

1 **Eutrophication and acidification: do they induce changes in the**
2 **dissolved organic matter dynamics in the coastal**
3 **Mediterranean Sea?**

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14
15 **Abstract**

16 Two mesocosms experiments were conducted in winter 2010 and summer 2011
17 to examine how increased pCO₂ and/or nutrient concentrations potentially
18 perturbate dissolved organic matter dynamics in natural microbial assemblages.
19 The fluorescence signals of protein- and humic-like compounds were used as a
20 proxy for labile and non-labile material, respectively, while the evolution of
21 bacterial populations, chlorophyll a (Chl a) and dissolved organic carbon (DOC)
22 concentrations were used as a proxy for biological activity. For both seasons,
23 the presence of elevated pCO₂ did not cause any significant change in the DOC
24 dynamics (p-value < 0.05). The conditions that showed the greatest changes in
25 prokaryote abundances and Chl a content were those amended with nutrients,
26 regardless of the change in pH. The temporal evolution of fluorophores and
27 optical indices revealed that the degree of humification of the organic molecules
28 and their molecular weight changed significantly in the nutrient-amended
29 treatment. The generation of protein-like compounds was paired to increases in
30 the prokaryote abundance, being higher in the nutrient-amended tanks than in
31 the control. Different patterns in the magnitude and direction of the generation
32 of humic-like molecules suggested that these changes depended on initial
33 microbial populations and the availability of extra nutrient inputs. Based on our
34 results, it is expected that in the future projected coastal scenarios the
35 eutrophication processes will favor the transformations of labile and recalcitrant
36 carbon regardless of changes in pCO₂.

37 **Keywords:** Acidification; eutrophication; mesocosms, DOM, Mediterranean
38 Sea, optical indices

1. – Introduction

As a result of human activities, atmospheric CO₂ levels have increased from approximately 280 ppm in pre-industrial times to 395 ppm in 2013 (Le Quéré et al., 2015 and references therein). A large portion of the atmospheric CO₂ is dissolved in the ocean and, thanks to the ‘solubility pump’, it is transported from the ocean’s surface to its interior in form of dissolved inorganic carbon (Volk and Hoffert, 1985). In addition to this passive diffusion of CO₂ into the ocean, marine biota plays an active role in the uptake of carbon dioxide from the atmosphere in what is known as the ‘biological pump’ which refers to the processes that involve the biologically-mediated uptake and transport of carbon from the upper to the deep ocean (Volk and Hoffert, 1985; Passow and Carlson, 2012). Thus, marine ecosystems play an important role in regulating atmospheric CO₂ concentrations and, in this way, in moderating climate change. However, the physical, chemical, and biological mechanisms governing the fluxes between the different carbon compartments in the marine system are still poorly understood.

The diffusion of CO₂ into the ocean is determined by temperature and salinity that provide a dependent solubility coefficient (Henry’s law, Henry, 1803). When a CO₂ molecule is finally taken up by the ocean, two main paths may follow: i) it may remain in a dissolved inorganic form, altering the marine carbonate chemistry equilibrium and leading to ocean acidification (Hönisch et al., 2012; Zeebe, 2012), or ii) may be captured by a photosynthetic marine organism, fixing it in the form of organic carbon. The pathways that this new biologically generated organic molecule may follow within the trophic chain are very diverse and vary from being incorporated into a larger organism (reaching higher trophic levels) to being excreted or respired as part of a variety of metabolic processes. The size of the excreted compounds varies widely, contributing to both the particulate organic matter (POM) and the dissolved organic matter (DOM) fractions. Regarding the DOM, this pool is mainly produced by phytoplankton exudation (Hopkinson et al., 2002; Romera-Castillo et al., 2011b; Sarmiento et al., 2013), viral lysis (Brussaard, 2004; Motegi et al., 2009), the sloppy feeding carried out by protists and metazoans grazers and the POM solubilization by bacterial and archaeal hydrolases (Nagata et al., 2000; Sala and Güde, 2004).

1 These mechanisms determine the quantity and the complexity of the molecules
2 contained in the DOM, as well as their fate along the biogeochemical cycles.

3
4 The estimations of oceanic CO₂ assimilation by phytoplankton to generate
5 cellular structures or its subsequent release of C as exudates (particulate and
6 dissolved primary production, respectively) range between 3 and 4 Pmol C
7 year⁻¹ (Berger, 1989; Antoine and Morel, 1996; Behrenfeld and Falkowski,
8 1997; Chavez et al., 2011). Research undertaken in the context of the US Joint
9 Global Ocean Flux Study (Schlitzer et al., 2003) concluded that a fraction of this
10 carbon is rapidly removed from surface waters and exported to the ocean's
11 interior. In addition, Jiao et al. (2010) emphasized the role of oceanic in
12 transforming POM and DOM into recalcitrant DOM, material susceptible of
13 staying sequestered in the ocean for long periods of time. The processes that
14 transform labile organic matter into refractory compounds are termed 'microbial
15 carbon pump' (MCP, Jiao et al., 2010).

16
17 The chromophoric dissolved organic fraction (CDOM; Coble, 1996) of the DOM
18 pool absorbs light at both ultraviolet (UV) and visible wavelengths. A sub-
19 fraction of this CDOM, the fluorescent DOM (FDOM; (Coble, 2007, 1996),
20 fluoresces when irradiated with UV light. Since 1990, (Coble et al., 1990) the
21 characterization of marine DOM has been performed by applying fluorescence
22 excitation-emission matrices (EEM). Although this technique does not permit
23 the quantification of specific molecules, it has been extensively used to track the
24 origin and transformations of DOM (Coble et al., 1990; Cory and McKnight,
25 2005; Nieto-Cid et al., 2005; Romera-Castillo et al., 2011a, Catalá et al, 2015)
26 because it is relatively inexpensive, low-time consuming and provides valuable
27 information about the quality of the DOM.

28
29 As it has been shown over the last years, ocean acidification affects marine
30 organisms and ecosystems in several ways (Gattuso et al., 2015 and
31 references therein). In addition, nitrogen (N) and phosphorous (P) pollution has
32 increased over the past decades, primarily due to the utilization of active N and
33 P for fertilizer use (Galloway et al., 2004). This utilization has enhanced the
34 nutrient loads from land to coastal zones and may favor an increase of

1 eutrophication episodes in the near future (Howarth & Marino, 2006). Since the
2 beginning of the 20th century, eutrophication has been a persistent problem and
3 a subject of different studies. Bio-assay experiments in lake and coastal
4 systems were done to test the effect of eutrophication on phytoplankton
5 dynamics in the seventies and eighties (Pomeroy et al., 1972; Carpenter &
6 Capone, 1983). Since then, numerous studies have been addressed this issue
7 in different aquatic systems (Statham, 2012 and references therein).

8
9 A convenient procedure to gain insight on the possible changes that ocean
10 acidification and eutrophication may induce on marine systems is the
11 deployment of mesocosms experiments (Kim et al., 2011; Teeling et al., 2012;
12 Riebesell et al., 2007, 2013; Bunse et al., 2016). Three recent mesocosms
13 studies (Yamada et al., 2013; Riebesell et al., 2013; Zark et al., 2015) have
14 examined the effects of ocean acidification on DOM transformation processes.
15 Yamada et al. (2013) did not find a significant effect of increased CO₂
16 concentration on the short-term decomposition of labile DOM in Sagami Bay
17 (Japan), yet the study did not look at the possible changes in organic matter
18 quality. The study conducted by Riebesell and collaborators (2013) in Svalbard
19 (Norway) shed light on the pathways that the organic matter followed when the
20 system was amended with nutrients and increased in pCO₂. They found that the
21 combination of these two stressors triggered a synergistic effect inducing an
22 increase in the dissolved organic carbon fraction. The study of Zark et al. (2015)
23 tracked the transformations suffered by DOM molecules in a mesocosms study
24 using Fourier transform ion cyclotron resonance mass spectrometry (FT-ICR-
25 MS) and they concluded that ocean acidification alone did not induce changes
26 in the composition of the DOM pool in the Gullmar Fjord (Sweden).

