by the fact that the particle size distribution is finer in SP (20 wt.% vs. 4 wt.% of < 0.5 mm). The surface area of contact in SP is also greater, increasing the possibility of fissure and crack formation at the interface between the sawdust and the coal with a consequent decrease in mechanical strength [34]. In addition the density of the briquettes prepared with SP is lower than that of the briquettes prepared with SC. The CRI indices of the cokes produced increase with the amount of briquettes in the coking blend. Only slight differences were observed between the two sawdusts. The addition of the briquettes with SC produces a slightly more reactive coke than that with SP. This could be related to the higher reactivity of the SC char observed in the thermobalance and its higher basicity index. As a consequence of the increase in the CRI index due to the addition of briquettes the CSR diminishes.

The effect of the size of the briquettes on the quality of the cokes produced was tested by adding BM4 and comparing it to BM4_F and by adding BM1_{SP} and comparing it to BM1_{SP-F}. The coke quality indices are presented in Figure 7. In general the decrease in the JIS index up to 15 wt.% is lower than 5 points except in the case of BM1_{SP} where the decrease is 7 points. The incorporation of smaller pieces of briquette led a lower diminution of cold mechanical strength. In general size is important when using additives as it determines the incorporation of the inerts into coke matrix, small inerts being more easily incorporated than large inerts. In the case of coke reactivity to CO₂ the smaller briquettes produced a greater increase in the CRI index than the larger

ones. This is in agreement with results reported in the literature that relate the size of the biomass to the reactivity of the coke produced from its blend with coal [35]. As a consequence of the high CRI the mechanical strength after reaction (CSR index) is low compared to the base blend (13-15 points for a 15 wt.% addition of BM1_{SP-F} and BM4_F respectively). Although the charge characteristics are very important for coke quality, coking conditions also play an important role. In the present work the low bulk density values due to the low density of briquettes BM1 and BM1_{SP} led to the cokes with the lowest CSR values.

4. Conclusions

The addition of briquettes caused a decrease in the fluidity of the industrial coal blends irrespective of the amount added. However coke was produced by every mixture. The incorporation of briquettes containing sawdust, regardless of their origin impaired the cold mechanical strength, coke reactivity and post-reaction strength. Nevertheless slight differences between the two sawdusts were apparent at 15 wt.% addition. The JIS index values improved in the case of BM1 due to the lower density of the BM1_{SP} briquettes while CRI and consequently CSR improved with the addition of briquettes containing pine sawdust (BM1_{SP}) due to the amount and composition of the ashes. Coal tar sludge was only better than tar as binder when briquettes with sawdust were added and as regards cold mechanical strength (JIS index). Additions of up to 10 wt.% of briquettes containing biomass prepared with tar as binder yielded good results because the variation in the CSR was lower than 2 points. The inclusion of briquette fines produced cokes with lower CSR than when full size briquettes were used. When evaluating the effect of biomass containing briquettes the environmental benefits should be also considered.

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Table 1. Briquettes composition

	B1	B4	BM1	BM4	BM1 _{SP}
T	15	15	-	-	-
M	-	-	15	15	15
SC	15	-	15	-	-
SP	-	-	-	-	15
C	70	85	70	85	70

Table 2. Main characteristics of the raw materials

	CB2	CB3	CB4	C	SC	SP	T	M
Ash (% db) ^b	8.8	8.5	7.8	10.2	1.3	0.3	0.8 ^a	2.2 ^a
Volatile Matter (% db)	23.9	23.0	26.2	14.5	78.5	85.3	--	--
Gieseler MF ^c (ddpm)	682	--	1016	n/a	n/a	n/a	n/a	n/a
C (% db)	80.3	82.1	82.4	80.8	50.2	50.7	90.3	89.1
H (% db)	4.6	4.7	4.9	4.0	5.7	6.1	4.7	4.2
N (% db)	1.8	1.9	1.9	1.7	0.5	0.5	0.8	1.1
S (% db)	0.58	0.51	0.62	0.45	0.01	0.00	0.38	0.52
O (% db) ^c	3.9	2.1	2.4	2.9	42.3	42.4	3.0	5.1
Particle size (wt.%)								
>3 mm	12.0	20.4	26.6	14.9	1.0	0.2	n/a	n/a
2-3 mm	7.0	9.8	10.6	7.7	4.5	1.3	n/a	n/a
1-2 mm	14.8	16.6	17.0	15.3	24.6	19.8	n/a	n/a
0.5-1 mm	14.8	16.6	15.3	19.4	65.8	59.6	n/a	n/a
<0.5 mm	51.4	36.6	30.5	42.7	4.1	19.1	n/a	n/a

^aObtained in a thermobalance. ^b: dry basis. ^c: Maximum fluidity. n/a: not applicable. ^d:

Calculated by difference.

There is a discontinuity in the text in line 334.pag 5

Table 3. Apparent density of the briquettes

	B1	BM1	BM1 _{SP}	B4	BM4
ρ_{H_2O} (kg/m ³)	1312	1179	873	1415	1149

Table 4. Textural characteristics of the sawdusts.

	ρ_{Hg} (g/cm ³)	ρ_{He} (g/cm ³)	V_{total} (cm ³)	ϵ (%)	V_{macro} (cm ³)	V_{meso} (cm ³)
SC	0.642	1.378	0.830	53.3	0.778	0.052
SP	0.688	1.407	0.742	51.1	0.271	0.010

ρ_{Hg} : Apparent density. ρ_{He} : True density. V_{total} : Total pore volume. V_{macro} : Macropore volume. V_{meso} : Mesopore volume.

Table 5. Parameters derived from the gasification tests in thermobalance.

	$t_{x=50\%}$ (min)	$r_{x=50\%}$ (%/min)	$t_{r\ max}$ (min)	$r_{\ max}$ (%/min)	$x_{r\ max}$ (%)
C	8.5	5.2	4.0	8.1	19.3
SC	4.1	25.9	5.1	28.5	78.1
SP	4.1	25.3	4.7	26.5	66.3
T	36.7	1.3	16.0	1.6	18.7
M	15.7	2.8	6.7	4.4	16.8

Table 6. Ash composition of the two sawdusts.

	SiO ₂ (wt.%)	Al ₂ O ₃ (wt.%)	Fe ₂ O ₃ (wt.%)	MgO (wt.%)	CaO (wt.%)	Na ₂ O (wt.%)	K ₂ O (wt.%)	TiO ₂ (wt.%)	P ₂ O ₅ (wt.%)	BI*
SC	28.38	7.10	3.18	3.79	41.21	0.60	4.09	0.29	1.72	1.93
SP	27.51	4.67	1.33	10.71	23.08	1.38	14.00	0.11	2.52	0.47

$$*BI = Ash \cdot \frac{(Fe_2O_3 + MgO + CaO + Na_2O + K_2O)}{(SiO_2 + Al_2O_3)}$$

Table 7. Coke quality of the base blends.