27
28 We investigated the effects of increasing pCO₂ and its synergy with increasing
29 nutrient availability on the dynamics of organic matter in a Mediterranean
30 coastal area. We have particularly examined the optically active fractions of the
31 DOM as they have been used as indicators of recalcitrant material and can
32 provide useful information about DOM transformations. In addition, the study of
33 these fractions is of remarkable interest in the Mediterranean waters where
34 CDOM to chlorophyll ratio is higher than the global average (Morel & Gentili,

1 2009; Claustre et al., 2002). We enclosed coastal water in mesocosms and
2 performed two experimental studies in which we manipulated pCO₂ and nutrient
3 concentrations. In order to assess the importance of the initial conditions in
4 regulating the responses to reducing pH and increasing nutrients, one
5 mesocosm experiment was performed in winter and the other in summer,
6 displaying contrasting initial oceanographic and biological characteristics.

7 8 **2. - Materials and methods**

9 **2.1. - Experimental setup and initial conditions at the sampling site**

10 Two experiments were conducted in winter 2010 and summer 2011 (W and S,
11 respectively) to examine the dynamics of microbial communities and organic
12 matter under different pH conditions and nutrient levels. Natural seawater from
13 the Blanes Bay Microbial Observatory, NW Mediterranean (BBMO; 41°40'0" N,
14 21°48'0" E; Gasol et al., 2012), was enclosed in eight 200 L tanks and
15 maintained in a temperature-controlled chamber, with a 12:12 h light:dark cycle.
16 Gro-lux and cool-white lamps were positioned in the walls of the chamber
17 surrounding the tanks. Light intensity inside the containers was 121.3 ± 3.5
18 $\mu\text{mol m}^{-2} \text{s}^{-1}$ during the winter experiment and $230 \pm 25 \mu\text{mol m}^{-2} \text{s}^{-1}$ during the
19 summer experiment, measured using a spherical radiometer (Biospherical
20 Instruments Inc., Model QSL 2100, San Diego, CA).

21
22 Four experimental conditions were randomly assigned to duplicated containers:
23 K1 and K2 (controls), KA1 and KA2 (reduced pH) N1 and N2 (nutrient
24 amended), and NA1 and NA2 (nutrient amended and reduced pH). The pH in
25 the KA and NA treatments was manually adjusted by bubbling CO₂ every
26 morning in a controlled way, to lower their pH in approximately 0.2 units respect
27 to the controls, so as to simulate future conditions in a medium-level mitigation
28 scenario such as the Representative Concentration Pathway (RCP) 4.5 (Taylor
29 et al., 2015). For reproducibility, the control tanks were also bubbled with
30 compressed air at current atmospheric CO₂ concentrations.

31
32 The seasonal cycle in Blanes Bay is characterized by a late winter
33 phytoplankton bloom dominated mostly by diatoms (Guadayol et al., 2009). In
34 contrast, during summer, when nutrient concentrations are lower,

1 picophytoplankton is the most representative group (Alonso-Sáez et al., 2008).
2 Moreover, DOC accumulates during summer, while annual minimum
3 concentrations are found in winter (Vila-Reixach et al., 2012; Romera-Castillo et
4 al., 2013). Due to this seasonality, the initial seawater of the winter experiment
5 was relatively rich in inorganic nutrients and poor in DOC. On the contrary, in
6 summer the water was depleted of inorganic nutrients and enriched in DOC,
7 generated via metabolic pathways during the bloom phase within the previous
8 spring season (Romera-Castillo et al., 2013). The summer nutrient depletion
9 limits the bacterial activity, reducing the microbial degradation of DOM and
10 leading to a DOC accumulation in this season (Thingstad et al., 1997) Thus, the
11 starting point conditions of the experiments differed in the original
12 concentrations of organic matter and inorganic nutrients.

13

14 **2.2. – Measured variables**

15 Measurements of the following variables were taken every day during 9 days.
16 Duplicate containers for each of the four treatments were simultaneously and
17 independently sampled. Temperature was monitored daily using a digital
18 thermometer *VWR 8202-156* (VWR International, LLC). This variable was set to
19 14 ± 1 °C and to 22 ± 1 °C for W and S experiments, respectively. The pH in the
20 mesocosms was determined every morning by spectrophotometry in the
21 laboratory, following standard procedures (Clayton and Byrne, 1993). In
22 addition, pH was continuously recorded using glass electrodes (Ecotrode Plus,
23 Metrohm) connected to a D130 data logger (Consort, Belgium) that were
24 calibrated on a daily basis with a Tris buffer following standard procedures
25 (SOP6a of Dickson et al., 2007). Chlorophyll a (Chl a) was measured according
26 to Yentsch and Menzel (1963): seawater (50 mL) was filtered through Whatman
27 GF/F glass fiber filters, which were subsequently placed in 90% acetone at 4 °C
28 for 24 h and the fluorescence of the extract measured using a fluorometer
29 (Turner Designs, Sunnyvale, CA).

30

31 Dissolved inorganic nutrient concentrations, nitrate (NO_3^-), phosphate (PO_4^{3-})
32 and silicate (SiO_2), were determined by standard segmented flow analyses with
33 colorimetric detection (Hansen & Grasshoff 1983) using a CFA Bran + Luebbe
34 autoanalyser. Precisions were $\pm 0.01 \mu\text{mol kg}^{-1} \text{NO}_3^-$, $\pm 0.02 \mu\text{mol kg}^{-1} \text{PO}_4^{3-}$, and

1 $\pm 0.01 \mu\text{mol kg}^{-1} \text{SiO}_2$. Inorganic nutrients were added to N and NA treatments to
 2 reach a final P:N:Si molar concentration of 1:16:30 and 0.25:4:8 in the winter
 3 and summer experiments, respectively. Initial and post-addition nutrient
 4 concentrations are summarized in Table 1. In both cases, the nitrogen
 5 enrichment was increased at least eight times from the seasonal average
 6 concentration measured in the BBMO during the last 10 years. Nitrogen and
 7 phosphorus were added at a Redfield ratio, whereas silicate was added in
 8 excess, so diatom growth was not affected by lack of elemental compounds.

9
 10 **Table 1.** Concentrations of the main inorganic nutrients measured before and
 11 after the additions. Values are expressed in μM . The standard deviations were
 12 calculated using duplicated containers of the same experimental condition.

	Winter		Summer	
	Before addition	After addition	Before addition	After addition
NO_3^-	2.55 ± 0.00	17.38 ± 0.45	0.10 ± 0.01	4.70 ± 0.06
PO_4^{3-}	0.11 ± 0.00	1.14 ± 0.07	0.03 ± 0.00	0.24 ± 0.03
SiO_2	2.33 ± 0.00	31.41 ± 0.79	0.37 ± 0.03	6.51 ± 0.13

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 14
 15 Samples for dissolved organic carbon (DOC), FDOM and CDOM were
 16 prefiltered under reduced pressure through precombusted (450 °C, 4h)
 17 Whatman glass fiber filters (GF/F). DOC samples were collected in 10 mL
 18 precombusted (450 °C, 24 h) glass ampoules, acidified with 50 μL 25% H_3PO_4
 19 to pH <2 and heat-sealed and stored in the dark at 4 °C until analysis. A
 20 Shimadzu TOC-CSV organic carbon analyzer was used to carry out analysis.
 21 Three to five injections of 150 μL per sample were performed, and DOC
 22 concentrations were calculated by subtracting a Milli-Q blank and dividing by
 23 the slope of a daily standard curve of potassium hydrogen phthalate. The
 24 precision of these measurements was $\pm 0.7 \mu\text{M}$. All samples were checked
 25 against deep Sargasso Sea reference water (2600 m).