	JIS D150/15	CRI	CSR
CB2	75	26	60
CB3	73	28	60
CB4	79	26	60

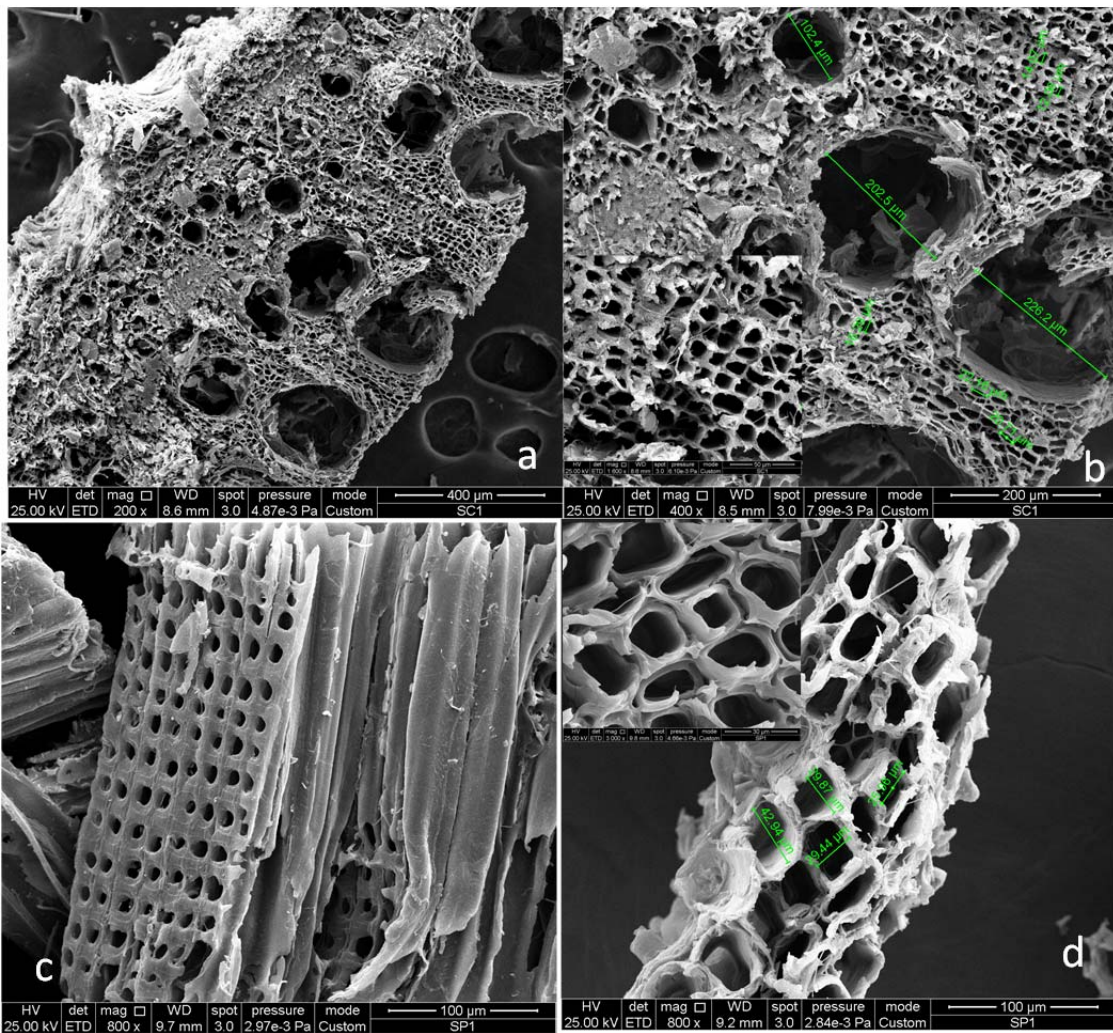


Figure 1. SEM images of sawdust showing different structure of chestnut and sawdust. SC (a and b), SP (c and d).

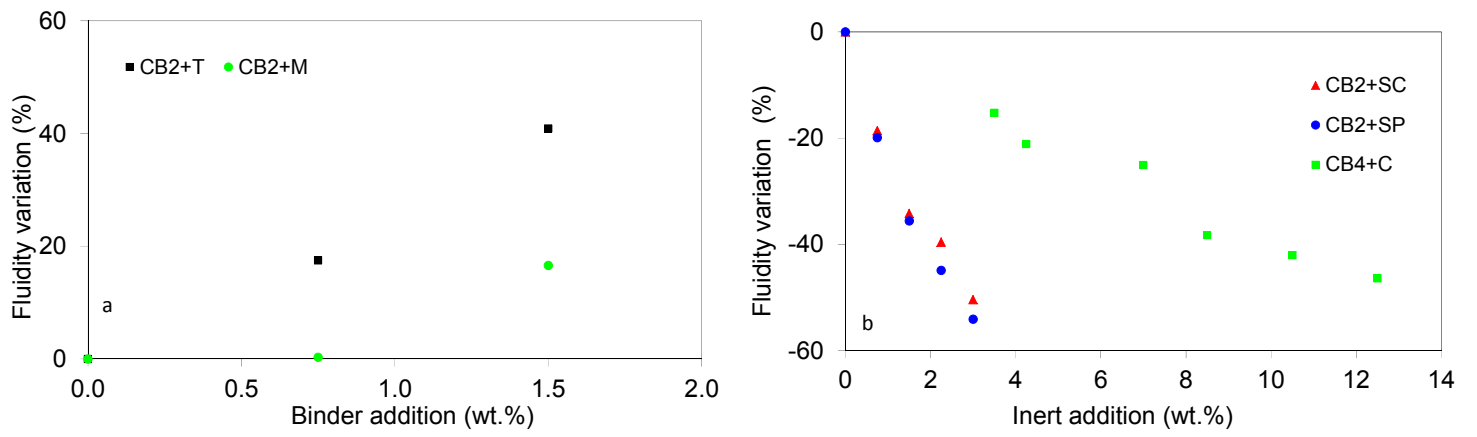


Figure 2. Variation of maximum fluidity of coal blends with the addition of the briquette components.

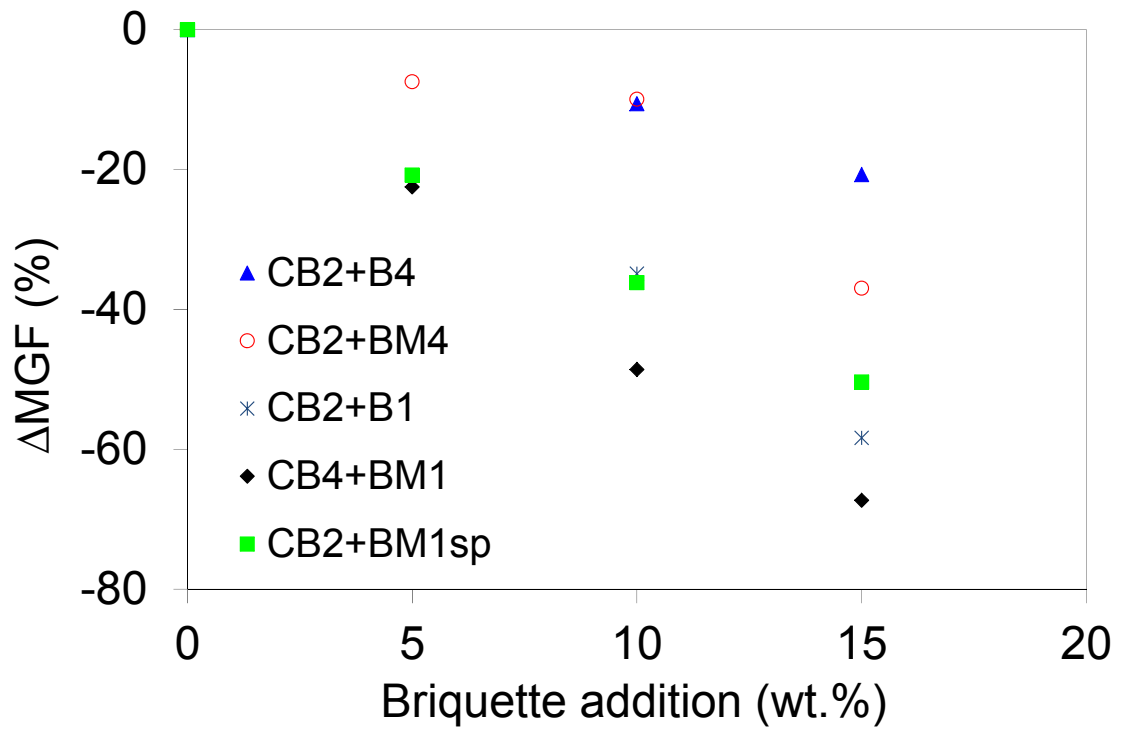


Figure 3. Influence of briquette addition on the maximum Gieseler fluidity of coal blends.