26
 27 CDOM absorption spectra were determined from 250 to 600 nm using a Varian
 28 Cary 100 Bio spectrophotometer equipped with 10 cm quartz-cells. Milli-Q water

1 was used as a blank. Absorbance was converted into napierian absorption
2 coefficient (a_λ , Green and Blough, 1994) using the equation:

$$3 \quad a_\lambda = \frac{2.303 \cdot \text{Abs}_\lambda}{l} \quad (1)$$

4
5
6 where Abs_λ is the absorbance at a given wavelength, the factor 2.303 converts
7 from decadic to natural logarithms, and l is the cell path-length in meters. In
8 addition, the UV absorption at 254 nm was also normalized to the dissolved
9 organic carbon (DOC) concentration to obtain the specific UV absorbance
10 coefficient (SUVA, in $\text{m}^{-1} \text{mg}^{-1} \text{L}$) following Weishaar et al. (2003). SUVA values
11 are correlated with DOM aromaticity and provide information on the complexity
12 of molecules (Helms et al., 2008; Weishaar et al., 2003). Furthermore, the
13 dimensionless slope ratio of a short wavelength region (275-295 nm) to that of a
14 longer wavelength region (350-400 nm) was determined (S_R ; Helms et al.,
15 2013). S_R is inversely correlated to SUVA and is related to the molecular weight
16 of the DOM (Helms et al., 2013).

17
18 A Perkin Elmer LS55 luminescence spectrometer was used to measure FDOM.
19 This instrument was equipped with a xenon discharge lamp equivalent to 20 kW
20 for an 8- μs duration. Both, single point measurements and emission excitation
21 matrices of the samples were acquired. The scan speed was set at 250 nm min^{-1}
22 and slit widths for the excitation and emission wavelengths were fixed at 10
23 nm. Measurements were performed in a 1 cm quartz fluorescence cell.
24 Following Coble 1996, the Ex/Em wavelengths used for the single point
25 measurements were: Ex/Em 280 nm/350 nm (peak-T) indicative of the
26 presence of protein-like compounds, Ex/Em 320 nm/410 nm (peak-M) as
27 indicator of marine humic-like substances, Ex/Em 340 nm/440 nm (peak-C) to
28 trace terrestrial humic-like substances and Ex/Em 250 nm/435 nm (peak-A) to
29 track humic materials in general. Additionally, EEMs were obtained by
30 concatenating 21 excitation/emission spectra of the sample. The fluorescence
31 intensities were reported as quinine sulfate units (QSU) by calibrating the
32 instrument at Ex/Em: 350 nm/450 nm against a quinine sulfate dehydrate (QS)
33 standard made up in 0.05 mol L^{-1} sulfuric acid. Optical analyses of tryptophan

1 (Try) dissolved in seawater at different levels of pH were performed to test the
2 pH influence in the fluorescence properties of the protein-like substances.

3
4 The humification index (HIX) describes the diagenetic state of the DOM and it
5 was calculated by dividing the peak area under the emission spectra at 435-480
6 nm by the peak area under the emission spectra at 300-345 nm, at an excitation
7 of 254 nm. The aromatic humic acids are known to have high HIX values
8 (Zsolnay, 2003; Giering et al., 2014).

9
10 Heterotrophic prokaryotes were enumerated with a FACSCalibur (Becton
11 Dickinson) flow cytometer equipped with a 15 mW argon-ion laser (488 nm
12 emission) as described by Gasol and del Giorgio (2000). Samples (1.8 mL)
13 were immediately fixed with 1% paraformaldehyde plus 0.05% glutaraldehyde
14 (final concentrations), incubated for 10 min at room temperature, frozen in liquid
15 nitrogen and stored at -80 °C. Before analysis, samples were unfrozen, stained
16 with SYBRGreen I (Molecular Probes) at a final concentration of 10 µM and left
17 in the dark for about 15 min. Each sample was then run at low speed (~12 µL
18 min⁻¹) for 2 minutes with Milli-Q water as a sheath fluid. We added 10 µL per
19 sample of a solution of yellow-green 0.92 µm Polysciences latex beads (10⁶
20 beads mL⁻¹) as an internal standard. Bacteria were detected by their signature
21 in a plot of side scatter versus FL1 (green fluorescence). Data analysis was
22 performed with the Paint-A-Gate software (Becton Dickinson).

23 24 **2.3. – Statistical analyses**

25 The software SigmaPlot v11.0 (Systat Software Inc.) was used to perform the
26 two-way ANOVA and the t-tests. Two-way ANOVA was carried out to test if
27 differences between conditions and experiments were significant and t-tests to
28 discriminate if the temporal evolution of the different variables measured in an
29 experiment (winter or summer) could be considered significant. The software
30 XLSTAT 2016 (Addinsoft ©) was used to perform Mantel tests. The Pearson
31 correlation implemented in the Mantel tests was performed to discriminate if
32 changes in the intensity of the fluorophores were significant between conditions.
33 p-values were set to $p < 0.05$ for all types of test.

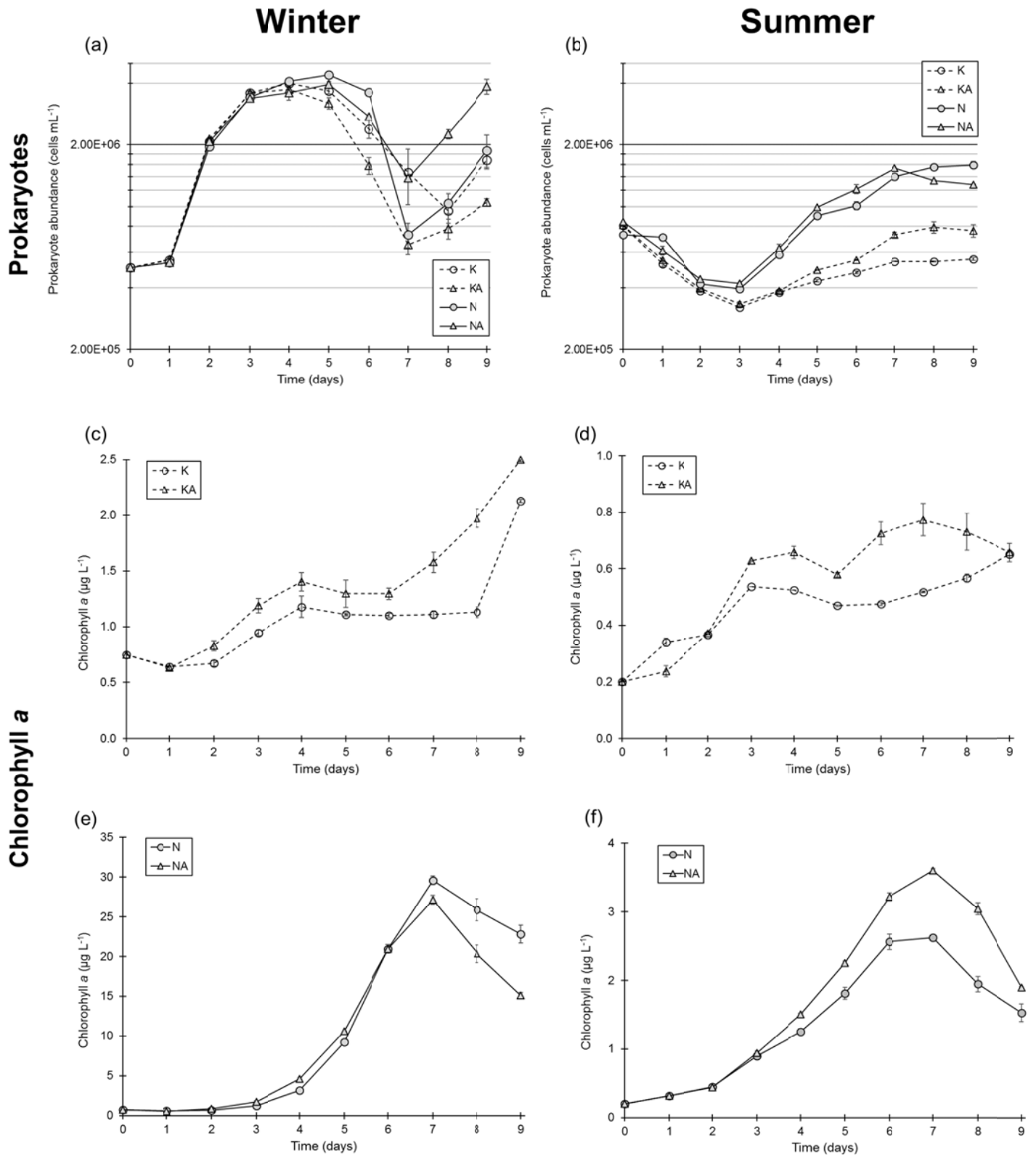
1 **3.- Results**

2 **3.1. – Plankton dynamics**

3 Figure 1 shows the differences between prokaryotic abundances and Chl *a*
4 levels for the different treatments during winter and summer experiments.
5 Prokaryotic abundances started at $5.0 \pm 0.1 \cdot 10^5$ cells mL⁻¹ at t_0 in the winter
6 experiment. Between t_3 and t_4 , the abundances in K and KA conditions reached
7 the highest numbers 4.0 ± 0.2 and $3.7 \pm 0.4 \cdot 10^6$ cells mL⁻¹, respectively. Within
8 a short time lag, slightly higher values were reached in N and NA tanks ($4.4 \pm$
9 0.3 and $4.0 \pm 0.3 \cdot 10^6$ cells mL⁻¹, respectively). Between t_5 and t_7 , prokaryotic
10 abundances decreased markedly and, by the end of the experiment the
11 abundances increased again in all conditions (Figure 1a). Prokaryotic
12 abundances at t_0 ranged from 7.3 ± 0.2 to $8.4 \pm 0.2 \cdot 10^5$ cells mL⁻¹ in the
13 summer experiment (Figure 1b). An initial drop was observed in all conditions,
14 reaching the lowest values at t_3 . After this time point, the prokaryotic
15 populations started to increase in all treatments. Prokaryotic numbers in K and
16 KA treatments were lower than those in treatments N and NA.

17

18 Chl *a* concentrations varied in a similar way in both the winter and summer
19 cases: Chl *a* under N and NA experimental conditions (Figure 1e, f) reached
20 higher concentrations than under K and KA treatments (Figure 1c, d). The Chl *a*
21 values were about three to eight times higher in the winter experiment, which
22 relates to the higher nutrient enrichment induced in that experiment.



7 **Figure 1.** Temporal dynamics of prokaryote abundances (cells mL⁻¹) in (a) winter and
 8 in (b) summer; chlorophyll a pigment concentration (µg L⁻¹) for K and KA treatments in
 9 (c) winter and (d) summer; chlorophyll a pigment concentration (µg L⁻¹) for N and NA
 10 treatments in (e) winter and (f) summer. Note that, in (c), (d), (e) and (f) panels, the
 11 scales of the vertical axes are different. Error bars indicate the standard error of 2
 12 replicates.

1 **3.2. – DOC**

2 DOC concentration in the winter experiment increased in the four treatments
3 (Figure 2a) reaching maximum values at t_7 in treatments K and KA, and t_9 in N
4 and NA. From t_7 to t_9 , DOC decreased in both K and KA treatments, while DOC
5 concentration kept increasing in N and NA conditions, coinciding with the decay
6 of phytoplankton bloom. During the summer experiment, small variations in
7 DOC concentration were observed (Figure 2b). In general, a positive trend to
8 higher concentrations was identified during the entire incubation. The starting
9 point conditions were about $80\text{-}85 \mu\text{mol L}^{-1}$ and the final concentrations
10 increased to $90\text{-}95 \mu\text{mol L}^{-1}$. Regarding nitrate concentrations, they generally
11 decreased from the beginning of the experiment in the N and NA tanks. Similar
12 patterns were found for phosphate concentrations in these enriched conditions,
13 while small variations occurred during the whole experimental period in the
14 control treatments.

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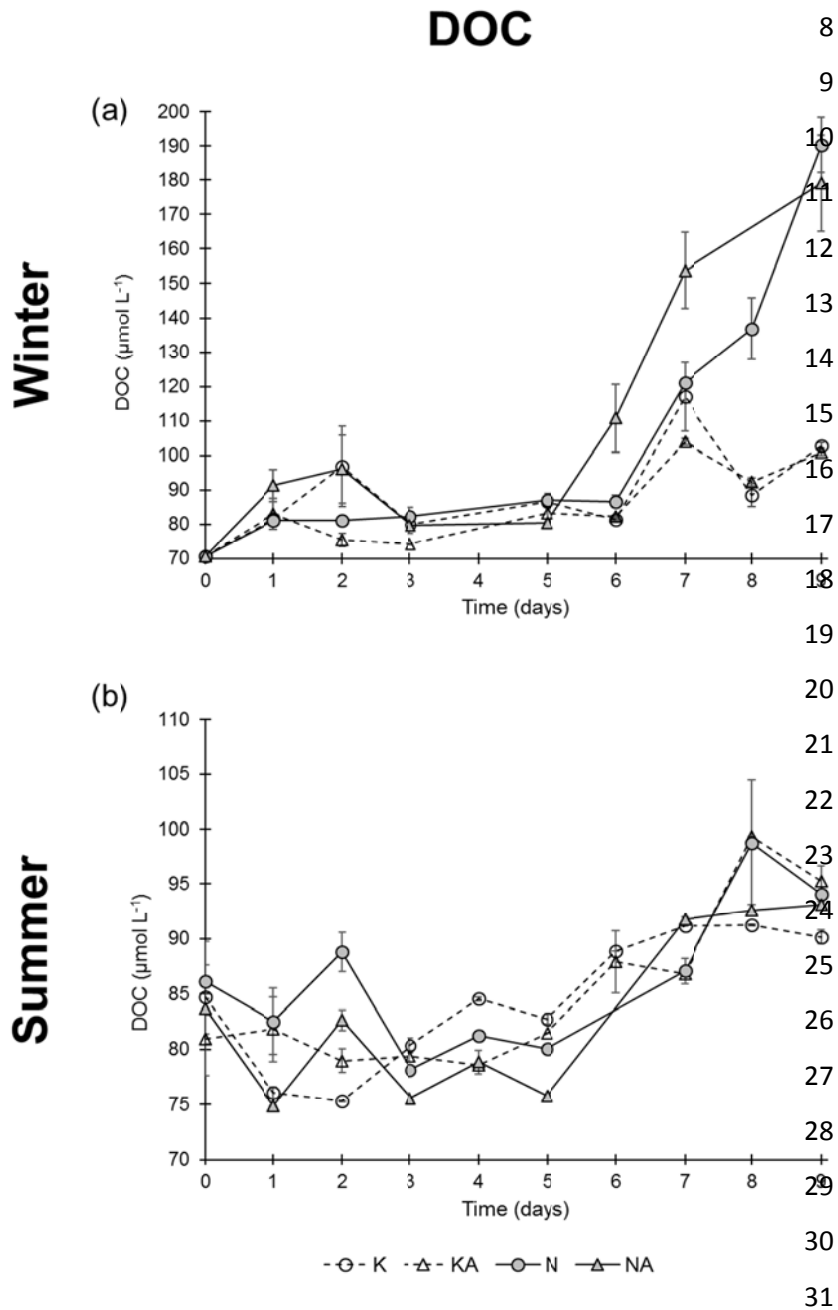
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31 **Figure 2.** Changes in DOC concentrations ($\mu\text{mol L}^{-1}$) in the (a) winter and in the (b)

32 summer experiments. Open symbols represent control (K) conditions and filled colored

33 symbols nutrient-amended (N) conditions. Note the change of scale in vertical axes.