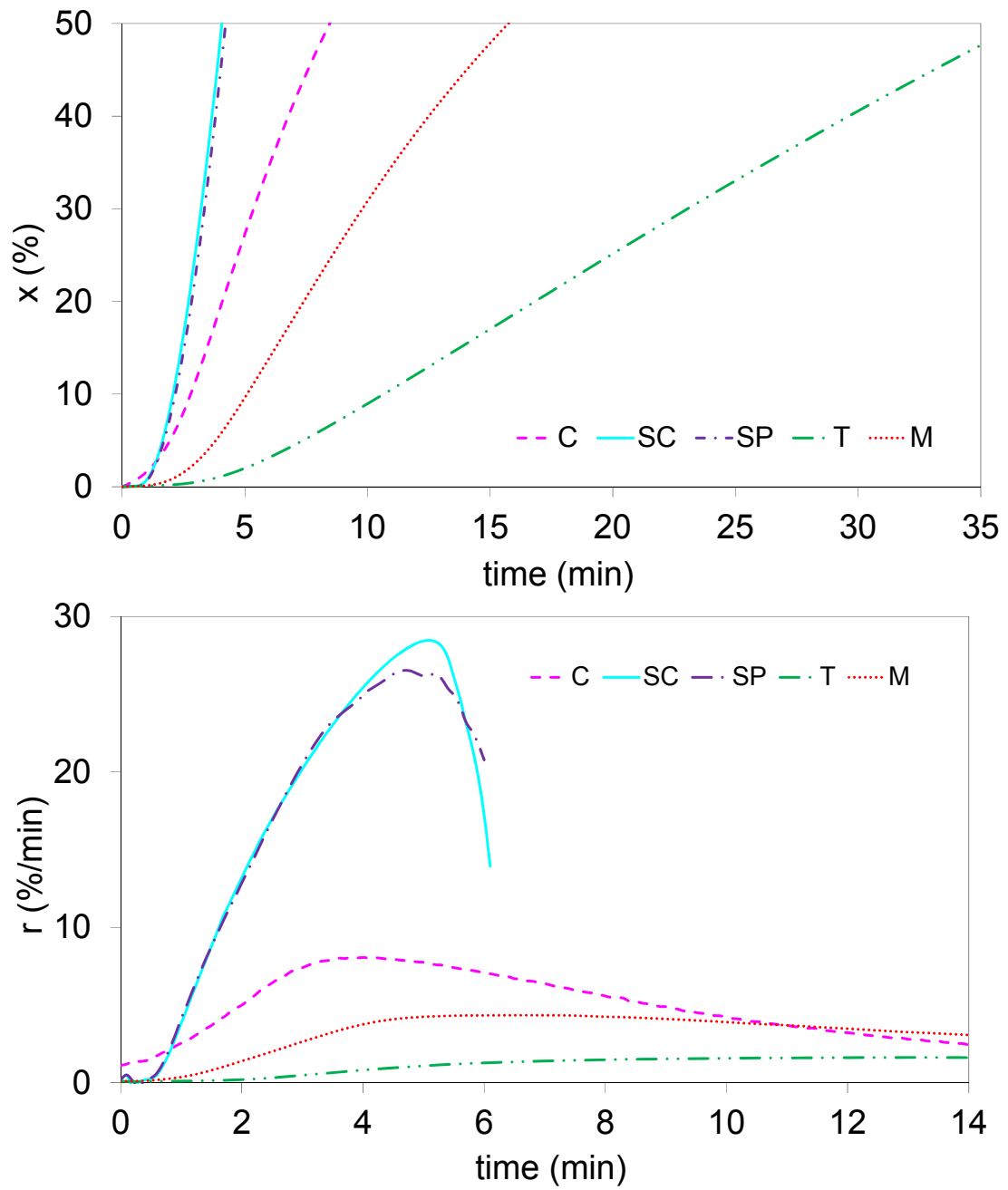


Figure 4. Variation of conversion (x) and reactivity (r) with gasification time.