34 Error bars indicate the error of 2 replicates.

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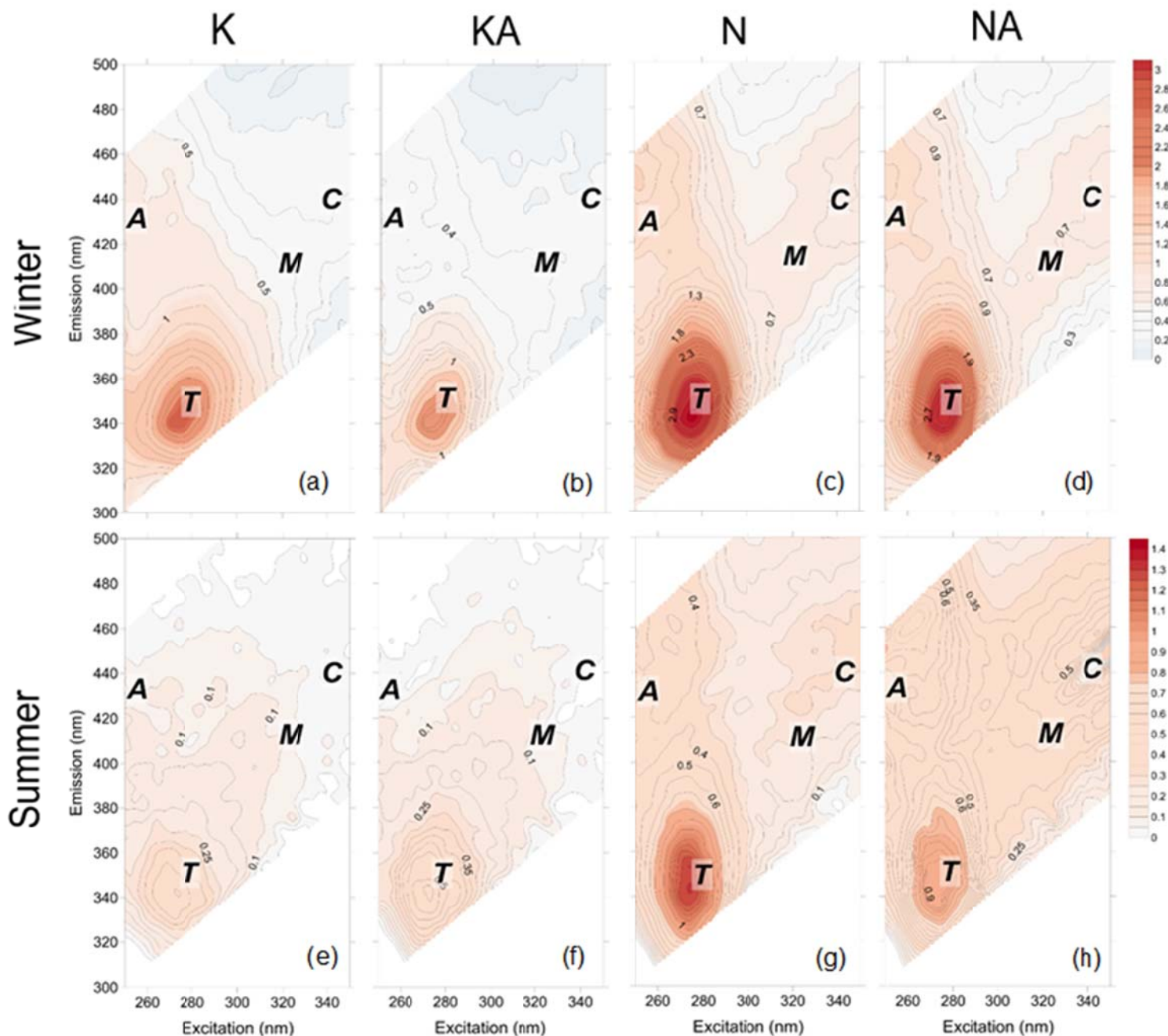
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1 **3.3. – Optical analyses of the DOM**

2 Fluorescence intensities during the experiments were measured to track
3 changes in the quality of organic matter. We visualized the differences by
4 subtracting the EEMs at t_0 from those at t_9 (Figure 3). In the winter experiment,
5 the most remarkable feature was the increase in the fluorescence signal of the
6 protein-like material (peak-T) in all treatments, including the control, which
7 reached 2.3 QSU (Figure 3a). However, in the acidified scenario (Figure 3b),
8 the increase was slightly smaller in the non-enriched treatments (~1.6 QSU).
9 Regarding the enriched mesocosms (N and NA, Figure 3c, d), we also
10 observed slighter increases of the fluorescence signal of the humic-like
11 compounds (A, C and M regions), in addition to those of the peak-T.

12

13 The patterns identified in summer and winter were similar: the main changes
14 were found around the protein-like fluorescence region, which increased in all
15 four conditions (~0.5 QSU in K and KA to ~1.4 in N and ~1.0 QSU in NA).
16 Again, slight increases of humic-like fluorescence were detected during the
17 experiments, mainly in the nutrient-enriched treatments (Figure 3g, h).



2

7 **Figure 3.** EEMs showing increases/decreases in fluorescence intensity over the 9 days
 8 of the mesocosm experiments ($\Delta\text{EEM} = \text{EEM}_{t_9} - \text{EEM}_{t_0}$) for the different fluorophores
 9 in winter (a-d) and summer (e-h). Values reported as quinine sulfate units (QSU).
 10 Humic-like fluorophores indicated as A, C and M; protein-like fluorophore indicated as
 11 T. Note the change in scale between the W and S experiments.