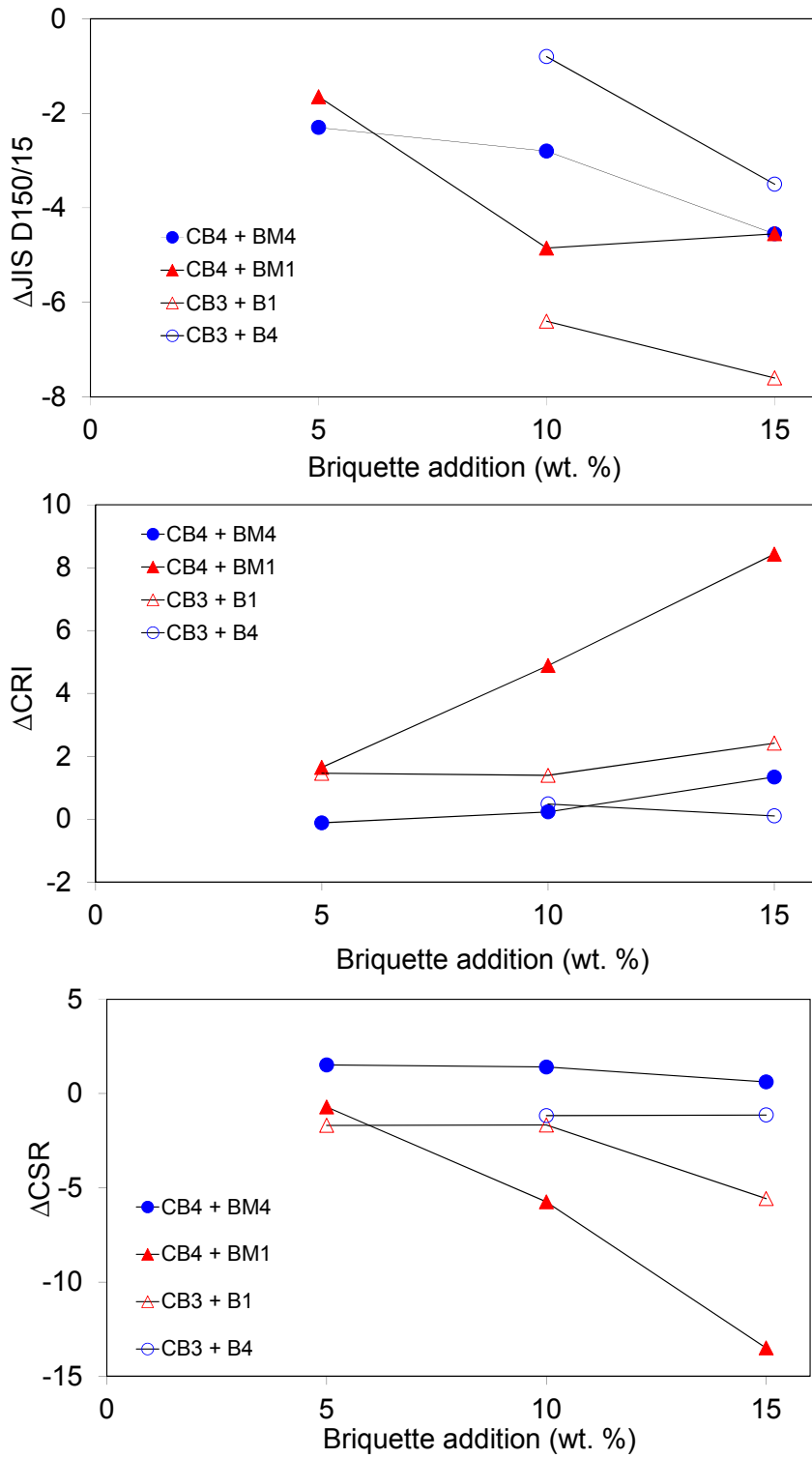


Figure 5. Variation of coke quality with briquette addition with sawdust (B1 and BM1) and without it (B4 and BM4) and using two binders coal tar (B1 and B4) and coal tar sludge (BM1 and BM4)).