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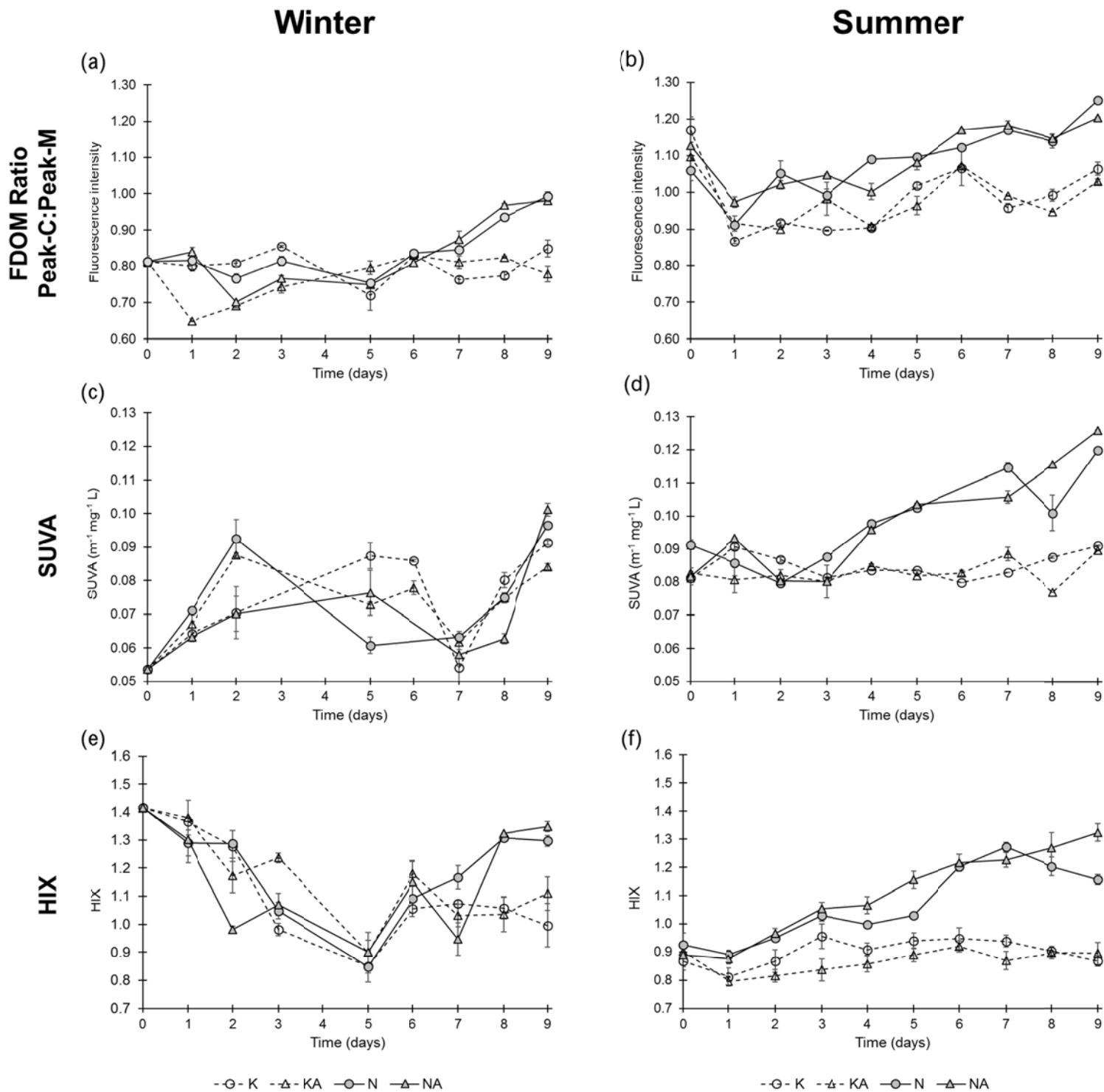
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1 The temporal evolution of peak-C to peak-M ratio (peak-C:peak-M) helped us to
2 explore, in more detail, the changes experienced by the humic-like substances
3 (Figure 4a, b). In general, the ratios were lower in winter than in summer, but
4 both experiments showed the highest ratios in the N and NA tanks at the end of
5 the experiment. In K and KA conditions, no clear trends were identified in
6 winter, while the evolution of the non-nutrient-enriched and nutrient-enriched
7 tanks was relatively parallel in summer. After a decrease observed from t_0 to t_1 ,
8 the values tended to increase until the end of the experiment reaching higher
9 values in the enriched conditions. The SUVA evolution (Figure 4c, d) did not
10 show a clear temporal pattern in the winter experiment, although by t_9 NA and N
11 conditions presented the highest increase in relation to the initial values. In
12 summer, after t_3 , the enriched- and non-enriched treatments diverged, reaching
13 significantly higher values in the N-conditions (p -value < 0.05) and, again, the
14 highest increase at the end of the experiment occurred in the NA treatment
15 ($0.040 \pm 0.004 \text{ m}^{-1} \text{ mg}^{-1} \text{ L}$) followed by the N treatment ($0.030 \pm 0.005 \text{ m}^{-1} \text{ mg}^{-1}$
16 L). In winter the K treatment displayed higher increases in SUVA than the
17 acidified control condition (KA) but this fact was not observed in summer. The
18 temporal evolution of HIX differed during the first days of the experiment
19 between the winter and summer scenarios, this index decreased until t_5 and
20 then increased until t_9 in winter, whereas it increased during the whole
21 experiment in summer. The HIX values reached at the end of the incubation
22 were always higher in the nutrient enriched conditions than in the non-enriched
23 ones. Furthermore, the values of the NA treatments were higher than the N
24 ones in both experiments.

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6 **Figure 4.** Time evolution of the quotient between peak-C and peak-M in (a) winter and
 7 (b) summer; specific UV absorbance at 254 nm (SUVA) in (c) winter and (d) summer;
 8 and humification index (HIX) in (e) winter and (f) summer. Peak-C:peak-M ratio and
 9 HIX are dimensionless variables. Error bars indicate the standard error of 2 replicates.

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1 **4. – Discussion**

2 **4.1. - DOC dynamics**

3 DOC net accumulation occurred always after the end of the exponential-growth
4 phase (coinciding with the phytoplankton post-bloom phase) either with high or
5 low pCO₂ levels (Figures 1, 2). Thus, the production of DOC, without distinction
6 of the seasonality or the addition of nutrients, was not significantly different
7 between the non-acidified and acidified tanks (K and N with respect to KA and
8 NA, p-value < 0.05). In the same way, MacGilchrist et al. (2014) found no
9 significant effect of pCO₂ on the DOC dynamics in five shipboard bioassay
10 experiments in the northwest European shelf seas. These results are also in
11 agreement with the mesocosm study by Maugendre et al. (2014) in the Bay of
12 Villefranche (France, NW Mediterranean Sea), where no significant effects of
13 elevated temperature and/or CO₂ were found on most biological parameters
14 and processes, including the generation of DOM. On the other hand, Yoshimura
15 et al. (2010) conducted incubation experiments with sea surface water
16 (depleted in nutrients) from the Sea of Okhotsk and detected a decrease in the
17 generation of DOC when pCO₂ levels were >480 μatm.

18

19 The evolution of DOC and nutrient dynamics in previous mesocosm
20 experiments, can be contradictory. In 2007, Riebesell and collaborators found
21 that although the CO₂ uptake was higher in conditions with elevated pCO₂, no
22 differences in the phytoplankton POC flux were observed. Thus, they suggested
23 that the extra CO₂ incorporated was lost as DOC or respiration. More recently,
24 in 2013, a mesocosm experiment was conducted in Svalbard to examine the
25 influence of high pCO₂ and nutrient availability on microbial activities (Riebesell
26 et al., 2013). In that experiment, pico-phytoplankton growth and DOC exudation
27 increased at elevated CO₂ concentrations after inorganic nutrients were
28 supplied. Another mesocosm experiment conducted in waters off the Baltic Sea
29 during 4 weeks in the summer season (Paul et al., 2015) revealed that under
30 high pCO₂ an important percentage of the organic matter production was in
31 dissolved form. In our study, the abundance of small phytoplankton (pico- and
32 nanoeukaryotes) was stimulated in the enriched conditions of the summer
33 experiment (Sala et al., 2016). However, this stimulation was not accompanied
34 by a net increase of DOM. In the nutrient-enriched conditions of the winter

1 scenario, we found an increase of DOC due to the phytoplankton bloom
2 (dominated by diatoms) reaching discrete higher values (although not
3 significant) during the bloom phase when high pCO₂ were induced. Moreover,
4 the experiments performed by Kim et al. (2011) with mesocosm enclosures in
5 Korean coastal waters, showed that when the pCO₂ and temperature increased,
6 the production of DOC was enhanced. A different study conducted by
7 Yoshimura et al. (2013) in the sub-Arctic Pacific obtained higher concentrations
8 of DOC in the lowest pCO₂ treatment (300 µatm) over the first 10 days of
9 incubation. Thus, in discordance with Kim et al. (2011), Riebesell et al. (2013),
10 Yoshimura et al. (2013) and Paul et al. (2015), no significant differences were
11 observed in DOC dynamics between acidified and non-acidified conditions.

12

13 Looking in further detail to the relationship between DOC and phytoplankton
14 biomass (DOC:Chl *a* ratio, Figure 5), the highest values of this ratio were found
15 under non-enriched conditions in both experiments. This fact could be due to a
16 nutrient limitation of prokaryote growth as it has been described in open and
17 coastal Mediterranean waters during low nutrient concentration episodes
18 (Thingstad et al., 1997; Sala et al., 2002). Besides, for the non-enriched
19 conditions, we found higher values of this ratio in the acidified conditions in
20 winter, while the opposite pattern was observed in summer. The high values
21 found for this quotient in summer have been previously discussed in different
22 studies conducted in Mediterranean waters (Morel and Gentili, 2009; Organelli
23 et al., 2014). As described in the work of Romera-Castillo et al. (2013), at the
24 Blanes Bay sampling site, the DOC accumulates during summer when the
25 degradation of organic matter by heterotrophic prokaryotes is reduced due to
26 the depletion of inorganic nutrients. In that scenario, DOC accumulates and the
27 ratio DOC/Chl *a*, increases with respect to winter.

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DOC:Chl a

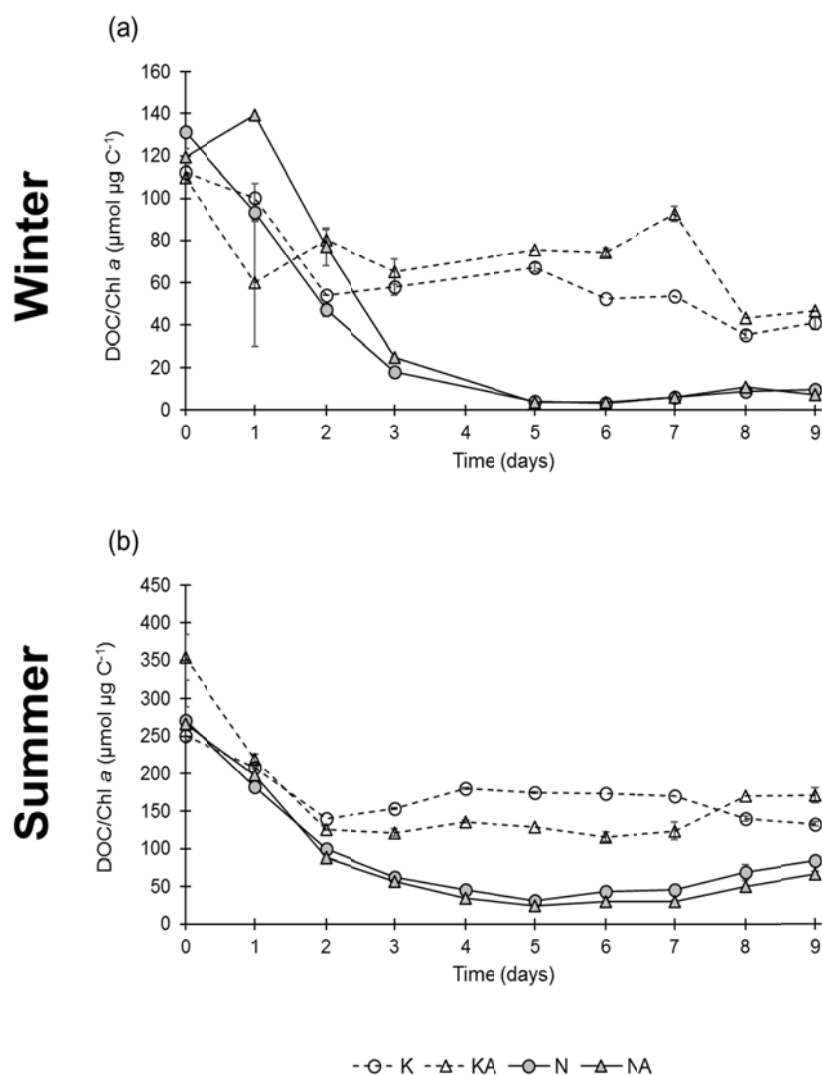


Figure 5. Time evolution of the quotient between DOC and Chl a in (a) winter and (b) summer. Units are expressed in $\mu\text{mol } \mu\text{g C}^{-1}$. Error bars indicate the standard error of 2 replicates.

4.3. – FDOM dynamics

Since fluorescence excitation-emission matrices are spectral signatures of the FDOM, they are useful to track the changes of different DOM constituents over time. As explained above, previous studies have hypothesized that the increasing concentration of CO_2 in seawater could channel the extra organic carbon fixed by photosynthesis into the dissolved fraction. Despite not finding accumulation of DOC in the treatments amended with high pCO_2 , we

1 consistently observed changes in the quality of DOM (Figure 3). The temporal
2 evolution of the four main fluorophores revealed an increase with time, so no
3 net consumption of FDOM was detected in any of the treatments. Data depicted
4 in Figure 3, confirms that the increase in the humic-like fluorescent signal was
5 greater in the tanks where nutrients were added than in the non-enriched tanks.
6 This would agree with the increase of CDOM compounds after enrichment with
7 nitrate found in the experiments performed by Lekunberri et al. (2012) and by
8 Yuan et al. (2016) in the Mediterranean Sea and the South China Sea,
9 respectively.

10
11 Acidification conditions could potentially alter the optical properties of the
12 protein- and humic-like portions of FDOM. Previous studies indicated that only
13 pH levels above and below specific values (i.e., a pH <3 or >8) have potential to
14 significantly change the structure of the DOM and thus induce a reduction of the
15 fluorescence efficiency of the humic-like molecules (Laane, 1982; Dryer et al.,
16 2008; Yan et al., 2013). Equally, it has been reported that the fluorescence
17 signal of protein-like compounds as tyrosine and tryptophan can be only altered
18 above and below specific pH values (i.e., a pH < 3 or >9; White, 1959).
19 Consequently, given that the pH values achieved in KA and NA conditions were
20 ~7.81 and ~7.76 for winter and summer, respectively, and that the original
21 seawater pH values were 7.99 in winter and 8.02 in summer, we can discard
22 that the changes in fluorescence were due to the alterations caused directly by
23 the pH levels reached. In addition, we also tested the possible effects of pH on
24 FDOM measurements (see Material & Methods section). And, as expected, we
25 found no differences of FDOM intensities within the range of pH observed in our
26 experiments. Therefore, we can assume that the changes in FDOM intensity
27 were induced only by biological activity (i.e. FDOM intensities were not affected
28 by pH).

29
30 The increment in peak-T fluorescence (Figure 3) indicated the generation of
31 protein-like compounds in all the experimental conditions. It is common to find
32 this type of fluorescence increases when studying microbial assemblages
33 because it is associated to high productivity periods (Coble, 1996, 2007). Also,
34 since microbial cells have protein-like fluorescence themselves (Determann et

1 al., 1998), several authors have found a positive direct relationship between
2 fluorescence intensities and microbial biomass, in estuaries (Boyd and Osburn,
3 2004; Chen et al., 2004; Nieto-Cid et al., 2006; Huguet et al., 2009), coastal
4 waters (Para et al., 2010; Romera-Castillo et al., 2011a, 2010), open ocean
5 (Yamashita and Tanoue, 2003; Aparicio et al., 2015) or lakes (Yao et al., 2011;
6 Catalán et al., 2013). The increase in peak-T intensity was lower in acidified
7 conditions (Figure 3) except for the non-enriched summer conditions. However,
8 the Pearson correlation performed in the Mantel tests revealed that differences
9 between acidified and non-acidified treatments were not significant (p-value <
10 0.05). We found a significant correlation between prokaryote abundance and
11 peak-T in summer experiment but not in winter, probably due to a larger
12 influence of other variables not measured here, such as grazing on bacteria.

13

14 Although the production of protein-like material was the most relevant
15 fluorescence feature in both experiments, the generation of humic-like
16 substances (peak-A, -C and -M) was also notable. Nutrient enriched
17 mesocosms (N and NA) presented an important increase in humic signals
18 regardless of the experiment. This is consistent with these humic signals being
19 by-products of the microbial respiration processes (Nieto-Cid et al., 2005;
20 Coble, 2007; Romera-Castillo et al., 2011b, 2013; Jørgensen et al., 2011, 2014;
21 Catalá et al., 2015). In order to elucidate whether the presence of higher
22 fluorescent signals in the humic-like substances area was accompanied by a
23 change in the quality of the FDOM, we made use of three different fluorescent
24 indices that are related to the quality of the DOM. The temporal trend of the
25 peak-C:peak-M ratio (Figure 4a, b) indicated that nutrient additions affected the
26 quality of the fluorescent organic matter, however changes in pCO₂ did not
27 induce significant changes in the quantity nor in the quality of the fluorescent
28 organic material (p-value < 0.05). It has been previously demonstrated that the
29 fluorescence in the peak-C region is associated with prokaryote respiration
30 (Lønborg et al., 2010) and exudation of prokaryote by-products (Romera-
31 Castillo et al., 2011b). Therefore the high peak-C:peak-M ratios found at the
32 end of the nutrient enriched experiments, compared to the non-nutrient
33 amended, could be linked to an increase of prokaryote respiration which would
34 be induced by the elevated nutrient availability.