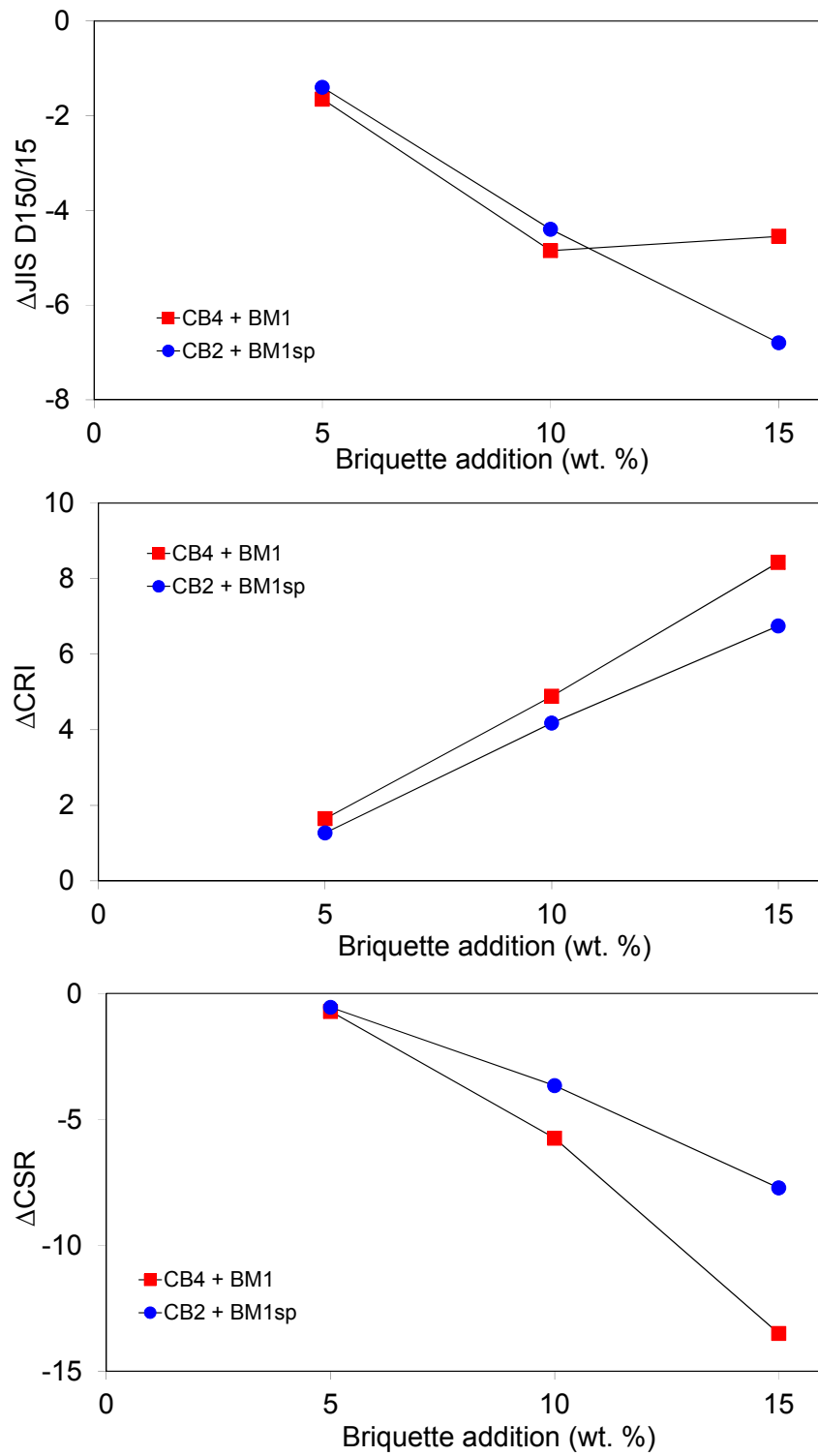


Figure 6. Variation of coke quality with briquette addition containing chestnut sawdust (BM1) and pine sawdust (BM1_{sp})

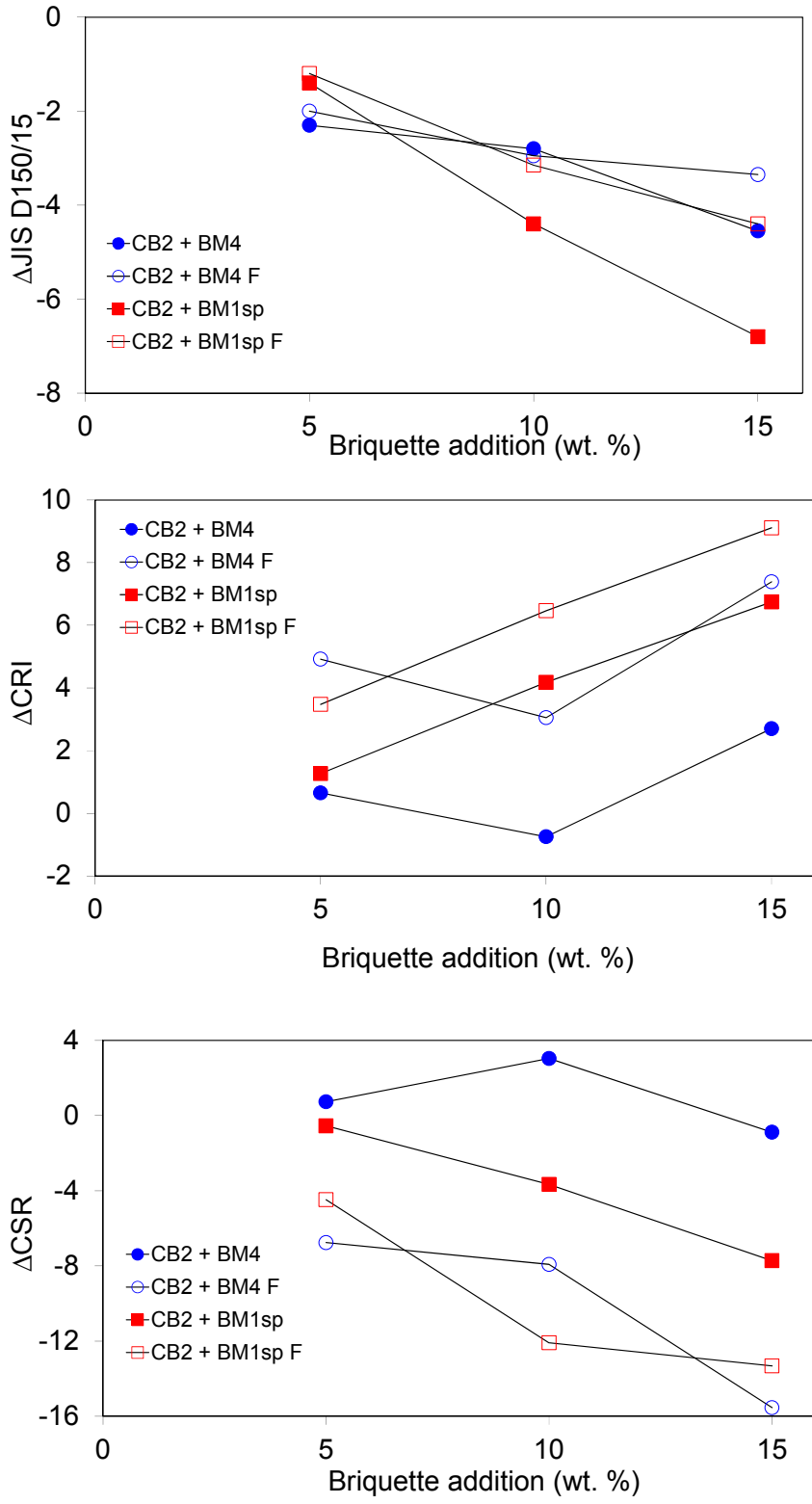


Figure 7. Variation of coke quality with briquette addition of two sizes (BM4 and BM4_F, and BM1_{SP} and BM1_{SP-F}).