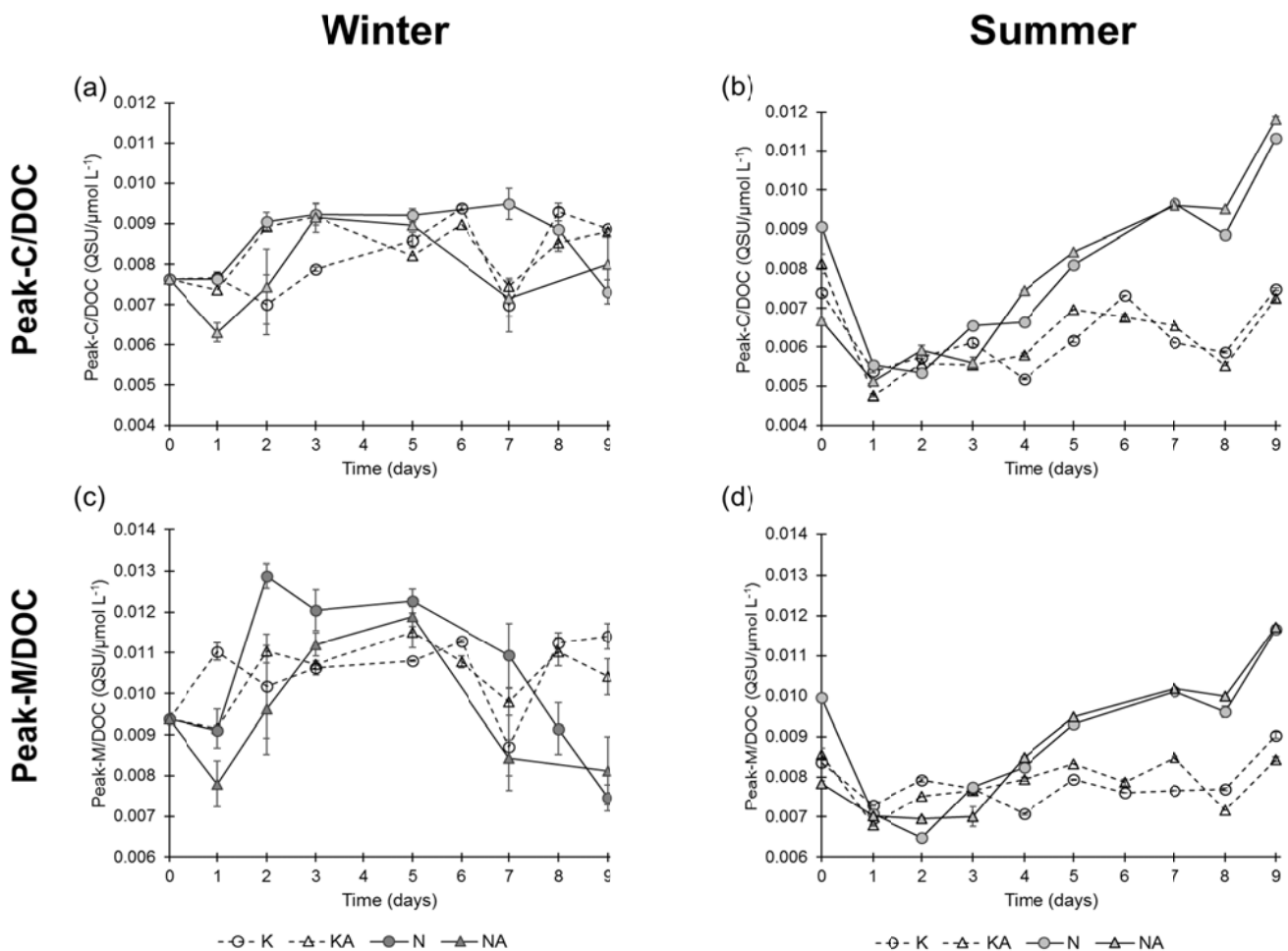
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The high pCO₂ did not seem to influence the SUVA index. However, the nutrient addition generated an increase of the SUVA values with respect to the non-amended mesocosms, but only in the summer season. This contrasting response to nutrient additions in summer could have resulted from differences in the initial quality of the DOM (note the initial values of the three indices tested, Figure 4). The generation of high molecular weight aromatic humic acids in the enriched mesocosms was distinguishable by the increase in HIX values at the end of the experiment. The HIX starting values in the two experiments differed between seasons and, most likely due to the intense solar radiation, we found lower values of HIX in the summer experiment. It has been demonstrated that photobleaching of humic-like materials results a loss of aromaticity and a decrease in the molecular weight of irradiated organic matter (Moran and Zepp, 1997; Osburn et al., 2001; Rochelle-Newall and Fisher, 2002; Helms et al., 2008; Para et al., 2010; Catalán et al., 2013). Nevertheless, because our mesocosms were not exposed to UV radiation (only photosynthetically active radiation -PAR- was provided) changes in HIX values were most parsimoniously ascribed to differences in biological activity.

It is clear that the evolution of the optical indices differed between seasons (Figure 4). In winter, nutrient and control conditions showed differences, but only in the last days of the experiment. In summer these differences were observed already at the beginning of the incubations, although the differences were statistically significant (p-value < 0.05) from t₄ (for SUVA and HIX indices) and from t₆ (peak-C:peak-M ratio). In this way, the normalization of the FDOM to DOC (peak-C/DOC and peak-M/DOC; Figure 6) also revealed that the quality of the DOM was highly influenced by the initial conditions. Although the starting values were similar for all the conditions, the different initial microbial populations could have conditioned the evolution of the DOC and FDOM dynamics. Diatoms clearly predominated over other phytoplanktonic group during winter, in the four tanks, whereas, during the summer experiment, this group was only present at small proportions in N and NA conditions and almost inexistent in the control conditions (Sala et al., 2016). Thus, a synergistic effect can be extracted from our results regarding the composition of the initial

5 microbial population and the nutrient availability. These results agreed with
 6 those obtained in the work of Zark and collaborators (2015), indicating that the
 7 ocean acidification *per se* does not influence the accumulation of DOM in
 8 coastal environments.

6



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8

11 **Figure 6.** Time evolution of the quotient between peak-C and DOC in (a) winter and (b)
 12 summer; quotient between peak-M and DOC in (c) winter and (d) summer. Units are
 13 expressed in QSU/μmol L⁻¹. Error bars indicate the standard error of 2 replicates.

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1 **5. – Conclusions**

2 The results extracted from this study highlight the value of mesocosm
3 experiments as a way to assess possible responses of DOM dynamics to future
4 environmental changes. To our knowledge, this is the first study that quantifies
5 the influence of high levels of pCO₂ on the fluorescent properties of DOM in the
6 Mediterranean Sea.

7
8 Although we found, in general, higher phytoplankton biomass under high-pCO₂
9 conditions, we did not observe differences in the DOC dynamics between
10 acidified and non-acidified treatments.

11
12 Transformations of DOM composition tracked by changes in its optical
13 properties (absorbance and fluorescence) indicated that eutrophication modified
14 the structure of the organic matter into more complex material, while a weak
15 aromatization of the DOM was observed under higher pCO₂ conditions.

16
17 The effects of eutrophication, in terms of quantity and quality of organic matter,
18 varied depending on the initial conditions, which highlights the importance of
19 conducting experiments under different seasons/regimes to account for
20 temporal variability in the response of the ecosystem to the studied variables.

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